





# Application of Metal Magnetic Memory Testing Technology to the Detection of Stress Corrosion Defect

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**Abstract:** The damage of equipment manufactured with ferromagnetic materials in service can be effectively detected by Metal Magnetic Memory Testing (MMMT) technology, which has received extensive attention in various industry fields. The effect of stress or strain on Magnetic Flux Leakage (MFL) signals of ferromagnetic materials has been researched by many scholars for assessing stress concentration and fatigue damage. However, there is still a lack of research on the detection of stress corrosion damage of ferromagnetic materials by MMMT technology. In this paper, the electrochemical corrosion system was designed for corrosion experiments, and three different experiments were performed to study the effect of corrosion on MFL signals. The distribution of MFL signals on the surface of the specimen was investigated. The results indicated that both the normal component  $H_n$  and tangential component  $H_t$  of MFL signals presented different signal characteristics when the specimen was subjected to different working conditions. Finally, two characterization parameters,  $S_n$  and  $S_t$ , were defined to evaluate the corrosion degree of the specimen, and  $S_t$  is better. The direct dependence of corrosion depth on the parameter was developed and the average error rates between the predicted and measured values are 8.94% under the same working condition. Therefore, the expression can be used to evaluate the corrosion degree of the specimen quantitatively. The results are significant for detecting and assessing the corrosion defect of ferromagnetic materials.

**Keywords:** metal magnetic memory testing technology; stress corrosion; ferromagnetic materials; characterization parameters

# 1. Introduction

Many types of equipment manufactured with ferromagnetic materials in service are working under the condition of stress corrosion. Most of this equipment is susceptible to corrosion defects. The effect of pitting corrosion on pipeline steel has been studied by many scholars [1–4]. Chen and Zhao reported the failure of oil and gas pipelines due to stress corrosion cracking [5,6]. It is necessary to conduct Non-Destructive Testing (NDT) for stress corrosion defects. The common NDT method for the quantitative evaluation of corroded defects, such as Electrochemical Noise Analysis (ENA), has been studied by many scholars [7–9]. It was used to investigate the corrosion behavior of mild steel (Q235) [10], and Wang reported that pitting corrosion of high strength steel is quantitatively evaluated by combining 3-D measurement and image-recognition-based statistical analysis [11]. However, the reliability for assessing the sizes of corrosion defects needs to be further discussed. The traditional magnetic testing technologies [12–14], such as magnetic particle testing [15,16], magnetic flux leakage testing [17–19], usually require a large external magnetic field [20–22], which is inconvenient to operate

and even impractical under some conditions. Although the non-destructive testing of stress corrosion defects has been widely studied, the traditional NDT methods have some limitations. Therefore, it is necessary to develop a more simple and effective NDT technology.

The Metal Magnetic Memory Testing (MMMT) technology, which was first presented at the 50th International Welding Conference in 1997 by a Russian researcher, is an emerging testing method in the field of non-destructive testing technology [23,24]. It is a weak-field detecting method, which uses the geomagnetic field instead of an artificial magnetic field as the stimulus source. MMMT technology is used to diagnose the early damage by measuring the self-magnetized leakage field on the surface of ferromagnetic materials [25,26]. MMMT technology has the advantages of simple equipment, simple operation, and no treatment on the surface of the tested part. Nowadays, the MMMT technique has attracted the attention of scholars in many industry fields, such as machinery, pressure vessels, pipelines, etc. [27,28]. It is widely studied by many scholars for its advantages over other magnetic testing methods [29–31], and the stress-magnetism effect has also been widely studied [32–34]. The impact of stress or strain on Magnetic Flux Leakage (MFL) signals of ferromagnetic materials has been researched for assessing stress concentration [35,36] and fatigue damage [37–39]. The previous study [40] showed that the ambient stress conditions during corrosion defect formation would affect the MFL signals based on magnetic flux leakage testing. In recent years, MMMT was applied to the detection of rebar corrosion in concrete and the commonly used steel strand in structural engineering to locate the corrosion area and evaluated the corrosion degree [41,42]. However, there is still a lack of research on stress corrosion. Therefore, it is very promising to detect stress corrosion defects by MMMT technology.

In this paper, three experiments were performed to study the effect of corrosion on magnetic leakage signals. The electrochemical corrosion system was designed for corrosion experiments. The distribution regularity of MFL signals during the corrosion process was investigated. The feasibility of evaluating the degree of corrosion of ferromagnetic materials using the MFL signals was discussed.

#### 2. Experimental Section

To study the effect of corrosion on MFL signals, three different experiments were performed. Experiment 1 is a static tensile experiment; experiment 2 and experiment 3 are corrosion experiments: Experiment 2 is a corrosion experiment without loading; experiment 3 is a corrosion experiment after loading. In addition, the Scanning Electron Microscope (SEM) and Energy Dispersive X-Ray Spectroscopy (EDX) were used to observe the surface of the specimen and analyze the corrosion layer, respectively.

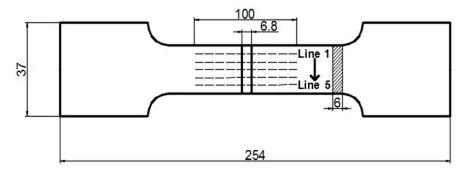
#### 2.1. Specimen Preparation

The specimens were made by 0.45% C steel in this study, which is medium-carbon structural steel and widely used in the structural engineering field. Composition of 0.45% C steel, % by weight is C (0.42–0.50%), Si (0.17–0.37%), Mn (0.50–0.80%), P ( $\leq$ 0.035%), S ( $\leq$ 0.035%). Its mechanical properties are listed in Table 1. Flat dog-bone specimens designed based on GB/T228-2002 were prepared without (used in corrosion experiments) and with a groove (used in the static tensile experiment). The rectangular groove (6.8 mm width and 2 mm depth) was carefully made by a linear cutting machine in the middle of the specimens. The specimen was designed to be a special shape (dog bone) to obtain a uniform distribution of stress in the corrosion area.

Table 1. Mechanical p	properties of 0.45% C steel.
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Material	Yield Strength $\sigma_s$ (MPa)	Ultimate Strength $\sigma_b$ (MPa)	Elongation Rate $\delta_5$ (%)
0.45% C steel	≥355	≥600	16

The shape and size of the 6-mm-thick specimens used in the experiments were drawn in Figure 1. Five scanning lines with a length of 100 mm, denoted as Line 1, Line 2, Line 3, Line 4, and Line 5, were parallel to each other at intervals of 4.5 mm. Line 3 is the centerline of the specimen.

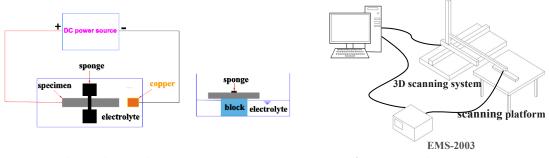


**Figure 1.** Geometry and dimensions of the specimen used in the experiments and scanning lines (in mm).

Before experiments, all of the specimens were polished to grade P1000 grit, followed by rinsing with isopropyl alcohol, drying in warm air, and weighing with an electronic balance. The specimen was polished only to remove the corrosion on the specimen surface and measure the mass loss of the specimen more accurately. In addition, every specimen was demagnetized before it was loaded or corroded to eliminate the effect of mechanical processing on the magnetic field.

#### 2.2. Description of Corrosion and Detection System

The schematic of the set-up for the corrosion system is shown in Figure 2a. The electrochemical method was used to accelerate the corrosion of the specimen. The working electrode is the specimen connected to the positive electrode of the direct current (DC) power source (UTP3303, produced by UNI-TREND Technology (China) CO. LTD). The copper sheet was served as a counter electrode connected to the negative electrode of the DC power source. There is no direct contact between the electrolyte and the specimen. The electrolyte contacts the surface of the specimen placed on a plastic block above the electrolyte level through a specially shaped sponge (Corrosion area =  $175 \text{ mm}^2$ ). The electrolyte for the corrosion process was 3.5% (wt.%) NaCl solution. In corrosion experiments, the specimen is corroded by galvanostatic corrosion of 0.4 A.



(a) electrochemical corrosion system

(**b**) 3D MFL detection system

**Figure 2.** Schematic of the electrochemical corrosion system and the 3D Magnetic Flux Leakage (MFL) detection system.

Two Faraday's laws of electrolysis govern the mass of anodic material corroded or consumed when the current passes through the circuit.

$$\Delta m = KQ = KIt \tag{1}$$

Hence, the mass of substance ( $\triangle m$ , theoretical):

$$\triangle m, \text{ theoretical} = \frac{3600MIt}{nF} \tag{3}$$

where *K* is electrochemical equivalent; *n* is the valence of the substance; *M* is the atomic mass or the molecular weight of the substance; *F* is Faraday's constant, 96,494 C/mole; *I* is the current in the electrochemical reaction; *t* is time in hours.

The corrosion experiments were divided into four stages: Accelerated corrosion was performed by the galvanostatic corrosion method using 0.4 A current for 2 h in the first corrosion stage, and the corrosion products of specimens were rinsed off carefully. Then the specimens were dried in warm air and weighed with an electronic balance to calculate the mass loss  $\Delta m$ . The geometric dimensions (width and depth) of corrosion defects at every scanning line of specimens were measured to calculate the average value of them. After that, the specimens were corroded again for 2 h and repeated the above procedure until the specimens were corroded for 8 h.

Corrosion at different stages are given in Table 2. Photographs of the corroded specimen, at various corrosion stages during the experiment, and the schematic of corrosion defect are shown in Figure 3. Variations of mass loss of the specimen ( $\Delta m$ , measured and  $\Delta m$ , theoretical) and defect feature (depth *h*) with corrosion time (*t*) are shown in Figure 4. It is found that the defect depth is associated with mass loss ( $\Delta m$ , measured). With the increase of corrosion time, the defect depth and mass loss (measured and theoretical) have the same variation tendency.

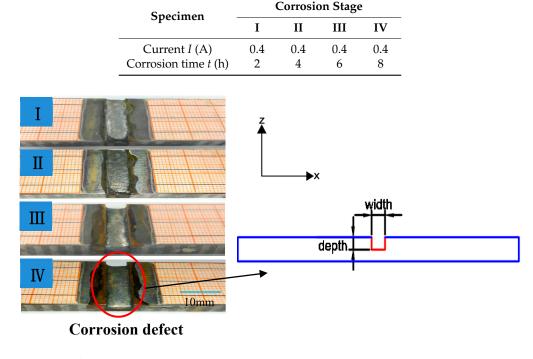
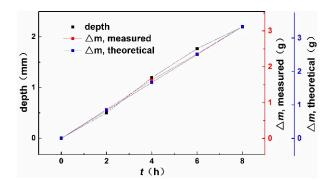


Table 2.	Corrosion	at different	stages.
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(a) Photographs at various corrosion stages of the corroded specimen

(b) Schematic of corrosion defect

**Figure 3.** (a) Photographs at various corrosion stages of the corroded specimen: (I) 2 h after an experiment is started; (II) 4 h after an experiment is started; (III) 6 h after the start of the experiment; (IV) 8 h after the start of experiment; (b) schematic of corrosion defect.



**Figure 4.** Variations of mass loss of the specimen ( $\triangle m$ , measured and  $\triangle m$ , theoretical) and defect feature (depth) with corrosion time (*t*).

The three-dimensional (3D) MFL detection system involves three main parts: a 3D scanning console system, a commercial EMS2003 metal magnetic memory testing apparatus, and a non-magnetic scanning platform. A Hall probe with a resolution of 1 A/m is a part of the testing apparatus, which was fixed on a non-magnetic 3D scanning console system controlled by a computer moving automatically at a uniform detection speed (90 mm/min). The specimens were placed on the scanning platform along the south-north direction when the MFL signals were detected. The data were stored in the EMS2003 metal magnetic memory testing apparatus. The schematic of the 3D MFL detection system is shown in Figure 2b.

#### 2.3. Testing Method

#### 2.3.1. Static Tension Experiment

The tensile tests were performed with a universal testing machine named SANS-100kN, whose load error is within  $\pm 0.5\%$ . Before loading, all specimens were demagnetized and put on the platform along the south-north direction to eliminate the influence of the sample orientation. The normal and tangential components of the demagnetization signals, namely  $H_n$  and  $H_t$  signals, were measured along the south-north direction using the 3D MFL detection system under the condition of room temperature. Tensile loads were applied at an increment of 5 kN. The specimens were measured offline to eliminate any possible effect of the machine, and the MFL signals of specimens were measured in the same way as the demagnetization signals. Then the specimens were loaded to a specific value, and the procedure was repeated until the specimen was loaded to 25 kN.

#### 2.3.2. Corrosion Experiment without Loading

The experiment was divided into four stages. Before the experiment, the specimens were demagnetized and put on the platform along the south-north direction. The demagnetization signals of each scanning line were measured in the same way as the static tension experiment. Then the specimens were corroded for 2 h, and the corrosion products were rinsed off carefully. The specimens were dried in warm air and weighed. The geometric dimensions (width and depth) of corrosion defects were measured, and the MFL signals of every scanning line were measured in the same way as the demagnetization signals. After that, the specimens were corroded again for 2 h, and the procedures were repeated until the specimens were corroded for 8 h.

#### 2.3.3. Corrosion Experiment after Loading

The measurement obtained yield stress of 50 kN for material before the experiment, so the specimens were loaded to a pre-determined value 25 kN. Before the experiment, the specimens were demagnetized and put on the platform along the south-north direction. The demagnetization signals of each scanning line were measured in the same way as the static tension experiment before load.

Then the specimens were loaded on the machine, and the MFL signals of every scanning line were measured offline in the same way as the demagnetization signals. After that, the specimens were corroded in the same way as experiment 2 (corrosion without loading).

#### 2.3.4. Surface Characterization

A sample was prepared for SEM to observe the surface of the uncorroded sample. The dimensions of this sample after being cut were  $10 \times 5 \times 3$  mm. The surface of the sample was wet ground through successive grades of SiC papers from P180 to P2000 and then polished until there were no scratches. Finally, the sample was immersed in etching reagent, 4mL HNO3 + 96 mL ethanol, for 5–10 s. The surface of the specimen was observed by LEO-1450 SEM equipped with KEVEX-Superdry EDX.

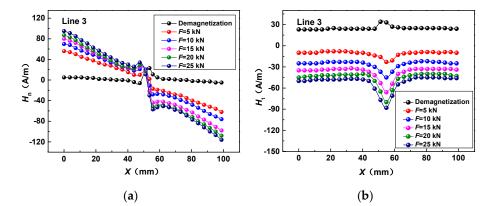
After the corrosion experiment, the dimensions of corrosion sections being cut were  $10 \times 5 \times 3$  mm to observe the surface of the sample by SEM. In addition, the cross-section of the sample was found to analyze the thickness and composition of the corrosion layer. To prevent the corrosion layer on the surface of the sample from being damaged artificially during grinding and polishing, the sample was cast into epoxy resin. Then the cross-section of the sample was wet ground and polished by the above treatment method. SEM and EDX observed the cross-section of the sample.

### 3. Results and Discussion

It should be noted that three specimens in each experiment (experiment 1, 2 and 3) were tested. MFL signals on all testing lines are the same. Thus, the MFL signals on Line 3 are shown here.

#### 3.1. Experimental Results

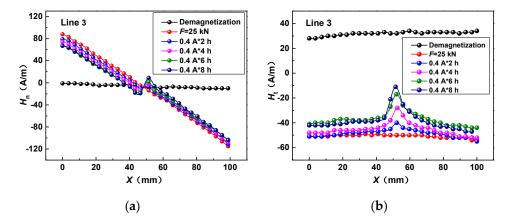
The distributions of MFL signals ( $H_n$  and  $H_t$ ) during experiment 1 (static tension) under different loads were plotted in Figure 5. It is found that the normal component  $H_n$  and the tangential component  $H_t$  presented different signal characteristics at the defect during the load process from Figure 5. As can be seen in Figure 5, the demagnetization signals at the defect presented prominent distortion characteristics.  $H_n$  presented a "trough-peak" shape, while  $H_t$  presented a "peak" shape. Compared with the demagnetization signals, the MFL signals at the defect occurred significant changes after loading. The normal component  $H_n$  changed from "trough-peak" to "peak-trough," while the tangential component  $H_t$  changed from "peak" to "trough"; and the baseline amplitude value of  $H_n$  and  $H_t$  besides the defect increased dramatically. In addition, with the increase of the applied load, the abnormal peak amplitude of  $H_n$  and  $H_t$  increased. The theory of the magnetic charge can explain the change of MFL signals during the load process [39,41].



**Figure 5.** MFL signals distributions of normal and tangential components during experiment 1 (static tension) at different loads measured on Line 3 (**a**) and (**b**), respectively.

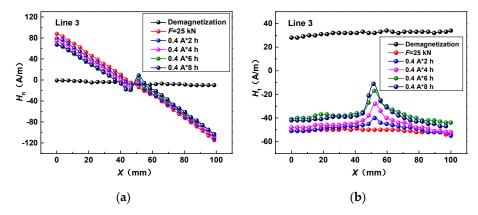
The distributions of MFL signals ( $H_n$  and  $H_t$ ) during experiment 2 (corrosion without loading) at different corrosion degrees were plotted in Figure 6. It is found that  $H_n$  and  $H_t$  of the demagnetization

signal presented other characteristics.  $H_n$  showed an approximately linear change along Line 3 with a shallow gradient, while  $H_t$  kept a nearly constant value. As can be seen in Figure 6,  $H_n$  and  $H_t$ presented different distortion characteristics at the defect during the corrosion process.  $H_n$  presented a "trough-peak" shape, while  $H_t$  presented an apparent "peak" shape. With the increase of corrosion degree, the abnormal peak amplitude of  $H_n$  and  $H_t$  increased. The results indicated that both  $H_n$  and  $H_t$  presented abnormal peak features during the corrosion process, showing the location of corrosion defect accurately.



**Figure 6.** MFL signals distributions of normal and tangential components during experiment 2 (corrosion without loading) at different corrosion degrees measured on Line 3 (**a**) and (**b**), respectively.

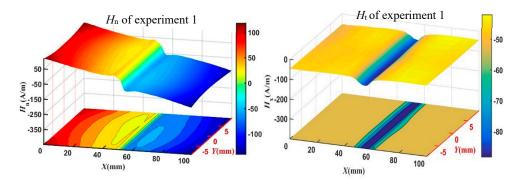
The distributions of MFL signals ( $H_n$  and  $H_t$ ) during experiment 3 (corrosion after loading) at different corrosion degrees were plotted in Figure 7. It is found that  $H_n$  and  $H_t$  of the demagnetization signal also presented different characteristics.  $H_n$  showed an approximately linear change along Line 3 with a shallow gradient, while  $H_t$  kept an almost constant value. After the specimen was loaded, the gradient and amplitude of  $H_n$  increased significantly, while changes only in amplitude could be found for  $H_t$ , still almost horizontal linear distribution. Such a phenomenon is due to the piezomagnetic effect [43]. It is known that applied stress can lead to the reorientation of magnetic domains along the tensile direction [44]. As can be seen in Figure 7,  $H_n$  and  $H_t$  presented different distortion characteristics at the defect during the corrosion process.  $H_n$  presented a visible "trough-peak" shape, while  $H_t$ presented an apparent "peak" shape. With the increase of corrosion degree, the abnormal peak amplitude of  $H_n$  and  $H_t$  increased significantly. At the same time, the gradient of  $H_n$  and the baseline amplitude value of  $H_n$  and  $H_t$  decreased. The results of experiment 3 (corrosion after loading) indicated that both  $H_n$  and  $H_t$  presented abnormal signal characteristics during the corrosion process, showing the location of the corrosion defect accurately.



**Figure 7.** MFL signals distributions of normal and tangential components during experiment 3 (corrosion after loading) at different corrosion degrees measured on Line 3 (**a**) and (**b**), respectively.

#### 3.2. Comparison of Experimental Results

The normal components  $H_n$  and the tangential component  $H_t$  of MFL signals in the last stage in three experiments were selected to compare the characteristics of signals under different working conditions. The surface and contour maps of signals in the chosen stage were drawn in Figure 8. As can be seen in Figure 8a,  $H_n$  and  $H_t$  of experiment 1 at the defect showed apparent "peak-trough" shape and "trough" shape, respectively. As can be seen in Figure 8b,*c*,  $H_n$  and  $H_t$  at the defect presented a "trough-peak" shape, and "peak" shape, respectively. Figure 9a shows the comparison of unprocessed  $H_n$  in three experiments. For the convenience of comparison,  $H_n$  and  $H_t$  were further processed.  $H_n$  subtracts the baseline amplitude, and the baseline amplitude of  $H_t$  returns to zero. Figure 9b shows the comparison of  $H_n$  subtracting the baseline amplitude value in three experiments. Figure 9c shows the comparison of  $H_t$  where the baseline amplitude returns to zero in three experiments. It is found that  $H_n$  and  $H_t$  presented different signal characteristics under different working conditions: When the specimen was subjected to tensile load and corrosion, respectively,  $H_n$  and  $H_t$  both exhibited opposite signal characteristics. When the specimen was corroded without and after loading,  $H_n$  and  $H_t$  both showed the same signal characteristics, and a more substantial amplitude value could be found in corrosion experiment after loading.





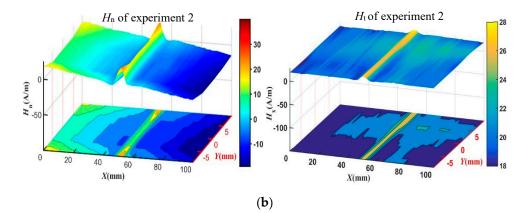
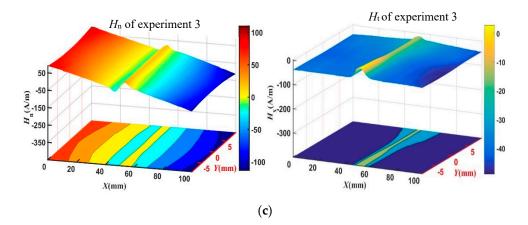
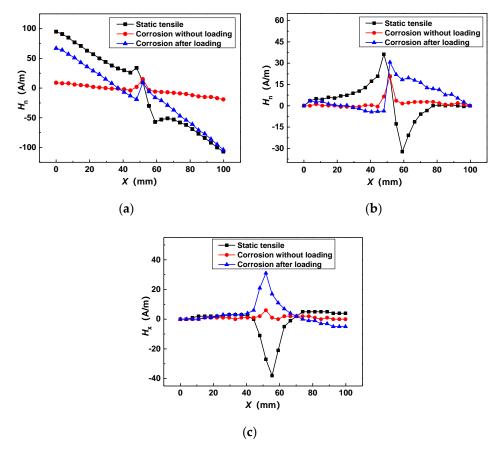


Figure 8. Cont.



**Figure 8.** Surface and contour maps of the normal and tangential components of MFL signals. (a) Loading to 25 kN in experiment 1 (static tension); (b) 8 h of corrosion in experiment 2 (corrosion without loading); (c) 8 h of corrosion in experiment 3 (corrosion after loading).



**Figure 9.** Comparison of  $H_n$  and  $H_t$  in the last stage in three experiments. (a) The normal components  $H_n$ ; (b) the normal components  $H_n$  (subtracting baseline amplitude value); (c) the tangential component  $H_t$  (zero in baseline amplitude value).

# 3.3. Characterization of the Corrosion Layer

The SEM micrographs of the surface of the samples are shown in Figure 10. As can be seen in Figure 10a, the microstructure of the uncorroded sample was composed of lamellar pearlite and white ferrite. As can be seen in Figure 10b, the surface of the sample was covered with relatively uniform corrosion products after 8 h of corrosion. Figure 11 shows the cross-section SEM image, analytical line, and corresponding X-ray maps of Fe and O for the sample corroded for 8 h. The thickness

of the corrosion layer is relatively uniform, approximately 16  $\mu$ m. As can be seen in X-ray maps, the corrosion layer mainly contained oxygen and iron elements. Compared with the iron matrix, the content of the oxygen element in the corrosion layer increased obviously. In contrast, the content of iron decreased obviously, indicating that the corrosion layer was mainly constituted of iron oxides. Therefore, the permeability of the corrosion layer is much less than iron [45], and the corrosion layer reduced the permeability of the sample.

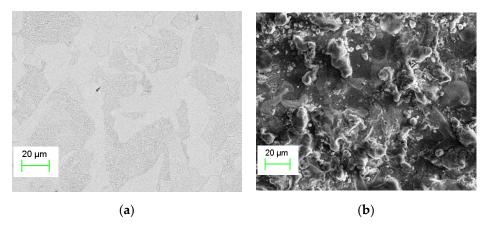
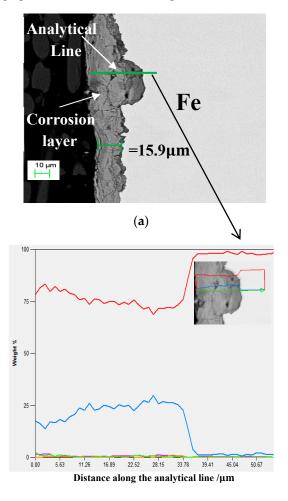
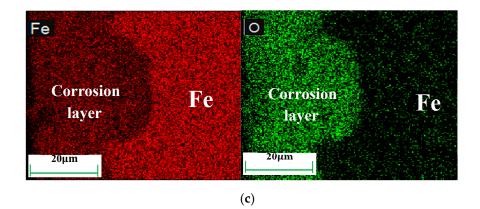


Figure 10. SEM micrographs of the surface of the samples: (a) no corrosion; (b) 8 h of corrosion.



<sup>(</sup>**b**)

Figure 11. Cont.



**Figure 11.** SEM and EDX observed the cross-section of the sample corroded for 8 h: (**a**) Cross-section SEM image; (**b**) analytical line; (**c**) corresponding X-ray maps of Fe and O.

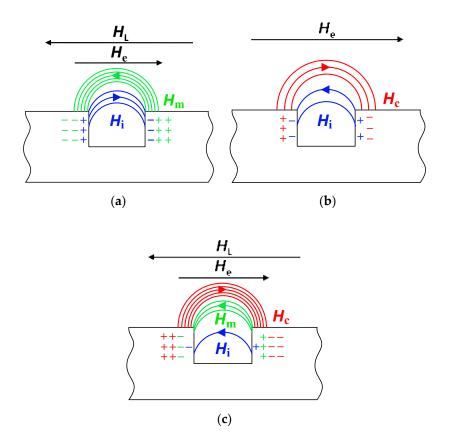
#### 3.4. Analysis and Discussion

The magnetic charge is an equivalent model, which has been developed to interpret the magnetic field leakage of ferromagnetic materials. For a ferromagnet, its external magnetic field would be considered to originate from the magnetic charge [41]. For the convenience of discussion, it is assumed that the magnetic flux leakage was caused by the equivalent magnetic charge on the two interfaces of the defect zone. Then, the MFL *H* at the measuring point caused by equivalent magnetic charge *q* is given as

$$H = \frac{q}{4\pi\mu_0 r^3} r \tag{4}$$

where  $\mu_0$  is the permeability of air, *r* is the position vector from equivalent magnetic charge to measuring point.

The schematic diagram of MFL signals in experiment 1 (static tensile) is shown in Figure 12a. The initial leakage magnetic field  $H_i$  is formed by the magnetic polarity generated due to the accumulation of a small number of magnetic charges at both sides of the defect after the specimen was demagnetized [39,41,46], which coincides with the initial magnetic field  $H_e$ . The demagnetization signals are the vector sum of  $H_e$  and  $H_i$ . Thus,  $H_n$  presents a "trough-peak" shape, and  $H_t$  gives a "peak" shape at the defect. After being loaded, the magnetization of the specimen changes under the combined action of the geomagnetic field and stress [43,44], results in the formation of leakage magnetic field  $H_{\rm L}$ . The accumulation of more magnetic charges at both sides of the defect due to the stress concentration at the defect after the specimen was loaded, resulting in a strong leakage magnetic field  $H_{\rm m}$ , which is opposite to the leakage magnetic field  $H_{\rm i}$ . In addition, the leakage magnetic field  $H_{\rm m}$  is stronger than  $H_{\rm i}$ , causing the reversal of MFL signals at the defect. Under such conditions, the MFL signals are the vector sum of the initial magnetic field  $H_{\rm e}$ , leakage magnetic field  $H_{\rm L}$ ,  $H_{\rm m}$ , and  $H_i$ . Thus, the normal component  $H_n$  of MFL signals at the defect presents a "peak-trough" shape, and the tangential component  $H_t$  shows a "trough" shape. With the increase of the load, the stress concentration on both sides of the defect increases, increasing the number of magnetic charges, which leads to the strengthening of leakage magnetic field  $H_m$ , causing the increase of the MFL signals at the defect (see Figure 5).



**Figure 12.** Schematic diagram of MFL signals in three experiments: (**a**) experiment 1 (static tensile); (**b**) experiment 2 (corrosion without loading); (**c**) experiment 3 (corrosion after loading).

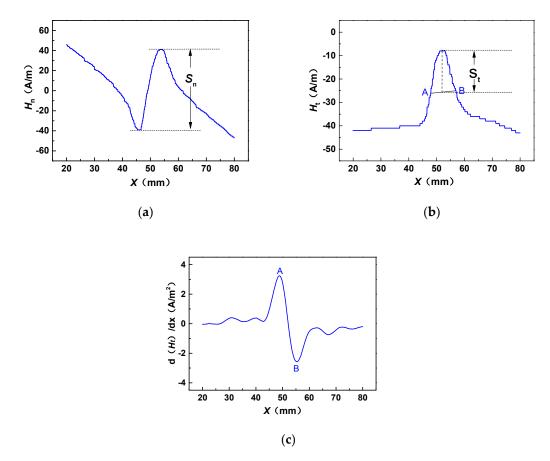
The schematic diagram of MFL signals in experiment 2 (corrosion without loading) is shown in Figure 12b. The defect formed on the surface of the specimen due to corrosion, results in the formation of an initial leakage magnetic field  $H_i$ . In addition, magnetic charges accumulated at both sides of the defect due to the effect of corrosion, result in the formation of leakage magnetic field  $H_c$ , which is opposite to the initial magnetic field  $H_i$ . Under such conditions, the MFL signals are the vector sum of the initial magnetic field  $H_e$ , leakage magnetic field  $H_c$  and  $H_i$ . Thus, the normal component  $H_n$  of MFL signals presents a "trough-peak" shape, and the tangential component  $H_t$  presents a "peak" shape. With the increase of corrosion degree, the leakage magnetic field  $H_c$  is gradually enhanced, causing the MFL signals increasing at the defect (see Figure 6).

The schematic diagram of MFL signals in experiment 3 (corrosion after loading) is shown in Figure 12c. Different from experiment 2, the specimen in experiment 3 was corroded after having been loaded, so a strong leakage magnetic field  $H_c$  was generated at the defect after the specimen was corroded, which is stronger than the leakage magnetic field  $H_m$  and  $H_i$ . Under such conditions, the MFL signals are the vector sum of the initial magnetic field  $H_e$ , leakage magnetic field  $H_L$ ,  $H_c$ , Hm, and  $H_i$ . Thus, the normal component  $H_n$  of MFL signals presents a "trough-peak" shape, and the tangential component  $H_t$  presents a "peak" shape. With the increase of corrosion degree, the vector sum of leakage magnetic field  $H_c$ ,  $H_m$ , and  $H_i$  increases, causing the MFL signals at the defect gradually increasing (see Figure 7).

#### 4. Quantitative Evaluation Parameters

To further analyze the relationship between the MFL signals and corrosion degree, two characteristic magnetic parameters  $S_n$  and  $S_t$  were defined. The parameters  $S_n$  and  $S_t$  were extracted from the MFL signals of representative scanning line (Line 3) in experiment 3 (corrosion after loading), and the calculation methods of magnetic characteristic parameters are shown in Figure 13. For the

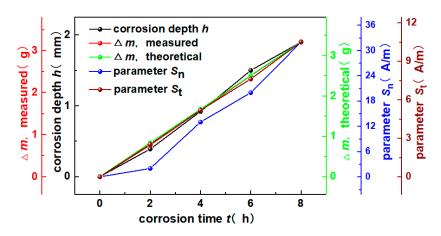
normal component presented in Figure 13a, the parameter  $S_n$  can be obtained directly. The parameter  $S_n$  is calculated by taking the signal difference between the peak and trough of the  $H_n$ . However, the tangential component  $H_t$  presented in Figure 13b and the parameter  $S_t$  cannot be obtained straightforwardly. Figure 13c shows the gradient of  $H_t$ , the position of the peak and trough of the gradient, A and B, can be obtained directly. Therefore, the parameter  $S_t$  can be calculated by taking the distance from the peak position of  $H_t$  to line AB.



**Figure 13.** The calculation methods of magnetic characteristic parameters: (**a**) calculation method of  $S_n$ ; (**b**) calculation method of  $S_t$ ; (**c**) gradient of  $H_t$ .

Variations of characteristic magnetic parameters ( $S_n$  and  $S_t$ ), mass loss of the specimen ( $\Delta m$ , measured and  $\Delta m$ , theoretical), and defect feature (corrosion depth h) with corrosion time (t) are shown in Figure 14. It is found that magnetic characteristic parameters, mass loss, and corrosion depth have the same variation tendency.  $S_t$  is considered to be the best parameter for quantitatively evaluating the degree of the corrosion among the four characterization parameters.

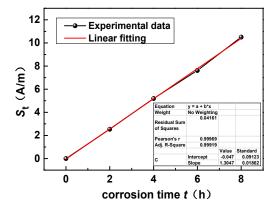




**Figure 14.** Variations of characteristic magnetic parameters ( $S_n$  and  $S_t$ ), mass loss of the specimen ( $\Delta m$ , measured and  $\Delta m$ , theoretical), and defect feature (corrosion depth *h*) with corrosion time (*t*).

The relationship between the parameter  $S_t$  and the corrosion time t is shown in Figure 15. It is found that the parameter  $S_t$  and the corrosion time t have established a good linear relationship. After linear fitting, the equation defining the dependence of  $S_t$  on t was found, and the correlation coefficient  $R^2 = 0.999$ :

$$S_{\rm t} = -0.0476 + 1.3047t \tag{5}$$



**Figure 15.** The relationship between the parameter  $S_t$  and the corrosion time t.

The direct dependence of mass loss ( $\triangle m$ , theoretical) on the parameter  $S_t$  was obtained by Equations (3) and (5) as follows:

$$\Delta m, \text{ theoretical} = \frac{3600(S_t + 0.0476)}{1.3047nF} MI$$
(6)

At the same time, the width change of the corrosion defect is small in the corrosion test, so the area of the corrosion area *S* is considered unchanged. The corrosion depth *h*, theoretical can be calculated from the mass loss  $\triangle m$ , theoretical:

$$h$$
, theoretical =  $\frac{\Delta m$ , theoretical  $\rho S$  (7)

where  $\rho$  is the density of the specimen. The direct dependence of corrosion depth (*h*, theoretical) on the parameter *S*<sub>t</sub>, *h*, theoretical = *f*(*S*<sub>t</sub>), was obtained by Equations (6) and (7) as follows:

h, theoretical = 
$$\frac{3600(S_{\rm t} + 0.0476)}{1.3047nF\rho S}MI$$
(8)

In order to verify the reliability of the expression, the comparison between corrosion depth h, theoretical obtained by parameter  $S_t$  and h, measured in experiment 3 (corrosion after loading) is shown in Figure 16. Meanwhile, corrosion depth h, measured in experiment 2 (corrosion without loading), is also shown in Figure 16. The average error rates of experiment 2 and experiment 3 were calculated to be 15.65% and 8.94%, respectively. It is found that the expression can predict the corrosion depth well and has a smaller error under the same working condition.

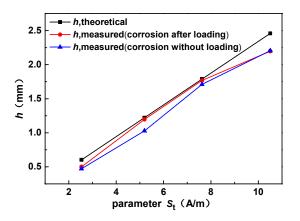


Figure 16. Comparison between measured and theoretical values of the corrosion depth *h*.

## 5. Conclusions

In this paper, static tensile experiments, corrosion experiments without and after loading were performed to study the effect of corrosion on MFL signals. The normal and tangential components of the MFL signals, namely  $H_n$  and  $H_t$ , were measured using the 3D MFL detection system, and the distribution regularity of them during the corrosion process was investigated. The following conclusions are obtained:

- (1) The normal and tangential components of the MFL signals in corrosion experiments without and after loading presented a "trough-peak", and "peak" shape, respectively. It is known from experimental results that when the specimen was corroded without and after loading, the normal and tangential components both showed the same signal characteristics, and a more substantial amplitude value could be found in the corrosion experiment after loading.
- (2) The normal and tangential components of the MFL signals in the static tensile experiment presented a "peak-trough", and "trough" shape, respectively. It is found that the normal and tangential components showed different signal characteristics under different working conditions: When the specimen was subjected to tensile load and corroded, respectively, the normal and tangential components both exhibited opposite signal characteristics. According to the analysis, the critical reason for different signal characteristics is that the leakage magnetic field  $H_c$  generated at the defect after the specimen has been corroded is opposite to the leakage magnetic field  $H_m$  caused by force.
- (3) The characterization parameters of MFL signals,  $S_n$  and  $S_t$ , are capable of evaluating the corrosion degree of the specimen, and  $S_t$  is better. The characterization parameter of the tangential component of the MFL signals is more sensitive to the corrosion, which is different from the traditional research results on the MFL signals caused by stress where the characterization parameter of the normal component of the MFL signals is more sensitive to the force. The direct dependence of corrosion depth (*h*, theoretical) on the parameter  $S_t$ , *h*, theoretical =  $f(S_t)$  was developed and the average error rates between the predicted and measured values are 8.94% under the same working condition. Therefore, the expression can be used to evaluate the corrosion degree of the specimen quantitatively.

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