

Supporting Information

Electrospinning Synthesis of Carbon-Supported Pt₃Mn Intermetallic Nanocrystals and Electrocatalytic Performance towards Oxygen Reduction Reaction

Lechao Peng ¹, Lan Zhou ¹, Wenjun Kang ¹, Rui Li ¹, Konggang Qu ¹, Lei Wang ¹, and Haibo Li ^{1,*}

¹ Shandong Provincial Key Laboratory of Chemical Energy Storage and Novel Cell Technology, School of Chemistry and Chemical Engineering, Liaocheng University, Liaocheng 252059, PR China; lechaopeng@163.com (L.P.); lan_zhou_z@163.com (L.Z.); kangwenjun@lcu.edu.cn (W.K.); lirui@lcu.edu.cn (R.L.); qukonggang@lcu.edu.cn (K.Q.); wanglei@lcu.edu.cn (L.W.)

* Correspondence: haiboli@mail.ustc.edu.cn (H.L.)

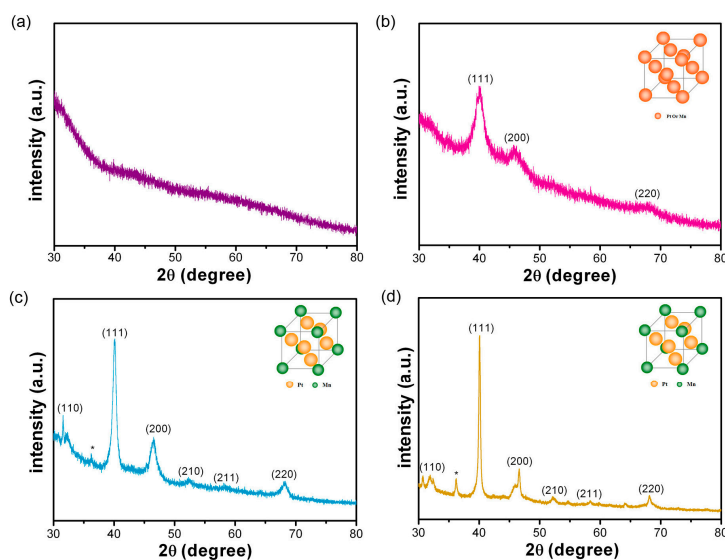


Figure S1. XRD patterns of (a) PtMn-650/CNFs, (b) PtMn-750/CNFs, (c) PtMn-850/CNFs, and (d) PtMn-1050/CNFs.

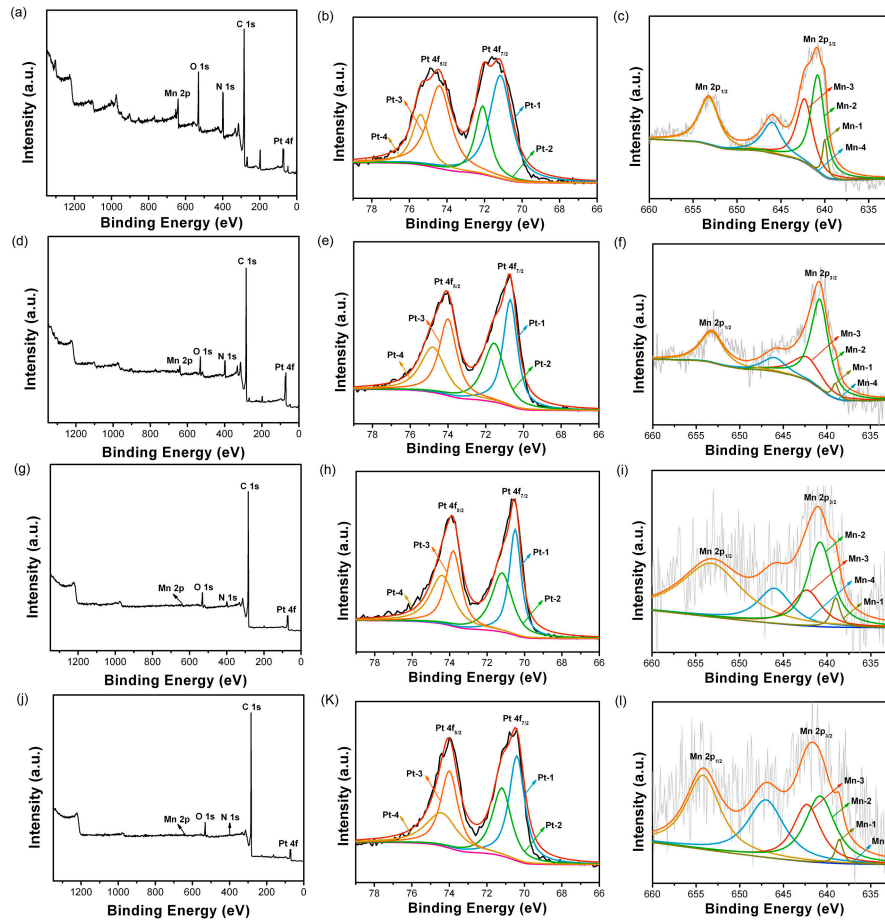


Figure S2. XPS survey spectra of (a) PtMn-650/CNFs, (d) PtMn-750/CNFs, (g) PtMn-850/CNFs, and (j) PtMn-1050/CNFs. Peak-fitting Pt 4f XPS spectra of (b) PtMn-650/CNFs, (e) PtMn-750/CNFs, (h) PtMn-850/CNFs, and (k) PtMn-1050/CNFs. Peak-fitting Mn 2p_{3/2} spectra of (c) PtMn-650/CNFs, (f) PtMn-750/CNFs, (i) PtMn-850/CNFs, and (l) PtMn-1050/CNFs.

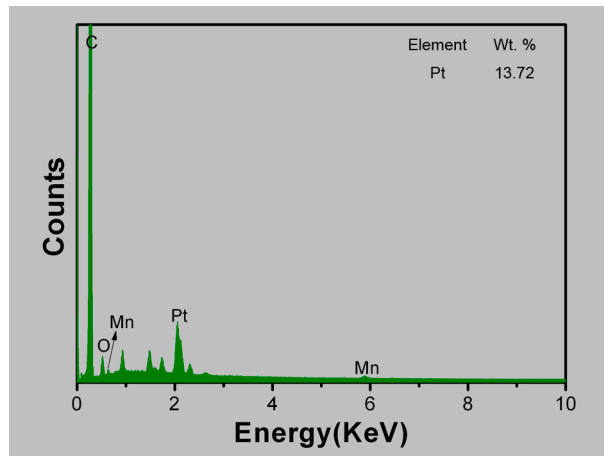


Figure S3. EDS of PtMn-950/CNFs.

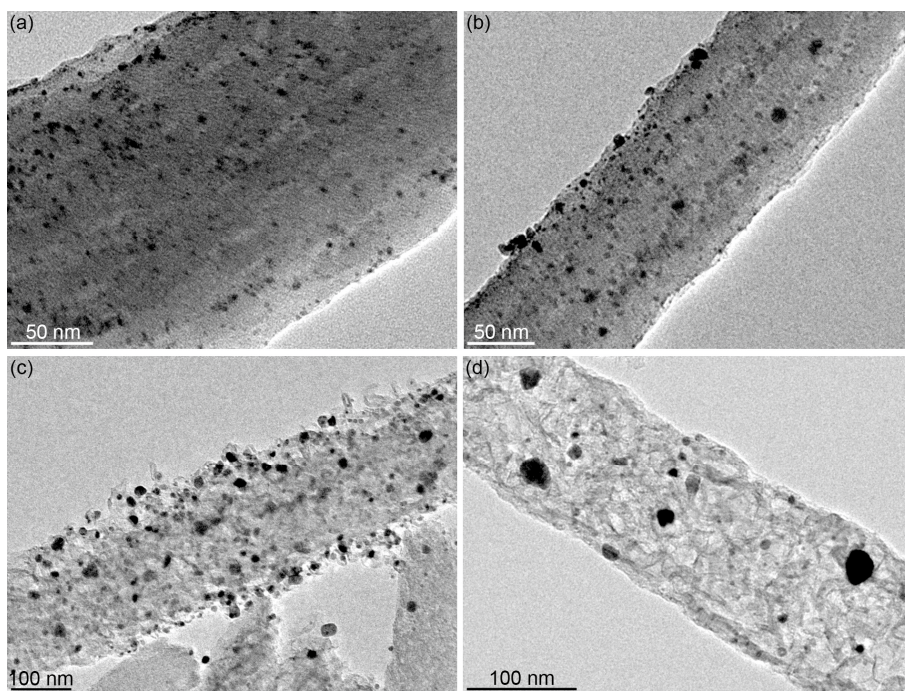


Figure S4. TEM images of (a) PtMn-650/CNFs, (b) PtMn-750/CNFs, (c) PtMn-850/CNFs, and (d) PtMn-1050/CNFs.

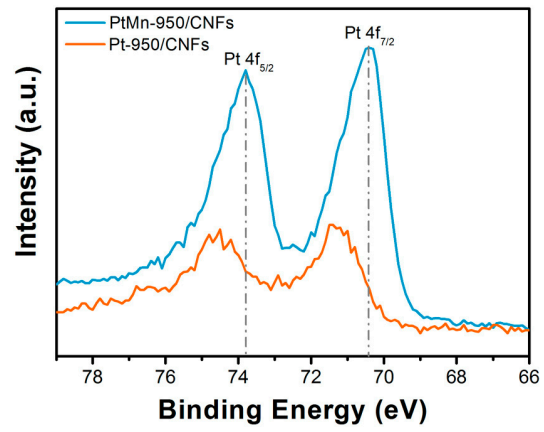


Figure S5. Pt 4f XPS spectra of PtMn-950/CNFs and Pt-950/CNFs.

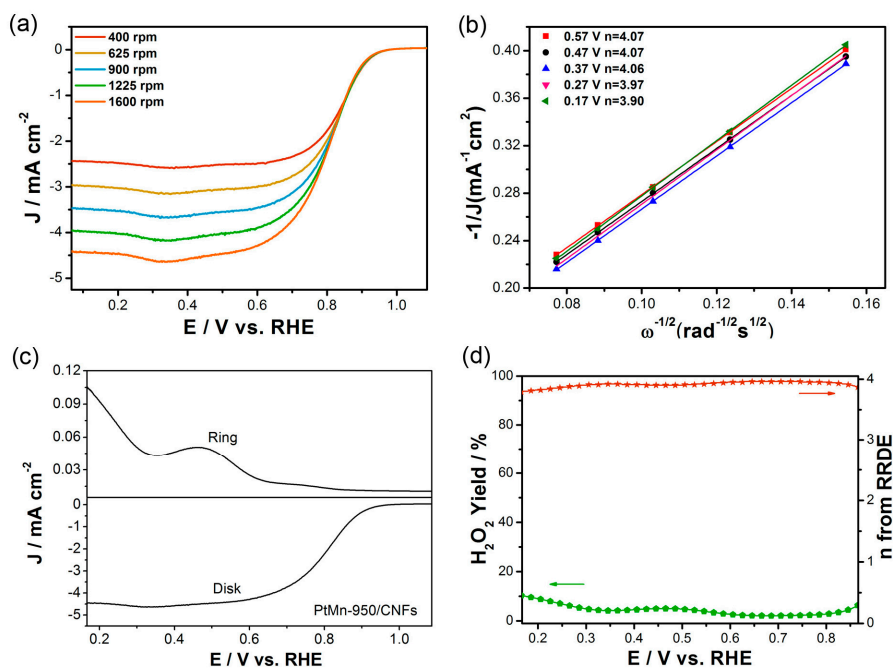


Figure S6. Electrochemical characterization of PtMn-950/CNFs after ADTs (5000 cycles): (a) ORR polarization curves at different rotation rates in O₂-saturated 0.10 M KOH; (b) Koutecky–Levich plots at different potentials; (c) ORR polarization curves recorded on the RRDE in O₂-saturated 0.10 M KOH solution at 1600 rpm; (d) the calculated H₂O₂ yield and electron transfer numbers (n).

Table S1. The calculation of the kinetic current densities at 0.832 V (vs. RHE)

Sample	PtMn-650/CNFs	PtMn-750/CNFs	PtMn-850/CNFs	PtMn-950/CNFs	PtMn-1050/CNFs
$J / \text{mA cm}^{-2}$	0.063	0.485	0.675	1.925	0.718
$1/J / \text{mA}^{-1} \text{cm}^2$	15.80	2.06	1.48	0.52	1.39
$J_L / \text{mA cm}^{-2}$	2.19	2.65	2.64	4.31	3.20
$1/J_L / \text{mA}^{-1} \text{cm}^2$	0.458	0.377	0.378	0.232	0.313
$J_K^* / \text{mA cm}^{-2}$	0.07	0.59	0.91	3.48	0.93

$$*J_K = 1/(1/J - 1/J_L)$$