



Supporting Information

Highly Efficient Antimicrobial Activity of $Cu_xFe_yO_z$ Nanoparticles Against Important Human Pathogens

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S1. Summary of different bacteria strains used in this work

Table S1. Bacterial strains used in this study

Strain	Relevant characteristics	Source or Reference
<i>Gram-negative</i>		
<i>Escherichia coli</i>		
strain B	Nonpathogenic	ATCC (11303)
strain O157:H7 E009	Nalidixic acid resistant	Ref. [1]
<i>Helicobacter pylori</i>		
strain X47	Mouse-colonizing strain	Ref. [2]
<i>Klebsiella pneumoniae</i>		
strain 4/484	Nosocomial, Multidrug-Resistant	Ref. [3]
strain 1100975	Multidrug resistant (New Delhi metallo-beta-lactamase 1)	ATCC (BAA2472)
<i>Salmonella enterica</i> Typhimurium		
strain 700408	Multidrug resistant	ATCC (700408)
<i>Shigella flexneri</i>		
strain 2457T	Serotype 2a	ATCC (700930)
<i>Gram-positive</i>		
<i>Listeria monocytogenes</i>		
strain Scott A	Human isolated from an outbreak linked to milk	Ref. [4]
<i>Staphylococcus aureus</i>		
Strain Rosenbach	Pathogenic	ATCC (6538)

ATCC: American Type Culture Collectio

S2. Size characterization of different $\text{Cu}_x\text{Fe}_y\text{O}_z$ NP samples

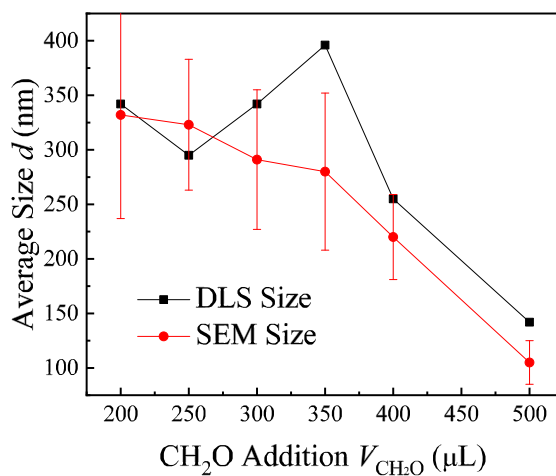


Figure S1. Size comparison between SEM data and dynamic light scattering (DLS). Different $\text{Cu}_x\text{Fe}_y\text{O}_z$ NP samples are designated by formaldehyde (μL) addition during synthesis.

Table S2 Summary of the properties of $\text{Cu}_x\text{Fe}_y\text{O}_z$ NP samples

Sample of $\text{Cu}_x\text{Fe}_y\text{O}_z$ as (L) of Formaldehyde	XRD Crystallite Size (nm)	SEM Average Size (nm)	Zeta Potential (mV)	10 hour MO Concentration <u>Reduction</u> (Dark)	24 hour MO Concentration <u>Reduction</u> (Dark)
200	35.3	300	20.9	3%	22%
250	35.3	270	-13.3	38%	71%
300	42.4	220	-8.7	35%	76%
350	21.2	220	13.8	35%	85%
400	23.6	230	14	28%	61%
500	53	130	-10.5	83%	90%

S3. Catalytic activity of S500 against MO and MG

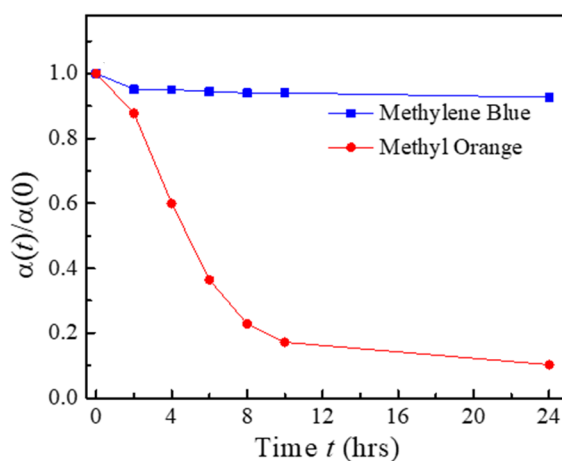


Figure S2. The normalized and time dependent optical absorption $\alpha(t)/\alpha(0)$ of the characteristic peaks of MO and MB after mixed with S500. The experiments were performed under dark.

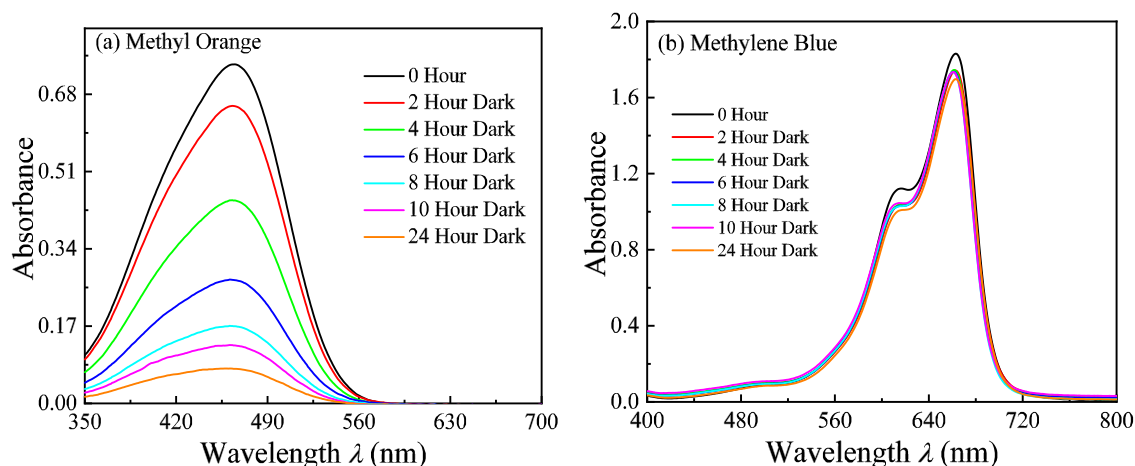


Figure S3. The time dependent UV-Vis spectra of (a) methyl orange and (b) methylene blue after mixed with S500 $\text{Cu}_x\text{Fe}_y\text{O}_z$ NPs in dark for 24 hours.

S4. Antimicrobial tests for different concentrations of S200 and S500

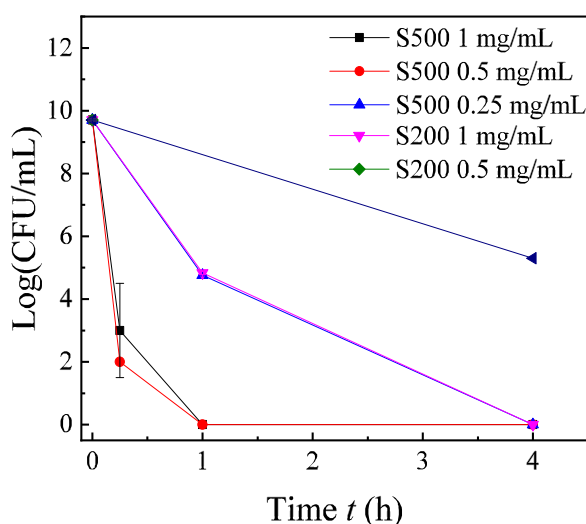


Figure S4. Concentration-dependent antimicrobial activity of S200 and S500 samples against *E. coli* B.

S5. Control samples and their properties

The control Fe_2O_3 and Cu_2O NPs were synthesized under the same microwave assisted hydrothermal condition with 500 μL 37% formaldehyde (Unfortunately we could not obtain CuFeO_2 NPs under similar conditions or other conditions). We kept the ion molar amount the same for the synthesis, i.e., for Fe_2O_3 NPs, 1 mmol $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ (0.404 g) was used; while for Cu_2O NPs, 1 mmol $\text{Cu}(\text{NO}_3)_2 \cdot 3\text{H}_2\text{O}$ (0.242 g) was used. The average sizes of the Fe_2O_3 and Cu_2O NPs were determined to be 632 and 832 nm by a dynamic light scattering measurements. The composition of the resulted NPs was determined by XRD, as shown in Fig. S5. The control Fe_2O_3 NPs are confirmed as pure Fe_2O_3 while the Cu_2O NPs contain small amount of CuO .

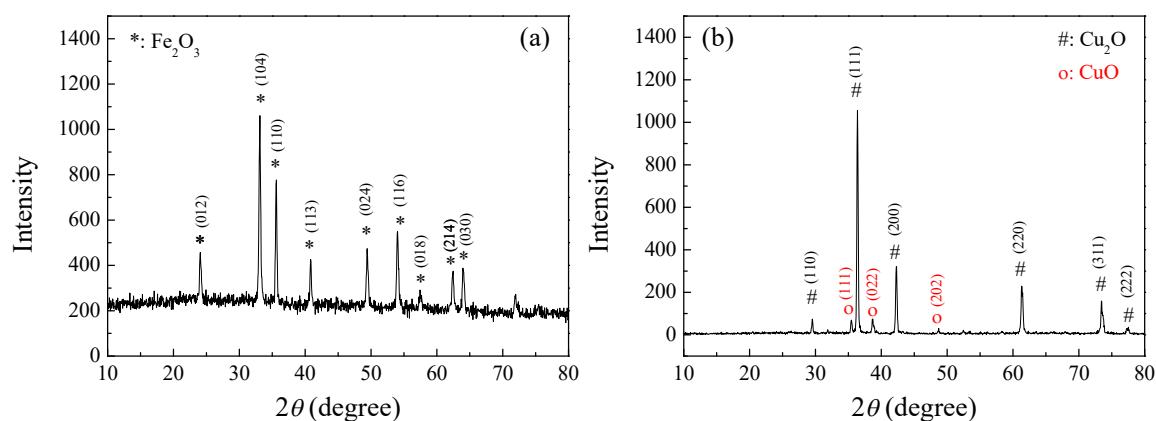


Figure S5. XRD patterns of the control (a) Fe_2O_3 and (b) Cu_2O NPs.

References

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