

Supplementary Materials

Construction and Mechanism Analysis of a Self-Assembled Conductive Network in DGEBA/PEI/HRGO Nanocomposites by Controlling Filler Selective Localization

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Table S1. Formulation of samples.

Sample	DGEBA (phr) ¹	PEI (phr) ¹	HRGO (phr) ¹	C-A solution
DGEBA	100	0	0	80
DP0H	100	0	(100 + 0) × 0.5%	80
DP5H	100	5	(100 + 5) × 0.5%	80
DP10H	100	10	(100 + 10) × 0.5%	80
DP15H	100	15	(100 + 15) × 0.5%	80
DP20H	100	20	(100 + 20) × 0.5%	80
DP25H	100	25	(100 + 25) × 0.5%	80
DP30H	100	30	(100 + 30) × 0.5%	80

¹phr: parts per hundred in the DGEBA.

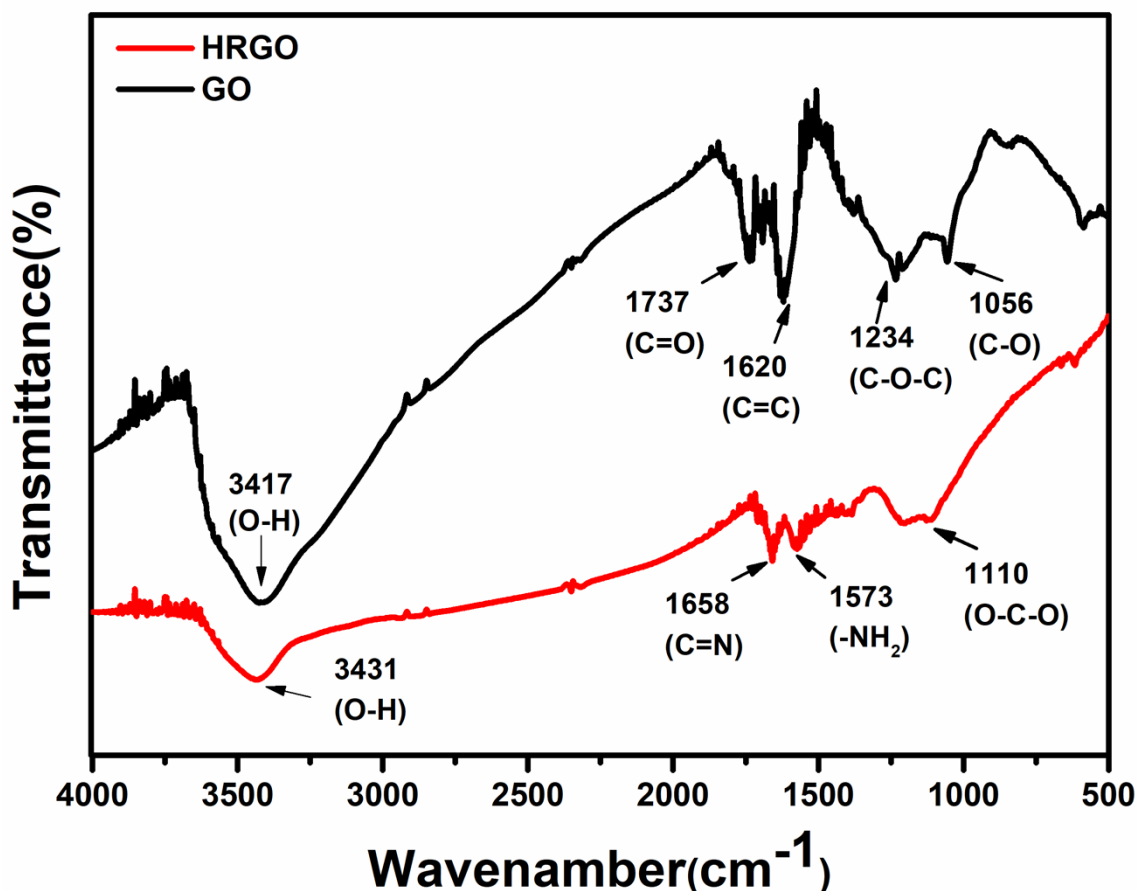


Figure S1. The FTIR spectra for HRGO and GO.

The reduction of GO was determined using FTIR technique. The FTIR spectra of HRGO and GO are represented

in Figure S1. The broad peak from ~ 3000 to ~ 3600 cm^{-1} was associated to the hydroxyl group (from $-\text{COOH}$ and H_2O), which reduces in case of HRGO [S3]. The peaks at ~ 1737 , ~ 1620 , ~ 1234 and 1056 cm^{-1} are associated with the stretching modes of ketone $\text{C}=\text{O}$, sp^2 hybridized carbon in plane $\text{C}=\text{C}$, epoxide $\text{C}-\text{O}-\text{C}$ and alkoxy $\text{C}-\text{O}$ [S4, S5]. However, in the case of HRGO, a significant decrement in peak intensities was observed and few peaks at ~ 1658 cm^{-1} and 1573 cm^{-1} were observed. The peaks of ~ 1658 cm^{-1} and 1573 cm^{-1} are formed due to hydrazine reduction and basically associated with $\text{C}=\text{N}$ and $-\text{NH}_2$ stretching[S6,S7].

Table S2. Surface tensions and components (mJ/m^2) suggested by van Oss et al.[S1,S2]

Liquid	γ_{L}	$\gamma_{\text{L}}^{\text{LW}}(\gamma_{\text{L}}^{\text{d}})$	$\gamma_{\text{L}}^{\text{AB}}(\gamma_{\text{L}}^{\text{P}})$	γ_{L}^+	γ_{L}^-	Polarity $[\gamma_{\text{L}}^{\text{P}}/\gamma_{\text{L}}^{\text{d}}]$
Deionized water (DI)	72.8	21.8	51	25.5	25.5	2.3
Glycerol (GL)	64.0	34.0	30	3.9	57.4	0.9
Formamide (FA)	58.0	39.0	0	2.28	39.6	0

Table S3. The static contact angles for the test liquids DGEBA, PEI, and HRGO.

Liquid	Contact angle(θ)		
	DGEBA	PEI	HRGO
Deionized water (DI)	88.4	84.2	80.5
Glycerol (GL)	84.3	75.0	55.0
Formamide (FA)	54.1	58.8	27.1

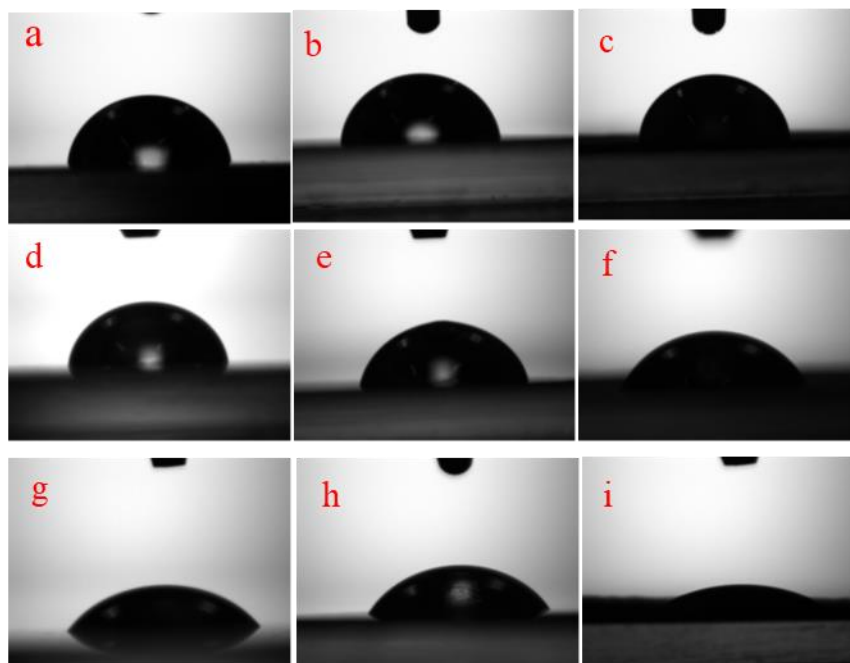


Figure S2. The contact angle images of samples given in Table S3. (a) DI on DGEBA, (b) DI on PEI, (c) DI on HRGO, (d) GL on DGEBA, (e) GL on PEI, (f) GL on HRGO, (g) FA on DGEBA, (h) FA on PEI, (i) FA on HRGO.

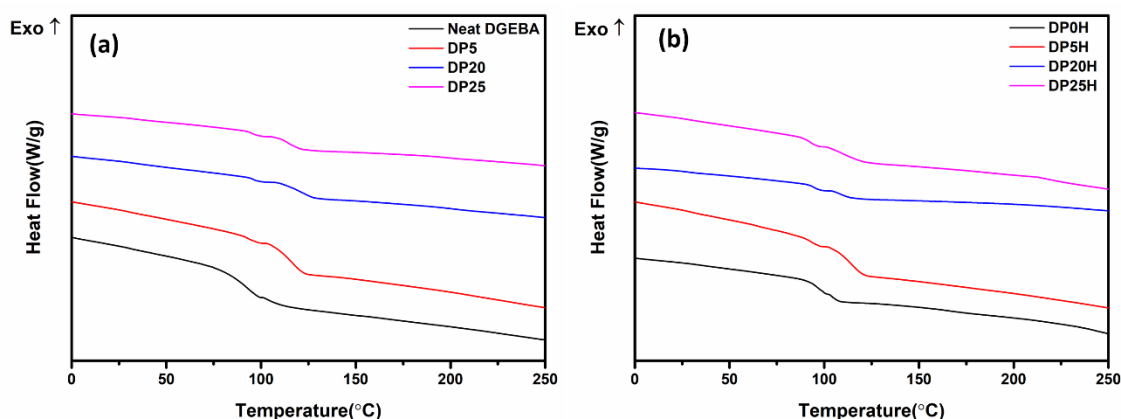


Figure S3. The differential scanning calorimeter (DSC) curves of (a) DGEBA/PEI polyblends and (b) DGEBA/PEI/HRGO polyblends.

References

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