

Supplementary Materials: A Clean Process to Prepare High-Quality Acid-Soluble Titanium Slag from Titanium Middling Ore

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Table S1. List of the preparation methods of synthetic rutile.

Feedstock	Method	Reference
Sulfated ilmenite	Selective thermal decomposition (500–560 °C)-sulfuric acid leaching (15 wt %)-NaOH leaching (5 wt %)	[7]
Ilmenite	Roasting with $(\text{NH}_4)_2\text{SO}_4$ (360 °C)-selective thermal decomposition (480 °C)-hydrochloric acid leaching (2.5 wt %)-NaOH leaching (5 wt %)	[9]
Titinium slag	Microwave roast (700–1100 °C)	[10–12]
Ilmenite	Anion-exchange (IRA 410 Cl)	[13]
Titinium slag	Molten modification (2% B_2O_3 amount)-hydrochloric acid leaching (4 wt %)-NaOH leaching (8 wt %)	[16]
Titinium slag	Soda ash roasting (850–900 °C)-water leaching-acid leaching-alkali leaching	[8,15]
Titinium slag	H_3PO_4 oxidation (1000 °C)-reduction (800 °C)-hydrochloric acid (20 wt %)-NaOH leaching	[14]
Titinium slag	Oxidizing with additives- hydrochloric acid leaching (6–8 wt %)	[17]
Ilmenite	Partial reduction-acid leaching	
Ilmenite	Full reduction-corrosion	

Table S2. Verification experiments of predominant species (L/S 4).

pH of Leaching Solution	Leaching Temperature (°C)	Leaching Time (h)	Reactant	Whether Reactant Dissolves
1	25	1	Al_2O_3	Yes
3	25	1	Al_2O_3	No
0.5	100	1	Al_2O_3	Yes
1.5	100	1	Al_2O_3	No
14	100	1	Al_2O_3	Yes
10	25	1	MgO	Yes
12	25	1	MgO	No
7	100	1	MgO	Yes
9	100	1	MgO	No
1	25	1	VO_2	Yes
3	25	1	VO_2	No
0.5	100	1	VO_2	Yes
1.5	100	1	VO_2	No
8	25	3	CaTiO_3	No
10	25	3	CaTiO_3	No
6	100	3	CaTiO_3	No
8	100	3	CaTiO_3	No
3	25	3	FeTiO_3	No
5	25	3	FeTiO_3	No
2	100	3	FeTiO_3	No
4	100	3	FeTiO_3	No
5	25	3	MnTiO_3	No
7	25	3	MnTiO_3	No
4	100	3	MnTiO_3	No
6	100	3	MnTiO_3	No

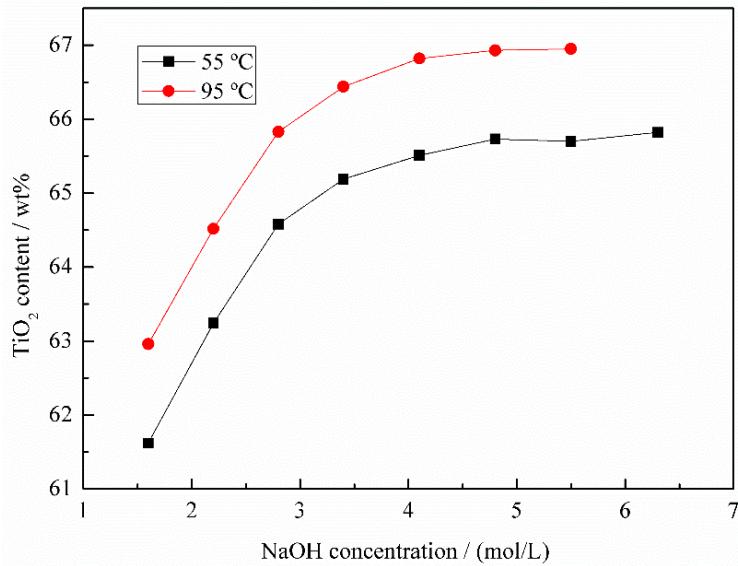


Figure S1. Effects of NaOH concentration on TiO₂ content at 55 °C and 95 °C.

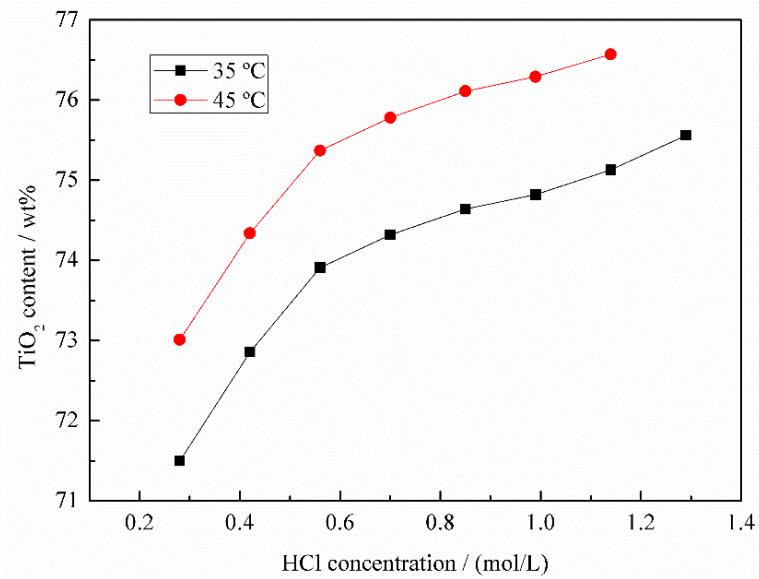


Figure S2. Effects of HCl concentration on TiO₂ content at 35 °C and 45 °C.

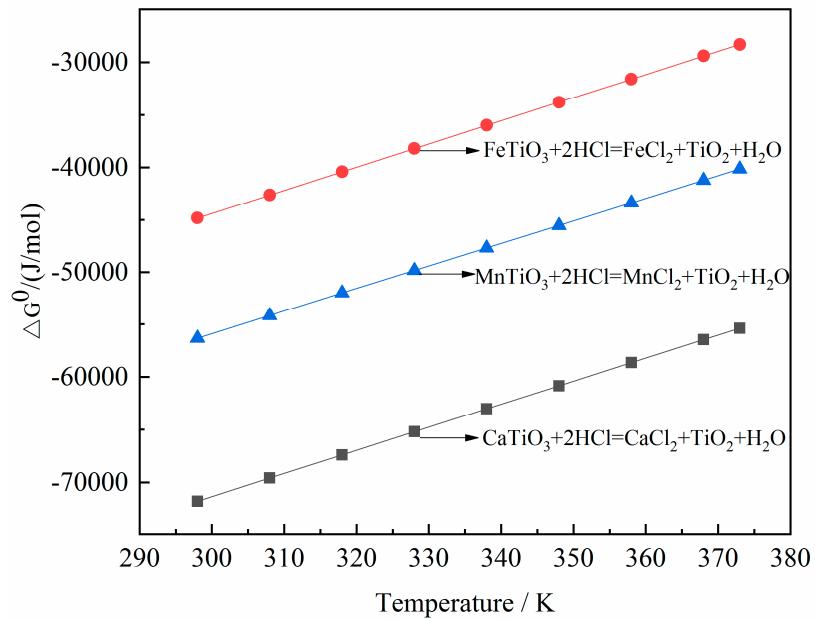


Figure S3. Standard Gibbs free energy change of reactions.



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