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Thermomagnesium: A By-Product of Ni Ore Mining as a Clean Fertilizer Source for Maize

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Abstract: This study explores whether Thermomagnesium (TM), a by-product of Ni ore mining, is an efficient fertilizer for maize. The effects of TM on soil pH, the supply of Si and Mg to the soil and plants, carbohydrate metabolism, grain filling, and yield were assessed in two simultaneous experiments performed in greenhouse conditions. Five TM doses were applied to two soil textures—clayey (0, 55, 273, 709, and 2018 mg kg⁻¹) and sandy (0, 293, 410, 645, and 1260 mg kg⁻¹). In general, the best results in soil and maize plants occurred at the highest TM dose for both soil textures (clayey 2018 mg kg⁻¹ and sandy 1260 mg kg⁻¹). The results demonstrated that in both soils, the concentrations of Mg and Si in the maize leaves increased with the dose of TM, similarly to that which occurred in the soil. Interestingly, in clayey soil, the soil pH increased linearly, whereas in sandy soil, the pH reached its maximum value between the two largest TM doses. The concentration of reducing sugars increased at the highest TM dose, whereas the concentrations of sucrose and starch decreased. The enhancement of carbohydrate partitioning led to higher maize growth, grain filling, and yield. Overall, the results clearly demonstrate that TM is a sustainable alternative fertilizer for maize and can be used for countless other crops and soil classifications, thus providing a suitable destination for this by-product of Ni ore mining.



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Keywords: *Zea mays* (L); mining waste; magnesium silicate; carbohydrate partitioning; grain filling

1. Introduction

From an economic perspective, maize (*Zea mays* L.) is a global standout among crops because of its wide range of uses in human food, animal feed, and industrial applications [1]. This broad utility has also led to an increasing demand for maize [2]. Enhancing crop productivity is key for improving agricultural production and guaranteeing global food and energy supplies [3]. Fertilizers are the main inputs for increasing crop productivity quickly [4,5] and for replacing nutrients removed from the soil by crops [6]. Between 30 and 50% of crop productivity is attributable to fertilization, and thus fertilizer use is essential for food production in order to meet the demands of the growing world population [4].

Brazil is the fifth largest consumer of fertilizer in the world, and maize accounts for approximately 16% of its fertilizer use [7,8]. However, Brazil imports approximately 81% of its fertilizer [9,10]. A potential alternative source of nutrients is mining by-products [11,12], which would reduce the environmental problem of rock mining while providing a cleaner, domestic option for soil fertilization [13]. The use of by-products as a fertilizer in agriculture is an effective means of improving soil quality, increasing crop yields, and mitigating environmental impacts, while generating a new source of revenue [14].

One such by-product is Thermomagnesium (TM; or magnesium silicate (MgSiO₃)), which originates from the extraction of nickel (Ni) and has potential for use as a soil remineralizer to provide silicon (Si) and magnesium (Mg) to crops [12]. Si is a beneficial mineral element commonly found in the soil that increases plant resistance to stress by

accumulating in the roots, stem, leaves, and bark [15]. In particular, supplying Si to plants can increase tolerance to water stress, thereby improving production stability, especially in regions subject to drought like the Brazilian Cerrado, a biome similar to the African savanna [16]. Mg is an important nutrient that is involved in several vital functions in plants [17,18]. Mg deficiency in plants may be caused not only by the depletion of soil reserves, but also by low Mg assimilation by roots as a result of competitive inhibition caused by the imbalance of cations such as calcium (Ca) and potassium (K) [19,20]. The reduced transport and accumulation of carbohydrates in Mg-deficient leaves alter the photosynthetic carbon metabolism and restrict CO₂ fixation [17,21]. Mg is closely linked to partitioning among starch synthesis, triose phosphate transport in the cytosol, and sucrose formation, and low Mg availability leads to a reduction of glucose–fructose bonds in cytosol [22].

To support the further development of TM as a fertilizer, the objective of the present study was to evaluate the effects of TM application on the supply of Si and Mg in sandy and clayey soil for the cultivation of maize, the partitioning of carbohydrates in the plants, grain yield, and the chemical attributes of the soil.

2. Materials and Methods

2.1. Thermomagnesium Processing

Thermomagnesium was produced by melting Ni ore in an electric furnace, which granulates nonmetallic elements, such as silicon dioxide (SiO₂), magnesium oxide (MgO), and ferric oxide (Fe₂O₃), in the presence of water. The ore was then crushed to alter its granulometry, followed by calcination in rotary kilns to remove all moisture, including chemically bound water. The calcined ore was subsequently charged in an electric furnace and was reduced at high temperature, which removed O from Fe and Ni oxides to produce an iron–nickel alloy (Fe–Ni) containing approximately 20% Ni, as well as a large amount of by-product [12].

The by-product was immediately cooled from 1600 °C to room temperature under pressure using several water jets. The solidified MgSiO₃-rich by-product did not exhibit a granulometric uniformity and thus underwent a forming process. First, the by-product was dried in a rotary dryer at a beneficiation plant [12]. After drying, the particle size distribution (100 mesh size, ~149 μm) was obtained by sieving, and the mineralogical phases of the by-product were determined by X-ray diffraction (XRD) in a Philips X-ray diffractometer (Table 1). The crystalline phases were identified by comparing the sample using the PDF2 Database of the International Center for Diffraction Data and the Inorganic Crystal Structure Database (ICSD; Figure 1). The values were calculated following the method of Rietveld [23], using the standard ICSD crystalline structures and internal fluorite (CaF₂) to determine the amorphous phase.

Table 1. Chemical composition of Thermomagnesium from nickel ore mining in São Paulo State, Brazil.

Mineral Species	Chemical Formula	Abundance	Unit
Chemical composition			
Arsenic	As ³⁺	<5	mg·kg ⁻¹
Cadmium	Cd ²⁺	<5	mg·kg ⁻¹
Lead	Pb ²⁺	<5	mg·kg ⁻¹
Mercury	Hg ²⁺	<0.20	mg·kg ⁻¹
Selenium	Se ²⁺	<0.10	mg·kg ⁻¹
Chromium	Cr ³⁺	0.88	%
Nickel	Ni ²⁺	0.11	%
Mineralogical phases			
Amorphous phase	-	74	%
Forsterite	Mg _{1.8} Fe _{0.2} SiO ₄	24	%
Lizardite	Mg ₃ Si ₂ (OH) ₄ O ₅	2	%
Mineral groups			
Magnesium oxide	MgO	28	%
Silicon dioxide	SiO ₂	48	%

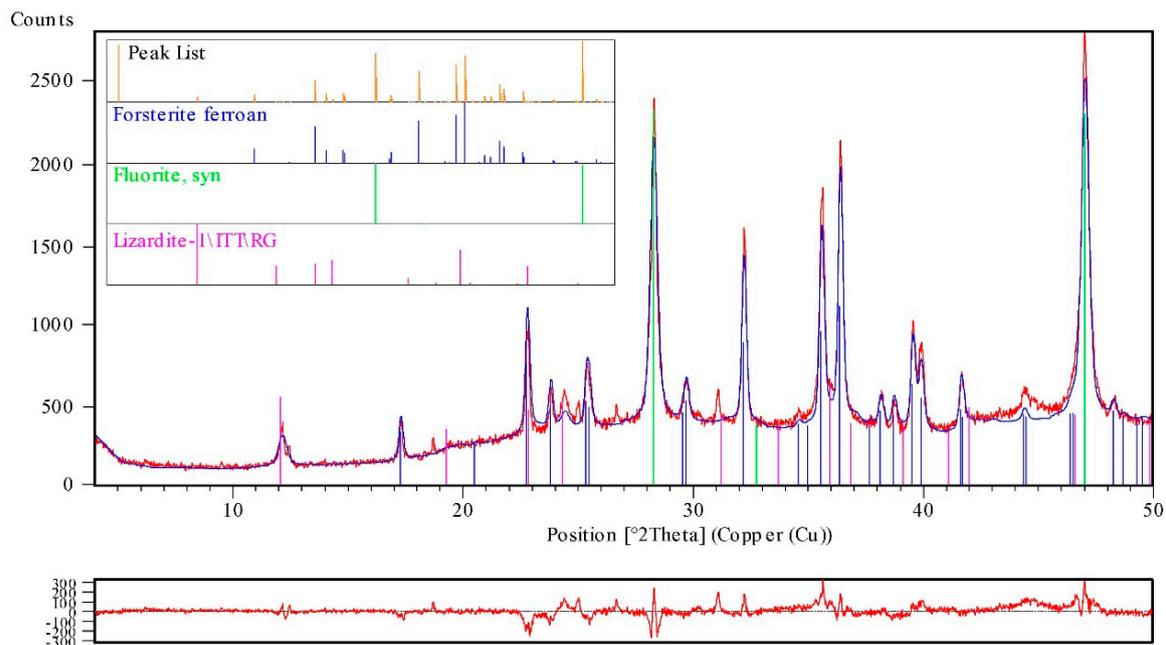


Figure 1. X-ray diffractogram for quantifying the mineralogical phases of Thermomagnesium from Ni ore mining. To calculate the amorphous material, 10% fluorite (ICDD 01-077-2093) was added.

According to the maximum limits of potentially toxic elements (PTEs; arsenic (As), cadmium (Cd), chromium (Cr), lead (Pb), nickel (Ni), mercury (Hg) and selenium (Se)) in soil remineralizers allowed by the Brazilian National Environmental Council, as established by Annexe I of the Normative Instruction SDA n. 27 [24], TM does not represent an environmental risk (Table 1) and can be safely reused in agriculture.

2.2. Experimental Conditions

This study was conducted in an experimental greenhouse with a climate-controlled environment at the Lageado Experimental Farm of São Paulo State University (UNESP), which is located in the municipality of Botucatu in the southeastern region of São Paulo State, Brazil (48°26' W, 22°51' S, elevation of 786 m above sea level). The greenhouse had an internal heating/cooling air circulatory system to maintain the temperature between 22 °C and 32 °C, and had a 70% relative humidity. The experiment was performed using polyethylene pots with a capacity of 25 kg of soil.

According to the Brazilian Soil Classification System [25], the clayey soil used in the experiment was classified as a dystrophic Red Latosol, corresponding to a typical Ferralsol [26], and the sandy soil was a kaolinitic, thermic Typic Haplorthox [27]. The soil texture and bulk density [28], as well as the chemical properties [29], were determined at a 0.0–0.2 m depth (Table 2). For the soil acidity correction, base saturation (aiming to obtain 60% base saturation) was conducted by adding calcium carbonate (CaCO₃; >99%, p.a., Sigma-Aldrich, Inc., St. Louis, MO, USA). Each pot was filled with air-dried soil mixed homogeneously with CaCO₃ and the specified TM dose, and was incubated for ~40 days to allow for a reaction with the soil.

Prior to the experiment, the soil water retention capacity was determined using a tension table and a Richards extraction chamber [30] in order to calculate the soil water potential (ψ_w).

Table 2. Physicochemical properties of soil types (0.00–0.20 m depth) in São Paulo, Brazil.

Soil Properties	Unit	Clayey	Sandy	
		Value		
Clay	$\text{g}\cdot\text{kg}^{-1}$	602	251	
Silt	$\text{g}\cdot\text{kg}^{-1}$	281	25	
Sand	$\text{g}\cdot\text{kg}^{-1}$	117	724	
Bulk density	$\text{g}\cdot\text{cm}^{-3}$	1.19	1.26	
pH (CaCl_2)	–	5.20	5.40	
Soil organic matter	$\text{g}\cdot\text{kg}^{-1}$	25.0	13.0	
Available phosphorus (P_{resin})	$\text{mg}\cdot\text{kg}^{-1}$	6.00	3.00	
Exchangeable	Calcium ($\text{Ca}^{2+}_{\text{resin}}$)	$\text{mmol}_c\cdot\text{kg}^{-1}$	35.0	18.0
	Mg ($\text{Mg}^{2+}_{\text{resin}}$)	$\text{mmol}_c\cdot\text{kg}^{-1}$	8.00	2.00
	Potassium ($\text{K}^{+}_{\text{resin}}$)	$\text{mmol}_c\cdot\text{kg}^{-1}$	0.70	0.50
	Aluminum ($\text{Al}^{3+}_{\text{KCl}}$)	$\text{mmol}_c\cdot\text{kg}^{-1}$	0.00	0.00
Potential acidity (H+Al)	$\text{mmol}_c\cdot\text{kg}^{-1}$	29.0	18.0	
S–Sulfate ($\text{S}-\text{SO}_4^{2-}_{\text{Ca}(\text{H}_2\text{PO}_4)_2}$)	$\text{mg}\cdot\text{kg}^{-1}$	9.00	6.00	
Boron ($\text{B}_{\text{Hot water}}$)	$\text{mg}\cdot\text{kg}^{-1}$	0.40	0.30	
Copper ($\text{Cu}_{\text{DTPA-TEA}}^1$)	$\text{mg}\cdot\text{kg}^{-1}$	4.00	1.00	
Iron ($\text{Fe}_{\text{DTPA-TEA}}$)	$\text{mg}\cdot\text{kg}^{-1}$	12.0	17.0	
Manganese ($\text{Mn}_{\text{DTPA-TEA}}$)	$\text{mg}\cdot\text{kg}^{-1}$	13.2	2.20	
Zinc ($\text{Zn}_{\text{DTPA-TEA}}$)	$\text{mg}\cdot\text{kg}^{-1}$	0.90	0.40	
Base saturation (BS)	%	60.0	53.3	
Cation exchange capacity ($\text{CEC}_{\text{pH } 7.0}$)	$\text{mmol}_c\cdot\text{kg}^{-1}$	72.7	38.5	

¹ DTPA–TEA—diethylenetriaminepentaacetic acid–triethanolamine.

2.3. Experimental Design and Treatments

The experimental design was a completely randomized block with four replicates comprising four TM doses plus an absolute control for both soil textures. The treatments involved TM application to the soil at the following five doses: 0, 55, 273, 709, and 2018 $\text{mg}\cdot\text{kg}^{-1}$ for clayey soil, and 0, 293, 410, 645, and 1260 $\text{mg}\cdot\text{kg}^{-1}$ for sandy soil. For both soils, doses were calculated respecting the Ca:Mg ratios (4:1, 3:1, 2:1, and 1:1). Maize seeds (hybrid P3707VYH; DuPont Pioneer®, Johnston, IA, USA) for grain production purposes were treated with fungicides (carboxin and thiram at 100 g + 100 g a.i. 100 kg^{-1} seeds) prior to planting. Nutrients were provided in the pots as needed for cultivation (Table 3) [31]. Five seeds were sown per pot, and the seedlings were thinned to three plants 10 d after sowing.

Table 3. Fertilization performed in the greenhouse experiment in Botucatu, São Paulo, Brazil.

Nutrient	Dose	Unit	Source
N	50	$\text{mg}\cdot\text{kg}^{-1}$	NH_4NO_3
	150 ¹	$\text{mg}\cdot\text{kg}^{-1}$	NH_4NO_3
P	200	$\text{mg}\cdot\text{kg}^{-1}$	$\text{Ca}(\text{H}_2\text{PO}_4)_2$
K	50	$\text{mg}\cdot\text{kg}^{-1}$	KCL
	100 ¹	$\text{mg}\cdot\text{kg}^{-1}$	K_2SO_4
Zn	5.0	$\text{mg}\cdot\text{kg}^{-1}$	$\text{ZnSO}_4\cdot 7\text{H}_2\text{O}$
Mn	5.0	$\text{mg}\cdot\text{kg}^{-1}$	$\text{MnSO}_4\cdot 3\text{H}_2\text{O}$
Fe	2.5	$\text{mg}\cdot\text{kg}^{-1}$	$\text{FeSO}_4\cdot 2\text{H}_2\text{O}$
Cu	1.5	$\text{mg}\cdot\text{kg}^{-1}$	$\text{CuSO}_4\cdot 7\text{H}_2\text{O}$
B	0.5	$\text{mg}\cdot\text{kg}^{-1}$	H_3BO_3

¹ Topdressing at the V₃ maize phenological stage.

The available water capacity was corrected to -0.01 MPa (field capacity), which corresponded to 110 and 129 $\text{g}\cdot\text{kg}^{-1}$ for the clayey and sandy soils, respectively, corresponding to 60%–100% field capacity. Four reference pots were weighed every four days, and ster-

ile distilled water was supplied as needed to restore the water content to field capacity (-0.01 MPa).

2.4. Soil–Plant Sampling and Analysis

At the R_1 phenological stage [32] (silking), 20 leaves were collected from 20 different plants of each plot and were milled in a Wiley-type mill coupled to a 1-mm sieve to determine the Mg and Si contents [33]. The same milled leaf samples were also subjected to sugar fractionation by high-performance liquid chromatography (HPLC) on a Shimadzu model 10A chromatograph with an RID-10A refractive index detector and model LC-10AD isocratic pump. Purified water was used as the mobile phase at a flow rate of 0.6 mL min^{-1} . Prior to HPLC, 1.0 g of leaf sample and 8.0 g of purified water were weighed and incubated in a Marconi metabolic bath model MA095 at 60 °C for 40 min with constant agitation. After subsequent centrifugation at $12,000$ rpm, the supernatant was filtered through a millipore polyvinylidene fluoride (PVDF) membrane with a porosity of 0.22 μm and diameter of 13.0 mm. The samples were then injected into the HPLC system, and the resulting chromatograms were compared with those of defined concentrations of sucrose, glucose, and fructose (reducing sugars). The concentrations of sucrose and reducing sugars were calculated by comparing their areas with those of the standards, and then multiplying by the dilution of each sample. The starch concentration in the leaves [34] was determined using a spectrophotometer at 535 nm.

Simultaneous with the leaf analysis, two plants were collected for dry matter analysis. The remaining plants were grown until the end of the cycle, and the weight of 100 grains (W100G) was converted to values on a dry weight basis by correcting for 13% moisture. The moisture was determined with an automatic measuring device (Gehaka G650i, Brazil). The grain yield was expressed as grams per plant.

After harvest, a soil sample from each pot was collected, and the pH and Mg and Si availability were analyzed. The soil pH was determined in calcium chloride (CaCl_2) suspensions at 0.01 mol L^{-1} (at a soil:solution ratio of $1:2.5$). Mg availability was determined using resin [35], and Si availability was determined by spectrophotometry after extraction with 0.01 mol L^{-1} CaCl_2 [36].

2.5. Statistical Analysis

All data were initially analyzed via the Shapiro–Wilk test [37] for normality and the Levene’s test for homoscedasticity [38], both at $p < 0.05$; the UNIVARIATE procedure of SAS version 9.4 was used for the analysis (SAS Institute, 2015). The data were also tested for sphericity using the Bartlett test [39] via the FACTOR procedure of SAS version 9.4 [40]. The results indicated that all data were distributed normally ($W \geq 0.90$) and exhibited no sphericity. The data were then subjected to analysis of variance and polynomial regression analysis to construct TM dose–response curves for the measured soybean and soil traits using significant regression equations with the highest coefficients of determination. A heatmap was built using the Pearson correlation coefficients ($p \leq 0.05$) among variables, and only significant correlations were shown. Principal component analysis (PCA) was performed using the statistical software Canoco v. 4.5.

3. Results

As the TM dose increased, the Mg and Si concentrations in maize leaves also increased ($p \leq 0.05$) in both clayey and sandy soils (Figure 2). When maize was established in clayey soil at the highest TM dose, the leaf Mg concentration reached 3.44 g kg^{-1} (Figure 2A), and the leaf Si concentration reached 15.42 g kg^{-1} (Figure 2B). In sandy soil, at the highest TM dose, the leaf concentrations of Mg and Si reached 3.76 and 15.55 g kg^{-1} (Figure 2C,D), respectively.

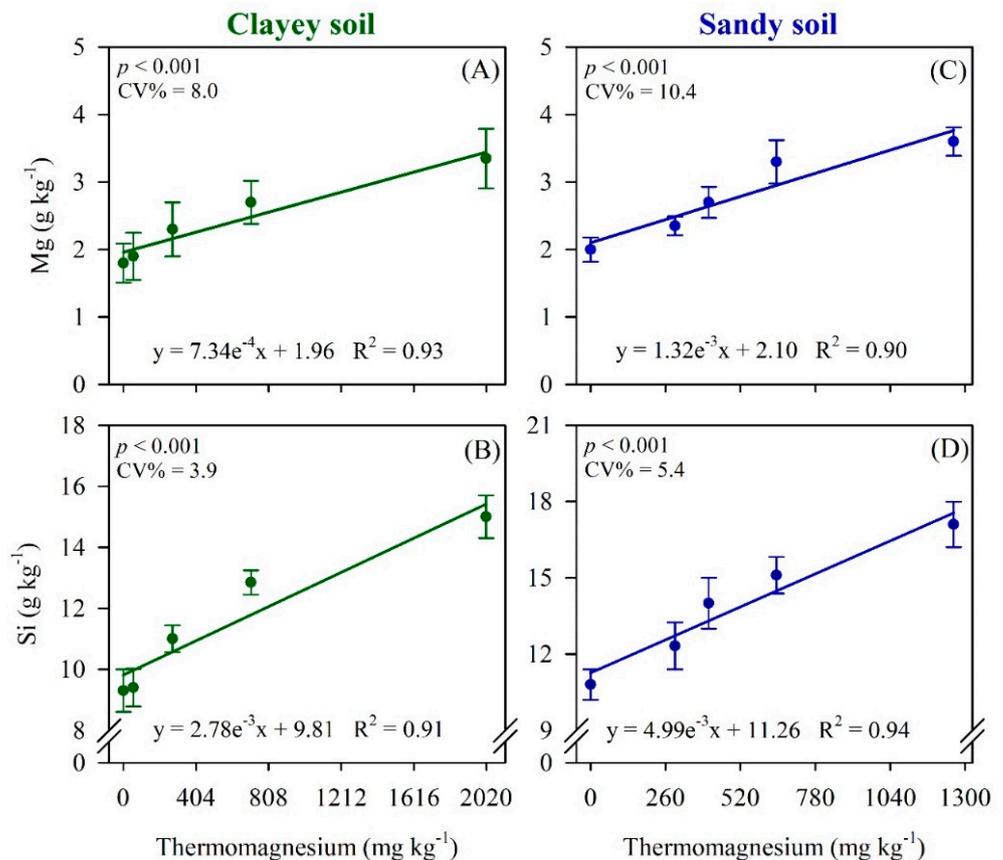


Figure 2. Leaf concentrations of (A,C) Mg and (B,D) Si in maize grown in clayey soil and sandy soil as a function of Thermomagnesium dose.

The concentrations of sugars in maize leaves also responded to the TM dose (Figure 3). The reducing sugars increased ($p \leq 0.05$) linearly up to the highest TM dose, regardless of soil texture, reaching concentrations of 76.76 g kg^{-1} in clayey soil (Figure 3A) and 83.76 g kg^{-1} in sandy soil (Figure 3D). By contrast, the concentrations of sucrose and starch decreased ($p \leq 0.05$) proportionally with the increasing TM dose, reaching their lowest values of 42.17 g kg^{-1} and 44.25 g kg^{-1} , respectively, in clayey soil (Figure 3B,C) and 54.07 and 53.21 g kg^{-1} , respectively, in sandy soil (Figure 3E,F) at the highest TM dose.

The yield parameters and grain yield of the maize increased ($p \leq 0.05$) with increasing the TM dose in both soils (Figure 4). The shoot dry matter reached 139 g plant^{-1} in clayey soil (Figure 4A) and 138 g plant^{-1} in sandy soil (Figure 4E). In both soils, the mean number of grains as a function of TM dose fit a quadratic equation ($p \leq 0.05$). In clayey soil, the maximum grain yield per ear (418 grains) of maize occurred at a maximum technical efficiency dose (MTED) of 349 mg kg^{-1} (Figure 4B), whereas in sandy soil, the MTED was 490 mg kg^{-1} to produce a maximum of 421 grains per ear (Figure 4F). The maize W100G and grain yield increased ($p \leq 0.05$) linearly with the TM dose in both soils. At the highest TM dose, the W100G was 28.04 g in clayey soil, and 28.93 g in sandy soil, while the grain yield was $115.4 \text{ g plant}^{-1}$ and 124 g plant^{-1} (Figure 4C,G,D,H, respectively).

Increasing the TM dose also altered the chemical attributes ($p \leq 0.05$) of both soils (Figure 5). The pH increased linearly with the TM dose in clayey soil, reaching pH 5.61 at the highest dose (Figure 5A), whereas in sandy soil, the maximum pH was 5.75 at an MTED of 965 mg kg^{-1} (Figure 5D). Similar to the patterns of the leaf concentrations of Mg and Si, the levels of these elements in soil increased linearly with the TM dose. At the highest TM dose, the available Mg^{2+} and Si levels reached $20.98 \text{ mmol}_c \text{ kg}^{-1}$ and 14.5 mg kg^{-1} , respectively, in clayey soil (Figure 5B,C), and $8.95 \text{ mmol}_c \text{ kg}^{-1}$ and 8.56 mg kg^{-1} , respectively, in sandy soil (Figure 5E,F).

Pearson's correlation analysis revealed different patterns of significant correlations ($p \leq 0.05$) of the parameters between clayey and sandy soils (Figure 6). For maize established in clayey soil, significant correlations were observed among all parameters, except the number of grains per ear (Figure 6A). Only sucrose and starch were negatively correlated with the other parameters, but were positively correlated with each other. However, in sandy soil, soil available Si correlated significantly only with reducing sugar and available Mg^{2+} (Figure 6B), and Mg^{2+} did not correlate with shoot dry matter and W100G. Interestingly, in both soils, the grain yield was significantly correlated with W100G, but not the number of grains per ear.

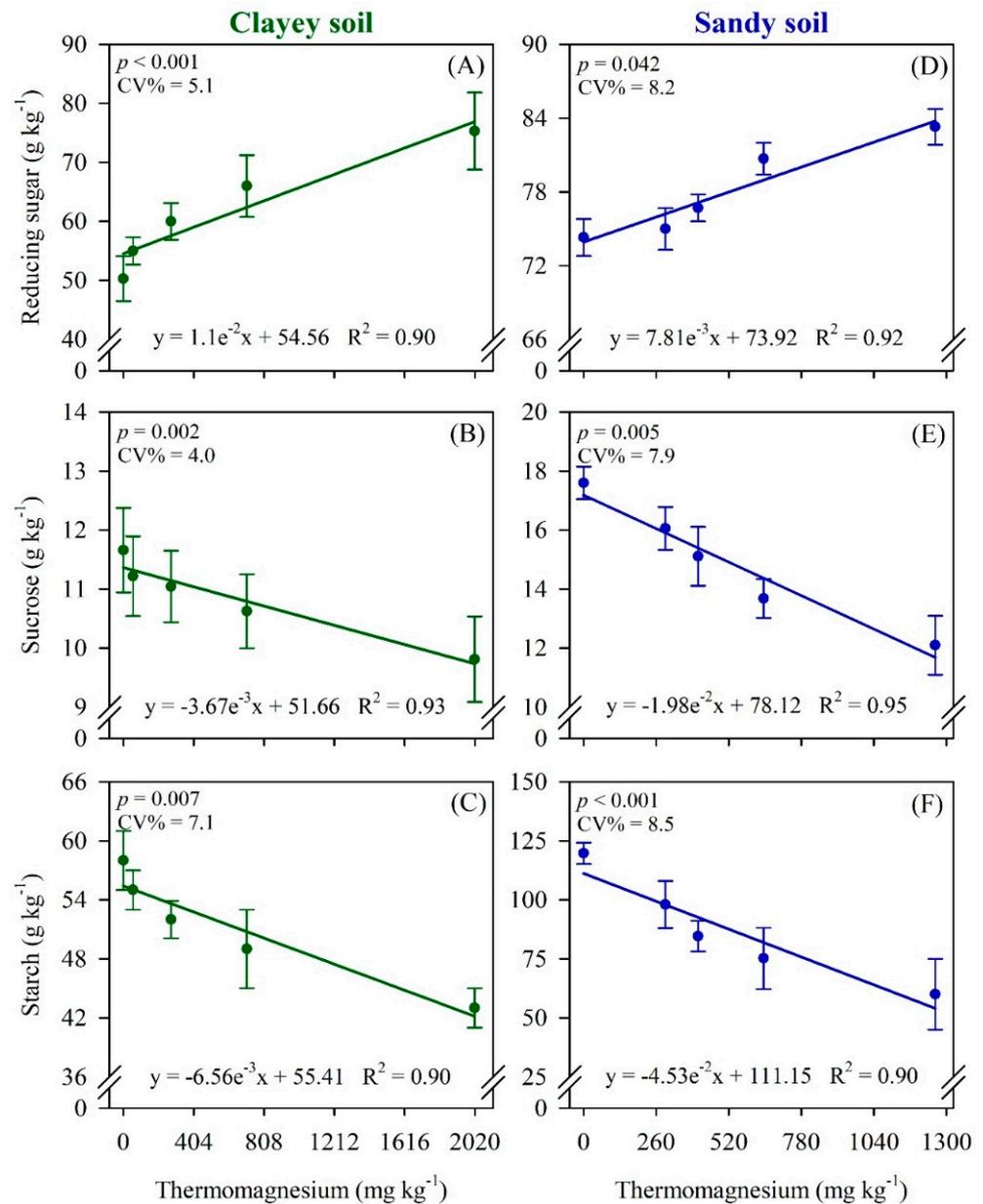


Figure 3. Leaf concentrations of (A,D) reducing sugars, (B,E) sucrose, and (C,F) starch in maize grown in clayey soil and sandy soil as a function of the Thermomagnesium dose.

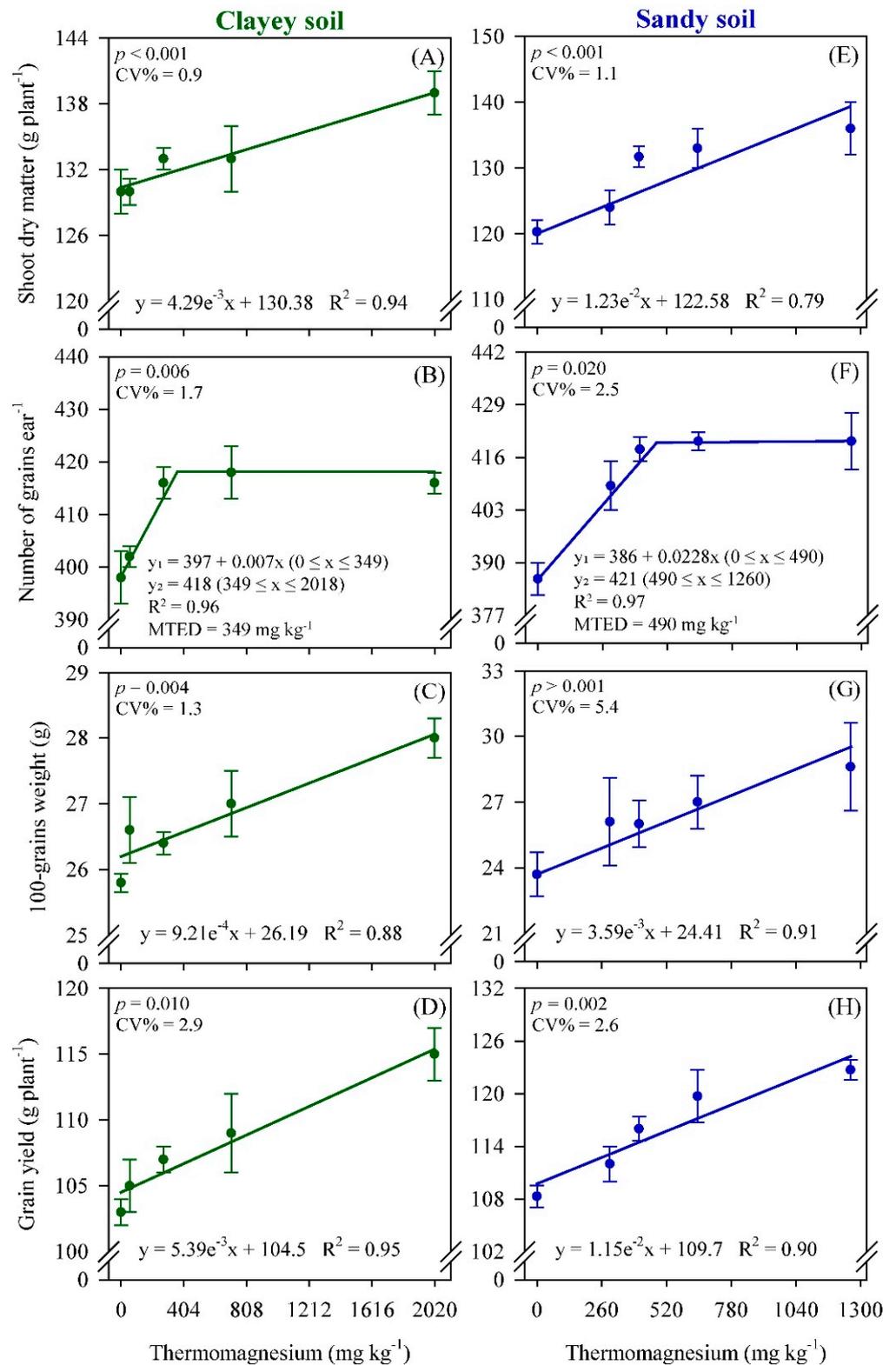


Figure 4. (A,E) Shoot dry matter, (B,F) number of grains per ear, (C,G) 100-grains weight, and (D,H) grain yield per plant of maize grown in clayey soil and sandy soil as a function of the Thermomagnesium dose.

PCA clearly segregated the TM doses according to the general similarity of effects on all of the evaluated parameters. In clayey soil, both PCA and PERMANOVA indicated a greater similarity between the two lowest TM doses (control and 55 mg kg⁻¹) and between the two subsequent doses (273 and 709 mg kg⁻¹). The highest dose (2018 mg kg⁻¹) showed unique behavior and was superior to the other doses, with the best responses of maize and soil parameters (Figure 6C). Similarly, in sandy soil, PCA showed a gradient of responses to TM dose (Figure 6D). PERMANOVA segregated the TM doses into three groups according to the general similarity of their effects on the following parameters: the control treatments of 293, 410, and 645 mg kg⁻¹, and the highest TM dose, 1260 mg kg⁻¹.

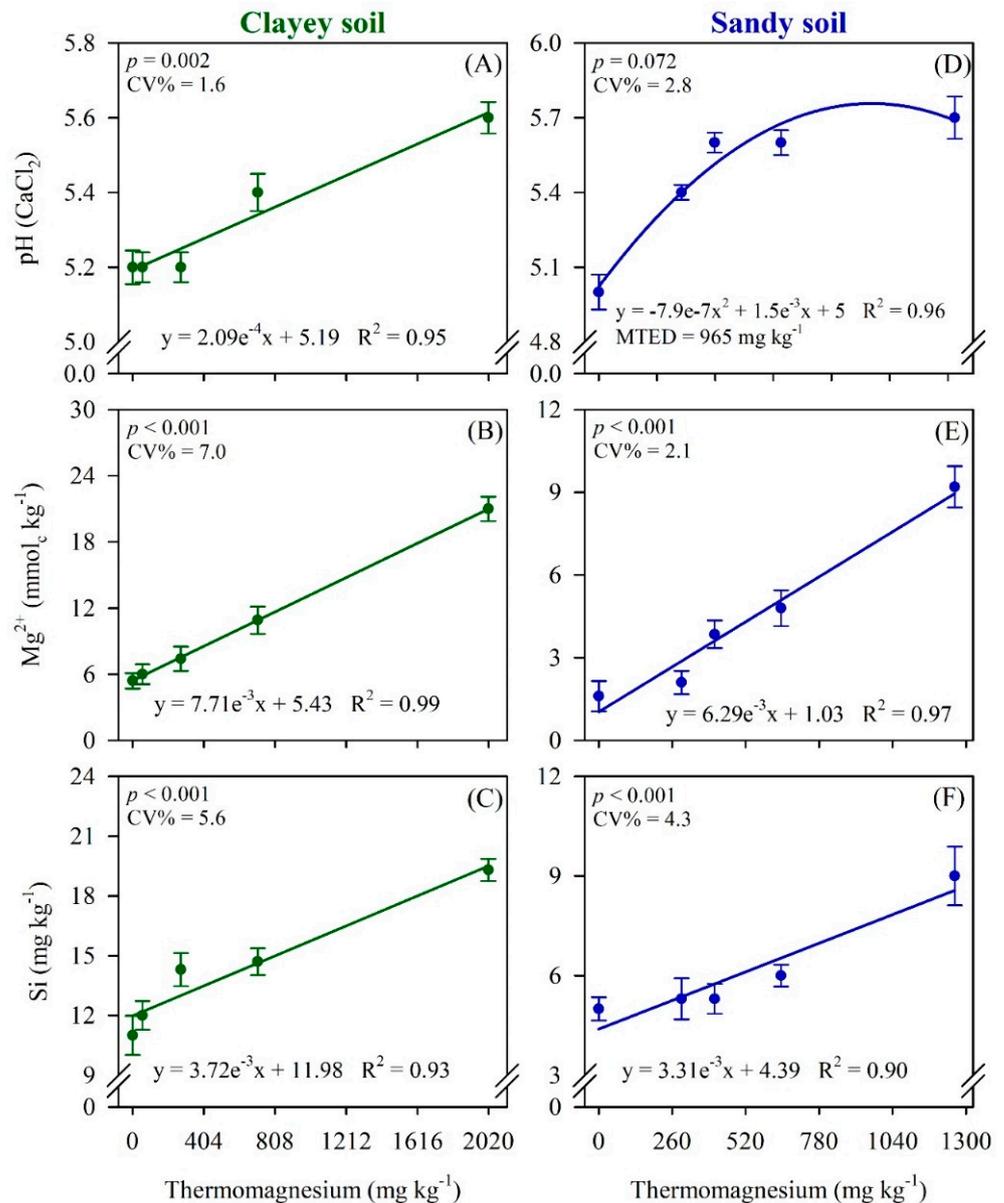


Figure 5. (A,D) Soil pH, (B,E) exchangeable Mg²⁺, and (C,F) Si availability in clayey and sandy soils as a function of the Thermomagnesium dose.

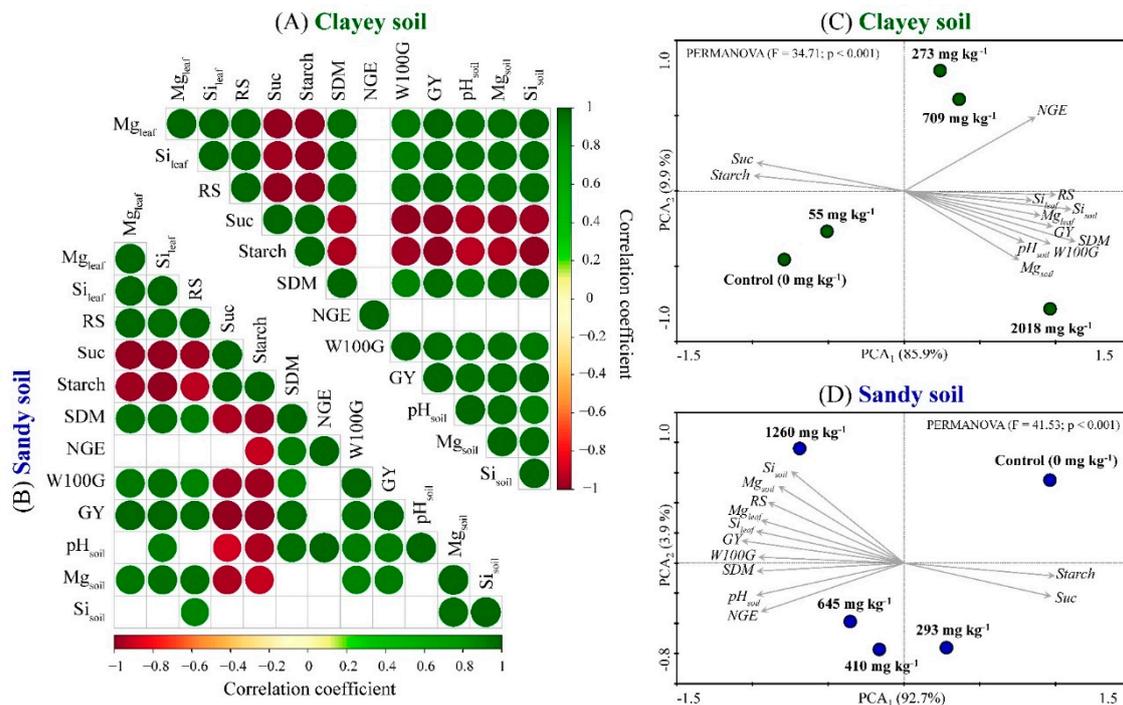


Figure 6. Heatmap of Pearson's correlation (A—Clayey soil, and B—Sandy soil) and principal component analysis (PCA) (C—Clayey soil, and D—Sandy soil) of soil and maize parameters. In the heatmap, only significant correlations at $p \leq 0.05$ are shown. Reducing sugar (RS), sucrose (Suc), shoot dry matter (SDM), number of grains per ear (NGE), weight of 100 grains (W100G), and grain yield (GY).

4. Discussion

The vast majority of studies of the reuse of industrial waste are limited to examining the petrography and mineralogy of these by-products and/or their potential use as nutrient remineralizers in soil [13,41–43]. By contrast, the present study investigated the direct impacts of supplying a by-product of Ni ore mining (TM) as a clean and sustainable fertilizer capable of improving plant metabolism and, consequently, maize grain yield.

Agricultural repurposing of TM as a Si–Mg fertilizer efficiently increased the levels of these elements in both the soil and maize leaves, particularly at the highest dose, as confirmed by PCA. Mg is considered a “forgotten” element in agriculture [44,45], and almost all Mg applied for agriculture comes from liming [46–48]. Thus, by-products such as TM provide a sustainable alternative for supplying Mg for cropping systems [12]. Supplying Mg to maize may improve crop growth and yield because of the important roles of Mg in plant metabolism, especially in photosynthesis and the activation of numerous key chloroplast enzymes [22,44,49,50].

Unfortunately, the impact of supplying Si to crops has received little attention [51]. Si is an important nutrient for plant growth and is the only nutrient that is not harmful when over-absorbed by plants [43]. Si relieves abiotic stress [52] by forming physical barriers through deposition in the cell wall of the plant, and may also activate plant defense enzymes to reduce the production of reactive oxygen species (ROS) [53], in addition to increasing crop productivity and quality [54,55]. The benefits of Si for crops are most evident in accumulating plants, such as maize, rice, wheat, barley, and pumpkin [53,56–58]. Si application is particularly beneficial for these crops, as the active uptake of Si occurs via transporters located on the plasma membrane [59,60]. TM could be an important alternative source of silicate for production systems, especially in acidic soils, which are typically Si-deficient [61].

Thermomagnesium also has a high neutralizing power for soil acidity [12], as it is composed of 51% SiO_3^{2-} and 28% MgO, which have respective neutralization capacities

relative to CaCO_3 of 1.0- and 2.48-fold, respectively [62]. Moreover, TM is more soluble than lime and thus may mitigate soil acidity more efficiently, in addition to providing Mg and Si to the soil [63], as observed in the soil and maize leaf analyses. Soil pH is an important determinant of Mg and Si availability in soil. As the TM level increased, soil pH increased linearly in clayey soil, but according to a quadratic model in sandy soil. The latter behavior can be explained by the low buffering power of sandy soil; as the pH increases, the effect of the acidity-correcting agent decreases [64]. Because of the low buffering power of sandy soils [65], caution is required in order to avoid excessive applications of soil amendments that drastically change the soil pH, such as carbonate- and silicate-based amendments. Increasing the soil pH above the range considered suitable for most crops (pH (H_2O) 5.5–6.5) reduces the availability of some nutrients, especially cationic micronutrients, for plant uptake as result of hydroxylation and subsequent precipitation [48,66–68]. As clayey soil naturally has a higher colloidal cation retention capacity, the Mg and Si concentrations were higher in the clayey soil than in the sandy soil.

The improvements in soil pH and Si–Mg availability and in Si–Mg plant nutrition also enhanced carbohydrate metabolism, mainly at higher doses, as supported by PCA. Like potassium (K), Mg plays prominent roles in processes that are strongly associated with the photosynthesis and translocation of photosynthates [20,69]. Plant growth and metabolism are dependent on the translocation of carbohydrates from source-to-sink organs [20,45]. Sucrose is the main form of carbohydrate transported in plants, via phloem, and is greatly affected by Mg levels [22,44]. Mg scarcity reduces the efficiency of long-distance carbon transport [17], resulting in the accumulation of high levels of carbohydrates in leaves. The accumulation of carbohydrates leads to an increase in the production of reactive oxygen species (ROS) in chloroplasts, which limits photosynthetic efficiency through a negative feedback effect, further depriving sink organs of carbohydrates [17,44].

The dynamics among the concentrations of reducing sugars, sucrose and starch, in the leaves of the plant change depending on the availability of intracellular Mg [70]. In the presence of light, sucrose is transported from the leaf's cytosol to vascular tissues, and starch accumulates in the chloroplasts; at night, this starch serves as a carbon skeleton source for sucrose synthesis [71]. Reducing sugars are produced via carbon fixation, which can be increased by supplying Mg [70]. Our results suggest that along with the increase in the partitioning of photoassimilates, the photosynthetic activity and the production of reducing sugars increased. The enhanced partitioning of more complex sugars (sucrose and starch) to vascular organs reduces the concentrations of these sugars in the source leaves [17]. Our findings indicate that the translocation of sugars to vascular organs occurs efficiently, and that the plant's photosynthetic metabolism quickly replenishes these sugars to maintain high levels of reducing sugars in the cytosol. Similarly, the starch formed in the chloroplasts reduces the translocation of photosynthates because of their consumption as a source of carbon skeletons during respiration [22,71].

The observations of the vegetative and reproductive development of maize plants corroborated the results of the nutritional and carbohydrate analyses, with better results at the highest TM dose. This conclusion was also supported by PCA. Energy reserves are essential for cell multiplication by the plant. Thus, Si–Mg supplementation of maize ensures greater efficiency of the photosynthetic metabolism. Magnesium acts directly in these processes, starting as a component of the chlorophyll molecule, through the activation of Rubisco in the process of carboxylation, and finally the partitioning of photoassimilates by the plant organs [17,22,45]. The plant converts this energy into dry matter accumulation during the vegetative stage and in grain filling in the reproductive stage. The number of grains per ear, W100G, and grain yield per plant were strongly affected by the TM application. A low response to the highest TM doses occurred for the number of grains per ear. The increase in these values may be related to the greater transport of photoassimilates to the ear during grain formation; however, the grain weight seemed to be the most determining factor of the grain yield. In both soil textures, the correlation analysis showed a strong correlation of grain yield with W100G, which is determined by the assimilation

transport rate during the grain filling period [72,73]. The direct translocation of assimilates and the redistribution of the reserve pool of assimilates contribute to grain filling [22]. Starch is the most abundant component of cereal grains, and a reduced carbohydrate content or low translocation of this carbohydrate to the reproductive organs will reduce the grain weight [74]. Thus, reduced carbohydrate translocation to grains as a result of a low Mg supply will directly reduce the grain yield. In addition to the benefits of TM as a result of the supply of Mg, the supply of Si by TM significantly enhances plant productive responses [53,56]. Si substantially improves the foliar architecture of plants, especially monocotyledons [59], favoring light interception, with a consequent increase in photosynthesis [56].

5. Conclusions

Our study confirmed that TM is yet another suitable and clean alternative fertilizer for agricultural systems that can improve soil pH and actively provide Si and Mg to soil and plants, regardless of soil texture, especially when applied at larger doses. Maize plants established in soils amended with TM were more metabolically active, redistributed their photosynthates appropriately among plant organs, and consequently reached a higher grain yield potential. Given these promising results, the application of TM as a new form of fertilizer should be considered for a variety of crops, soil conditions, and agricultural systems worldwide in order to help solve problems related to the disposal of by-products of mining, while simultaneously decreasing the consumption of highly soluble fertilizers.

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