

Table of experimental methods to assess redox status.

Method/Main Component	Input or general Information	Output	Challenge	Reference
ROS				
Ferricytochrome c reduction to ferrocycytochrome c	ferricytochrome c reduction to ferrocycytochrome c Absorbance based	Indicator for ROS formation rate	Outdated - indirect + less sensitive compared to current methods	(19)
Lucigenin (redox cycling) Coelenterazine	Chemilumenscent reactions to determine ROS production	Indicator for ROS formation	Lucinogen - not ideal for quantitative measures Coelenterazine does not redox cycle - not specific to ROS alone - also to peroxnitrite	(10)
Hydrocyanine dyes	Fluorogenic sensors for ROS and hydroxyl radicals, functions both <i>in vitro</i> and <i>in vivo</i>	Indicator for ROS and hydroxyl radicals	High auto-oxidation Low strokes shift Low solubility	(16)
Dihydroethidium	Fluorescence until oxidized by ROS and RNS Spectroscopy	Marker product of ROS and hydroethidine	Differentiation between ROS and RNS oxidative function not possible - only semi quantitative (if conditions are compared)	(36)
ROS	<ul style="list-style-type: none"> • redox-sensitive fluorophores • fluorescent reporters • electron paramagnetic resonance spectroscopy • HPLC 	ROS quantity		(32)
Mitochondrial ROS				
MitoSOX™ assay	Cationic derivative of dihydroethidium - permeates live cells and targets mitochondria - In live cells - fluorescence	Mitochondrial ROS	Cannot differentiate between different ROS species	(15)
CellROX®	Cell-permeant dyes that are weakly fluorescent in reduced state and exhibit stable fluorescence upon oxidation by ROS	Dependent on specific CellROX type either mitochondrial ROS	Conflicts with H ₂ O ₂ responses	(33)
Mitochondrial membrane potential indicator for OS	In vitro method Usually coupled with oxygen consumption rate Time-series TMRM fluorescence measurement	Indicator for ROS mediated cell injury		(31)
H₂O₂ detection				

Amplex ® Red	Amplex Red is oxidized by H ₂ O in presence of HRP and then converted to resorufin Detection colorimetric or fluorescence	H ₂ O ₂ detection	Highly sensitive to light exposure and then less sensitive to H ₂ O ₂	(35)
Homovanillic acid	Oxidizes with H ₂ O ₂ through HRP Detection via fluorescence	H ₂ O ₂ detection	Sensitive to background signal especially when polystyrene microplates are used	(23)
TMB	Oxidizes with H ₂ O ₂ through HRP	H ₂ O ₂ detection	Less sensitive, but cheaper General drawback with HRP catalysis - thiols skew the result	(34)
H2DCF	Is reduced to DCF by H ₂ O ₂ (and other ROS) Detection via fluorescence	H ₂ O ₂ detection	Originally used for H ₂ O ₂ detection until it was shown not to be specific to H ₂ O ₂ H ₂ O ₂ specific extensions: Peroxy Green1 and Peroxy Crimson1	(20)
Nitric Oxide				
Nitrate and nitrite Follow up with nitrite only assay “Griess assay”	NO ₃ reduction to NO ₂ by nitrate reductase. Nitrite forms nitrous acid in presence of hydrogen ions - nitrous acid reacts with sulfanilamide and produces diazonium ion Detection colorimetric	Indirect NO detection	Indirect and unspecific	(11)
NADH and NADPH	Detection via cellular autofluorescence + FLIM	NAD(P)H redox state (separation of NADH and NADPH)		(3)
Fluorometric probe: DAF-2	DAF-2 reacts with either N ₂ O ₃ or NO ₂ to form highly-fluorescent triazole	Detection of NO and RNS	Susceptible to experimental artifacts (increased fluorescence by Ca ²⁺ ions, incident light and non-neutral pH). Cross reactivity with peroxynitrite, nitroxyl, and ascorbic acid	(13)
Fluorometric probe: DAN	DAN reacts with nitrous acid and forms 2,3 naphthotriazole Detection via fluorescence Works both <i>in vitro</i> and <i>in vivo</i>	Indicator of NO formation	Not able to inform about area of NO production	(4)
Bioassays cGMP level Vessel relaxation	Details stated in method title	Measurement of downstream actions of NO	Indirect measurement Do not adjust for other confounding factors	(21)

Inhibition of platelet aggregation				
Glutathione				
HPLC	Uses iodoacetic acid to modify thiols and 1-fluoro-2,4-dinitrobenzene to add a chromophore to the amino groups, allowing direct measurement of GSH, GSSG, Cys, CySS, and CySSG	GSH to GSSG ratio	Not sensitive enough to measure GSSG in plasma	(14)
<ul style="list-style-type: none"> • Capillary electrophoresis • Free zone CE • Micellar electrokinetic chromatography • Isotachopheresis • Capillary gel electrophoresis • electrochromatography 	Electrophoretic separation by application of high electric field to sample in capillary - background electrolyte (BGE)	GSH to GSSG ratio	Limited analyte loading leads to diminished sensitivity	(6)
Microplates		GSH to GSSG ratio	Prone to human error	(2)
Luciferase assay (example Promega)	Luciferin derivative glutathione s-transferase - GSH amount proportional to luminescent signal when luciferase is added	GSH quantity	Prone to human error	(25)
Colorimetric assay:	GSH + DTNB + glutathione reductase	Reduced GSH quantity	Prone to human error	(12)
Monobromobimane	Monobromobimane and monochlorobimane react with low molecular weight thiols (incl glutathione) and form fluorescent adducts	GSH ratio before and after chemical reaction	Monochlorobimane more thiol selective however, accumulates in the nucleus	(17)
ThiolTracker Violet	Reacts with reduced thiols in intact cells → follow up flow cytometry and fluorescence microscopy	Glutathione detection	Measures thiols in general, GSH is approximately measured	(18)
Lipid peroxidation				
MDA	Reaction of membrane unsaturated fatty acids with free radicals generates lipid peroxides - MDA and TBA adduct can be measured	TBA	Highly sensitive but not specific to MDA	(26)

IsoP Elisa	Stated in method title	IsoP quantity	Requires purification of samples	(22)
IsoP GC/MS <i>In vivo</i>	Plasma or urinary samples GC/MS	IsoP quantity	High cost	(28)
BODIPY® Image-IT®	Live cells - fluorescent-Mass spectrometry	IsoP ratio	Sensitive to photo bleaching in high-intensity illumination conditions	(8)
LAA Click-iT® Alexa Fluor ®	LAA incorporates into cellular membrane and is oxidized in response to lipid peroxidation - LAA then produces HPODE that modifies surrounding proteins - the resulting alkyne-containing proteins can be detected using AlexaFluor or Click-iT	Lipid peroxidation reporter	Cells must be fixed	(37)
•HPLC •HPLC-MS	Used to separate lipid mixtures in fractions, based on molecular weight and polarity.. Gold standard for quantification of known biomarkers	<ul style="list-style-type: none"> • Lipid peroxidation biomarkers • Specific identification of unknown peroxidation derivatives of large nonvolatile lipids 	<p>In isolation, requires known standard.</p> <p>Quantification can be imprecise if the molecule of interest has similar molecular weight of other molecules in solution</p> <p>High financial and time cost</p>	(24)
•GC •GC-MS	Used to separate lipid mixtures in fractions, based on their volatility. Followed by MS allows identification of unknown peroxidation products	<ul style="list-style-type: none"> • Lipid peroxidation biomarkers • Specific identification of unknown peroxidation derivatives 	<p>Requires high temperatures</p> <p>Quantification can be imprecise if the molecule of interest has similar molecular weight of other molecules in solution</p> <p>High financial and time cost</p>	(9)
Redox Status				
Genetically encoded sensors	<i>In vivo</i> Change fluorescence in response to alteration in redox state i.e. roGFP - linked to Orp1 which forms disulfides after reacting with H ₂ O ₂	GSH redox potential	Monitoring redox changes in multiple compartments remains difficult	(27)
2D electrophoresis PAGE (DIGE)	Cysteine-labeling	Identification of redox sensitive proteins specific to user defined conditions	High cost	(7)

DNPH	DNPH reacts with all protein carbonyls	Protein carbonyl content	Interacts with heme groups	(1)
OxiBlot	<ul style="list-style-type: none"> • DNPH reacts with protein carbonyls • Followed by separation of oxidatively modified proteins by electrophoresis • Followed by westernblot with anti-DNP antibodies 	Protein carbonyl content	Time consuming	(29)
8-hydroxy-2' - deoxyguanosine	Detection methods: <ul style="list-style-type: none"> • HPLC • GC-MS • HPLC tandem • ELISA 	Indicator for DNA damage caused by OS	High cost and low throughput	(30)
<i>In vivo</i> Measurements of Ca ²⁺ , pH, redox state	<ul style="list-style-type: none"> • Genetically encoded fluorescent reporters • Two photon microscopy • nuclear magnetic resonance 	<ul style="list-style-type: none"> • Ca²⁺ • pH • redox state 	High expertise required Specialized equipment required High cost	(5)

Table of experimental methods to assess redox status, sorted by output. Included are the general name the method is referred to, as well as keywords describing the method and/or reactions the method is based on, outcome information, challenges and a reference referring to either a publication describing the method, or a product link. Methods that can be used for several biomarkers are only explained once.

Abbreviations			
DAF-2	: Diaminofluorescein	IsoP	: Isoprostane
DAN	: 2, 3-diaminonaphthalene	LAA	: Linoleamide alkyne
DNPH	: 2,4-dinitrophenylhydrazine	MDA	: Malondialdehyde
DTNB	: 5,5 dithio-bis-nitrobenzoic acid	N ₂ O ₃	: Dinitrogen trioxide
GC/MS	: Gas chromatography–mass spectrometry	NADH	: Nicotinamide adenine dinucleotide hydrogen
GSH	: Glutathione	NADPH	: Nicotinamide Adenine Dinucleotide Phosphate Hydrogen
GSSH	: Oxidized glutathione	NO ₂	: Nitrogen Dioxide
H2DCF	: 2,7-dichlorodihydrofluorescein diacetate	NO ₃	: Nitrate
H ₂ O ₂	: Hydrogenperoxyde	roGFP	: Reduction-oxidation sensitive green fluorescent protein
HPODE	: 3-hydroperoxyoctadecadienoic acid	TMB	: Tetramethylbenzidine

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