

## Supplementary Materials:

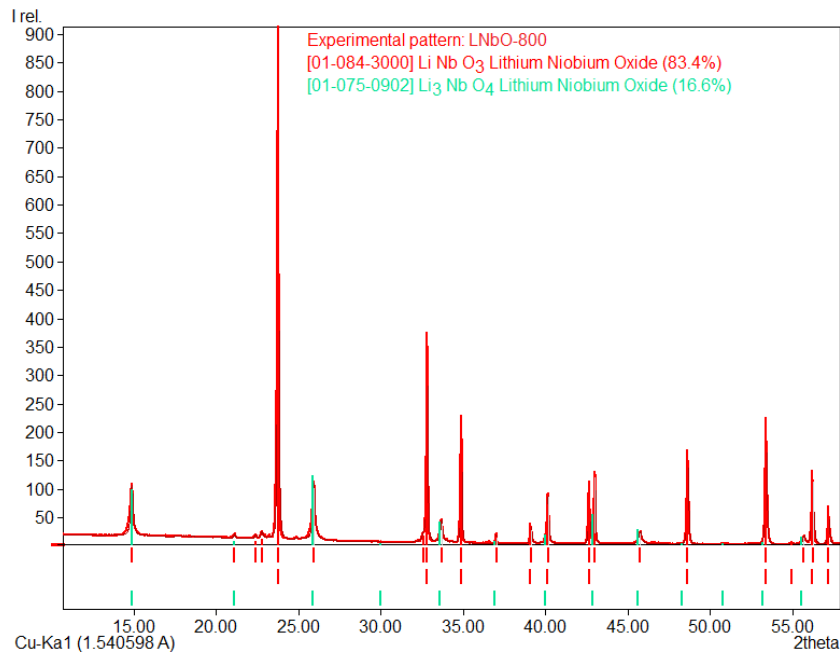
# Enhancing the Stability of $\text{LiNi}_{0.5}\text{Mn}_{1.5}\text{O}_4$ by Coating with $\text{LiNbO}_3$ Solid-State Electrolyte: Novel Chemically Activated Coating Process Versus Sol-Gel Method

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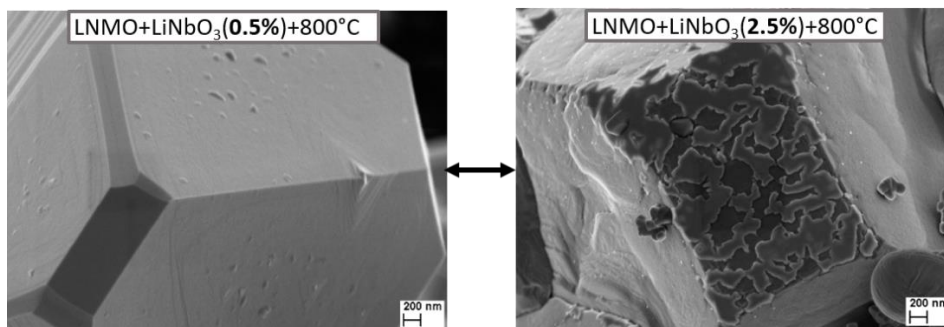
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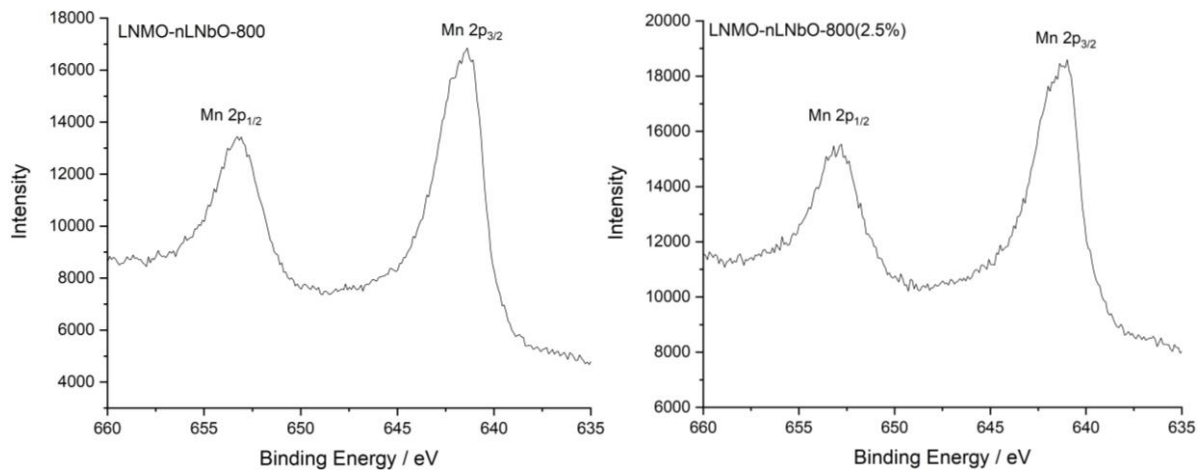
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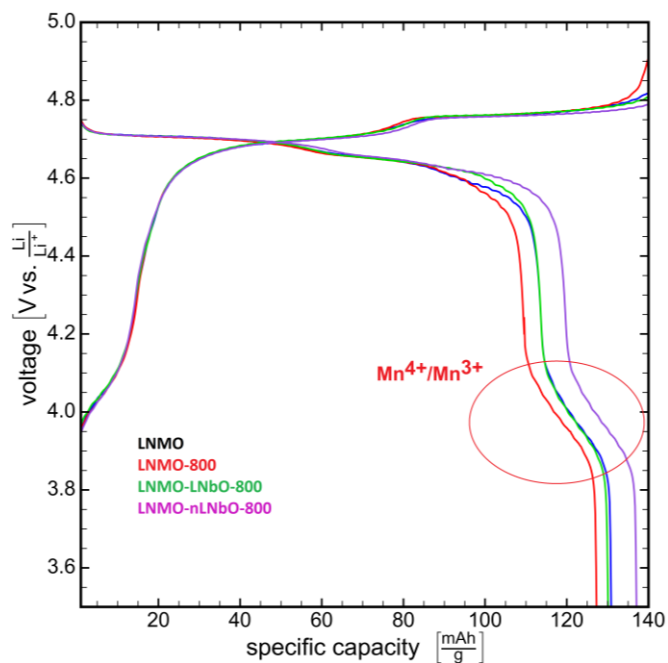
**Figure S1.** XRD patterns of  $\text{LiNbO}_3$  calcined at  $800^\circ\text{C}$  (LNbO-800).



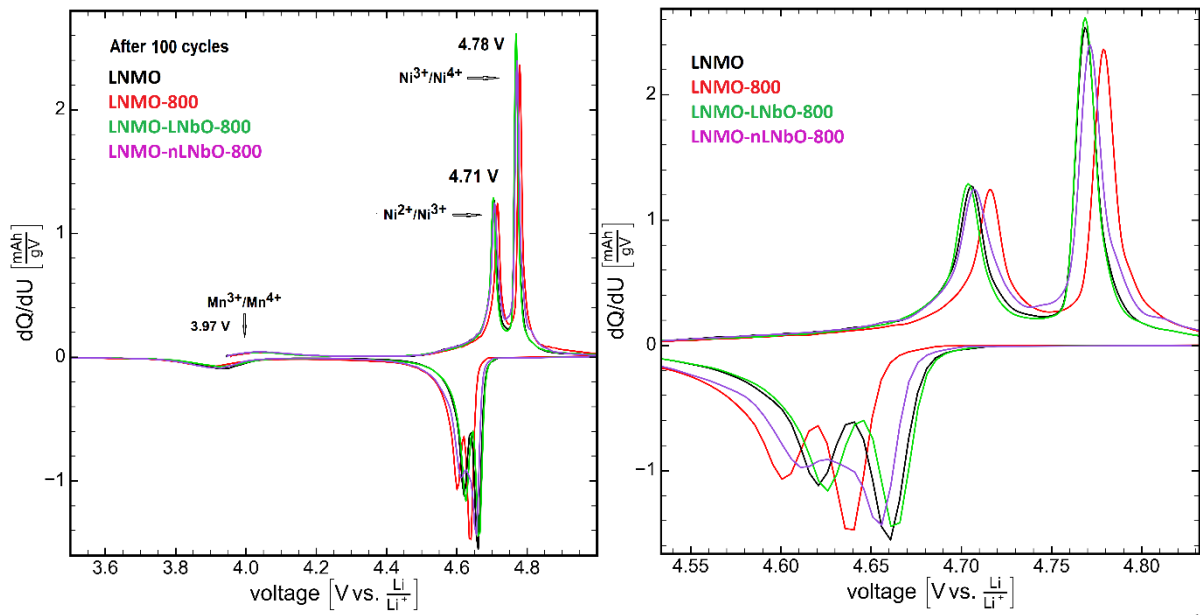
**Figure S2.** SEM diagrams of samples LMNO-nLNbO-800 with 0.5%  $\text{LiNbO}_3$  (left) and LMNO-nLNbO-800(2.5%) with 2.5%  $\text{LiNbO}_3$  (right).



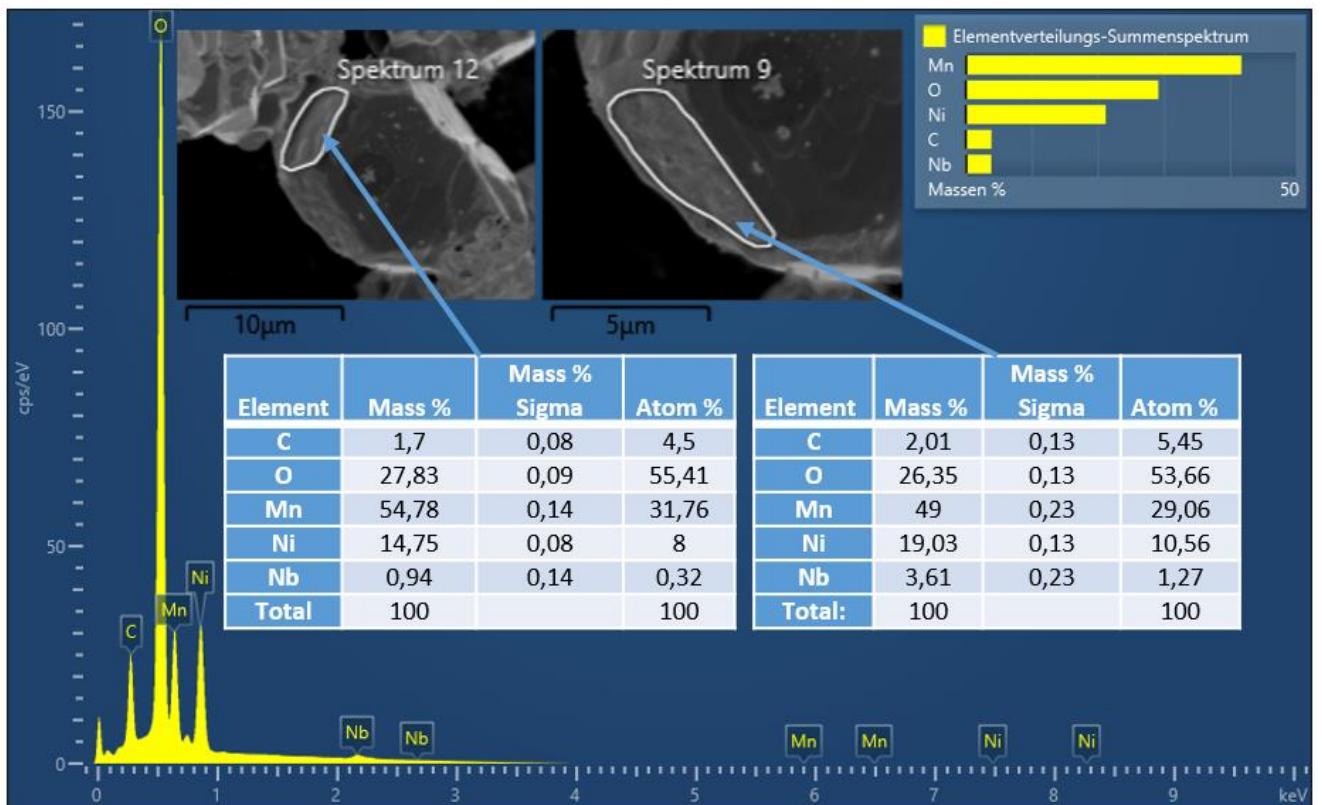
**Figure S3.** Mn  $2p_{1/2}$  and Mn  $2p_{3/2}$  XPS spectra of (left) LNMO-nLNbO-800 and (right) LNMO-nLNbO-800(2.5%).



**Figure S4.** Initial charge-discharge voltage profiles at 0.05 C of uncoated and LiNbO<sub>3</sub>-coated LNMO samples in the potential range of 3.5–5.0 V.



**Figure S5** (Left) Differential capacity versus potential ( $dQ/dV$  versus V) between 3.5 and 5 V (100<sup>th</sup> cycle). (Right) The zoomed 4.54 – 4.83 V region.



**Figure S6.** SEM diagrams with EDS results of sample LNMO-nLNbO-800(2.5%).