

Supporting Information

Homoserine lactone as a structural key element for the synthesis of multifunctional polymers

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1. Synthesis of the thiolactone-lactone coupler (1)

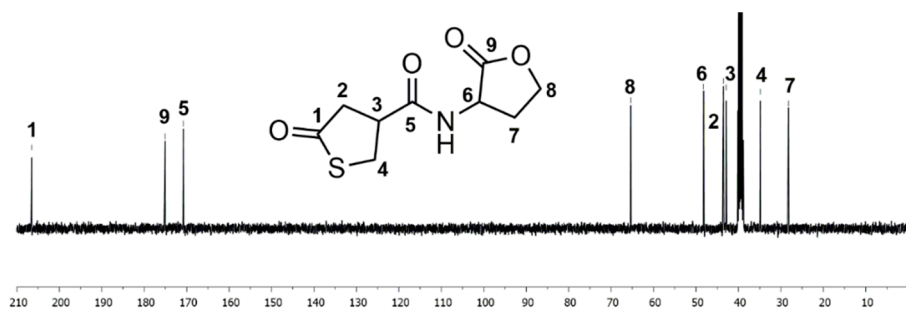


Figure S1. ¹³C NMR spectrum of thiolactone-lactone (1) measured in DMSO-*d*₆.

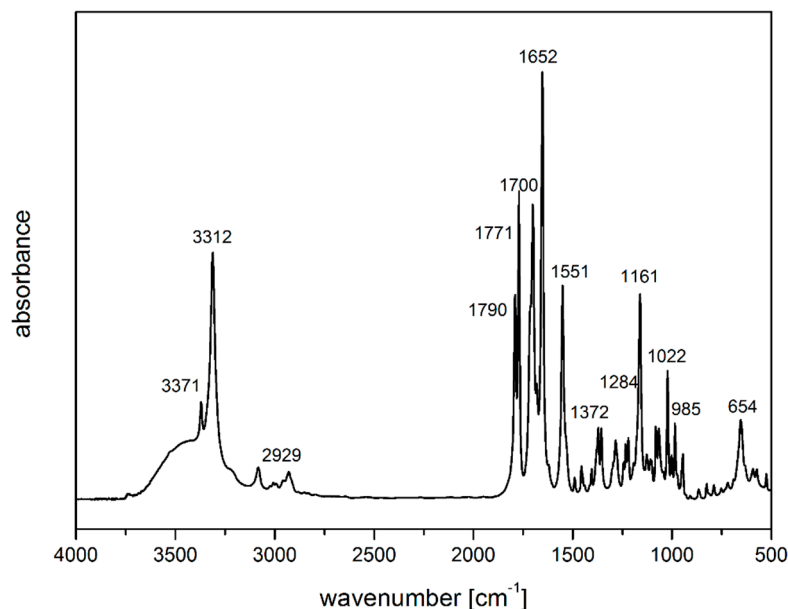


Figure S2. FTIR spectrum of thiolactone-lactone (1).

2. Model reaction of coupler 1 with hexylamine in a 1:1 molar ratio

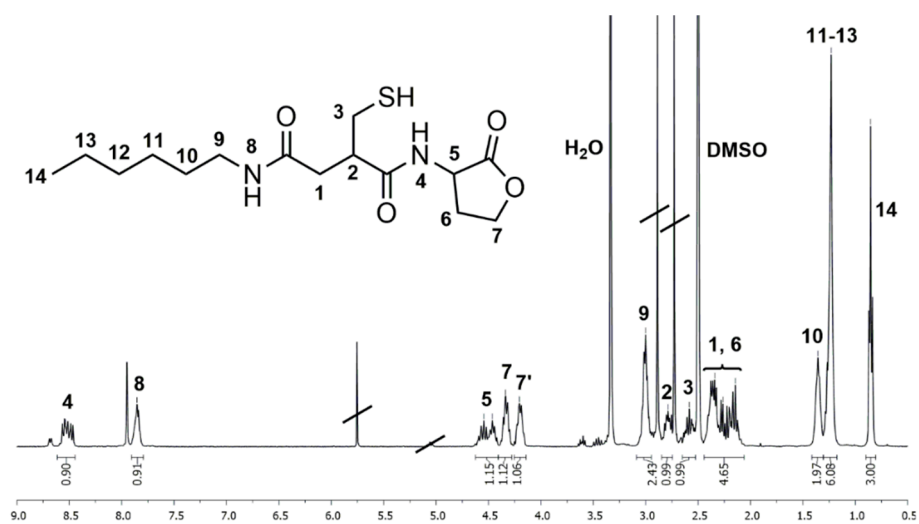


Figure S3. ^1H NMR spectrum of compound 2 measured in $\text{DMSO-}d_6$.

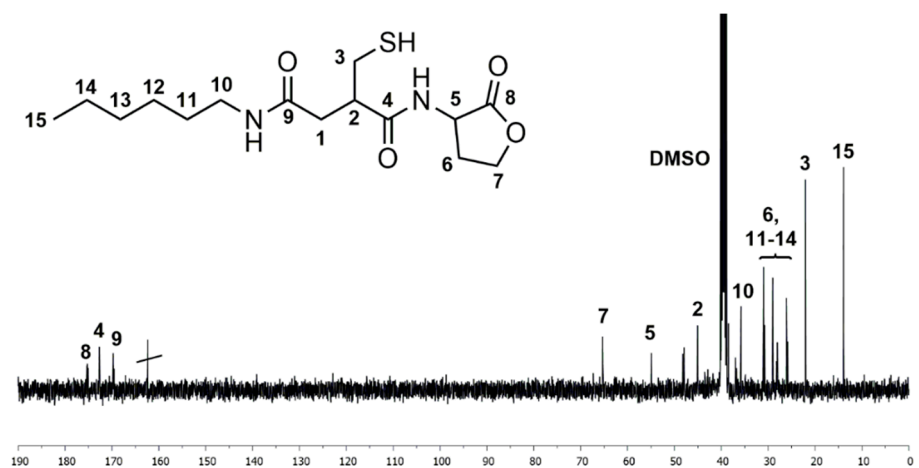


Figure S4. ^{13}C NMR spectrum of compound 2 measured in $\text{DMSO-}d_6$.

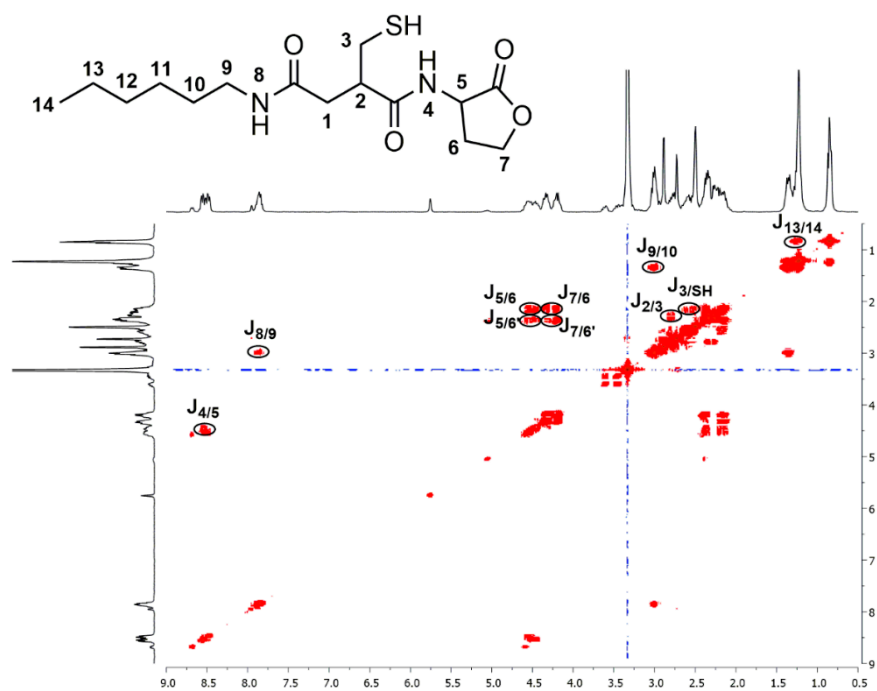


Figure S5. H,H-COSY NMR spectrum of compound 2 measured in $\text{DMSO-}d_6$.

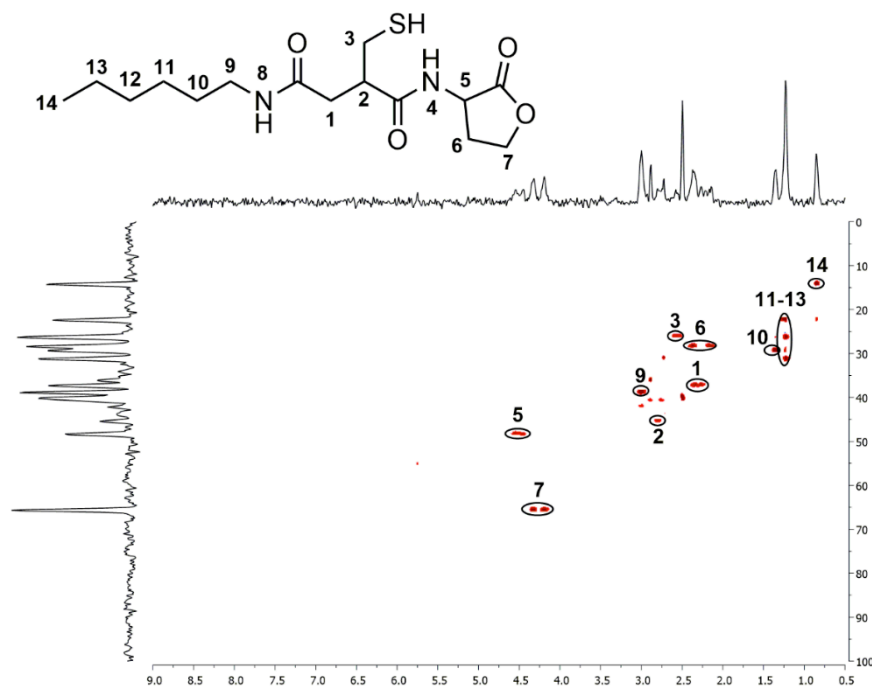


Figure S6. HSQC NMR spectrum of compound 2 measured in DMSO- d_6 .

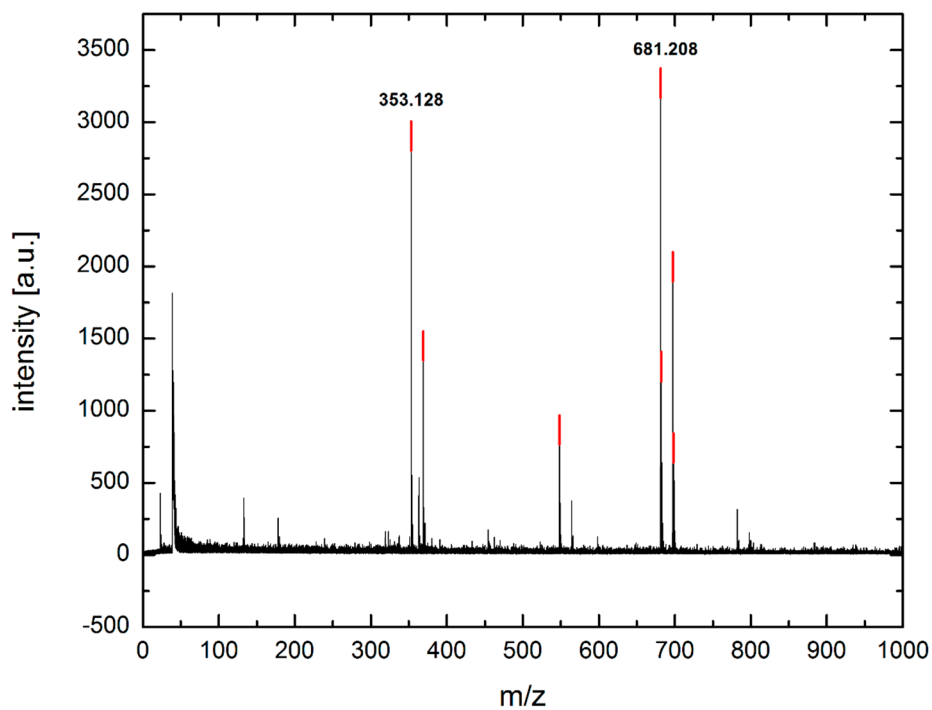


Figure S7. NALDI-TOF spectrum of compound 2.

Table S1. Identified signals of the NALDI-TOF measurement of the addition product of the addition one equivalent hexylamine to the thiolactone-lactone coupler (1). Reported signals describe masses of $[M + Na^+]$ ($MW = 22.99 \text{ g} \cdot \text{mol}^{-1}$).

Product	m/z (calculated)	m/z (found)	intensity [a.u.]
2	353.15	353.128	2902
Disulfide of 2	681.29	681.208	3270

3. Model reaction of coupler 1 with hexylamine in a 1:2 molar ratio

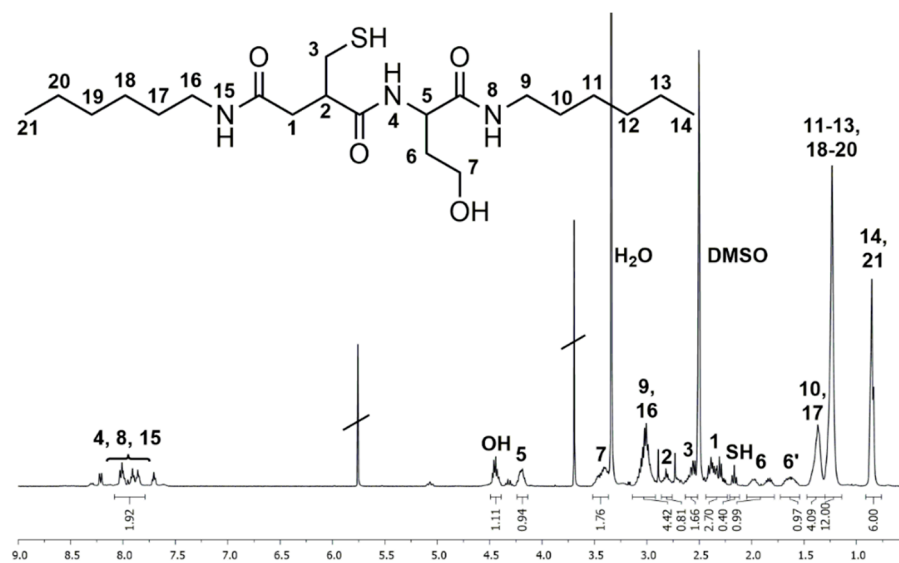


Figure S8. ^1H NMR spectrum of compound 3 measured in $\text{DMSO-}d_6$.

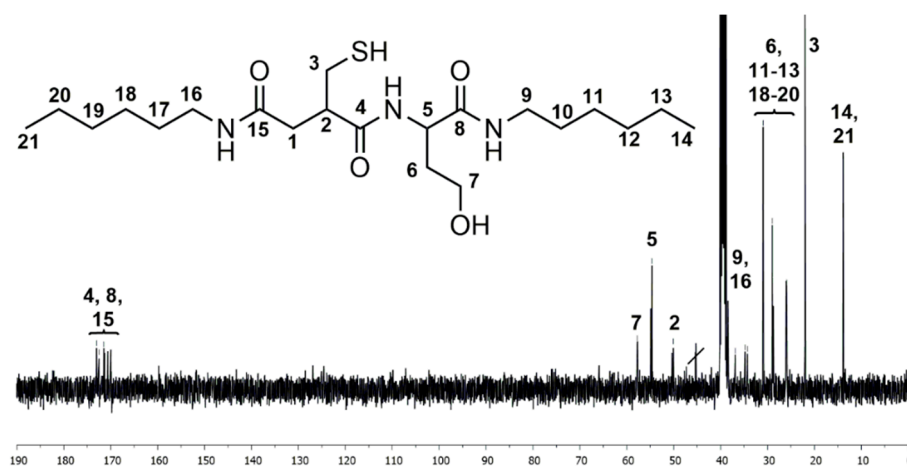


Figure S9. ^{13}C NMR spectrum of compound 3 measured in $\text{DMSO-}d_6$.

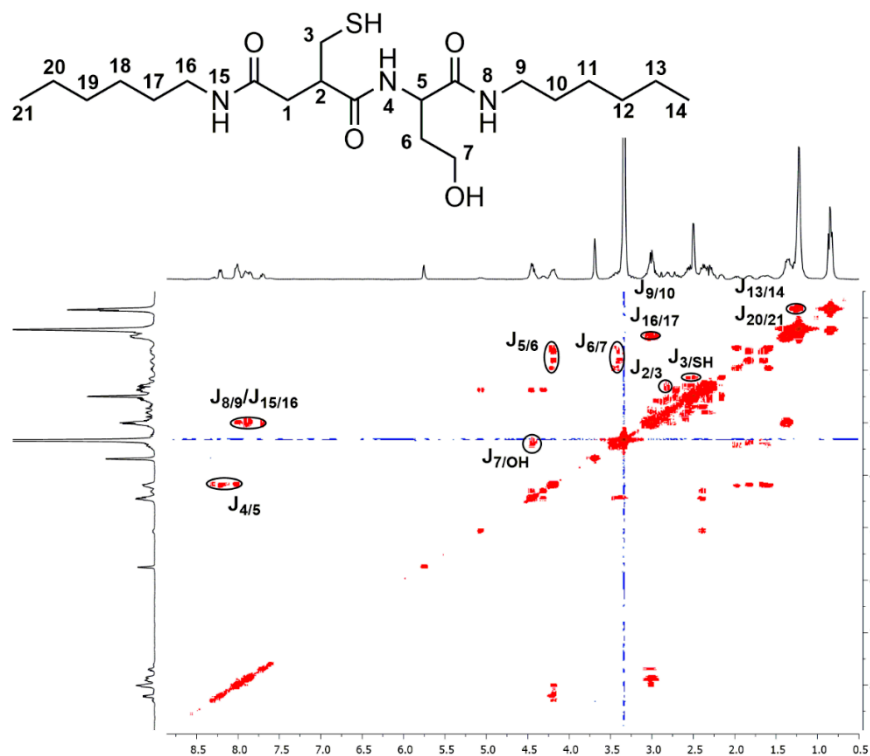


Figure S10. H,H-COSY NMR spectrum of compound 3 measured in DMSO-*d*₆.

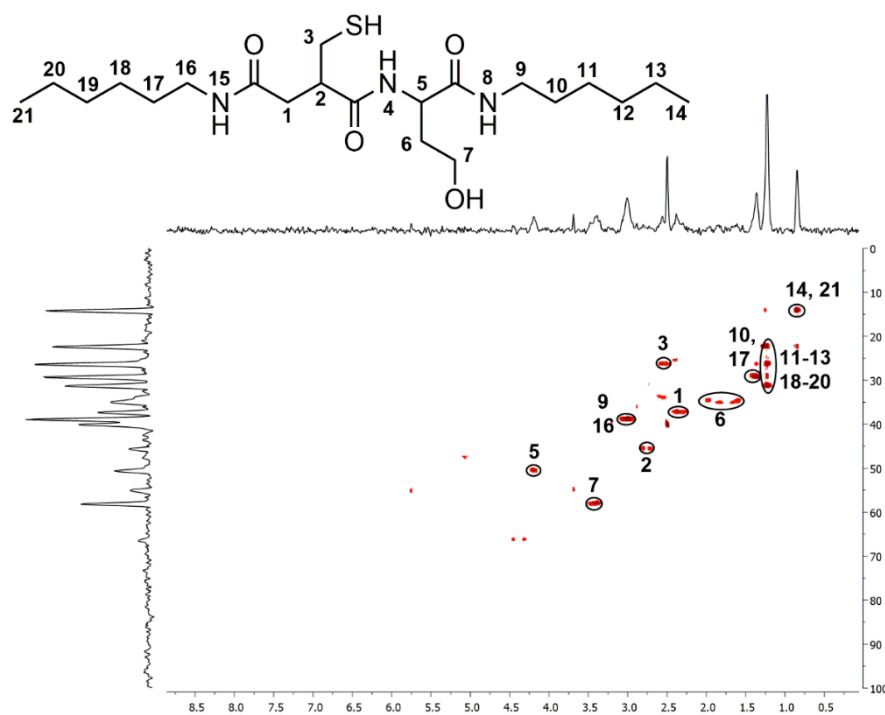


Figure S11. HSQC NMR spectrum of compound 3 measured in DMSO-*d*₆.

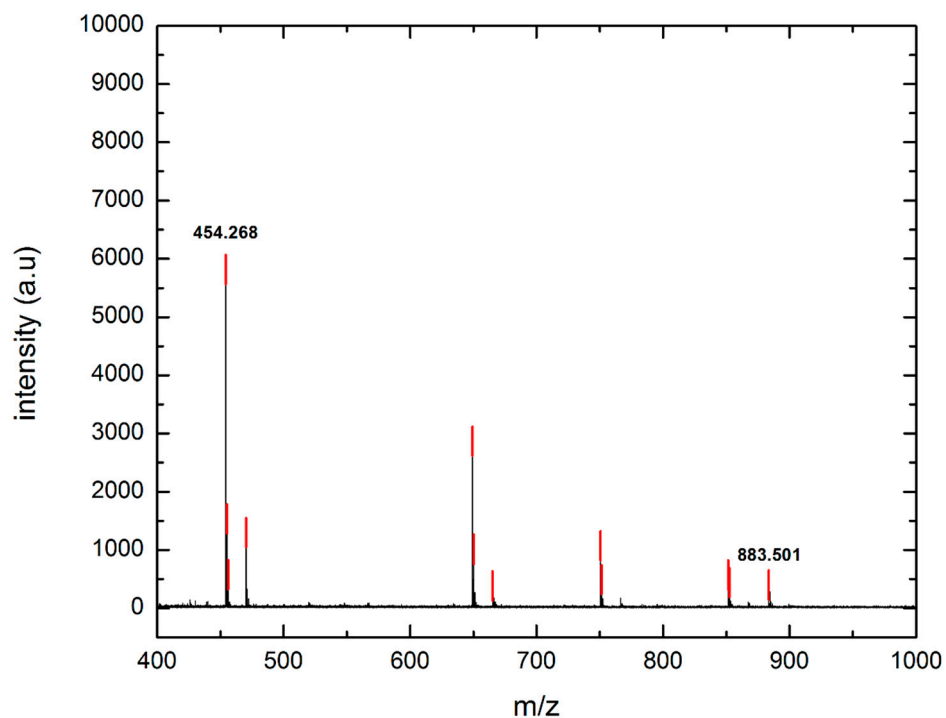


Figure S12. NALDI-TOF spectrum of compound 3.

Table S2. Identified signals of the NALDI-TOF measurement of the addition product of the addition of two equivalents hexylamine to the thiolactone-lactone coupler (**1**). Reported signals describe masses of $[M + Na^+]$ ($MW = 22.99 \text{ g} \cdot \text{mol}^{-1}$).

Product	m/z (calculated)	m/z (found)	intensity [a.u.]
3	454.27	454.268	5819
Disulfide of 3	883.53	883.501	403

4. Polyaddition reaction with of PEG-diamine to coupler 1

Table S3. Reagent ratios for the synthesis of **4a-d** ($T = 90 \text{ }^\circ\text{C}$, $t = 16 \text{ h}$).

polymer	m_1 [g]	mMMA [g]	mPEG-diamine [g]	V_{DMF} [mL]	c [g · mol ⁻¹]
7a	0.249	0.094	0.400	2.170	0.5
7b	0.100	0.038	0.161	0.436	1.0
7c	0.100	0.038	0.161	0.291	1.5
7d	0.100	0.038	0.161	-	-

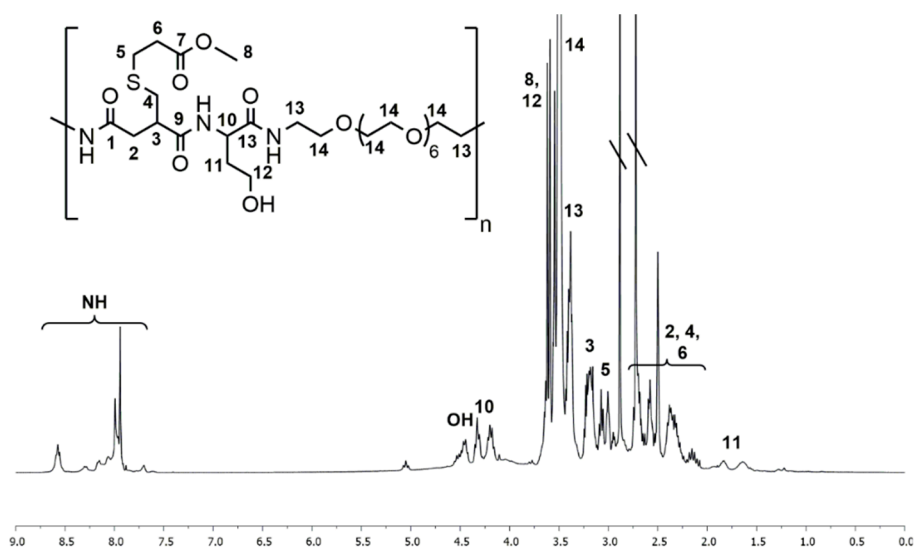


Figure S13. ^1H NMR spectrum of the reaction of thiolactone-lactone (4) with PEG-diamine measured in $\text{DMSO-}d_6$.

5. Synthesis of PG_{26} (5)

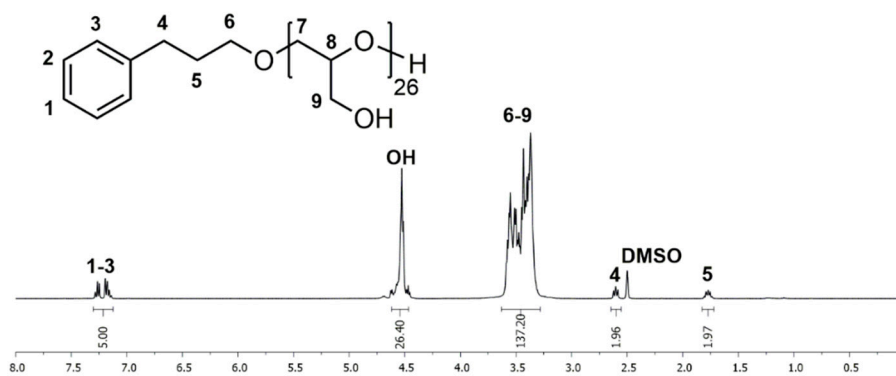


Figure S14. ^1H NMR spectrum of PG_{26} (5) measured in $\text{DMSO-}d_6$.

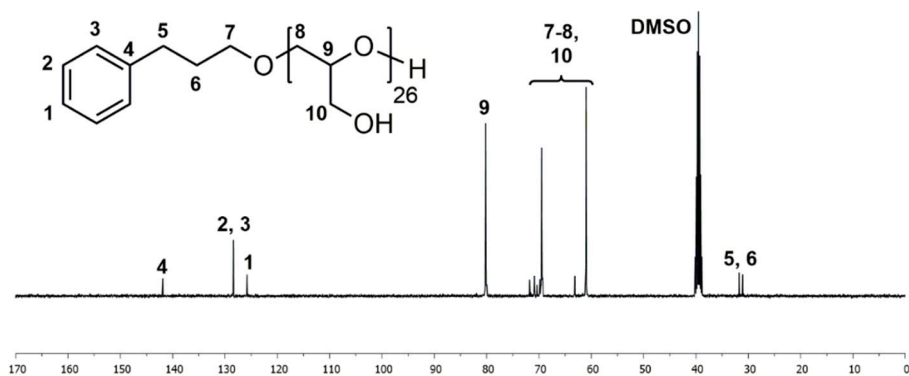
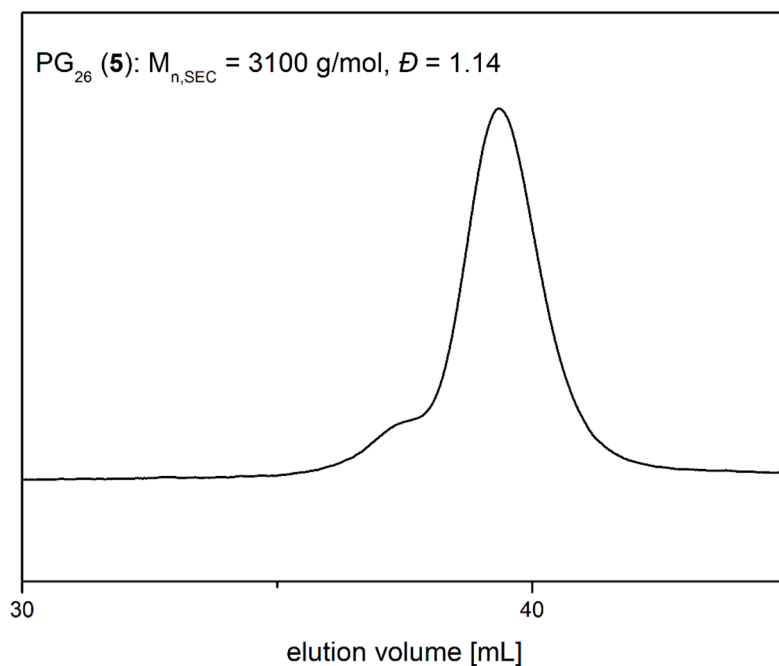
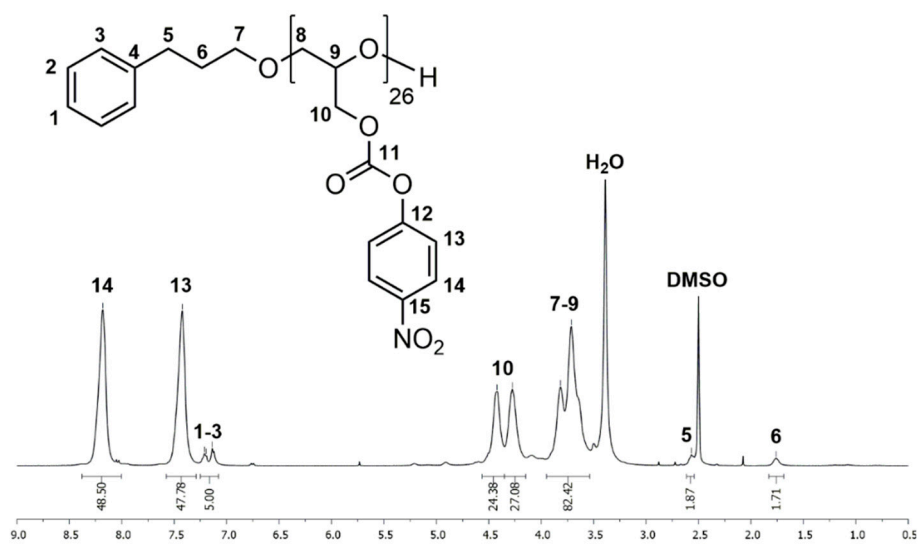


Figure S15. ^{13}C NMR spectrum of PG_{26} (5) measured in $\text{DMSO-}d_6$.

Figure S16. DMF-SEC traces of PG₂₆ (5).

6. Functionalization of polyglycidol (5) with DL-homoserine lactone hydrobromide

Figure S17. ¹H NMR spectrum of P(G^{NPC})₂₆ (6) measured in DMSO-*d*₆.

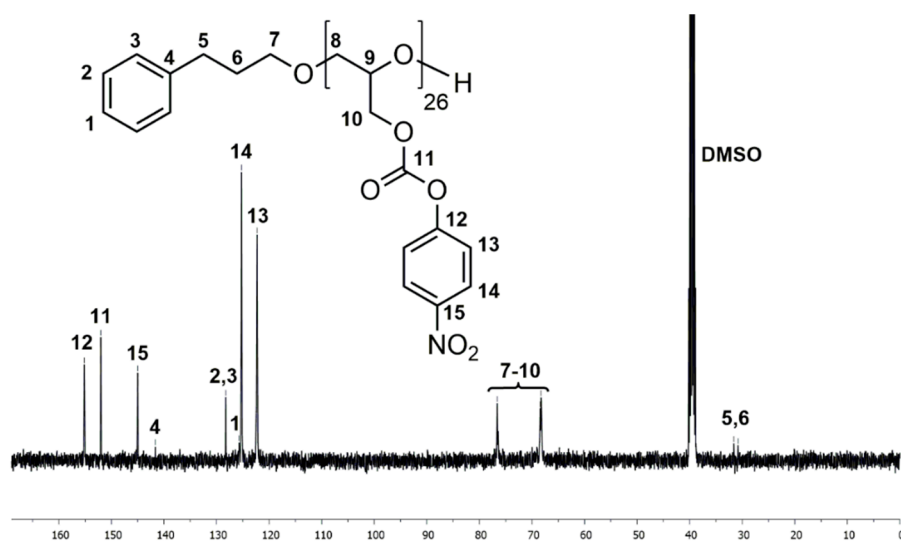


Figure S18. ¹³C NMR spectrum of P(G^{NP}C)₂₆ (6) measured in DMSO-*d*₆.

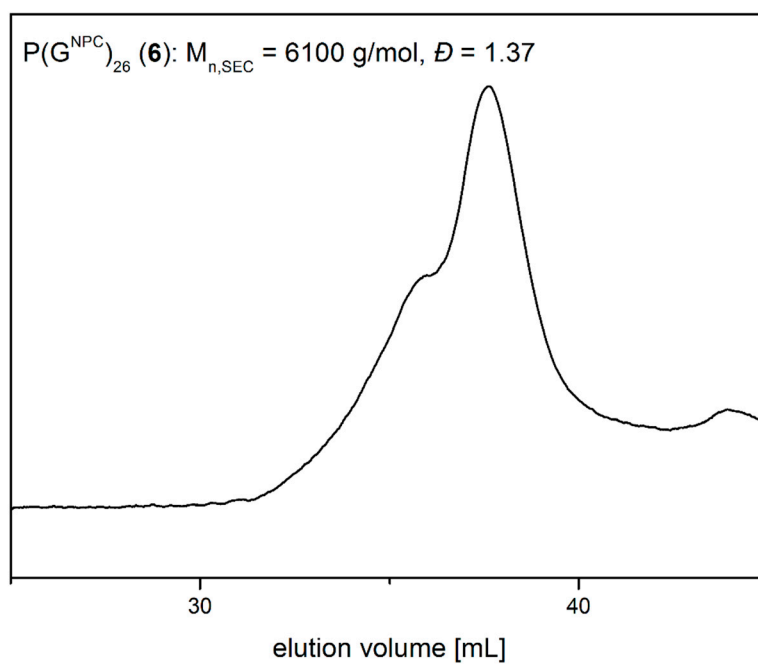


Figure S19. DMF-SEC traces of P(G^{NP}C)₂₆ (6).

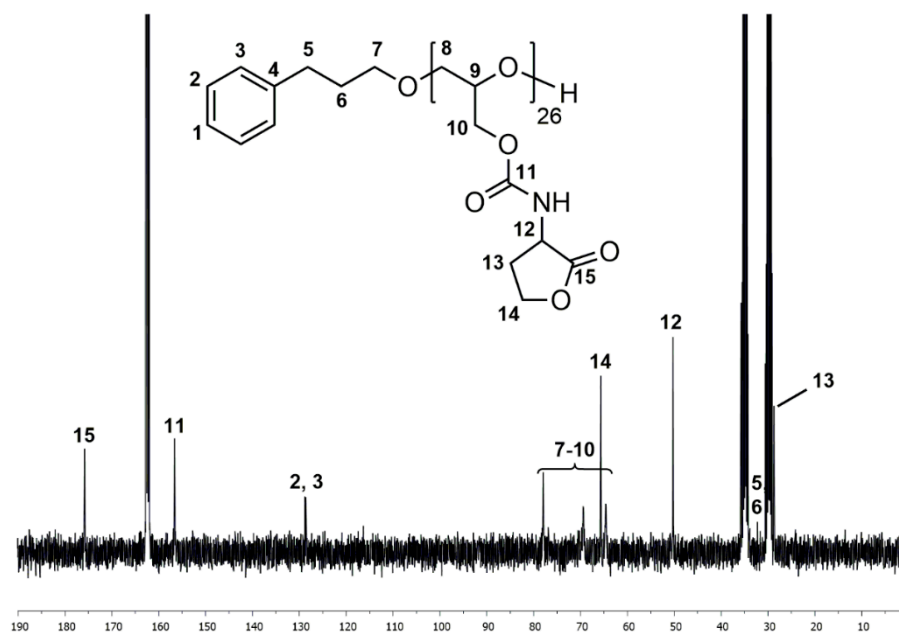


Figure S20. ¹³C NMR spectrum of P(G^{HSL})₂₆ (7) measured in DMF-*d*₇.

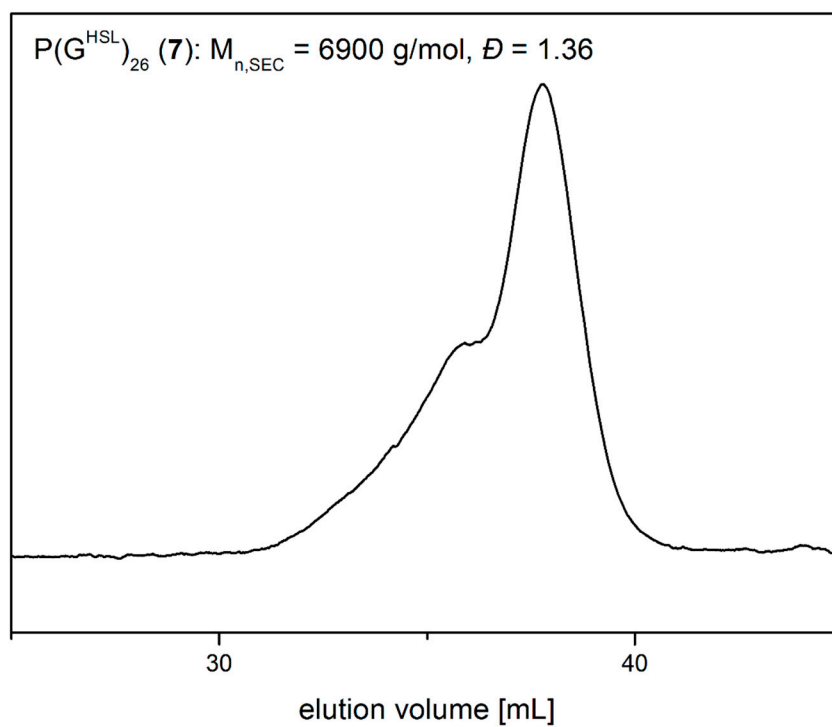


Figure S21. DMF-SEC traces of P(G^{HSL})₂₆ (7).

7. Ring-opening of P(G^{HSL})₂₆

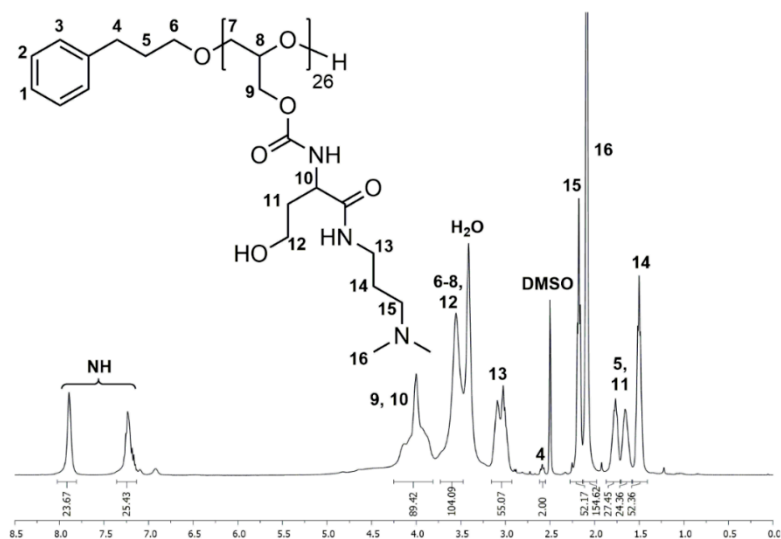


Figure S22. ^1H NMR spectrum of $\text{P}(\text{G}^{\text{HSL},0})_{26}$ (8) measured in $\text{DMSO-}d_6$.

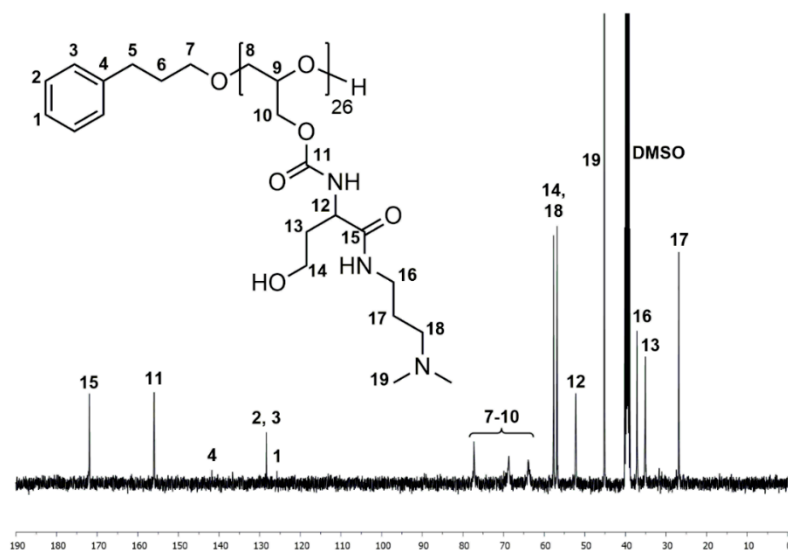


Figure S23. ^{13}C NMR spectrum of $\text{P}(\text{G}^{\text{HSL},0})_{26}$ (8) measured in $\text{DMSO-}d_6$.

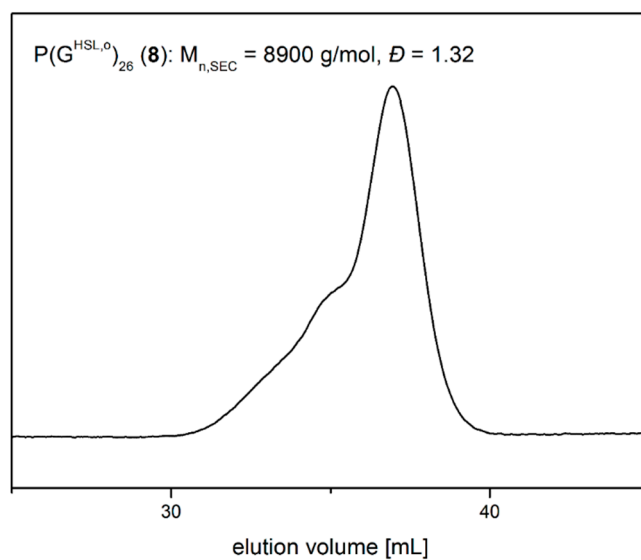
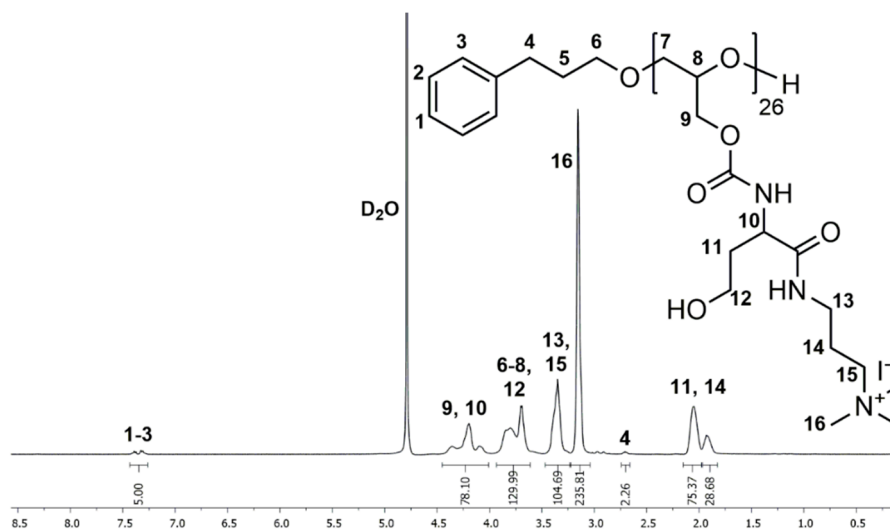
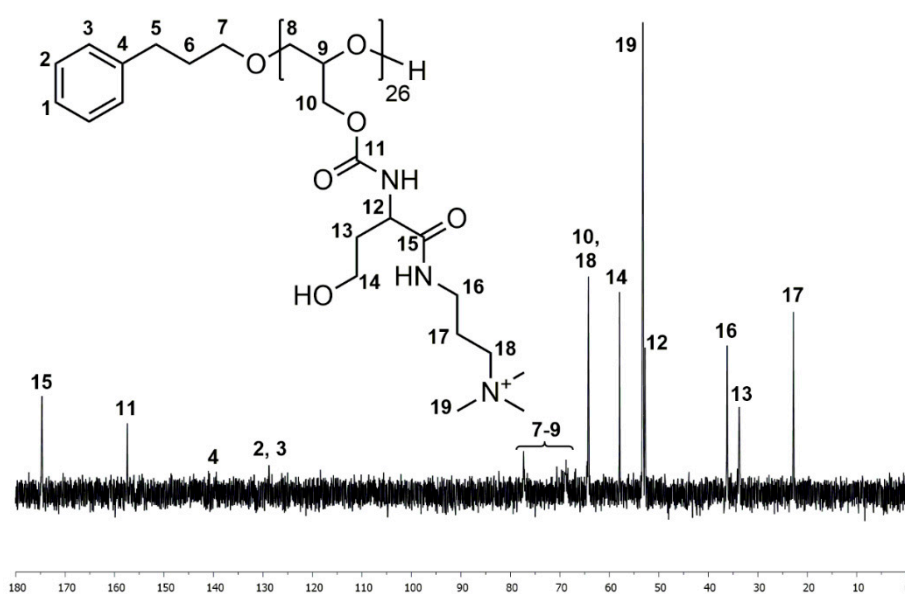


Figure S24. DMF-SEC traces of $\text{P}(\text{G}^{\text{HSL},0})_{26}$ (8).

8. Quaternization of P(G^{HSL,0})₂₆Figure S25. ¹H NMR spectrum of P(G^{HSL,0,q})₂₆ (9) measured in D₂O.Figure S26. ¹³C NMR spectrum of P(G^{HSL,0,q})₂₆ (9) measured in D₂O.