

Article

# Mechanochemically Synthesized PAN-Based Co-N-Doped Carbon Materials as Electrocatalyst for Oxygen Evolution Reaction

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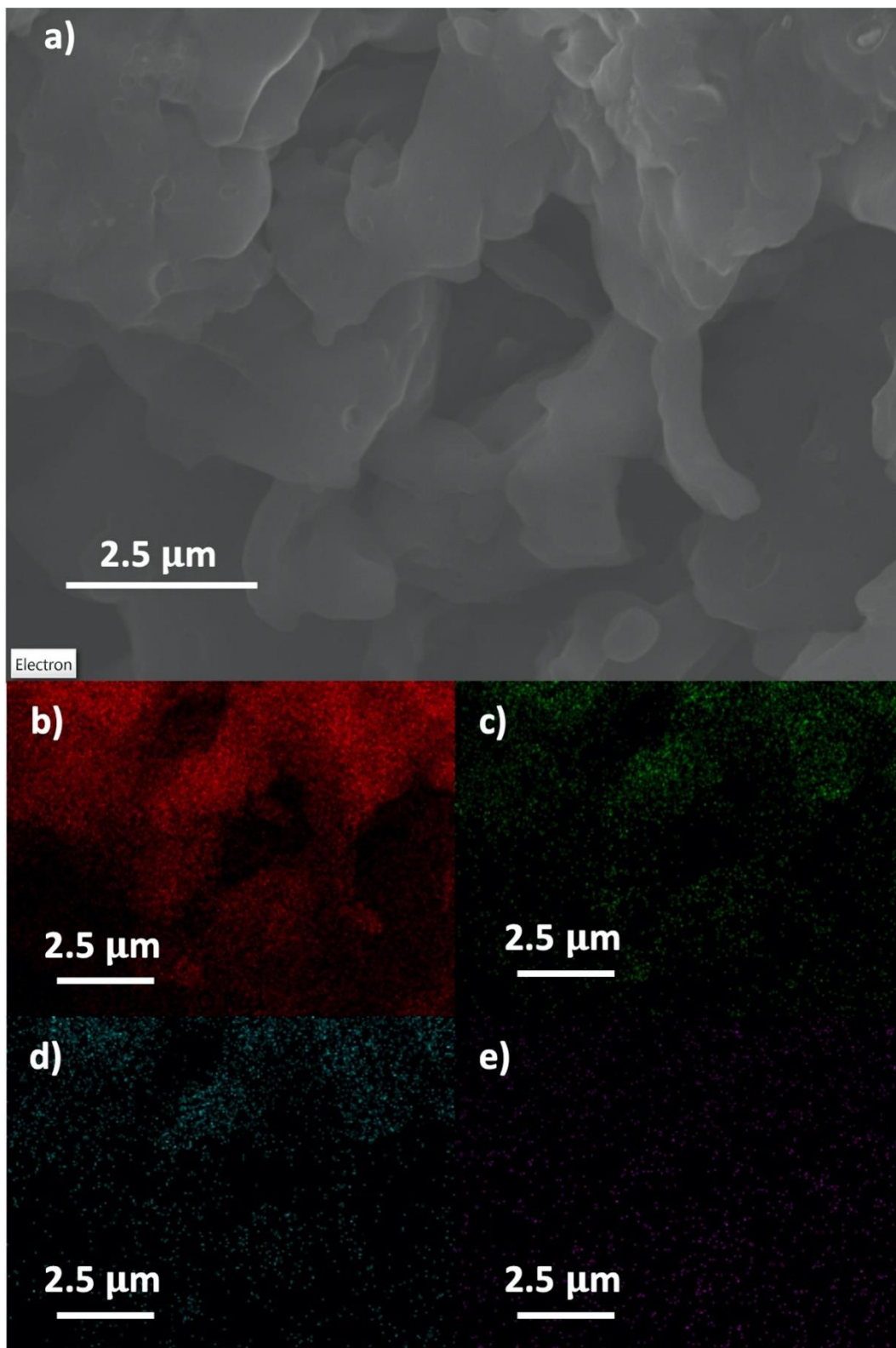
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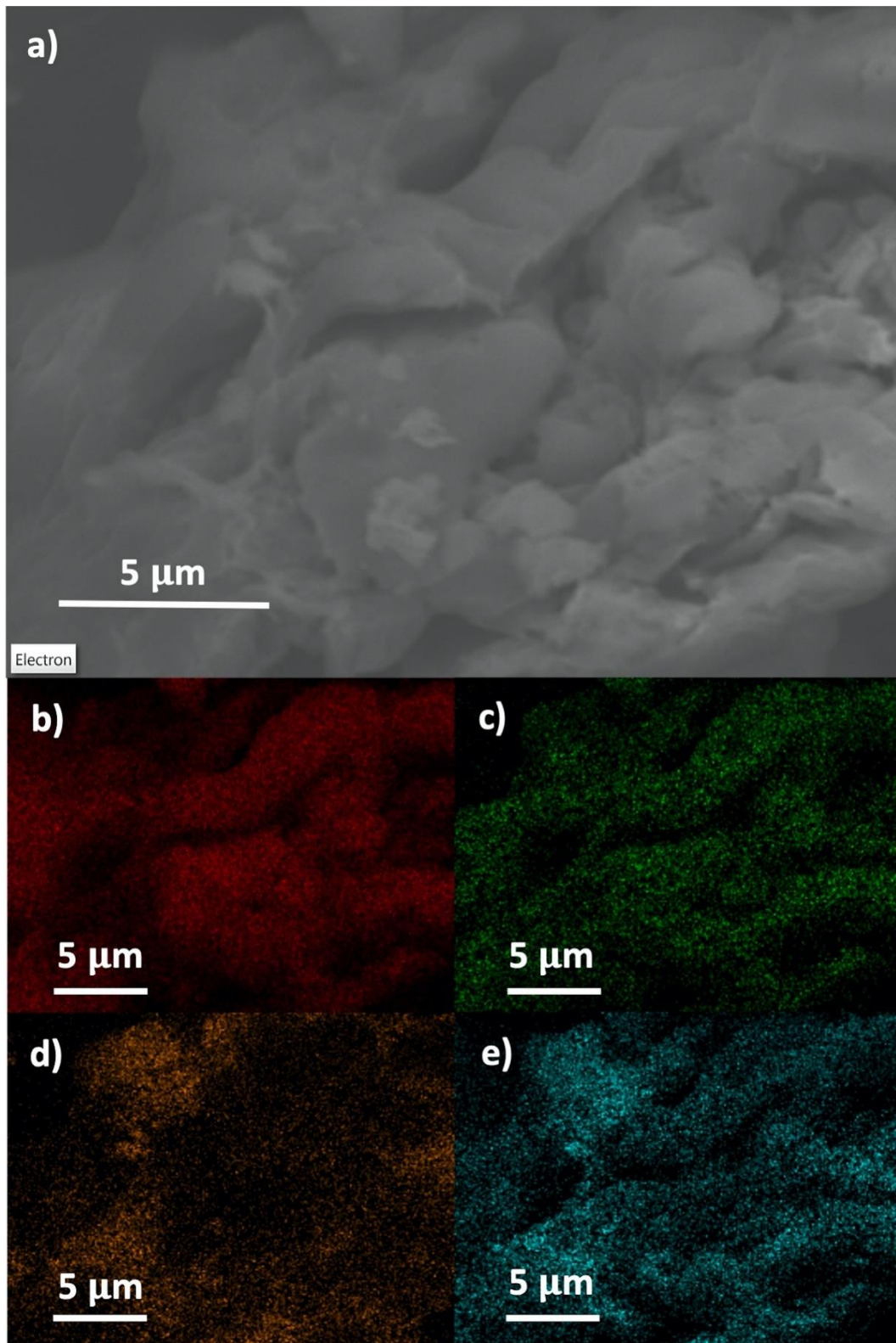
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**Abstract:** We report a new class of polyacrylonitrile (PAN)-based Co-N-doped carbon materials that can act as suitable catalyst for oxygen evolution reactions (OER). Different Co loadings were mechanochemically added into post-consumed PAN fibers. Subsequently, the samples were treated at 300 °C under air (PAN-A) or nitrogen (PAN-N) atmosphere to promote simultaneously the Co<sub>3</sub>O<sub>4</sub> species and PAN cyclization. The resulting electrocatalysts were fully characterized and analyzed by X-ray diffraction (XRD) and photoelectron spectroscopy (XPS), transmission (TEM) and scanning electron (SEM) microscopies, as well as nitrogen porosimetry. The catalytic performance of the Co-N-doped carbon nanomaterials were tested for OER in alkaline environments. Cobalt-doped PAN-A samples showed worse OER electrocatalytic performance than their homologous PAN-N ones. The PAN-N/3% Co catalyst exhibited the lowest OER overpotential (460 mV) among all the Co-N-doped carbon nanocomposites, reaching 10 mA/cm<sup>2</sup>. This work provides in-depth insights on the electrocatalytic performance of metal-doped carbon nanomaterials for OER.

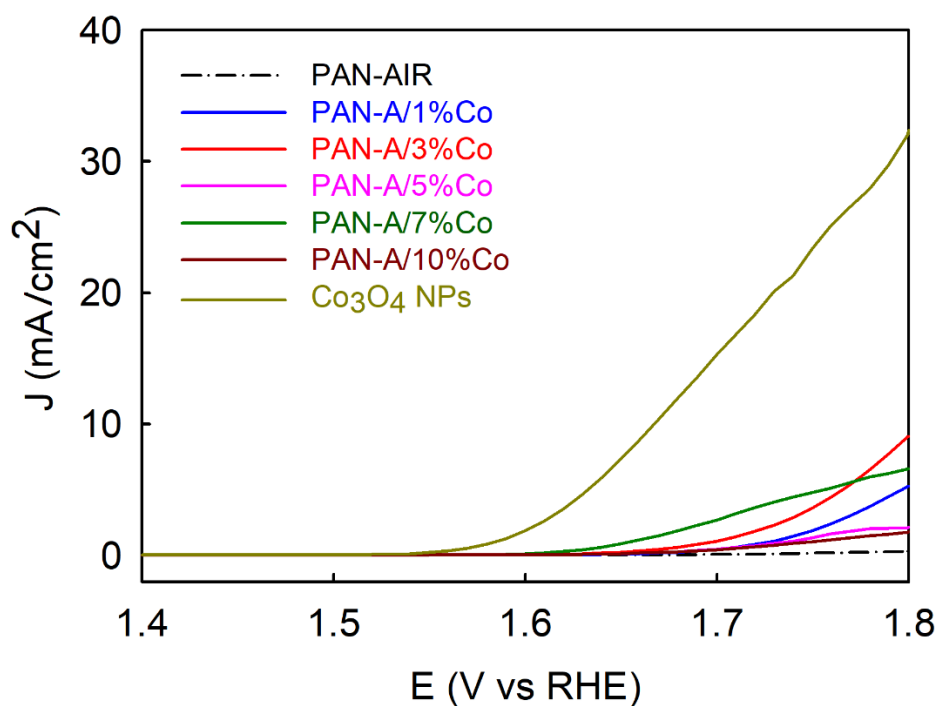
**Keywords:** mechanochemical synthesis; carbon N-doped; Co<sub>2</sub>O<sub>3</sub> nanoparticles; PAN; OER



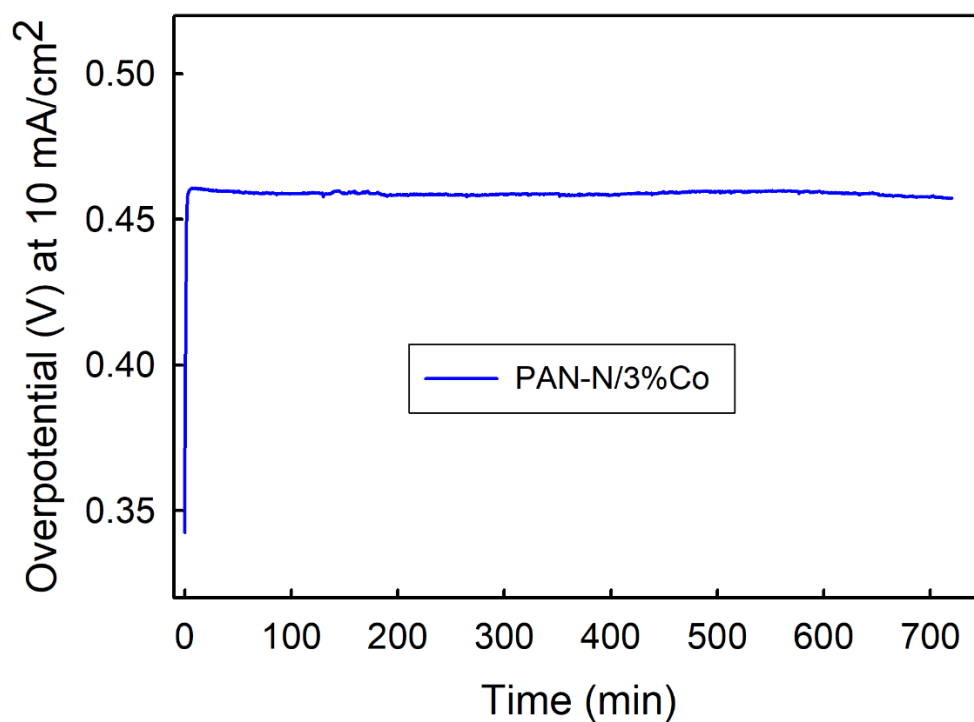
**Figure S1.** (a) SEM-EDX for PAN-N (0% Co) and their elemental mapping of (b) carbon, (c) nitrogen, (d) oxygen and (e) Iron, no detection, we can discard the impurity by iron ball milling.



**Figure S2.** (a) SEM-EDX for PAN-AIR/3%Co and their elemental mapping of (b) carbon, (c) nitrogen, (d) cobalt and (e) oxygen.



**Figure S3.** OER polarization curves obtained for the different PAN-AIR/Co samples, including the reference ( $\text{Co}_3\text{O}_4$  NPs) in 0.5 M KOH solution at 2 mV/s.



**Figure S4.** Chronopotentiometry measurements at  $J = 10 \text{ mA/cm}^2$  for the PAN-N/3%Co sample.

**Table S1.** A comparison on the OER electrocatalytic activity of our PAN-N/3%Co sample and other reported Co@Carbon-based OER catalysts.

Co@C Based Catalysts	$\eta_{10}$ (mV)	Tafel slope (mV/dec)	References
Co <sub>9</sub> S <sub>8</sub> /C nanosheets	434	122	[S1]
CoCMP	340	87	[S2]
Co@C600	277	46	[S3]
Co-B/C	320	75	[S4]
Co/CoO-C	340	84.5	[S5]
Co/CoO <sub>x</sub> @BS-NCNTs-700	360	112	[S6]
CoC <sub>x</sub> /FeCo@C/rGO	390	99.1	[S7]
MoS <sub>2</sub> /Co-N-CN <sub>2</sub>	410	74	[S8]
Co-Ni-P/MoS	300	71	[S9]
Co <sub>3</sub> S <sub>4</sub> @NiMoO <sub>4</sub>	320	102	[S10]
Co-Mo <sub>2</sub> C@NGCS-2	360	54	[S11]
ACTP <sub>2</sub> @Co,N-800	374	98	[S12]
Co <sub>2</sub> N <sub>0.67</sub> -BHPC	340	90	[S13]
Co/N-CNT (1500 SCCM)	310	66	[S14]
PBA-Pd-Co/C	500		[S15]
PAN-N/3%Co	460	60	This work

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