



Supporting Information

Tuning the Surface Structure of Polyamide Membranes Using Porous Carbon Nitride Nanoparticles for High-Performance Seawater Desalination

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Supplementary Figures

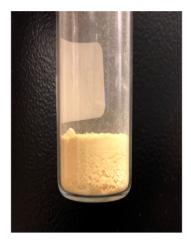
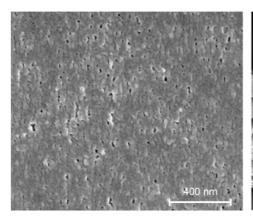




Figure S1. Optical photographs of C₃N₄ powders and the aqueous solution.



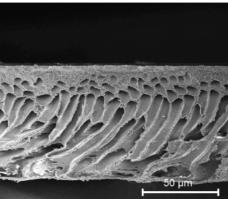


Figure S2. SEM images of the prepared PSf substrate.

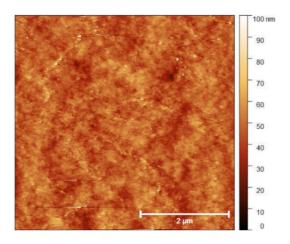


Figure S3. AFM image of the prepared PSf substrate.

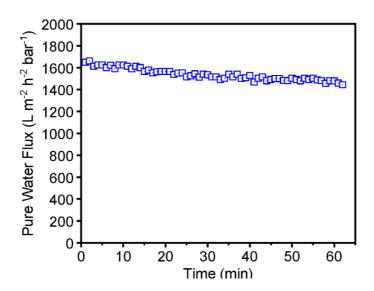


Figure S4. Pure water flux of the prepared PSf substrate.

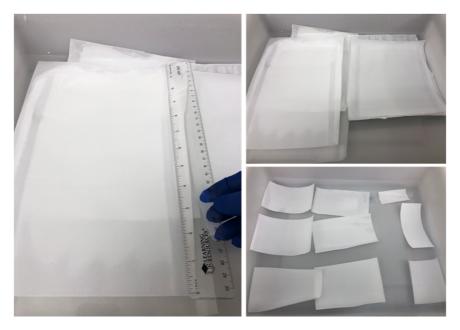


Figure S5. Optical photographs of the prepared pristine TFC and TFN membranes.

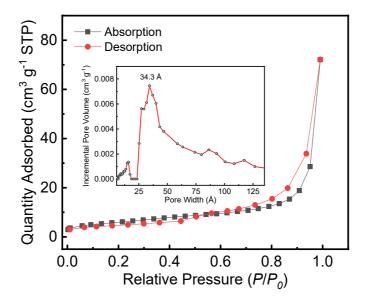


Figure S6. N2 adsorption isotherm (G), and pore size distribution (G inset) of the prepared C3N4.

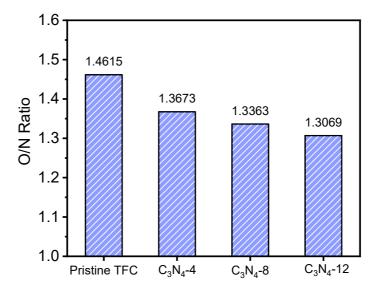


Figure S7. O/N ration of PA layer fabricated with various amounts C₃N₄ nanosheets. The degree of cross-linking was calculated by $\frac{X}{X+Y} \times 100\%$, where X and Y were calculated from the following equations, 3X+4Y = O1s and 3X+2Y=N1s.

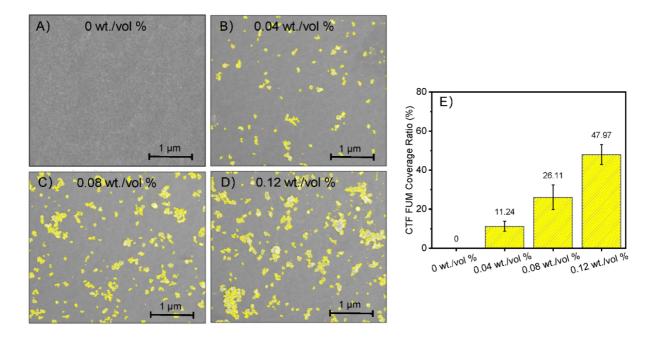


Figure S8. SEM images of C₃N₄ deposition on PSf support and surface coverage of the deposited particles on supports, as measured by *Image J* software.

We supposed that the distribution of C₃N₄ on PSf substrate after removing the excess aqueous solution has a great effect on the next step of interfacial polymerization. As shown in Fig. S8, C₃N₄ uniformly distributed on PSf support surface benefitting from their excellent dispersion in water and suitable particle size. The surface coverage ratio of the deposited C₃N₄ on the PSf surface obviously increased from 11.24% to 47.97% as the increased concentration of the particles in MPD solution from 0.04 wt./vol % (C₃N₄-4) to 0.12 wt./vol % (C₃N₄-12). Note that with increasing the C₃N₄ loading up to 0.12 wt./vol%, most of the nanoparticles were deposited well on PSf surface as a monolayer but some overlapped forming agglomeration. The interesting deposition features of the nanosheets indicate that an appropriate nanoparticles loading amount is significant.

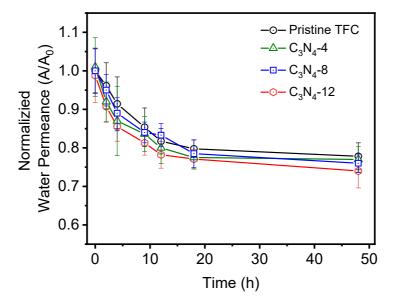


Figure S9. Normalized water permeance under long-term operation.

Supplimentary Tables

Table S1. Elementary composition of PA layer with various C₃N₄ amount.

RO Membranes	C1s [%]	O1s [%]	N1s [%]
Pristine TFC	77.6	13.3	9.1
C3N4-4	76.1	13.8	10.1
C3N4-8	77.1	13.1	9.8
C3N4-12	76.7	13.2	10.1

Table S2. *Jw*, *B* and *R* of TFC and TFN membranes.

RO Membranes	Jw [LMH/bar]	B [LMH]	R [%]
Pristine TFC	1.7 ± 0.4	0.5 ± 0.1	98.0 ± 0.4
C3N4-4	2.8 ± 0.2	0.3 ± 0.1	99.2 ± 0.3
C3N4-8	3.2 ± 0.2	0.2 ± 0.1	99.5 ± 0.2
C3N4-12	3.6 ± 0.2	1.2 ± 0.2	97.6 ± 0.4

Table S3. Comparison of the *A*, *B* and *R* of TFC and TFN membranes reported in references and in this work.

RO Membranes	ΔP [bar]	NaCl [ppm]	Jw [LMH/bar]	B [LMH]	R [%]	Ref.
PA-TFC	15.5	2000	1.7 ± 0.4	0.5 ± 0.1	98.0 ± 0.4	This work
C3N4-4	15.5	2000	2.8 ± 0.2	0.3 ± 0.1	99.2 ± 0.3	This work
C3N4-8	15.5	2000	3.2 ± 0.2	0.2 ± 0.1	99.5 ± 0.2	This work
C3N4-12	15.5	2000	3.6 ± 0.2	1.2 ± 0.2	97.6 ± 0.4	This work
Dow-SW30HR	15.5	2000	0.92 ± 0.12	0.03 ± 0.01	99.3 ± 0.1	1
Dow-BW30	15.5	2000	3.77 ± 0.13	0.53 ± 0.06	96.8 ± 0.3	1
Sepro-RO1	15.5	2000	4.90 ± 0.11	0.59	99.1 ± 0.1	2
Sepro-RO4	15.5	2000	0.97 ± 0.02	0.16	98.8 ± 0.1	2
TFN-ZIF-8	15.5	2000	3.35 ± 0.08	0.22	98.5 ± 0.5	2
TFC-1-GO	15.5	2000	5.42 ± 0.28	1.32	98.2 ± 0.7	3
TFC GO	15.5	2000	1.97	0.17	98	4
0.5 wt.% o-CNT TFC	15.5	2000	3.03	0.31	97.7	5
ASP-Silica RO	15.5	2000	4.16	2.07	96.4	6

References

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