

Supplementary Materials

Preparation of Fe₃O₄-Ag Nanocomposites with Silver Petals for SERS Application

Thi Thuy Nguyen ^{1,2,*}, Fayna Mammeri ³, Souad Ammar ³, Thi Bich Ngoc Nguyen ², Trong Nghia Nguyen ², Thi Ha Lien Nghiem ², Nguyen Thi Thuy ⁴ and Thi Anh Ho ⁵

¹ Graduate University of Science and Technology, Vietnam Academy of Science and Technology, 18 Hoang Quoc Viet, Cau Giay, Hanoi 10000 , Vietnam

² Institute of Physics, Vietnam Academy of Science and Technology, 18 Hoang Quoc Viet, Cau Giay, Hanoi 10000, Vietnam; ntbnngoc@iop.vast.ac.vn (T.B.N.N.); trongnghia@iop.vast.ac.vn (T.N.N.); halien@iop.vast.ac.vn (T.H.L.N.)

³ ITODYS, Université de Paris, CNRS, UMR 7086, 15 rue J-A de Baïf, 75013 Paris, France; Fayna.mammeri@univ-paris-diderot.fr (F.M.) ; ammarmer@univ-paris-diderot.fr (S.A.)

⁴ Institute of Materials Science, Vietnam Academy of Science and Technology, 18 Hoang Quoc Viet, Cau Giay, Hanoi 10000, Vietnam; ntthuy@ims.vast.ac.vn

⁵ Faculty of engineering physics and nanotechnology, VNU University of Engineering and Technology, 144 Xuan Thuy, Cau Giay, Ha Noi 10000, Vietnam; anhht2508@gmail.com

* Correspondence: thuynt@iop.vast.ac.vn; Tel.: +84-2438347953

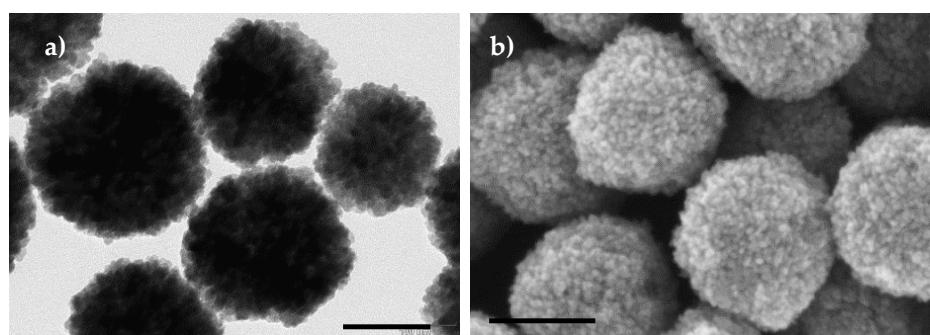


Figure S1. Representative TEM (a) and SEM (b) images of Fe₃O₄ particles. The scale bar is of 100 nm in all the micrographs.

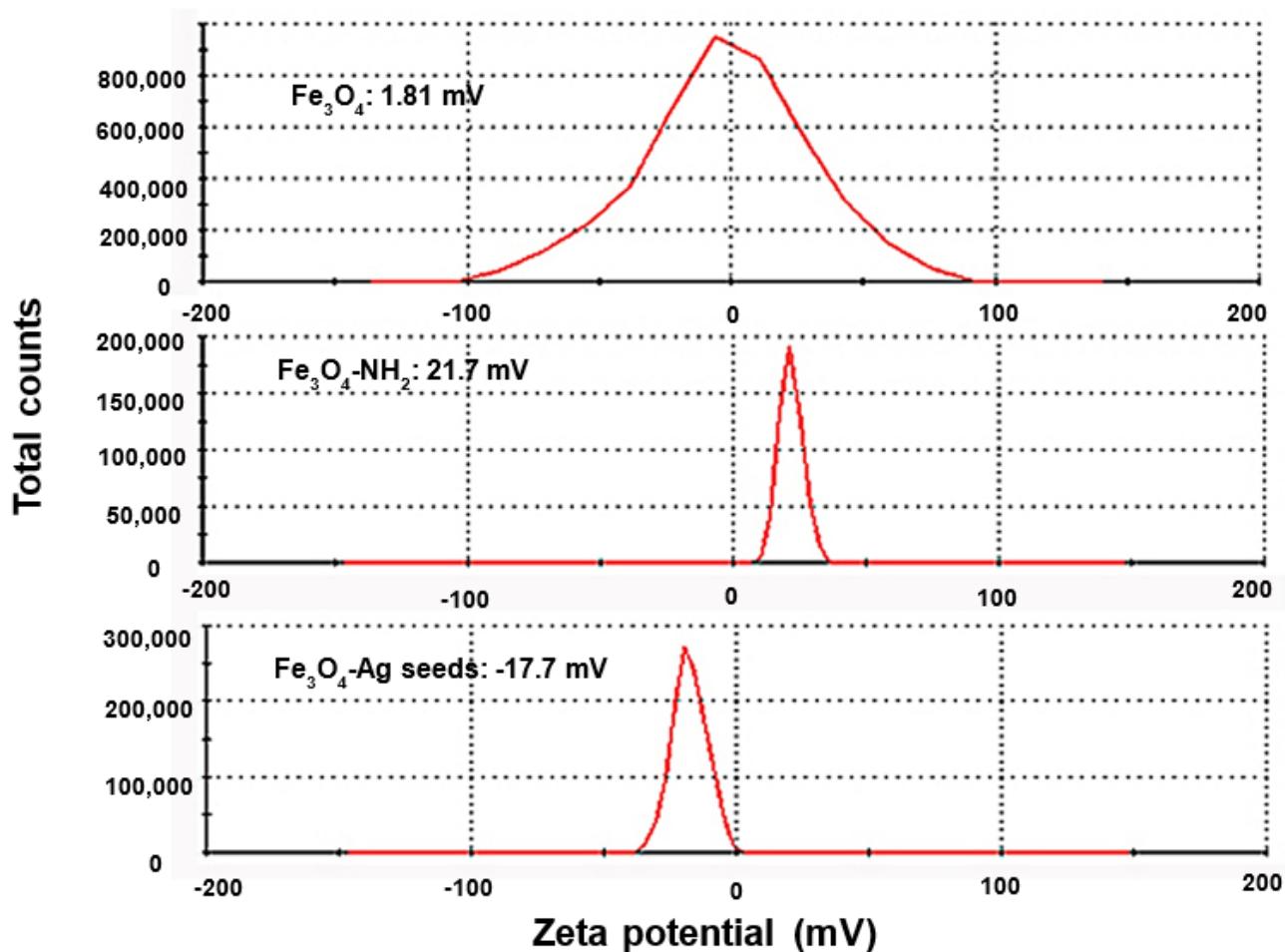


Figure S2. Zeta potential distribution of the $\text{Fe}_3\text{O}_4\text{-Ag}$ seeds compared to that of Fe_3O_4 particles coated with APTES and when dispersed in deionized water.

Table S1. EDX elemental analysis of Fe_3O_4 particles, $\text{Fe}_3\text{O}_4\text{-Ag}$ seeds, $\text{Fe}_3\text{O}_4\text{-Ag}$ nanocomposites produced with PVP for a r value of 10 and 2, and without PVP for a r value of 10.

Sample	% Weight		% Atomic	
	Fe	Ag	Fe	Ag
Fe_3O_4 particles	66.9	0	37.72	0
$\text{Fe}_3\text{O}_4\text{-Ag}$ seeds	64.14	0.24	34.86	0.07
$\text{Fe}_3\text{O}_4\text{-Ag}$ with PVP ($r=10$)	8.59	76.12	38.57	0.31
$\text{Fe}_3\text{O}_4\text{-Ag}$ with PVP ($r=2$)	17.83	57.92	21.98	13.07
$\text{Fe}_3\text{O}_4\text{-Ag}$ without PVP ($r=10$)	0	94.76	0	73.19

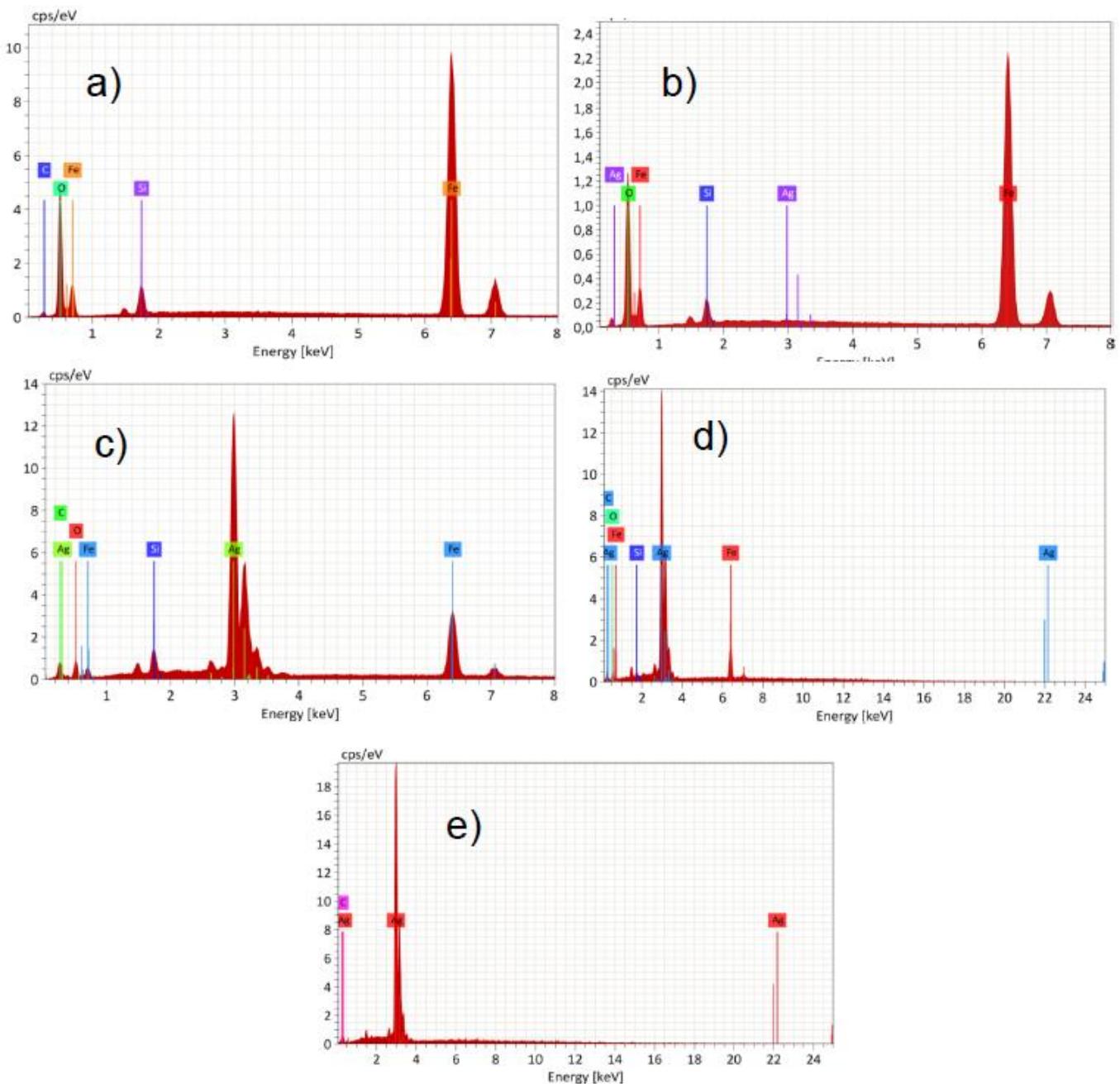


Figure S3. EDX analysis of: (a) Fe₃O₄ particles; (b) Fe₃O₄-Ag seeds; (c-e) Fe₃O₄-Ag nanocomposites produced with PVP for a r value of 10 and without PVP for a r value of 10, respectively.

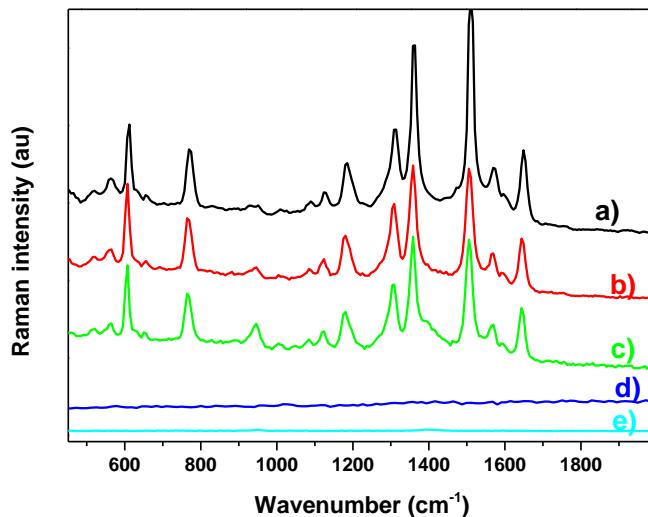


Figure S4. SERS spectra of R6G at 10^{-6} adsorbed on (a) Fe₃O₄-Ag nanocomposites with PVP ($r = 10$), (b) Fe₃O₄-Ag nanocomposites with PVP ($r = 2$), (c) Fe₃O₄-Ag nanocomposites without PVP ($r = 10$), (d) Fe₃O₄-Ag seeds, (e) Fe₃O₄ particles.

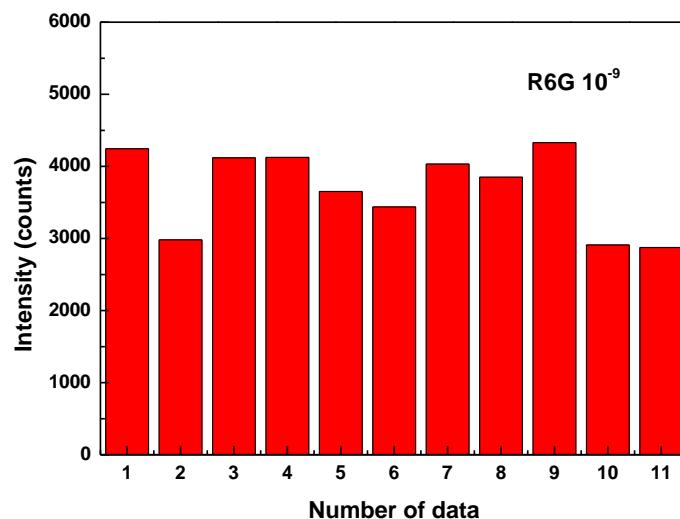


Figure S5. SERS intensity of R6G at 10^{-9} adsorbed on Fe₃O₄-Ag nanocomposites with PVP ($r = 10$).

Table S2. Assignment of Raman bands in spectra of R6G.

Raman Shift (cm ⁻¹)	Assignment
612	C-C-C ring in-plane bending
774	C-H out-of-plane bending
1185	C-H in-plane bending
1313	Aromatic C-C stretching
1363	Aromatic C-C stretching
1510	Aromatic C-C stretching
1574	Aromatic C-C stretching
1654	Aromatic C-C stretching