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Thermodynamic and Exergoeconomic Analysis of a Supercritical CO₂ Cycle Integrated with a LiBr-H₂O Absorption Heat Pump for Combined Heat and Power Generation

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Abstract: In this paper, a novel combined heat and power (CHP) system is proposed in which the waste heat from a supercritical CO₂ recompression Brayton cycle (sCO₂) is recovered by a LiBr-H₂O absorption heat pump (AHP). Thermodynamic and exergoeconomic models are established on the basis of the mass, energy, and cost balance equations. The proposed sCO₂/LiBr-H₂O AHP system is examined and compared with a stand-alone sCO₂ system, a sCO₂/DH system (sCO₂/direct heating system), and a sCO₂/ammonia-water AHP system from the viewpoints of energy, exergy, and exergoeconomics. Parametric studies are performed to reveal the influences of decision variables on the performances of these systems, and the particle swarm optimization (PSO) algorithm is utilized to optimize the system performances. Results show that the sCO₂/LiBr-H₂O AHP system performs better than sCO₂/DH system and sCO₂/ammonia-water AHP system do, indicating that the LiBr-H₂O AHP is a preferable bottoming cycle for heat production. The detailed parametric analysis, optimization, and comparison results may provide some references in the design and operation of sCO₂/AHP system to save energy consumption and provide considerable economic benefits.

Keywords: supercritical CO₂ cycle; absorption heat pump; LiBr-H₂O solution; parametric study; optimization

1. Introduction

In the past few decades, energy crisis and environment problems became serious because of the increasing energy demand and the rapid economic development all over the world. Many efforts have been devoted into developing advanced energy conversion technologies to relieve the current challenging energy situation. For the existing energy sources (such as fossil fuels, solar, biomass, geothermal energy, and nuclear energy) and industrial waste heat sources, various novel energy conversion systems were proposed in place of the conventional ones to improve energy utilization [1]. Among these proposed systems, the supercritical CO_2 power cycle (s CO_2) is considered to be a promising technology with great potential and competiveness owing to its advantages of a compact structure, environmental friendliness and high efficiency [2–4]. The s CO_2 power cycle operates above the critical point of CO_2 (31.3 °C, 7.39 MPa). Due to the dramatic changes of thermodynamic properties near the critical point, the inlet state of its compressor is always designed just above the critical point so that the compressor work can be reduced significantly. Hence, cooling the CO_2 before compression



process can be beneficial to the efficiency improvement [5–7]. However, a huge amount of heat is inevitably rejected by the cooler during the cooling process. So, it is worthwhile to reuse this low-grade heat energy through waste heat recovery systems to improve the performance of the sCO₂ cycle [8,9].

Various waste heat power generation systems were adopted to recover the waste heat of the sCO_2 cycle. Chacartegui et al. [10] applied an Organic Rankine Cycle (ORC) to reuse this waste heat for producing electric power, and compared the combined sCO₂/ORC system with two stand-alone closed CO_2 cycles. They concluded that the addition of ORC could improve the thermal efficiency by 7–12%. Akbari and Mahmoudi [11] conducted a thermoeconomic analysis for a combined recompression sCO₂/ORC system. Compared to the stand-alone sCO₂ cycle, the exergy efficiency increased by up to 11.7% and the total product unit cost decreased by up to 5.7% for the combined sCO_2/ORC system. Wang and Dai [12] compared a transcritical CO_2 cycle (t CO_2) with an ORC as the bottoming cycle for a recompression sCO₂ cycle. The sCO₂/tCO₂ system showed a better performance than the sCO₂/ORC system at a lower pressure ratio, while the latter had a slightly lower total product unit cost than the former. Besarati and Goswami [13] chose ORC as the bottoming cycle for a simple sCO₂ cycle, a recompression sCO_2 cycle, and a partial cooling sCO_2 cycle. It was concluded that the recompression sCO₂/ORC system presented the maximum combined cycle efficiency. Some researchers also integrated a Kalina cycle with the sCO₂ cycle to enhance the overall system performances. Li et al. [14] proposed a combined recompression sCO₂/Kalina cycle, and found out that the total product unit cost and exergy efficiency of the combined cycle were 5.5% lower and 8.02% higher than those of the sCO₂ cycle. Mahmoudi et al. [15] studied the thermodynamic and economic performances for a stand-alone sCO₂ cycle and a combined sCO₂/Kalina cycle. Results showed that combining the Kalina cycle with the sCO₂ cycle could reduce the exergy destruction significantly.

In addition to the combined power generation systems above, establishing combined power and heat/cooling systems can also enhance the overall performance for the sCO₂ cycle. In the combined power and heat/cooling systems, the low-grade waste heat of the topping cycle is transformed into cooling energy or to produce heat in bottoming cycle, which can obtain better gains than transforming high-grade electric energy into low-grade heat/cooling energy because of the lower energy conversion efficiency in the electric power generation process [16]. The absorption refrigeration cycle (ARC) is widely integrated with the sCO₂ cycle to provide electric power and cooling energy simultaneously. Wu et al. [17] investigated an combined sCO₂ cycle/ammonia-water based ARC. The thermal efficiency, exergy efficiency, and total product unit cost of the combined cycle were 26.12% and 2.73% higher, and 2.03% lower than those of the stand-alone sCO₂ cycle, respectively. Li et al. [18] coupled the lithium bromide-water ARC with a recompression sCO_2 cycle, and compared it with the recompression sCO₂ cycle/ammonia-water ARC. The single-objective optimization results showed that the sCO₂/LiBr-H₂O ARC system had a better performance than the sCO₂/ammonia-water ARC system. Recently, Balafkandeh et al. [19] developed a tri-generation system by using biomass energy based on a sCO₂ cycle and a LiBr-H₂O ARC. Compared to the sCO₂ cycle, the proposed combined cycle presented a large performance improvement in terms of efficiency and environmental impacts.

A number of studies proposed combined heat and power (CHP) systems based on sCO_2 cycle to achieve heat and electric power cogeneration. Zhang et al. [20] investigated a solar energy powered sCO_2 Rankine cycle, in which the thermal energy of recuperators was recovered by the heating water to provide heat for users directly. Moroz et al. [21] studied several types of sCO_2 cycles in a CHP plant, and compared these sCO_2 cycles with steam CHP systems. They concluded that the sCO_2 cycle should be considered as a base for future CHP plants due to its excellent performances. However, in previous sCO_2 -based CHP systems, the waste heat of the sCO_2 cycle was directly supplied to heat users. The temperature of waste heat is much higher than the heating temperature of heat users, which is inconsistent with the principle of energy cascade utilization. Thus, direct heating is not a high-efficiency method to sufficiently utilize the waste heat of the sCO_2 cycle. Compared with the conventional direct heating (DH) systems, the advantage of heat amplification makes absorption heat pump (AHP) a preferable choice that can produce more heat to satisfy the user demand [22], especially

when the waste heat of the sCO_2 cycle is limited or insufficient for users. However, very limited efforts were devoted to comprehensively analyzing the feasibility of AHP to recover the waste heat of the sCO_2 cycle based on the principle of energy cascade utilization. Consequently, the purpose of this study is to propose a combined sCO_2 /AHP system for high-efficiency heat and power cogeneration, and to compare it with existing systems to present its advantages quantitatively.

On the other side, the most frequently used working fluids for AHP are an ammonia-water solution and a LiBr-H₂O solution. Both of them were frequently considered for comparison and discussion [23,24]. The LiBr-H₂O AHP has a smaller operational pressure and can be easier to achieve than the ammonia-water AHP. Besides, if the working fluid leaks, the water vapor in LiBr-H₂O AHP is much safer than the ammonia vapor in ammonia-water AHP. Therefore, the LiBr-H₂O AHP will be the main focus of this study while the ammonia-water AHP will be adopted as a comparison. To the best of the authors' knowledge, the LiBr-H₂O AHP has never been applied as a bottoming cycle to recover the waste heat of the sCO₂ cycle. Thus, a combined sCO₂/LiBr-H₂O AHP system is first proposed in this study as a novel high-efficiency CHP system.

In this study, a preliminary design and analysis of a CHP system is carried out, in which the topping cycle is a sCO₂ cycle driven by a nuclear reactor to generate electric power and the bottoming cycle is a LiBr-H₂O AHP to recover the waste heat of topping cycle for producing heat. Firstly, the sCO₂/LiBr-H₂O AHP system is proposed and investigated in terms of energy, exergy, and exergoeconomics. A single sCO₂ cycle, a sCO₂/DH system, and a sCO₂/ammonia-water AHP system are compared with the sCO₂/LiBr-H₂O AHP system in order to show its advantages. Then, parametric analysis is performed to reveal the influences of several key system parameters, namely the turbine inlet temperature, compressor pressure ratio, generator temperature and evaporator temperature, on the power generation, heating, and overall performances for these systems. Finally, performances of these systems are optimized and then compared by utilizing the particle swarm optimization (PSO) algorithm.

2. System Description and Assumptions

Figure 1a,b depict the schematic diagrams of the proposed $sCO_2/LiBr-H_2O$ AHP system (supercritical CO₂ recompression Brayton cycle/LiBr-H₂O absorption heat pump system) and a sCO_2/DH system (supercritical CO₂ recompression Brayton cycle/direct heating system), respectively. In the sCO_2/DH system, a conventional direct heating system is applied to recover the waste heat in the cooler of the sCO_2 cycle. The waste heat is supplied to heat users through a direct heat exchanger (DHE). Thus, the sCO_2/DH system consists of a turbine, a main compressor (MC), a recompression compressor (RC), a high-temperature recuperator (HTR), a low-temperature recuperator (LTR), a cooler, and a DHE, as shown in Figure 1b.

The following Figure 1 shows the schematic diagrams of combined heat and power systems: (a) the sCO_2/AHP system; (b) the sCO_2/DH system.

Differently in the sCO₂/LiBr-H₂O AHP system, a part of the waste heat in the cooler is transferred to a LiBr-H₂O based AHP, so that the heat amount can be amplified to produce more heat for users, especially when the waste heat of the sCO₂ cycle is insufficient to satisfy the demand. The AHP includes a generator, an absorber, an evaporator, a condenser, a pump, and a solution heat exchanger (SHE), as shown in Figure 1a.

As both LiBr-H₂O solution and ammonia-water solution are frequently utilized as working fluids, a sCO₂/ammonia-water AHP system is established in which an ammonia-water AHP is coupled with the sCO₂ cycle, so as to conduct a comparative study with the proposed sCO₂/LiBr-H₂O AHP system. As the structure and main components are the same for both combined systems, Figure 1a also describes the sCO₂/ammonia-water AHP system.



Figure 1. Schematic diagrams of combined heat and power systems: (a) sCO₂/AHP system; (b) sCO₂/DH system.

2.1. Working Process of the sCO₂/AHP System

As can be seen in Figure 1a, the sCO_2/AHP system consists of four parts. The working process for each part is introduced below,

- 1. sCO₂ topping cycle
 - (a) The CO₂ working fluid (stream 4) absorbs heat from the reactor (stream 5) and expands in a turbine to drive an electric generator;
 - (b) After expansion, the vapor exhaust (stream 6) flows into the HTR to heat stream 3, and then the outlet working fluid (stream 7) flows into the LTR to heat stream 2;
 - (c) The working fluid after releasing heat (stream 8) is split into 8a and 8b;
 - (d) The stream 8a releases a part of heat to drive the AHP bottoming cycle, and then (stream 9) flows into the DHE (stream 10) and the cooler before compression (stream 1) in the main compressor and afterwards (stream 2) flows into the LTR to be heated (stream 3a);
 - (e) The stream 8b is compressed directly to stream 3b;
 - (f) The compressed stream 3a and stream 3b are mixed to stream 3 and then heated to stream 4 before moving into the reactor.

2. AHP bottoming cycle

- (a) The diluted solution A3 (or ammonia-rich solution) absorbs heat in the SHE to state A4, and then is heated by stream 8a in the generator and separated into a strong solution A7 (or ammonia-poor solution) and a vapor stream A8;
- (b) The stream A8 is cooled to the saturated liquid (stream A9) in the condenser, throttled by the valve 1, and then heated to saturated vapor (stream A1) in the evaporator. It is noted that the heat release in the condenser is applied to reheat the heating water;
- (c) The strong solution A7 (or ammonia-poor solution) releases heat in the SHE and decompresses through the valve 2 to a low-pressure solution A5;
- (d) The strong solution A5 (or ammonia-poor solution) absorbs the vapor A1 in the absorber. The merged diluted solution (or ammonia-rich solution) is cooled by the heating water, and pressured by the pump to the high-pressure solution A3.

3. Heating water cycle

- (a) The low-temperature heating water from heat users is divided into stream H1 and stream H4;
- (b) The stream H1 obtains heat from the absorber and the condenser of AHP to state H3;
- (c) The stream H4 is heated by the CO₂ working fluid (stream 9) to state H5, then mixed with stream H3 to the high-temperature heating water H6, and finally supplied to heat users.
- 4. Cooling water cycle

The cooling water (stream 11) absorbs heat in the cooler (stream 12), and then a part of it (stream 13) flows into the evaporator to heat the working fluid of AHP and finally back to the cooling tower, while another part of cooling water flows back directly to the cooling tower.

2.2. Working Process of the sCO₂/DH System

The working process of the sCO_2 cycle in the sCO_2/DH system is same as that in Section 2.1. The only difference is that the stream 8a in Figure 1b releases heat only in the DHE, and all the heat release before the cooler is supplied to the heat users directly through the DHE.

2.3. Assumptions

The main assumptions for this study are listed as below,

- (a) The systems are assumed to operate at a steady state, and the off-design performance or dynamic performance is not considered in this study;
- (b) The variation of kinetic and potential energy is neglected;
- (c) The pressure losses and heat losses of pipes and heat exchangers are neglected [25];
- (d) Isentropic efficiencies are assumed for the turbine, pump and compressors [26];
- (e) The liquid working fluid exiting the condenser (stream A9) and the generator (stream A7) is saturated liquid, while a subcooled degree of 3K is assumed at the outlet of absorber (stream A2) [27];
- (f) Vapor exiting the generator is assumed to be pure ammonia for the sCO₂/ammonia-water AHP system [27,28];
- (g) A temperature difference or an effectiveness is assumed in each heat exchanger [26].

3. System Modelings

3.1. Thermodynamic Model

The thermodynamic model for the combined cycle is developed according to the mass and energy balance equations of each component.

Thermodynamic relations and the effectiveness for HTR and LTR are expressed as Equations (1)-(4),

$$h_6 - h_7 = h_4 - h_3 \tag{1}$$

$$(h_3 - h_2)(1 - x) = h_7 - h_8 \tag{2}$$

$$\varepsilon_{HTR} = \frac{T_6 - T_7}{T_6 - T_3} \tag{3}$$

$$\varepsilon_{LTR} = \frac{T_7 - T_8}{T_7 - T_2}$$
(4)

where *x* is the mass separation ratio of stream 3b to stream 3, and ε_{HTR} and ε_{LTR} are the effectiveness of HTR and LTR, respectively.

For turbine, main compressor and recompression compressor, the isentropic efficiencies are calculated with Equation (5)–(7),

$$\eta_{\rm tur} = \frac{h_5 - h_6}{h_5 - h_{6s}} \tag{5}$$

$$\eta_{\rm mcom} = \frac{h_{2s} - h_1}{h_2 - h_1} \tag{6}$$

$$\eta_{\rm rcom} = \frac{h_{3s} - h_8}{h_3 - h_8} \tag{7}$$

where η_{tur} , η_{mcom} , and η_{rcom} are the isentropic efficiencies for the turbine, main compressor, and recompression compressor, respectively.

The thermodynamic relation for the cooler is calculated using Equation (8),

$$(1-x)m_{\rm CO_2}(h_{10}-h_1) = m_{cw1}(h_{12}-h_{11})$$
(8)

where m_{CO2} and m_{cw1} are the mass flow rates of CO₂ and the cooling water of the cooler, respectively.

The power output of the turbine and the consumptions of two compressors can be derived as Equations (9)–(11),

$$W_{\rm tur} = m_{\rm CO_2}(h_5 - h_6) \tag{9}$$

$$W_{\rm mcom} = (1 - x)m_{\rm CO_2}(h_2 - h_1) \tag{10}$$

$$W_{\rm rcom} = xm_{\rm CO_2}(h_3 - h_8).$$
 (11)

For the sCO₂/AHP system, the energy conservation equation of DHE is Equation (12),

$$Q_{\rm DHE} = (1 - x)m_{\rm CO_2}(h_9 - h_{10}) = m_{hw1}(h_{\rm H5} - h_{\rm H4})$$
(12)

while that for the sCO₂/DH system is Equation (13),

$$Q_{\rm DHE} = (1 - x)m_{\rm CO_2}(h_8 - h_{10}) = m_{hw1}(h_{\rm H6} - h_{\rm H4})$$
(13)

where m_{hw1} is the mass flow rate of the heating water flowing through the DHE. Q_{DHE} is the heat absorption of DHE from the topping cycle.

In the sCO_2/AHP system, the conservation equations for the AHP are expressed as follows. The thermodynamic relation for the generator is Equation (14),

$$Q_{\text{Gen}} = (1 - x)m_{\text{CO}_2}(h_8 - h_9) = m_{\text{A7}}h_{\text{A7}} + m_{\text{A8}}h_{\text{A8}} - m_{\text{A4}}h_{\text{A4}}$$
(14)

where Q_{Gen} is the heat absorption amount of generator from the topping cycle.

Assuming y as the concentration of LiBr or ammonia in the water, the concentration balance equation is obtained as Equation (15),

$$m_{\rm A4}y_{\rm A4} = m_{\rm A7}y_{\rm A7} + m_{\rm A8}y_{\rm A8}.$$
 (15)

The conservation equations for the condenser, evaporator, and absorber are obtained as Equations (16)–(18),

$$Q_{\text{Cond}} = m_{\text{A8}}h_{\text{A8}} - m_{\text{A9}}h_{\text{A9}} = m_{hw2}(h_{\text{H3}} - h_{\text{H2}})$$
(16)

$$Q_{\text{Eva}} = m_{\text{A1}}h_{\text{A1}} - m_{\text{A10}}h_{\text{A10}} = m_{cw2}(h_{13} - h_{14})$$
(17)

$$Q_{\rm Abs} = m_{\rm A1}h_{\rm A1} + m_{\rm A5}h_{\rm A5} - m_{\rm A2}h_{\rm A2} = m_{hw2}(h_{\rm H2} - h_{\rm H1})$$
(18)

where m_{hw2} is the mass flow rate of the heating water flowing through the AHP.

The conservation equation for SHE is obtained as Equation (19),

$$m_{A4}h_{A4} - m_{A3}h_{A3} = m_{A7}h_{A7} - m_{A6}h_{A6}.$$
(19)

For the pump, the isentropic efficiency and power consumption are defined according Equations (20) and (21),

$$\eta_{\rm pump} = \frac{h_{\rm A3s} - h_{\rm A2}}{h_{\rm A3} - h_{\rm A2}} \tag{20}$$

$$W_{\rm pump} = m_{\rm A3} h_{\rm A3} - m_{\rm A2} h_{\rm A2}.$$
 (21)

The net output power of the combined cycle can be calculated as Equation (22),

$$W_{\rm net} = W_{\rm tur} - W_{\rm rcom} - W_{\rm mcom} - W_{\rm pump}.$$
(22)

The heat amount supplied to the heat users, $Q_{h,users}$, and the heat absorption from the topping cycle for the purpose of heating, $Q_{h,absorb}$ were then defined for both combined systems. For the sCO₂/DH system, they are expressed as Equations (23) and (24),

$$Q_{h,users} = m_{hw1}(h_{\rm H6} - h_{\rm H4})$$
(23)

$$Q_{h,absorb} = (1 - x)m_{\rm CO_2}(h_8 - h_{10}).$$
⁽²⁴⁾

For the sCO₂/AHP system, they are expressed as Equations (25) and (26),

$$Q_{h,users} = (m_{hw1} + m_{hw2})(h_{\rm H6} - h_{\rm H1})$$
(25)

$$Q_{h,absorb} = Q_{\text{Gen}} + (1 - x)m_{\text{CO}_2}(h_9 - h_{10}).$$
(26)

Ignoring the variations in kinetic and potential exergies, the exergy of a stream consists of two parts, namely the chemical and physical exergy, as Equation (27),

$$E = E_{\rm ph} + E_{\rm ch}.\tag{27}$$

The physical exergy can be calculated with Equation (28),

$$E_{\rm ph} = m[(h - h_0) - T_0(s - s_0)].$$
⁽²⁸⁾

For the ammonia water in AHP, the chemical exergy can be determined with Equation (29) [29,30],

$$E_{\rm ph} = m \left[\left(\frac{y}{M_{\rm NH_3}} \right) e_{\rm ch, NH_3}^0 + \left(\frac{1-y}{M_{\rm H_2O}} \right) e_{\rm ch, H_2O}^0 \right]$$
(29)

where e_{ch,NH_3}^0 and e_{ch,H_2O}^0 are the standard chemical exergises of ammonia and water, respectively. For the LiBr-H₂O solution, the chemical exergy can be calculated according to reference [31].

Then, the exergy balance equation for each component in the combined systems can be obtained, as shown in Table 1. On this basis, the total exergy destruction of the combined system is expressed as Equation (30),

$$I_{\text{total}} = \sum I_{\text{components}}.$$
 (30)

The following Table 1 shows the exergy balance equations for sCO_2/AHP system and sCO_2/DH system.

System Components	sCO ₂ /AHP System	sCO ₂ /DH System
reactor	$E_{\rm core} + E_4 = E_5 + I_{\rm core}$	$E_{\rm core} + E_4 = E_5 + I_{\rm core}$
sCO ₂ turbine	$E_5 = E_6 + W_{\rm tur} + I_{\rm tur}$	$E_5 = E_6 + W_{\rm tur} + I_{\rm tur}$
HTR	$E_3 + E_6 = E_4 + E_7 + I_{\rm HTR}$	$E_3 + E_6 = E_4 + E_7 + I_{\rm HTR}$
LTR	$E_2 + E_7 = E_{3a} + E_8 + I_{LTR}$	$E_2 + E_7 = E_{3a} + E_8 + I_{LTR}$
main compressor	$E_1 + W_{\rm mcom} = E_2 + I_{\rm mcom}$	$E_1 + W_{\rm mcom} = E_2 + I_{\rm mcom}$
recompression compressor	$E_{8b} + W_{rcom} = E_{3b} + I_{rcom}$	$E_{\rm 8b} + W_{\rm rcom} = E_{\rm 3b} + I_{\rm rcom}$
cooler	$E_{10} = E_1 + I_{\text{cooler}}$	$E_{10} = E_1 + I_{cooler}$
DHE	$E_9 + E_{H4} = E_{10} + E_{H5} + I_{DHE}$	$E_{8a} + E_{H4} = E_{10} + E_{H6} + I_{DHE}$
generator	$E_{8a} + E_{A4} = E_9 + E_{A7} + E_{A8} + I_{Gen}$	/
absorber	$E_{A1} + E_{A5} + E_{H1} = E_{A2} + E_{H2} + I_{Abs}$	/
SHE	$E_{A3} + E_{A7} = E_{A4} + E_{A6} + I_{SHE}$	/
condenser	$E_{A8} + E_{H2} = E_{A9} + E_{H3} + I_{Cond}$	/
Evaporator	$E_{\rm A10} + E_{13} = E_{\rm A1} + E_{14} + I_{\rm Eva}$	/
Pump	$E_{A2} + W_{pump} = E_{A3} + I_{pump}$	/
Valve1	$E_{A9} = E_{A10} + I_{Valve1}$	/
Valve2	$E_{\rm A6} = E_{\rm A5} + I_{\rm Valve2}$	/

Table 1. Exergy balance equations for the sCO₂/AHP system and sCO₂/DH system.

3.2. Exergoeconomic Model

On the basis of the mass, energy and exergy balances above, the exergoeconomic analysis can be conducted to obtain the cost per unit exergy of product streams and to assess the combined systems. Firstly, all energy and exergy values should be determined, which has been achieved in the above section. Then, the fuel-production definition for the energy conversion system should be determined. For the sCO_2/AHP system and sCO_2/DH system in this study, the definitions are shown in Table 2. Finally, the cost balance equations can be constructed for all components. For the *k*th system component, the cost balance can be expressed with Equations (31)–(34) [32,33],

$$\sum \dot{C}_{out,k} + \dot{C}_{w,k} = \sum \dot{C}_{in,k} + \dot{C}_{q,k} + \dot{Z}_k$$
(31)

in which,

$$\dot{C} = c \cdot E \tag{32}$$

$$\dot{C}_{q,k} = c_{q,k} \cdot E_{q,k} \tag{33}$$

$$\dot{C}_{wk} = c_{wk} \cdot W \tag{34}$$

where C is the cost rate of the stream, while C_q and C_w are the cost rate of the heat transfer and power, respectively. Z_k represents the total cost rate of capital investment, operation and maintenance, which can be obtained as Equation (35),

$$\dot{Z}_k = \dot{Z}_k^{\text{CI}} + \dot{Z}_k^{\text{OM}}.$$
(35)

where \dot{Z}_k^{CI} denotes the annual levelized capital investment and \dot{Z}_k^{OM} denotes the annual levelized operation and maintenance cost. They can be obtained with Equations (36) and (37) [32],

$$\dot{Z}_{k}^{\rm OM} = \gamma_{k} Z_{k} / \tau \tag{36}$$

$$\dot{Z}_{k}^{\text{CI}} = \left(\frac{CRF}{\tau}\right) Z_{k} \tag{37}$$

where *CRF*, τ and γ represent the capital recovery factor, operating hours, and the maintenance factor, respectively. *CRF* is relative to the bank interest rate, which can be expressed as Equation (38),

$$CRF = \frac{i_r (1+i_r)^n}{(1+i_r)^n - 1}$$
(38)

where i_r is the interest rate, and n is the number of operation years. The values of i_r , n, τ , γ , and the expression of cost function Z_k for each component in this study are listed in Table 3.

The following Table 2 shows the fuel-product definitions for the combined systems.

Components	sCO ₂ /AH	P System	sCO ₂ /DH System		
components	Fuel Exergy	Product Exergy	Fuel Exergy	Product Exergy	
Reactor	$E_4 + E_{in}$	E_5	$E_4 + E_{in}$	E_5	
sCO_2 turbine	$E_{5} - E_{6}$	W_{tur}	$E_{5} - E_{6}$	W _{tur}	
HTR	$E_{6} - E_{7}$	$E_4 - E_3$	$E_{6} - E_{7}$	$E_4 - E_3$	
LTR	$E_7 - E_8$	$E_{3a} - E_2$	$E_7 - E_8$	$E_{3a} - E_2$	
Main compressor	W _{mcom}	$E_2 - E_1$	W _{mcom}	$E_2 - E_1$	
Recompression compressor	W _{rcom}	$E_{3b} - E_{8b}$	W _{rcom}	$E_{3b} - E_{8b}$	
Cooler	$E_{10} - E_1$	$E_{12} - E_{11}$	$E_{10} - E_1$	$E_{12} - E_{11}$	
DHE	$E_9 - E_{10}$	$E_{{ m H5}} - E_{{ m H4}}$	$E_{8a} - E_{10}$	$E_{\rm H6} - E_{\rm H4}$	
Generator	$E_{8a} - E_9$	$E_{\rm A7} + E_{\rm A8} - E_{\rm A4}$	/	/	
Absorber	$E_{\rm A1} + E_{\rm A5} - E_{\rm A2}$	$E_{\rm H2} - E_{\rm H1}$	/	/	
Condenser	$E_{A8} - E_{A9}$	$E_{\rm H3} - E_{\rm H2}$	/	/	
Evaporator	$E_{13} - E_{14}$	$E_{A1} - E_{A10}$	/	/	
SHX	$E_{A7} - E_{A6}$	$E_{A4} - E_{A3}$	/	/	
Pump	Wpump	$E_{A3} - E_{A2}$	/	/	
Valve1	\tilde{E}_{A9}	E_{A10}	/	/	
Valve2	E_{A6}	E_{A5}	/	/	

Table 2. Fuel-product definitions for the combined systems.

Table 3 shows the cost functions of the system components.

Component	Economic Parameters
Number of operation year (<i>n</i>)	20
Annual operation hours (τ)	8000
Interest rate (i_r)	0.12
Maintenance factor (γ)	0.06
Reactor	$Z_{\text{core}} = C_1 * Q_{\text{core}}, C_1 = 283 \text{\$}/\text{kW}_{\text{th}}$
sCO ₂ turbine	$Z_{\rm tur} = 479.34 \times m_{\rm in} \left(\frac{1}{0.93 - \eta_{\rm tur}}\right) \times \ln(PRc) \times \left(1 + e^{(0.036T_{\rm in} - 54.4)}\right)$
Compressor	$Z_{\text{mcom&rcom}} = 71.1 \times m_{\text{in}} \left(\frac{1}{0.92 - \eta_{\text{mcom&rcom}}}\right) \times PRc \times \ln(PRc)$
HTR, LTR, Cooler, DHE	$Z_k = 30 \times Mass_k$
Generator, Absorber, SHE, Condenser, Evaporator	$Z_k = Z_{ m ref} \Big(rac{A_k}{A_{ m ref}} \Big)^{0.6}$
Pump	$Z_{\text{pump}} = 1120 \times W_{\text{pump}}^{0.8}$

Table 3. Cost functions of the system components [11,32,34].

The mass and areas of heat exchangers in Table 3 are determined based on the heat exchanger design according to the mathematical models of different heat exchanger types. Compared to the conventional shell-and-tube heat exchanger, the Print Circuit Heat Exchanger (PCHE) shows higher heat transfer performances with a larger operation range (up to 980 °C and 96.5 MPa), which has been widely suggested and evaluated in previous studies [35–37]. Thus, PCHE is chosen as the heat exchangers (HTR, LTR, DHE, and cooler) in the SCO₂ topping cycle. Corresponding mathematical models for PCHE are established according some typical investigations [38,39], as listed in Table 4. Unlike in the SCO₂ topping cycle, the pressure and temperature in the AHP bottoming cycle are much lower, so that traditional high-efficiency heat exchangers are adopted in AHP for the generator, absorber, condenser, evaporator, and SHE. Table 4 describes the heat exchanger types and corresponding mathematical models, the heat exchange areas for all heat exchangers are calculated by using the enhanced logarithmic mean temperature difference method [40].

The following Table 4 shows the mathematical models for heat exchangers.

Components	Types	Fluids	Mathematical Models
HTR, LTR, DHE, cooler	PCHE	CO ₂	Gnielinski expression ^a
Generator	Nucleate pool boiling heat exchanger	CO ₂ LiBr-H ₂ O Ammonia water	Gnielinski expression ^a Jakob and Hawkin expression ^b Táboas's correlation ^c
Evaporator	Horizontal falling film heat exchanger	Cooling water Water (H ₂ O) Ammonia	Petukhov-Popov expression ^d Wilke's correlation ^e Lee expression ^f
Condenser	Horizontal tubes heat exchanger	Heating water Vapor	Petukhov-Popov expression ^d Nusselt expression ^g
Absorber	Horizontal falling film heat exchanger	LiBr-H ₂ O Ammonia water Heating water	Wilke's correlation ^e Lee expression ^f Petukhov-Popov expression ^d
SHE	Annular heat exchanger	Ammonia water or LiBr-H ₂ O	Gnielinski expression ^a

Table 4. Mathematical models for heat exchangers.

^a Gnielinski expression [38]; ^b Jakob and Hawkin expression [41]; ^c Táboas's correlation [42]; ^d Petukhov-Popov expression [43]; ^e Wilke's correlation [44]; ^f Lee expression [45]; ^g Nusselt expression [46].

Finally, Table 5 lists the cost balance equations for sCO₂/AHP system and sCO₂/DH system with auxiliary equations. The cost rate of cooling water is assumed to be zero as it is generally regarded

as a free resource. The electric power consumed by the pump in the AHP is assumed to come from the power production of the turbine. In addition, the capital investment cost at the present year is obtained according to the cost indices, namely the Chemical Engineering Plant Cost Index (CEPCI). The cost at present year can be obtained with Equation (39) [32],

Cost at present year = Original
$$\cos t \times \frac{\text{Cost index for present year}}{\text{Cost index for original year}}$$
. (39)

The Gauss-Seidel method is utilized to solve the linear system of equations in the Table 5 for two combined systems. After the solution, the cost rates for all exergy streams can be obtained. Table 5 shows the cost balance equations and auxiliary equations for the combined systems.

Components sCO₂/DH System sCO₂/AHP System $\dot{C}_5 = \dot{C}_4 + \dot{Z}_{reactor} + \dot{C}_{fuel}$ Reactor $\dot{C}_5 = \dot{C}_4 + \dot{Z}_{reactor} + \dot{C}_{fuel}$ $\dot{C}_6 + \dot{C}_{W_{\text{tur}}} = \dot{C}_5 + \dot{Z}_{\text{tur}}$ $\dot{C}_6 + \dot{C}_{W_{\text{tur}}} = \dot{C}_5 + \dot{Z}_{\text{tur}}$ sCO₂ turbine $\frac{\dot{C}_5}{E_5} = \frac{\dot{C}_6}{E_6}$ $\frac{\dot{C}_5}{E_5} = \frac{\dot{C}_6}{E_6}$ $\dot{C}_4 + \dot{C}_7 = \dot{C}_3 + \dot{C}_6 + \dot{Z}_{HTR}$ $\dot{C}_4 + \dot{C}_7 = \dot{C}_3 + \dot{C}_6 + \dot{Z}_{HTR}$ HTR $\frac{C_6}{E_6} = \frac{C_7}{E_7}, \dot{C}_3 = \dot{C}_{3a} + \dot{C}_{3b}$ $\frac{C_6}{E_6} = \frac{C_7}{E_7}, \dot{C}_3 = \dot{C}_{3a} + \dot{C}_{3b}$ $\dot{C}_8 + \dot{C}_{3a} = \dot{C}_2 + \dot{C}_7 + \dot{Z}_{\rm LTR}$ $\dot{C}_8 + \dot{C}_{3a} = \dot{C}_2 + \dot{C}_7 + \dot{Z}_{\text{LTR}}$ LTR $\frac{C_7}{E_7} = \frac{C_8}{E_8}, \dot{C}_{8a} = \dot{C}_8(1-x)$ $\frac{C_7}{E_7} = \frac{C_8}{E_8}, \dot{C}_{8a} = \dot{C}_8(1-x)$ $\dot{C}_{2} = \dot{C}_{1} + \dot{C}_{W_{\text{mcom}}} + \dot{Z}_{\text{mcom}}$ $\frac{\dot{C}_{W_{\text{mcom}}}}{W_{\text{mcom}}} = \frac{\dot{C}_{W_{\text{tur}}}}{W_{\text{tur}}}$ $\dot{C}_{2} = \dot{C}_{1} + \dot{C}_{W_{\text{mcom}}} + \dot{Z}_{\text{mcom}}$ $\frac{\dot{C}_{W_{\text{mcom}}}}{W_{\text{mcom}}} = \frac{\dot{C}_{W_{\text{tur}}}}{W_{\text{tur}}}$ Main compressor $\dot{C}_{3b} = \dot{C}_{8b} + \dot{C}_{W_{\text{rcom}}} + \dot{Z}_{\text{rcom}}$ $\frac{\dot{C}_{W_{\text{rcom}}}}{W_{\text{rcom}}} = \frac{\dot{C}_{W_{\text{tur}}}}{W_{\text{tur}}}, \dot{C}_{8b} = x\dot{C}_8$ $\dot{C}_{3b} = \dot{C}_{8b} + \dot{C}_{W_{\rm rcom}} + \dot{Z}_{\rm rcom}$ Recompression compressor $\frac{C_{W_{\rm rcom}}}{W_{\rm rcom}} = \frac{C_{W_{\rm tur}}}{W_{\rm tur}}, C_{\rm 8b} = xC_{\rm 8}$ $\dot{C}_{10} + \dot{C}_{H6} = \dot{C}_{8a} + \dot{C}_{H4} + \dot{Z}_{DHE}$ $\dot{C}_{10} + \dot{C}_{H5} = \dot{C}_9 + \dot{C}_{H4} + \dot{Z}_{DHE}$ DHE $\frac{\dot{C}_{8a}}{\overline{E}_{8a}} = \frac{\dot{C}_{10}}{\overline{E}_{10}}$ $\frac{\dot{C}_9}{E_9} = \frac{\dot{C}_{10}}{E_{10}}, \dot{C}_{H6} = \dot{C}_{H5} + \dot{C}_{H3}$ $\dot{C}_1 + \dot{C}_{12} = \dot{C}_{10} + \dot{C}_{11} + \dot{Z}_{cooler}$ $\dot{C}_1 + \dot{C}_{12} = \dot{C}_{10} + \dot{C}_{11} + \dot{Z}_{cooler}$ Cooler $\dot{C}_{11} = 0, \dot{C}_{12} = 0$ $\dot{C}_{11} = 0, \dot{C}_{12} = 0$ $\dot{C}_{A7} + \dot{C}_{A8} + \dot{C}_{9} =$ $\dot{C}_{A4} + \dot{C}_{8a} + \dot{Z}_{Gen}$ $\dot{C}_{8a} = \frac{\dot{C}_9}{E_9}, \\ \dot{C}_{A7} - \dot{C}_{A4} = \frac{\dot{C}_{A8} - \dot{C}_{A4}}{E_{A8} - E_{A4}} = \frac{\dot{C}_{A8} - \dot{C}_{A4}}{E_{A8} - E_{A4}}$ Generator / $\dot{C}_{A9} + \dot{C}_{H3} = \dot{C}_{A8} + \dot{C}_{H2} + \dot{Z}_{Cond}$ $\frac{\dot{C}_{A8}}{E_{A8}} = \frac{\dot{C}_{A9}}{E_{A9}}$ / Condenser $\dot{C}_{A1} + \dot{C}_{14} = \dot{C}_{A10} + \dot{C}_{13} + \dot{Z}_{Eva}$ Evaporator / $\dot{C}_{13} = 0, \dot{C}_{14} = 0$
$$\begin{split} \dot{C}_{A2} + \dot{C}_{H2} = \\ \dot{C}_{A1} + \dot{C}_{A5} + \dot{C}_{H1} + \dot{Z}_{Abs} \\ \frac{\dot{C}_{A1} + \dot{C}_{A5}}{E_{A1} + E_{A5}} = \frac{\dot{C}_{A2}}{E_{A2}}, \\ \frac{\dot{C}_{H1}}{E_{H1}} = \frac{\dot{C}_{H4}}{E_{H4}} \end{split}$$
/ Absorber $\dot{C}_{A4} + \dot{C}_{A6} = \dot{C}_{A3} + \dot{C}_{A7} + \dot{Z}_{SHE} \\ \frac{\dot{C}_{A6}}{E_{A6}} = \frac{\dot{C}_{A7}}{E_{A7}}$ / SHE $\dot{C}_{A3} = \dot{C}_{A2} + \dot{C}_{W_{pump}} + \dot{Z}_{pump}$ / Pump $\frac{\dot{C}_{W_{\text{pump}}}}{W_{\text{pump}}} = \frac{\dot{C}_{W_{\text{tur}}}}{W_{\text{tur}}}$

Table 5. Cost balance equations and auxiliary equations for the combined systems.

3.3. Performance Evaluation

In order to evaluate the performances of the proposed sCO₂/AHP system and the sCO₂/DH system, several criterions are considered in this study, including the coefficient of power performance (*COPP*), coefficient of heat performance (*COHP*), overall system exergy efficiency (η_{ex}) and total product unit cost (c_{total}). The *COPP* and *COHP* represent the electric power generation and heat production capacity, respectively, which are defined according to Equations (40) and (41) [18,47],

$$COPP = \frac{W_{net}}{Q_{core}} \tag{40}$$

$$COHP = \frac{Q_{h,users}}{Q_{h,absorb}}$$
(41)

where $Q_{h,users}$ is the heat amount supplied to the heat users, while $Q_{h,absorb}$ is the heat absorption from the topping cycle for the purpose of heating. The definitions of $Q_{h,users}$ and $Q_{h,absorb}$ can be found in Equations (23)–(26).

The exergy efficiency η_{ex} can be defined by Equations (42)–(44) [32,47],

$$\eta_{ex} = \frac{W_{net} + E_{h,users}}{E_{core}}.$$
(42)

 $E_{h,users}$ is the exergy that heat users obtain from the DHE and AHP. For the sCO₂/DH system,

$$E_{h,users} = E_{\rm H6} - E_{\rm H4} \tag{43}$$

while for the sCO₂/AHP system,

$$E_{h,users} = E_{\rm H6} - E_{\rm H4} - E_{\rm H1}.$$
(44)

The total product unit cost c_{total} presents the system performance in terms of economics, which is determined by Equation (45) [32],

$$c_{total} = \frac{\sum_{i=1}^{n_k} \dot{Z}_k + \sum_{i=1}^{n_f} c_{fi} E_{fi}}{\sum_{i=1}^{n_p} E_{pi}}$$
(45)

where E_{fi} and E_{pi} are the exergy flow rate of the fuel and the product, respectively. n_k , n_{f_i} and n_p represent the number of components, fuel components, and product components.

4. Model Verifications

In this section, model verifications are conducted to ensure the reliability and accuracy of the thermodynamic models for the following performance evaluation and analysis. The results of present model are compared to the reported data in published studies under the same conditions. As the overall system proposed in this study contains two parts: a recompression sCO₂ topping cycle and an AHP bottoming cycle, both of them should be validated. In addition, the verification of the AHP bottoming cycle is carried out for both LiBr-H₂O solution and ammonia water.

By using the software Matlab, the thermodynamic and exergoeconomic simulation platform is constructed according to the mass, energy, and cost balance equations in Section 3 to simulate the system performances under different conditions. Besides, the software REFPROP NIST is combined with Matlab to evaluate the physical and thermodynamic properties of CO_2 , water, ammonia, and ammonia-water solution, but those of the LiBr-H₂O solution are not included. Therefore, according to Pátek and Klomfar [48], a set of empirical formulas are utilized to calculate the pressure, enthalpy, entropy, density, and isobaric heat capacity of LiBr-H₂O solution in the temperature range of 273–500 K and the concentration range of 0–75 wt%. As the crystallization of LiBr-H₂O solution should be avoided during the system operation, the solubility curve of pure LiBr in water by Boryta [49] is utilized to examine whether the crystallization occurs for all LiBr-H₂O streams. These formulas and methods cover the application range of LiBr-H₂O solution in this study and provide a solid foundation for the following analysis.

4.1. Verification of Recompression sCO₂ Cycle Model

The first verification is conducted by comparing the results of the recompression sCO₂ cycle with the reported data by Sarkar and Bhattacharyya [6] under the same conditions to validate the sCO₂ topping cycle model, as shown in Table 6. Clearly, the simulated results of the present model show an excellent agreement with the published results. Thus, the developed thermodynamic model for the sCO₂ cycle is accurate and reliable enough to be applied for the following investigation and analysis.

Table 6 shows the comparison between the simulated results and the published data for the sCO₂ cycle.

	Paran	neters		x	η_{th}		
T_{\min} (°C)	T_{max} (°C)	P _{max} (MPa)	PRc	Present	Published	Present	Published
32	550	20	2.64	0.3332	0.334	41.18	41.18
32	550	30	3.86	0.3546	0.355	43.32	43.32
32	750	20	2.65	0.2212	0.223	46.07	46.07
32	750	30	3.94	0.2809	0.281	49.84	49.83
50	550	20	2.40	0.1842	0.184	36.71	36.71
50	550	30	2.80	0.2533	0.254	38.94	38.93
50	750	20	2.88	0.0962	0.109	43.50	43.50
50	750	30	3.08	0.1745	0.175	45.28	45.28

Table 6. Comparison between the simulated results and the published data for the sCO_2 cycle.

4.2. Verification of LiBr-H₂O AHP Model

Then the AHP model using LiBr-H₂O solution is examined. Cheng and Shih [47] conducted a detailed thermodynamic analysis for a LiBr-H₂O AHP and reported the main thermodynamic state points, as listed in Table 7. By setting the same thermodynamic conditions with those in Reference [47], the simulated results are obtained based on the present AHP model, and also listed in Table 7 for comparison. In addition, the calculated *COHP* is 1.695 here while the reported *COHP* is 1.69. Obviously, the results of present model match reasonably with the published data, which proves the accuracy of the LiBr-H₂O AHP model in this study.

The following Table 7 shows the comparison between the simulated results and the published data for LiBr-H₂O AHP.

Chata Dalata	T	(°C)	<i>h</i> (kJ/kg)			
State Points	Present	Published	Present	Published		
A7	164.85	162.75	369.49	365.80		
A4	142.63	140.76	313.86	310.30		
A2	91.99	92.94	216.03	217.87		
A6	106.75	106.90	280.22	265.63		
A8	160.52	162.75	2798.07	2797.36		
A1	51.85	51.85	2594.55	2595.19		
A9	96.85	96.85	405.88	404.90		

Table 7. Comparison between the simulated results and the published data for LiBr-H₂O AHP.

4.3. Verification of Ammonia-Water AHP Model

Finally, the ammonia-water AHP model is validated by using the reported results in Wang and Ferreira's work [50]. The same assumptions and conditions are adopted in the simulation platform:

(1) The temperatures of generator, condenser, absorber and evaporator are set to be 120 $^{\circ}$ C, 45 $^{\circ}$ C, 45 $^{\circ}$ C, and 10 $^{\circ}$ C; (2) The minimum temperature approach of SHE is set to be 5 K; (3) The solution leaving the absorber is assumed to have a subcooling of 3 K. Then, the comparison is shown in Table 8, which indicates a high agreement between two groups of results.

Table 8 shows the comparison between the simulated results and the published data for ammonia-water based AHP.

 Table 8. Comparison between the simulated results and the published data for ammonia-water based AHP.

Parameters	Present Results	Published Results
y _{A2}	0.481	0.481
<i>YA</i> 7	0.335	0.335
q_{A4}	0.023	0.024
circulation ratio	4.554	4.555
COHP	1.615	1.612

5. Results and Discussions

In this section, the energy, exergy, and economic performances of the proposed $sCO_2/LiBr-H_2O$ AHP system are analyzed. Performance comparisons are conducted between the $sCO_2/LiBr-H_2O$ AHP system with a single sCO_2 system, a sCO_2/DH system, and a $sCO_2/ammonia$ -water AHP system. Parametric studies are carried out to find out the influences of some key parameters on the system performance indicators, including *COPP*, *COHP*, η_{ex} , and c_{total} , as defined in Section 3.3. Then, the particle swarm optimization (PSO) algorithm is adopted to obtain the system optimal operation conditions.

The main input parameters and assumptions are listed in Table 9. By using these input conditions, the parameters of system state points are calculated and summarized in Tables 10-12 for the sCO₂/LiBr-H₂O AHP system, sCO₂/ammonia-water AHP system and sCO₂/DH system, respectively.

The following Table 9 shows the main input conditions for the simulation.

Items	Values	
<i>T</i> ₀ (°C)	25	
P_0 (MPa)	0.101325	
$Q_{\rm core}$ (MW)	600 ^a	
$T_{\rm core}$ (°C)	800 ^a	
<i>T</i> ₁ (°C)	35 ^a	
P_1 (MPa)	7.4 ^a	
$\eta_{mcom} \& \eta_{rcom}$	0.85 ^b	
$\eta_{ m tur}$	0.86 ^c	
$\eta_{ m pump}$	0.75 ^d	
$\varepsilon_{\rm HTR}$ & $\varepsilon_{\rm LTR}$	0.86 ^b	
$\Delta T_{\text{DHE,end}}$ (°C)	5 ^d	
$\Delta T_{\text{SHE,end}}$ (°C)	5 d	
$T_{\rm H6}$ (°C)	60 ^e	
<i>T</i> _{H1} (°C)	45 ^e	
P _{user} (MPa)	1.0 ^e	
Fuel cost (\$/MWh)	7.4 ^a	

Table 9. The main input conditions for the simulation.

^a Reference [11]; ^b reference [51]; ^c reference [52]; ^d reference [50]; ^e reference [53].

Table 10 shows the thermodynamic properties and costs of exergy streams for the $sCO_2/LiBr-H_2O$ AHP system.

State	Т	Р	h	s	т	e _{ph}	e _{ch}	Ċ	С
Points	(°C)	(MPa)	(kJ/kg)	(kJ/(kg·K))	(kg/s)	(kJ/kg)	(kJ/kg)	(\$/h)	(\$/GJ)
1	35.00	7.40	402.40	1.663	2309.66	216.60	/	26,077.65	14.480
2	110.06	20.72	444.87	1.680	2309.66	254.09	/	33,328.11	15.775
3	245.43	20.72	656.60	2.159	3099.68	322.95	/	57,628.04	15.991
4	393.02	20.72	841.06	2.473	3099.68	413.97	/	73,239.23	15.855
5	550.00	20.72	1034.63	2.734	3099.68	529.77	/	84,301.17	14.260
6	433.11	7.40	906.74	2.763	3099.68	392.97	/	62,533.52	14.260
7	271.71	7.40	722.28	2.467	3099.68	296.85	/	47,237.39	14.260
8	132.69	7.40	564.63	2.133	3099.68	238.91	/	38,017.79	14.260
9	106.60	7.40	532.82	2.052	2309.66	231.26	/	27,421.47	14.260
10	50.00	7.40	447.38	1.806	2309.66	218.94	/	25,959.82	14.260
11	25.00	0.1013	104.92	0.367	2485.38	0.00	/	0.00	0.000
12	35.00	0.1013	146.72	0.505	2485.38	0.69	/	0.00	0.000
13	35.00	0.1013	146.72	0.505	2287.10	0.69	/	0.00	0.000
14	29.00	0.1013	121.64	0.423	2287.10	0.11	/	0.00	0.000
A1	26.00	0.0034	2548.32	8.535	24.74	8.06	478.83	477.82	11.017
A2	55.00	0.0034	119.31	0.353	157.07	2.92	490.90	3252.42	11.648
A3	55.00	0.0158	119.31	0.353	157.07	2.92	490.90	3252.71	11.649
A4	91.60	0.0158	197.37	0.578	157.07	13.76	509.19	3462.53	11.710
A5	60.00	0.0034	190.79	0.387	132.32	3.41	628.81	3535.42	11.739
A6	60.00	0.0158	190.79	0.386	132.32	3.42	628.81	3535.42	11.739
A7	110.00	0.0158	283.45	0.646	132.32	18.84	649.17	3735.49	11.739
A8	110.00	0.0158	2705.90	8.288	24.74	239.39	478.83	653.86	10.221
A9	55.00	0.0158	230.26	0.768	24.74	5.83	478.83	441.23	10.221
A10	26.00	0.0034	230.26	0.787	24.74	0.31	478.83	441.23	10.338
H1	45.00	1.00	189.30	0.638	3329.79	3.58	/	0.00	0.000
H2	50.00	1.00	210.19	0.703	3329.79	5.05	/	803.80	13.280
H3	54.40	1.00	228.59	0.760	3329.79	6.59	/	1030.05	13.045
H4	45.00	1.00	189.30	0.638	1903.60	3.58	/	0.00	0.000
H5	69.79	1.00	292.97	0.952	1903.60	13.70	/	1726.18	18.385
H6	60.00	1.00	252.00	0.831	5233.39	8.87	/	2756.23	16.495

Table 10. Thermodynamic properties and costs of exergy streams for the sCO₂/LiBr-H₂O AHP system.

Table 11 shows the thermodynamic properties and costs of exergy streams for the $sCO_2/ammonia$ -water AHP system.

Table 11. Thermodynamic properties and costs of exergy streams for the sCO₂/ammonia-water AHP system.

State	Т	Р	h	S	т	e _{ph}	e _{ch}	Ċ	С
Points	(°C)	(MPa)	(kJ/kg)	(kJ/(kg·K))	(kg/s)	(kJ/kg)	(kJ/kg)	(\$/h)	(\$/GJ)
1	35.00	7.40	402.40	1.663	2309.66	216.60	/	26,077.65	14.480
2	110.06	20.72	444.87	1.680	2309.66	254.09	/	33 <i>,</i> 328.11	15.775
3	245.43	20.72	656.60	2.159	3099.68	322.95	/	57,628.04	15.991
4	393.02	20.72	841.06	2.473	3099.68	413.97	/	73 <i>,</i> 239.23	15.855
5	550.00	20.72	1034.63	2.734	3099.68	529.77	/	84,301.17	14.260
6	433.11	7.40	906.74	2.763	3099.68	392.97	/	62,533.52	14.260
7	271.71	7.40	722.28	2.467	3099.68	296.85	/	47,237.39	14.260
8	132.69	7.40	564.63	2.133	3099.68	238.91	/	38,017.79	14.260
9	108.26	7.40	534.91	2.057	2309.66	231.72	/	27,475.03	14.260
10	50.00	7.40	447.38	1.806	2309.66	218.94	/	25,959.82	14.260
11	25.00	0.1013	104.92	0.367	2485.38	0.00	/	0.00	0.000
12	35.00	0.1013	146.72	0.505	2485.38	0.69	/	0.00	0.000
13	35.00	0.1013	146.72	0.505	1801.09	0.69	/	0.00	0.000

State	Т	Р	h	s	т	e _{ph}	e _{ch}	Ċ	с
Points	(°C)	(MPa)	(kJ/kg)	(kJ/(kg·K))	(kg/s)	(kJ/kg)	(kJ/kg)	(\$/h)	(\$/GJ)
14	29.00	0.1013	121.64	0.423	1801.09	0.11	/	0.00	0.000
A1	26.00	1.0345	1627.18	5.779	44.37	323.24	19,805.29	71,903.43	22.363
A2	55.00	1.0345	190.82	1.400	255.36	46.26	10,889.39	225,020.13	22.383
A3	55.27	2.3111	193.01	1.402	255.36	47.95	10,889.39	225,116.94	22.390
A4	93.26	2.3111	390.83	1.970	255.36	76.45	10,889.39	225,820.58	22.401
A5	60.47	1.0345	178.12	1.331	210.99	22.81	9014.33	153,708.18	22.393
A6	60.27	2.3111	178.12	1.327	210.99	24.21	9014.33	153,708.18	22.389
A7	110.00	2.3111	417.55	1.996	210.99	64.20	9014.33	154,388.21	22.389
A8	110.00	2.3111	1810.74	5.970	44.37	449.85	19,805.29	72,304.83	22.347
A9	55.00	2.3111	609.26	2.347	44.37	328.52	19,805.29	71,871.72	22.347
A10	26.00	1.0345	609.26	2.376	44.37	319.83	19,805.29	71,871.72	22.357
H1	45.00	1.00	189.30	0.638	3144.32	3.58	/	0.00	0.000
H2	49.65	1.00	208.72	0.699	3144.32	4.94	/	615.28	11.012
H3	53.70	1.00	225.67	0.751	3144.32	6.33	/	1060.68	14.807
H4	45.00	1.00	189.30	0.638	1903.60	3.58	/	0.00	0.000
H5	70.39	1.00	295.50	0.959	1903.60	14.03	/	1778.22	18.491
H6	60.00	1.00	252.00	0.831	5047.92	8.87	/	2838.90	17.614

Table 11. Cont.

Table 12 shows the thermodynamic properties and costs of exergy streams for the sCO₂/DH system.

State	Т	Р	h	s	e _{ph}	m	Ċ	с
Points	(°C)	(MPa)	(kJ/kg)	(kJ/(kg·K))	(kJ/kg)	(kg/s)	(\$/h)	(\$/GJ)
1	35.00	7.40	402.40	1.663	216.60	2309.66	26,014.53	14.445
2	110.06	20.72	444.87	1.680	254.09	2309.66	33,257.10	15.741
3	245.43	20.72	656.60	2.159	322.95	3099.68	57,527.04	15.963
4	393.02	20.72	841.06	2.473	413.97	3099.68	73,115.84	15.828
5	550.00	20.72	1034.63	2.734	529.77	3099.68	84,177.77	14.240
6	433.11	7.40	906.74	2.763	392.97	3099.68	62,441.98	14.240
7	271.71	7.40	722.28	2.467	296.85	3099.68	47,168.25	14.240
8	132.69	7.40	564.52	2.132	238.88	3099.68	37,957.41	14.240
10	50.00	7.40	447.38	1.806	218.94	2309.66	25,921.82	14.240
11	25.00	0.1013	104.92	0.367	0.00	8957.55	0.00	0.000
12	35.00	0.1013	146.72	0.505	0.00	8957.55	0.00	0.000
H4	45.00	1.0	189.30	0.638	3.58	4314.5	1775.3	31.890
H6	60.00	1.0	252.00	0.831	8.87	4314.5	4393.2	31.890

Table 12. Thermodynamic properties and costs of exergy streams for the sCO₂/DH system.

5.1. Parametric Analysis

The influences of several key parameters on the thermodynamic and exergoeconomic performances of the overall system are revealed through parametric analysis. The parameters studied include the turbine inlet temperature (T_5), the compressor pressure ratio (*PRc*), the generator outlet temperature (T_{A8}), and the evaporator outlet temperature (T_{A1}). As the turbine outlet pressure is set as 7.4 MPa to keep the sCO₂ topping cycle operating in the supercritical state, the *PRc* also represents the turbine inlet pressure. Both turbine inlet temperature and pressure are the vital parameters of sCO₂ topping cycle that can influence the electric power generation significantly, while other two parameters are important to the AHP bottoming cycle that can greatly affect the heat production. The parameteric analysis is carried out by changing only one parameter at a time with all other parameters fixed.

5.1.1. Effects of the Compressor Pressure Ratio (PRc)

Figure 2 depicts the influences of PRc on the system performance indicators, namely COPP, COHP, exergy efficiency η_{ex} and total product unit cost c_{total} , for a single sCO₂ system, sCO₂/DH system, sCO₂/ammonia-water AHP system, and sCO₂/LiBr-H₂O AHP system, respectively. Obviously in Figure 2a, the thermodynamic and exergoeconomic performances for these four systems can be ranked according to η_{ex} and c_{total} as: sCO₂/LiBr-H₂O AHP > sCO₂/ammonia-water AHP > sCO₂/DH > sCO₂ (A > B means A is better than B). Thus, the combined cycles show better performances than the single sCO₂. This indicates that the performance of recompression sCO₂ cycle can be enhanced by assembling a waste heat recovery system based on the energy cascade utilization principle. Besides, by comparing three combined cycles, it can be concluded that AHP is a better choice than DH, and that the LiBr-H₂O solution is a better working fluid than the ammonia-water solution for AHP. Thus, the proposed sCO₂/LiBr-H₂O AHP system is a desirable waste heat recovery system for the sCO₂ cycle, with which the combined cycle could achieve a high exergy efficiency with a low cost.

Figure 2 shows the variation trends of system performance indicators with *PRc* for different thermal systems: (a) η_{ex} and c_{total} ; (b) *COPP* and *COHP*.



Figure 2. Variation trends of system performance indicators with *PRc* for different thermal systems: (a) η_{ex} and c_{total} ; (b) *COPP* and *COHP*.

explained using these two aspects. Figure 2b depicts the variations of *COPP* and *COHP* with *PRc*. *COPP* represents the power generation capacity while *COHP* reflects the heat production capacity. Clearly, *COHP* increases with *PRc*, but *COPP* increases firstly and then decreases. Figure 3 presents the turbine output power W_{tur} , compressor power consumption W_{com} and net output power W_{net} varying with *PRc* for the sCO₂/LiBr-H₂O AHP system. The increase of *PRc* (turbine inlet pressure) strengthens the power generation capacity of the unit mass working fluid, and leads to the increase of W_{tur} . At the same time, the increase of the turbine inlet pressure requires more power to drive the compressors and leads to the increase of W_{com} . So, with the increase of *PRc*, W_{net} increases firstly when the increment of W_{tur} is larger than that of W_{com} , and then decreases for the opposite situation. It explains why *COPP* and η_{ex} increase firstly but then decrease with *PRc* in Figure 2a,b.

Figure 3 shows the variation trends of W_{tur} , W_{com} , and W_{net} with *PRc* for the sCO₂/LiBr-H₂O AHP system.



Figure 3. Variation trends of W_{tur}, W_{com}, and W_{net} with PRc for the sCO₂/LiBr-H₂O AHP system.

5.1.2. Effects of the Turbine Inlet Temperature (T₅)

The effects of T_5 on η_{ex} , c_{total} , COPP, and COHP are displayed in Figure 4 for four thermal systems. Clearly, η_{ex} increases with T_5 while c_{total} shows an opposite variation trend. It indicates that increasing T_5 could be beneficial to the improvement of system performances for these thermal systems. Figure 5 depicts the net output power W_{net} , the temperature of stream 8 (T_8) and the mass flow rate of 8a (m_{8a}) varying with T_5 for the sCO₂/LiBr-H₂O AHP system. When T_5 ascends, the working capacity of unit mass working fluid is strengthened, and W_{net} increases correspondingly. As a result, *COPP* in Figure 4b also increases with T_5 . On the other side, the increase of T_5 enlarges the heat absorption amount of the unit mass flow rate CO_2 from the reactor, so that the mass flow rate of CO_2 decreases. Correspondingly, m_{8a} decreases with the increase of T_5 , as shown in Figure 5. It means that the mass flow rate of CO₂ flowing into AHP and DHE decreases, which in turn decreases the heat absorption amount of AHP and DHE from the topping cycle. At the same time, T_8 increases with T_5 , so that the generator temperature ascends and the heat absorption amount of AHP and DHE from the topping cycle will increase. The opposite effects between T_8 and m_{8a} makes the COHP of sCO₂/AHP systems change slightly, as shown in Figure 4b. Thus, the increase of η_{ex} with T_5 is dominated by the increase of *COPP*, i.e., the power generation capacity of sCO₂ topping cycle, while the increase of T_5 exerts slight influences on the performances of the bottoming cycle. As the increase of T_5 results in the decrement of

mass flow rate of working fluid, the capital investment costs for the turbine, compressors, and pressure vessels decrease correspondingly and lead to the decline of c_{total} , as shown in Figure 4a.

The following Figure 4 shows the variation trends of system performance indicators with T_5 for different thermal systems: (a) η_{ex} and c_{total} ; (b) *COPP* and *COHP*.



Figure 4. Variation trends of system performance indicators with T_5 for different thermal systems: (**a**) η_{ex} and c_{total} ; (**b**) *COPP* and *COHP*.

The following Figure 5 shows the variation trends of T_8 , m_{8a} , and W_{net} with T_5 for the sCO₂/LiBr-H₂O AHP system.



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Figure 5. Variation trends of T_8 , m_{8a} , and W_{net} with T_5 for the sCO₂/LiBr-H₂O AHP system.

5.1.3. Effects of the Generator Outlet Temperature (T_{A8})

Then, the influences of key parameters of AHP bottoming cycle on the overall system performances are investigated. As the single sCO₂ cycle and the sCO₂/DH system have no AHP bottoming cycle, only the sCO₂/ammonia-water AHP system and the sCO₂/LiBr-H₂O AHP system are analyzed here. Figure 6 shows the variation of η_{ex} and c_{total} with the generator outlet temperature T_{A8} for these two combined systems. As T_{A8} increases, η_{ex} increases to a peak and next goes down slightly, while c_{total} shows an opposite trend in Figure 6a, which indicates that there exists a T_{A8} to optimize the overall system performances for the sCO₂/ammonia-water AHP system. The COHP also ascends firstly and then descends with T_{A8} . In order to explain this phenomenon, Figure 7a presents the mass flow rate of stream A8 (m_{A8}), the total mass flow rate of ammonia water (m_{A4}), the pump power consumption (W_{pump}) , and the exergy that heat users obtain $(E_{h,users})$ varying with T_{A8} . Clearly, as T_{A8} increases, the heat absorption amount per unit mass flow rate from the topping cycle increases, so that the total mass flow rate m_{A4} descends and then leads to the decline of W_{pump} . m_{A8} increases firstly with T_{A8} as more working fluid is evaporated from the generator solution. More ammonia vapor flows into the condenser and releases the considerable latent heat to the heating water, so that $E_{h,users}$ increases correspondingly. Then, as the effect of decreasing m_{A4} dominates when T_{A8} is larger, both m_{A8} and $E_{h,users}$ decrease with T_{A8} . Therefore, COHP and η_{ex} increase and then decrease with T_{A8} according to Equations (41) and (42).

Figure 6 shows the variation trends of η_{ex} , c_{total} , and *COHP* with T_{A8} : (a) a sCO₂/ammonia-water AHP system; (b) a sCO₂/LiBr-H₂O AHP system.



Figure 6. Variation trends of η_{ex} , c_{total} and *COHP* with T_{A8} : (**a**) sCO₂/ammonia-water AHP system; (**b**) sCO₂/LiBr-H₂O AHP system.

The following Figure 7 shows the variation trends of m_{A8} , m_{A4} , W_{pump} , and $E_{h,users}$ with T_{A8} : (a) a sCO₂/ammonia-water AHP system; (b) a sCO₂/LiBr-H₂O AHP system.





Figure 7. Variation trends of m_{A8} , m_{A4} , W_{pump} , and $E_{h,users}$ with T_{A8} : (**a**) a sCO₂/ammonia-water AHP system; (**b**) a sCO₂/LiBr-H₂O AHP system.

For the sCO₂/LiBr-H₂O AHP system, the same trends can be observed for the *COHP*, m_{A4} , m_{A8} , W_{pump} , and $E_{h,users}$ in Figures 6b and 7b, which will not be explained again here. It should be noted that the operational pressure of LiBr-H₂O AHP is much lower than that of ammonia-water AHP, so the pump power consumption W_{pump} in LiBr-H₂O AHP is also lower than that in ammonia-water AHP, as shown in Figure 7.

5.1.4. Effects of the Evaporator Temperature (T_{A1})

Figure 8 describes the variations of η_{ex} , c_{total} and *COHP* with T_{A1} . When T_{A1} increases, more heat is transferred to the heating water in the absorber. The heat absorption amount of heating water ascends correspondingly, and results in the increase of *COHP*. As the pump work is very limited compared to the turbine output power, the variation of pump power consumption with T_{A1} has ignorable effects on the net output power. Thus, the power generation capacity is hardly affected by the change of T_{A1} . This means that the overall system performances are mainly affected by the

variation of heat production capacity when T_{A1} changes. Therefore, the exergy efficiency η_{ex} has the same variation trend of *COHP* and the total product unit cost c_{total} decreases with T_{A1} , indicating that the thermodynamic and economic performances of the sCO₂/AHP systems can be improved by increasing the evaporator temperature.

The following Figure 8 shows the variation trends of system performance indicators with T_{A1} for different thermal systems: (a) η_{ex} and c_{total} ; (b) *COHP*.



Figure 8. Variation trends of system performance indicators with T_{A1} for different thermal systems: (a) η_{ex} and c_{total} ; (b) *COHP*.

5.2. System Performance Optimization and Comparison

Finally, parameter optimizations are carried out in this section to provide references for the practical design and operation. The overall performances of a single sCO₂ system, sCO₂/DH system, sCO₂/ammonia-water AHP system and sCO₂/LiBr-H₂O AHP system are optimized by using the particle swarm optimization (PSO) algorithm, which was proposed by Kennedy and Eberhart [54]. The PSO algorithm shows higher accuracy and converges faster compared to other optimization algorithms,

and has been successfully applied to optimize performances for thermal systems [55,56]. A swarm of particles is included in the PSO algorithm to represent the candidate solutions. Each candidate solution has a velocity vector \vec{v} and a position vector \vec{u} . In addition, a best global position $\vec{g}(t)$ and a best personal position $\vec{p}(t)$ are stored for each particle. For each time step t, the particles move to a better position in the space so that the velocity vector \vec{v} and the position vector \vec{u} update correspondingly, as shown in Equations (46) and (47):

$$\vec{u}(t+1) = \vec{u}(t) + \vec{v}(t+1)$$
(46)

$$\vec{v}(t+1) = \omega \vec{v}(t) + U(0,\varphi_1) \left(\vec{p}(t) - \vec{u}(t) \right) + U(0,\varphi_2) \left(\vec{g}(t) - \vec{u}(t) \right)$$
(47)

where ϕ_1 and ϕ_2 denote the importance of $\vec{p}(t)$ and $\vec{g}(t)$, respectively. ϖ is the inertia weight that controls the velocity $\vec{v}(t)$.

In this study, the key parameters studied include the compressor pressure ratio *PRc*, the turbine inlet temperature T_{5} , the generator outlet temperature T_{A8} , and the evaporator temperature T_{A1} . The constraints of key parameters for the PSO algorithm are displayed in Table 13. The exergy efficiency η_{ex} and the total product unit cost c_{total} are chosen as the objective functions. The optimization is achieved by maximizing η_{ex} or minimizing c_{total} . So two optimal design cases are considered, namely, the exergy efficiency optimal design (EOD) case and the cost optimal design (COD) case. The inertia weight ϖ and the swarm size for the PSO algorithm are chosen as 0.6 and 25, respectively. Both ϕ_1 and ϕ_2 are chosen as 1.8. In addition, a maximal velocity v_{max} of 15% is adopted to constrain the velocity at each time step [57].

The following Table 13 shows the boundary conditions of decision variables for the PSO algorithm.

Items	sCO ₂	sCO ₂ /DH	sCO ₂ /Ammonia-Wa AHP	ter sCO ₂ /LiBr-H ₂ O AHP
PRc	2.2–3.3	2.2–3.3	2.2–3.3	2.2–3.3
T ₅ (°C)	500-600	500-600	500-600	500-600
T_{A8} (°C)	/	/	92-125	87-110
T_{A1} (°C)	/	/	23–28	23–28

Table 13. Boundary conditions of decision variables for the PSO algorithm.

Tables 14 and 15 show the optimized results of these four thermal systems for the EOD case and the COD case, respectively. Obviously, due to the different optimization targets, the exergy efficiencies η_{ex} for the EOD case are much higher than those for the COD case, while the total product unit costs *c*_{total} for the COD case are lower than those for the EOD case. The optimum *PRc* for the COD case is smaller than that for the EOD case, meaning that the capital investment, operation, and maintenance costs for the turbine, compressors and pressure vessels are lower, which leads to a lower c_{total} in the COD case. The total product unit costs c_{total} of a sCO₂ system, sCO₂/DH system, sCO₂/ammonia-water AHP system, and sCO₂/LiBr-H₂O AHP system for the COD case are about 4.00%, 4.39%, 4.39%, and 3.87% lower than those for the EOD case. As for the expense, the exergy efficiencies η_{ex} of sCO₂ system, sCO₂/DH system, sCO₂/ammonia-water AHP system, and sCO₂/LiBr-H₂O AHP system for the COD case dropped by about 3.14%, 3.35%, 4.09%, and 4.13%, respectively. In both the COD case and the EOD case, the sCO₂/LiBr-H₂O AHP system has the highest η_{ex} with a lowest cost c_{total} , indicating that the overall system performances are greatly improved by combining the LiBr-H₂O AHP with sCO₂ cycle. In the the EOD case, η_{ex} of sCO₂/LiBr-H₂O AHP system is 13.39%, 3.53% and 1.13% higher than those of the sCO₂ system, sCO₂/DH system, and sCO₂/ammonia-water AHP system, respectively. In the the COD case, c_{total} of sCO₂/LiBr-H₂O AHP system is 8.66%, 1.27%, and 0.42% lower than those of sCO2 system, sCO2/DH system and sCO2/ammonia-water AHP system, respectively. In addition, the COHP of the $sCO_2/LiBr-H_2O$ AHP system is much larger than those of the sCO_2/DH system and

the sCO_2 /ammonia-water AHP system. It indicates that the LiBr-H₂O AHP can maximize the recovery of waste heat from sCO_2 cycle to provide heat for users.

Table 14 shows the optimized results for the exergy efficiency optimal design case.

Items	sCO ₂	sCO ₂ /DH	sCO ₂ /Ammonia-Wa AHP	ter sCO ₂ /LiBr-H ₂ O AHP
PRc	3.1641	3.2599	3.2960	3.2803
T ₅ (°C)	600	600	600	599.31
T_{A8} (°C)	/	/	115.86	90.30
T_{A1} (°C)	/	/	28	27.92
COPP	0.3980	0.3979	0.3978	0.3975
COHP	/	1	1.282	1.410
η_{ex} (%)	55.11	60.36	61.79	62.49
c_{total} (\$/GJ)	19.603	18.210	18.054	17.881

Table 14. Optimized results for the exergy efficiency optimal design case.

Table 15 shows the optimized results for the cost optimal design case.

Items	sCO ₂	sCO ₂ /DH	sCO ₂ /Ammonia-Wa AHP	ter sCO ₂ /LiBr-H ₂ O AHP
PRc	2.3693	2.3692	2.3449	2.3992
T ₅ (°C)	600	600	600	600
T_{A8} (°C)	/	/	114.31	96.98
T_{A1} (°C)	/	/	28	28
COPP	0.3855	0.3855	0.3845	0.3866
COHP	/	1	1.131	1.235
η _{ex} (%)	53.38	58.34	59.26	59.91
c_{total} (\$/GJ)	18.819	17.410	17.262	17.189

Table 15. Optimized results for the cost optimal design case.

6. Conclusions

In this study, a novel combined sCO₂/LiBr-H₂O AHP system is first proposed and analyzed for heat and power cogeneration according to the principle of energy cascade utilization. Detailed thermodynamic and exergoeconomic analysis, parametric studies, optimizations, and comparisons are carried out among a stand-alone sCO₂ system, a sCO₂/DH system, a sCO₂/ammonia-water AHP system, and the proposed sCO₂/LiBr-H₂O AHP system. The main conclusions and contributions of this work can be summarized as follows:

- 1. Using the LiBr-H₂O AHP to recover the waste heat from sCO₂ cycle can significantly improve the thermodynamic and economic performances for the overall system. Based on the optimization results, the sCO₂/LiBr-H₂O AHP system can gain an improvement of 13.39% in the exergy efficiency and a reduction of 8.66% in the total product unit cost compared to the stand-alone sCO₂ cycle;
- 2. The exergy efficiency of sCO₂/LiBr-H₂O AHP system is 3.53% and 1.13% higher than those of sCO₂/DH system and sCO₂/ammonia-water AHP system, respectively, in the EOD case, and the total product unit cost of the sCO₂/LiBr-H₂O AHP system is 1.27% and 0.42% lower than those of the sCO₂/DH system and sCO₂/ammonia-water AHP system, respectively in the the COD case. In addition, *COHP* of sCO₂/LiBr-H₂O AHP system is much larger than those of sCO₂/DH system and sCO₂/ammonia-water AHP system. The the LiBr-H₂O AHP can maximize the recovery of waste heat from sCO₂ cycle to provide heat for users. Therefore, the LiBr-H₂O AHP is a desirable waste heat recovery system for the sCO₂ cycle, with which the combined system could achieve a higher exergy efficiency and a lower cost. Besides, the proposed sCO₂/LiBr-H₂O

AHP system is presented to be a high-efficiency CHP system. It provides a meaningful direction for the design and improvement of sCO2-based CHP systems to reduce energy consumption and to bring considerable economic benefits;

3. The parametric study presents the influences of decision variables on the system performances, and the PSO optimization finds optimal design conditions for different cases. The results of parametric study, optimization, and comparison analysis could provide useful references for designers and researchers attempting to obtain desirable system designs and operation conditions for sCO₂-based CHP systems.

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Nomenclature

Α	heat transfer area (m ²)	
Ċ	cost rate (\$/h)	
С	cost per unit exergy (\$/GJ)	
C _{total}	total product unit cost (\$/GJ)	
е	specific exergy (kJ/kg)	
Ε	exergy rate (kW)	
h	enthalpy (kJ/kg)	
<i>i</i> _r	interest rate	
Ι	exergy destruction (kW)	
т	mass flow rate (kg/s)	
п	number of operation year	
Р	pressure (MPa)	
PRc	compressor pressure ratio	
Q	heat transfer rate (kW)	
S	entropy (kJ/(kg·K))	
Т	temperature (°C)	
W	power (kW)	
x	recompressed mass flow ratio	
у	concentration of LiBr or ammonia in the solution	
Ζ	capital cost of a component (\$)	
Ż	capital cost rate (\$/h)	
Greek letters	3	
η	efficiency (%)	
ε	effectiveness	
γ	maintenance factor	
τ	annual operation hours (h)	
Subscripts and abbreviations		
0	ambient state	
1, 2, et al.	state points	
А	state points of absorption heat pump	
AHP	absorption heat pump	
Abs	absorber	

ch	chemical exergy
CI	capital investment
COD	cost optimal design
com	compressor
Cond	condenser
COHP	coefficient of heat performance
COPP	coefficient of power performance
core	reactor core
CRF	capital recovery factor
DH	direct heating
DHE	direct heating exchanger
EOD	exergy efficiency optimal design
ex	exergy
Eva	evaporator
Gen	generator
Н	state points of heating water
HTR	high temperature recuperator
LTR	low temperature recuperator
LMTD	logic mean temperature difference
MC	main compressor
mcom	main compressor
OM	operation and maintenance
ph	physical exergy
pump	pump
RC	recompression compressor
rcom	recompression compressor
SHE	solution heat exchanger
tur	turbine

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