

## SUPPLEMENTARY MATERIAL

### Producing magnetic nanocomposites from paper sludge for the adsorptive removal of pharmaceuticals from water - A fractional factorial design

Luciana S. Rocha<sup>1,\*</sup>, Érika M.L. Sousa<sup>1</sup>, María V. Gil<sup>2</sup>, João A.B.P. Oliveira<sup>1</sup>, Marta Otero<sup>3</sup>, Valdemar I. Esteves<sup>1</sup>, Vânia Calisto<sup>1</sup>

<sup>1</sup> *Department of Chemistry and CESAM, University of Aveiro, 3810-193 Aveiro, Portugal*

<sup>2</sup> *Instituto de Ciencia y Tecnología del Carbono, INCAR-CSIC, Francisco Pintado Fe 26, 33011 Oviedo, Spain*

<sup>3</sup> *Department of Environment and Planning and CESAM, University of Aveiro, 3810-193 Aveiro, Portugal*

**Table S1.** Physical and chemical properties of amoxicillin tri-hydrate (AMX), carbamazepine (CBZ) and sodium diclofenac (DCF).

**Table S2.** Values obtained for the specific surface area ( $S_{\text{BET}}$ ) and textural properties (total pore volume  $V_{\text{p}}$ , micropore volume  $V_{\text{mic}}$ , and average pore diameter  $D$ ) of PAC and the eighteen MACs.

**Table S3.** Results of ANOVA analysis: sum of square (SS) and mean of square (MS) values, degree of freedom (df), F-test and the  $p$ -value (confidence level of 95%).

**Table S4.** Values of  $A$  (%) obtained for AMX, CBZ and DCF using MACs prepared using different carbon precursors, along with the experimental conditions used in the adsorption experiments ( $C_i$  of each pharmaceutical, dose of MAC, pH, temperature and contact time).

**Figure S1.**  $N_2$  adsorption isotherms expressed as (A.) adsorption ( $\text{cm}^3 \text{ STP g}^{-1}$ ) vs relative pressure ( $p/p^0$ ) and (B.) adsorption ( $\text{cm}^3 \text{ STP g}^{-1}$ ) vs  $-\log(p/p^0)$  of the following materials: PAC (—), MAC 4 (—), MAC 7 (—), MAC 11 (—) and MAC 17 (—).

**Figure S2.** Pore size distribution of PAC, MAC 3, MAC6, MAC 13 and MAC 16.

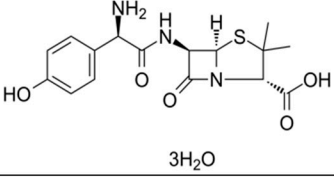
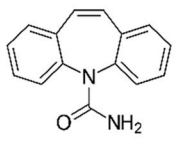
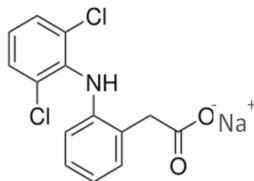
**Figure S3.** Values obtained for  $S_{\text{BET}}$  (A.) and  $M_s$  (B.) as a function of PAC:Fe salts ratio ( $w/w$ ).

**Figure S4.** SEM images of the MAC 7 using a magnification of 20000 (A.) and 40000 $\times$  (B.).

**Figure S5.** Overall XPS spectra of PAC (—), MAC 4 (—), MAC 7 (—), MAC 11 (—) and MAC 17 (—).

**Figure S6.** Deconvolution of C1s, O1s and Fe2p XPS peaks of MAC 7, MAC 11 and MAC 17: experimental peak (—), adjusted peak (—) and the component groups (—).

**Table S1.** Physical and chemical properties of amoxicillin tri-hydrate (AMX) <sup>1</sup>, carbamazepine (CBZ) <sup>2</sup> and sodium diclofenac (DCF) <sup>3</sup>.

Pharmaceutical		
Classe	Molecular structure	Properties
Antibiotic	 <p>3H<sub>2</sub>O</p>	<b>Amoxicillin tri-hydrate:</b> Mw: 419.5 g mol <sup>-1</sup> pK <sub>a1</sub> : 3.23, pK <sub>a2</sub> : 7.43 log K <sub>ow</sub> : 0.87
Antiepileptic		<b>Carbamazepine:</b> Mw: 236.3 g mol <sup>-1</sup> pK <sub>a</sub> : 13.9 log K <sub>ow</sub> : 2.45
Non-steroidal anti-inflammatory drug		<b>Sodium diclofenac:</b> Mw: 318.1 g mol <sup>-1</sup> pK <sub>a</sub> : 4.15 log K <sub>ow</sub> : 4.51

<sup>1</sup> <https://pubchem.ncbi.nlm.nih.gov/compound/Amoxicillin-trihydrate> (accessed in 15.07.2020)

<sup>2</sup> <https://pubchem.ncbi.nlm.nih.gov/compound/Carbamazepine> (accessed in 15.07.2020)

<sup>3</sup> <https://pubchem.ncbi.nlm.nih.gov/compound/Diclofenac-sodium> (accessed in 15.07.2020)

**Table S2.** Values obtained for the specific surface area ( $S_{\text{BET}}$ ) and textural properties (total pore volume  $V_{\text{p}}$ , micropore volume  $V_{\text{mic}}$ , and average pore diameter  $D$ ) of PAC and the eighteen MACs.

Material	N <sub>2</sub> adsorption at -196 °C			
	$S_{\text{BET}}$ (m <sup>2</sup> g <sup>-1</sup> )	$V_{\text{p}}$ (cm <sup>3</sup> g <sup>-1</sup> )	$V_{\text{mic}}$ (cm <sup>3</sup> g <sup>-1</sup> )	$D$ (nm)
PAC	1438	1.00	0.57	1.39
MAC1	782	0.68	0.31	1.74
MAC2	658	0.56	0.26	1.69
MAC3	532	0.46	0.21	1.73
MAC4	725	0.63	0.29	1.73
MAC5	653	0.56	0.26	1.70
MAC6	475	0.48	0.19	2.03
MAC7	794	0.74	0.32	1.87
MAC8	609	0.60	0.24	1.97
MAC9	672	0.62	0.27	1.86
MAC10	624	0.58	0.25	1.87
MAC11	741	0.76	0.30	2.04
MAC12	552	0.59	0.22	2.15
MAC13	828	0.86	0.33	2.08
MAC14	645	0.68	0.26	2.10
MAC15	538	0.57	0.21	2.13
MAC16	899	0.82	0.36	1.83
MAC17	767	0.72	0.31	1.87
MAC18	641	0.68	0.26	2.13

**Table S3.** Results of ANOVA analysis: sum of square (SS) and mean of square (MS) values, degree of freedom (df), F-test and the *p*-value (confidence level of 95%).

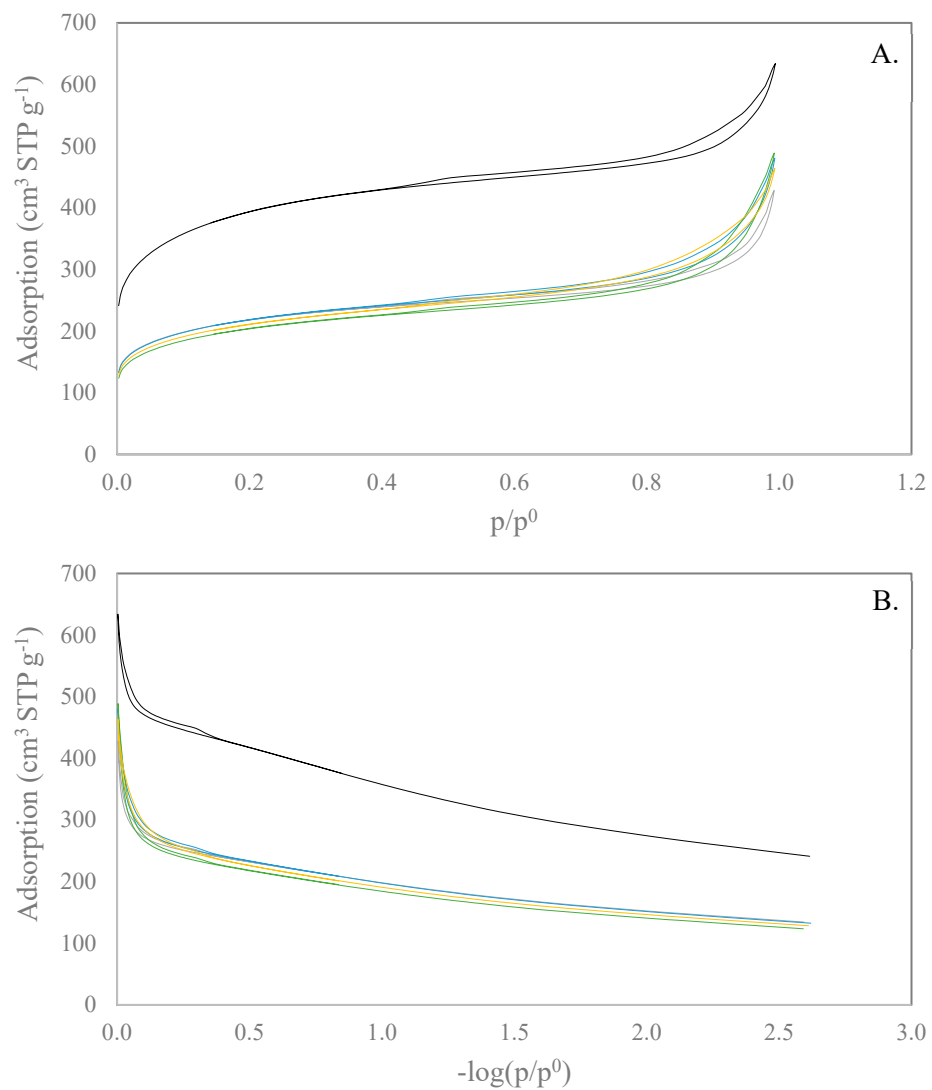
	Source	SS	df	MS	F	<i>p</i> -values*
$S_{\text{BET}}$ ( $\text{m}^2 \text{g}^{-1}$ )	$\chi_1$	$2.47 \times 10^4$	2	$1.24 \times 10^4$	2.29	0.152
	$\chi_2$	$1.43 \times 10^5$	2	$7.17 \times 10^4$	13.3	<b>0.002</b>
	$\chi_3$	$4.23 \times 10^3$	2	$2.11 \times 10^3$	0.39	0.686
	$\chi_4$	$3.89 \times 10^3$	1	$3.89 \times 10^3$	0.72	0.416
$M_s$ ( $\text{emu g}^{-1}$ )	$\chi_1$	$2.09 \times 10^1$	2	$1.04 \times 10^1$	0.17	0.846
	$\chi_2$	$1.26 \times 10^3$	2	$6.32 \times 10^2$	10.3	<b>0.004</b>
	$\chi_3$	$2.02 \times 10^2$	2	$1.01 \times 10^2$	1.64	0.241
	$\chi_4$	$7.31 \times 10^1$	1	$7.31 \times 10^1$	1.19	0.301
$A_{\text{AMX}}$ (%)	$\chi_1$	$3.30 \times 10^2$	2	$1.65 \times 10^2$	1.79	0.217
	$\chi_2$	$3.74 \times 10^2$	2	$1.87 \times 10^2$	2.03	0.182
	$\chi_3$	$1.11 \times 10^1$	2	5.56	0.06	0.942
	$\chi_4$	$6.50 \times 10^1$	1	$6.50 \times 10^1$	0.70	0.421
$A_{\text{CBZ}}$ (%)	$\chi_1$	$2.39 \times 10^2$	2	$1.19 \times 10^2$	1.2	0.342
	$\chi_2$	$1.92 \times 10^3$	2	$9.58 \times 10^2$	9.59	<b>0.005</b>
	$\chi_3$	$1.09 \times 10^1$	2	5.44	0.05	0.947
	$\chi_4$	$1.20 \times 10^2$	1	$1.20 \times 10^2$	1.2	0.299
$A_{\text{DCF}}$ (%)	$\chi_1$	$2.36 \times 10^1$	2	$1.18 \times 10^1$	0.29	0.757
	$\chi_2$	$7.86 \times 10^2$	2	$3.93 \times 10^2$	9.51	<b>0.005</b>
	$\chi_3$	$1.94 \times 10^2$	2	$9.72 \times 10^1$	2.35	0.146
	$\chi_4$	3.74	1	3.47	0.08	0.778

\* The bold *p*-values indicate the significant effect of factor on the response.

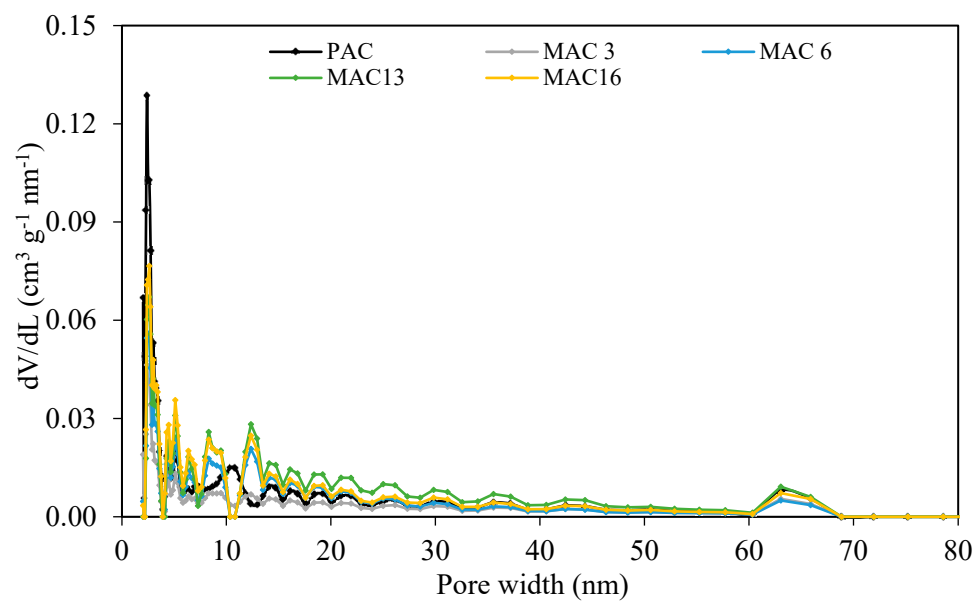
**Table S4.** Values of  $A$  (%) obtained for AMX, CBZ and DCF using MACs prepared using different carbon precursors, along with the experimental conditions used in the adsorption experiments ( $C_i$  of each pharmaceutical, dose of MAC, pH, temperature and contact time).

Carbon Precursor	Adsorption experiments	$A$ (%)	Reference
<b>AMX</b>			
Commercial PAC	$C_{AMX}=50 \text{ mg L}^{-1}$ ; Dose <sub>MAC</sub> $=1000 \text{ mg L}^{-1}$ pH=5; T=20 °C Contact time: 1.5 h	95	[49]
Waste-based PAC	$C_{AMX}=5 \text{ mg L}^{-1}$ ; Dose <sub>MAC</sub> $=35 \text{ mg L}^{-1}$ ; pH=6 <sup>(a)</sup> ; T=25 °C Contact time: 4 h	61-70	Present study (MAC 4, 7, 11, 17)
<b>CBZ</b>			
Commercial PAC	$C_{CBZ}=5.88 \text{ mg L}^{-1}$ ; Dos <sub>MAC</sub> $=248.5 \text{ mg L}^{-1}$ ; pH=6.6 <sup>(b)</sup> ; T=25 °C Contact time: 0.136 h	93	[25]
Commercial granular AC	$C_{CBZ}=30 \text{ mg L}^{-1}$ ; Dose <sub>MAC</sub> $=200 \text{ mg L}^{-1}$ pH=6 Contact time: 48 h	~80	[36]
Waste-based PAC	$C_{CBZ}=5 \text{ mg L}^{-1}$ ; Dose <sub>MAC</sub> $=35 \text{ mg L}^{-1}$ pH=6 <sup>(a)</sup> ; T=25 °C Contact time: 4 h	69-77	Present study (MAC 4, 7, 11, 17)
<b>DCF</b>			
Waste-based PAC	$C_{DCF}=5 \text{ mg L}^{-1}$ ; Dose <sub>MAC</sub> $=35 \text{ mg L}^{-1}$ pH=6 <sup>(a)</sup> ; T=25 °C Contact time: 4 h	80-84	Present study (MAC 4, 7, 11, 17)

<sup>(a)</sup> Unadjusted value. <sup>(b)</sup> The adsorption experiments were performed in wastewater treatment plant influent.

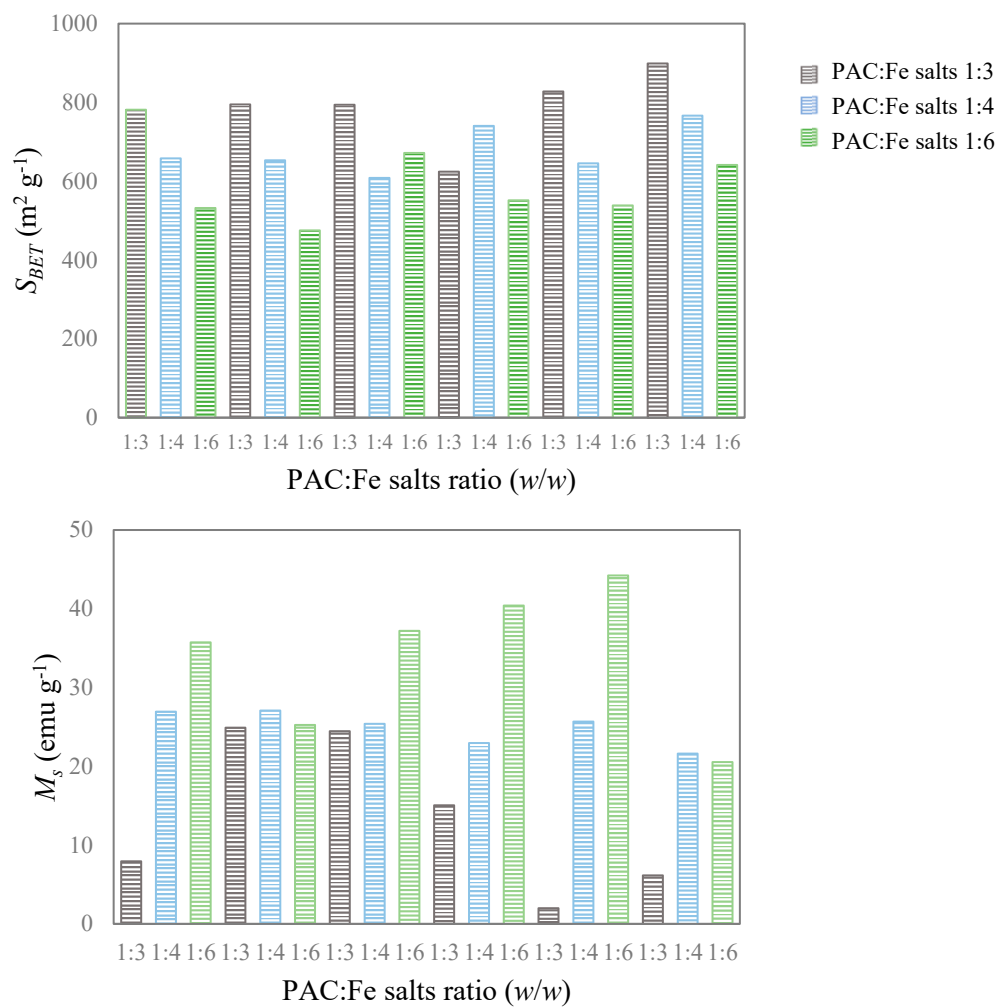


**Figure S1.**  $N_2$  adsorption isotherms expressed as (A) adsorption ( $\text{cm}^3 \text{ STP g}^{-1}$ ) vs relative pressure ( $p/p^0$ ) and (B) adsorption ( $\text{cm}^3 \text{ STP g}^{-1}$ ) vs  $-\log(p/p^0)$  of the following materials: PAC (—), MAC 4 (—), MAC 7 (—), MAC 11 (—) and MAC 17 (—).

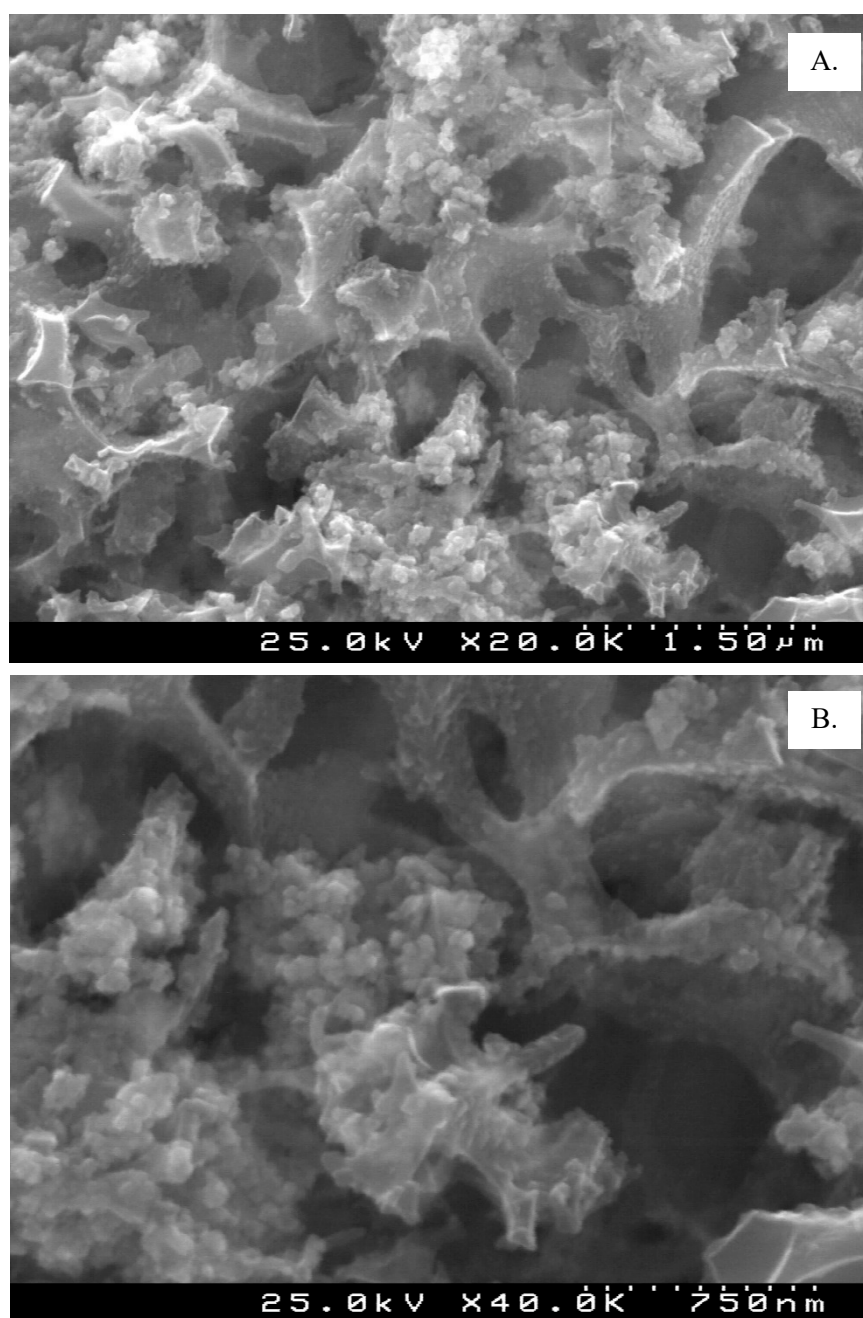


**Figure S2.** Pore size distribution of PAC, MAC 3, MAC6, MAC 13 and MAC 16.

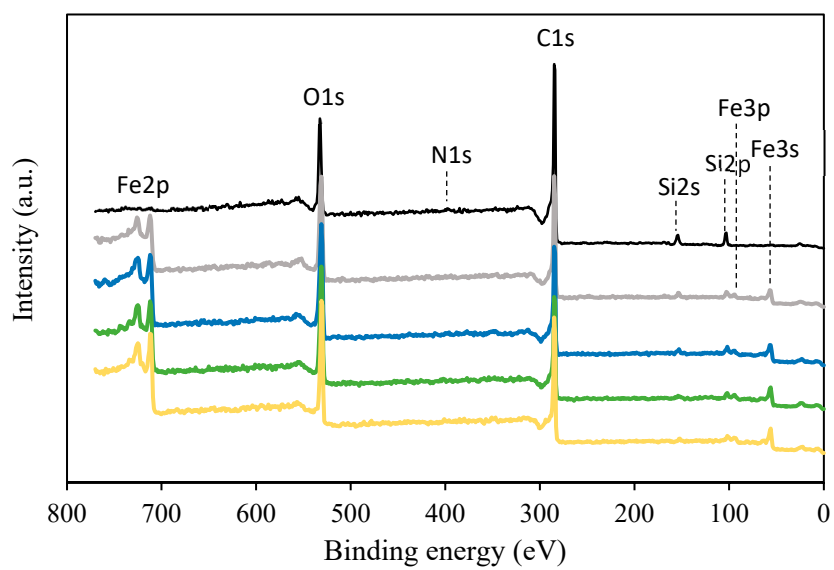




**Figure S3.** Values obtained for  $S_{BET}$  (A.) and  $M_s$  (B.) as a function of PAC:Fe salts ratio (w/w).

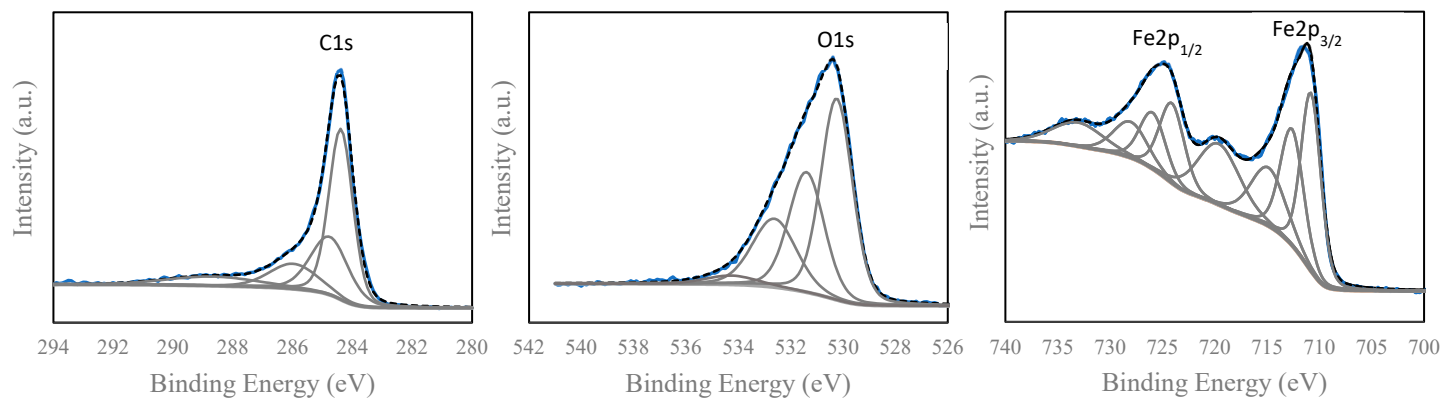


**Figure S4.** SEM images of the MAC 7 using a magnification of 20000 (A.) and 40000× (B.).

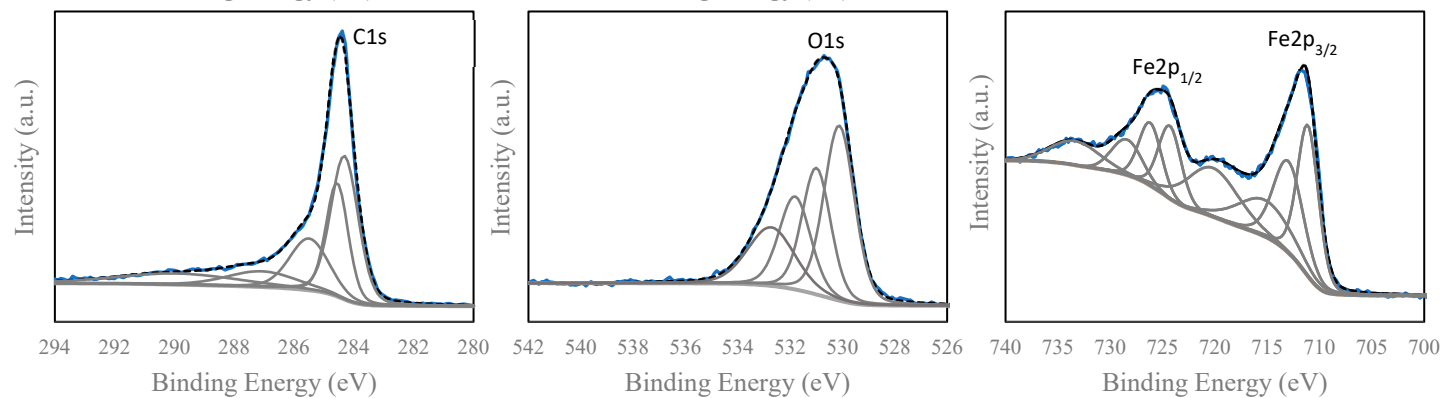


**Figure S5.** Overall XPS spectra of PAC (—), MAC 4 (—), MAC 7 (—), MAC 11 (—) and MAC 17 (—).

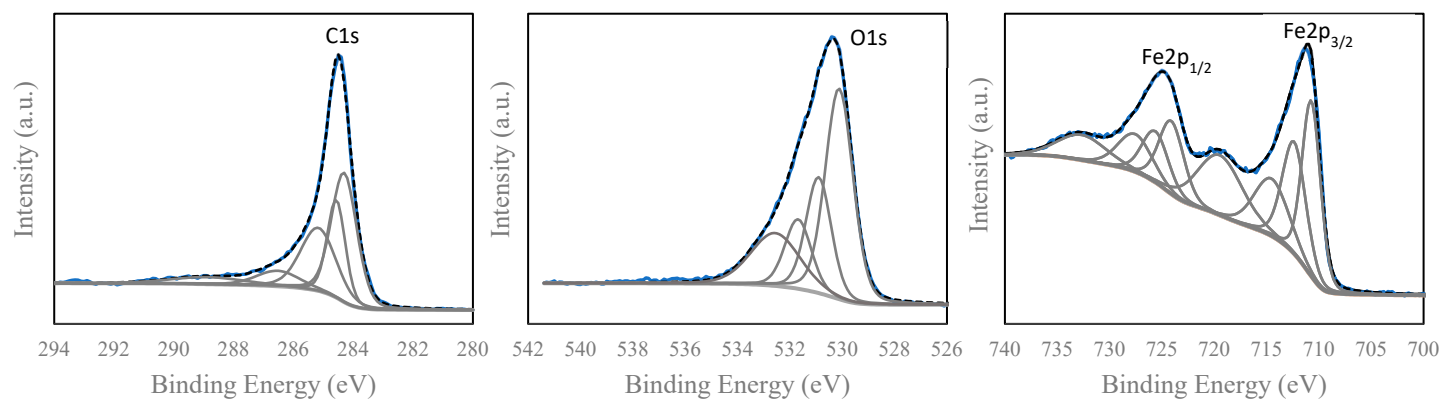
MAC 7



MAC 11



MAC 17



**Figure S6.** Deconvolution of C1s, O1s and Fe2p XPS peaks of MAC 7, MAC 11 and MAC 17: experimental peak (—), adjusted peak (—) and the component groups (—).