



Supplementary Material

Plasma Ag-Modified α -Fe₂O₃/g-C₃N₄ Self-Assembled S-Scheme Heterojunctions with Enhanced Photothermal-Photocatalytic-Fenton Performances

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Materials

Melamine (C₃H₆N₆, AR, ≥99.9%), Cyanuric acid (C₃H₃N₃O₃, AR, ≥99.9%), Iron(III) chloride hexahydrate (FeCl₃·6H₂O, AR, ≥99.9%), Silver nitrate (AgNO₃, AR, ≥99.9%), Sodium acetate (CH₃COONa, AR, ≥99.9%) were purchased from Aladdin Reagent (Shanghai) Co. Ltd, China. Ethanol absolute (C₂H₆O, ≥99.5%) was purchased from Shanghai Maclean Biochemical Technology Co. Ltd, China. All chemicals were sourced from commercial suppliers and were not further purified before being used, and deionized water was used throughout the experiments.

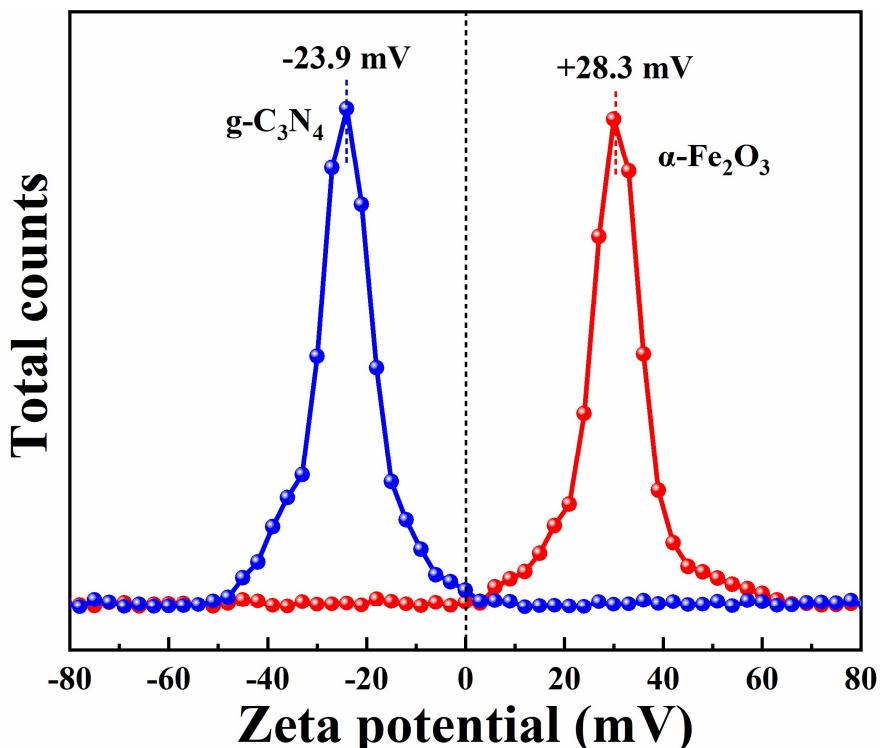


Figure S1. Zeta potential of the g-C₃N₄ and acidified α-Fe₂O₃ dispersed in deionized water at pH 7.

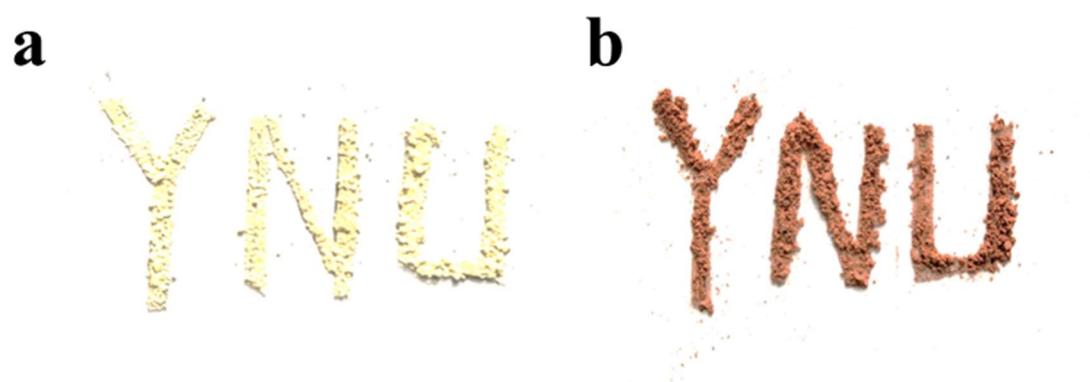


Figure S2. The sample patterns of g-C₃N₄ and Ag/α-Fe₂O₃/g-C₃N₄.

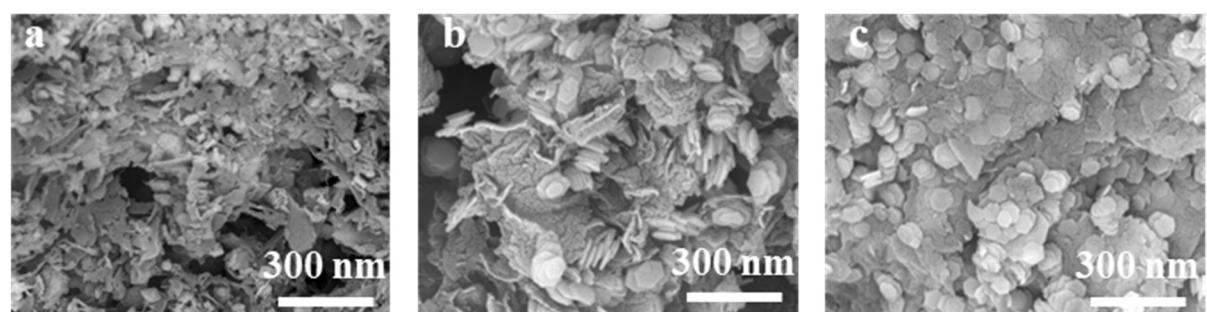


Figure S3. SEM images of α-Fe₂O₃/g-C₃N₄ for α-Fe₂O₃ weight ratios 20% (a), 30% (b), 40% (c).

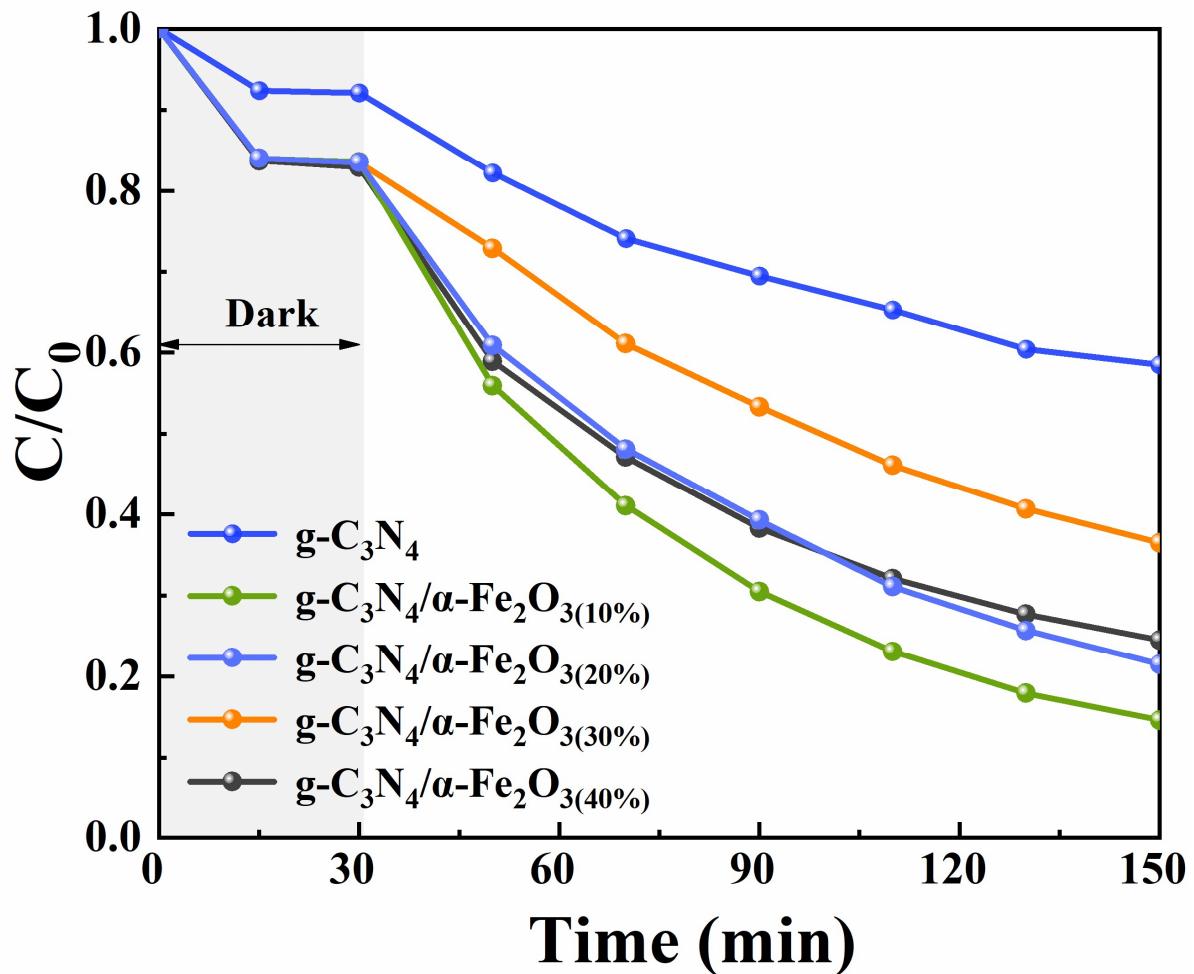


Figure S4. Degradation of TC by different weight ratios of $\alpha\text{-Fe}_2\text{O}_3/\text{g-C}_3\text{N}_4$.

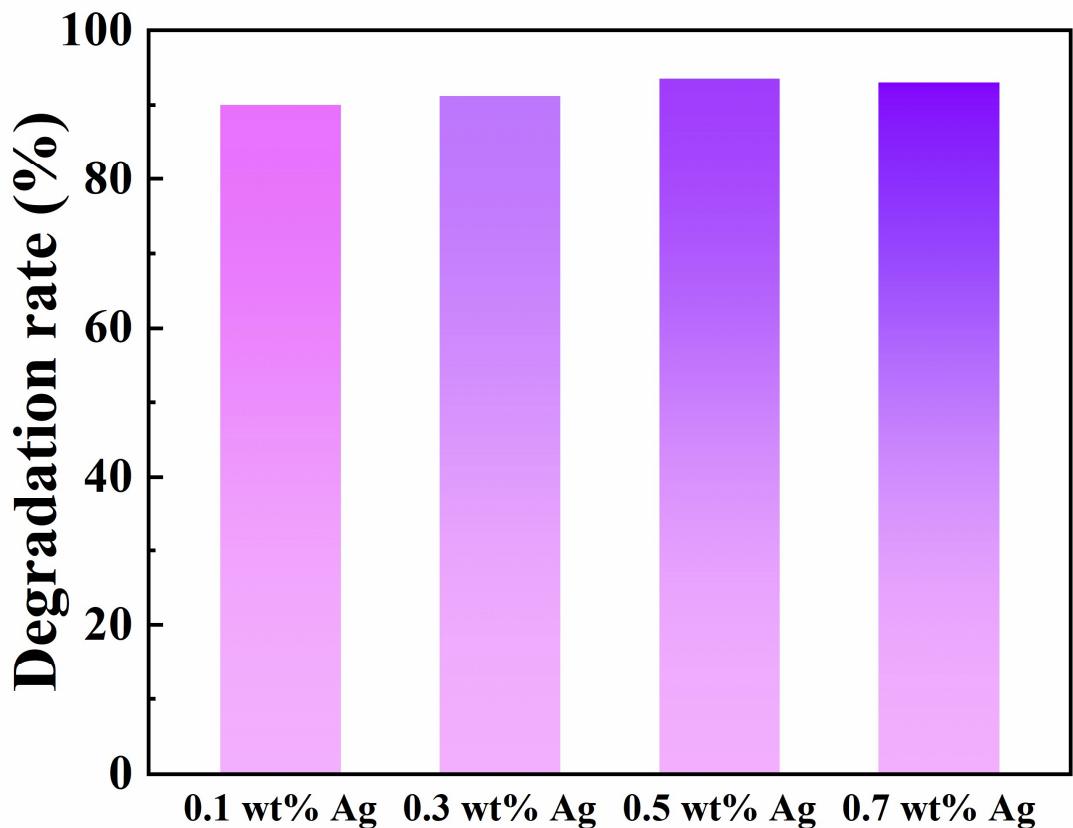


Figure S5. Degradation of TC by catalysts with different Ag loadings.

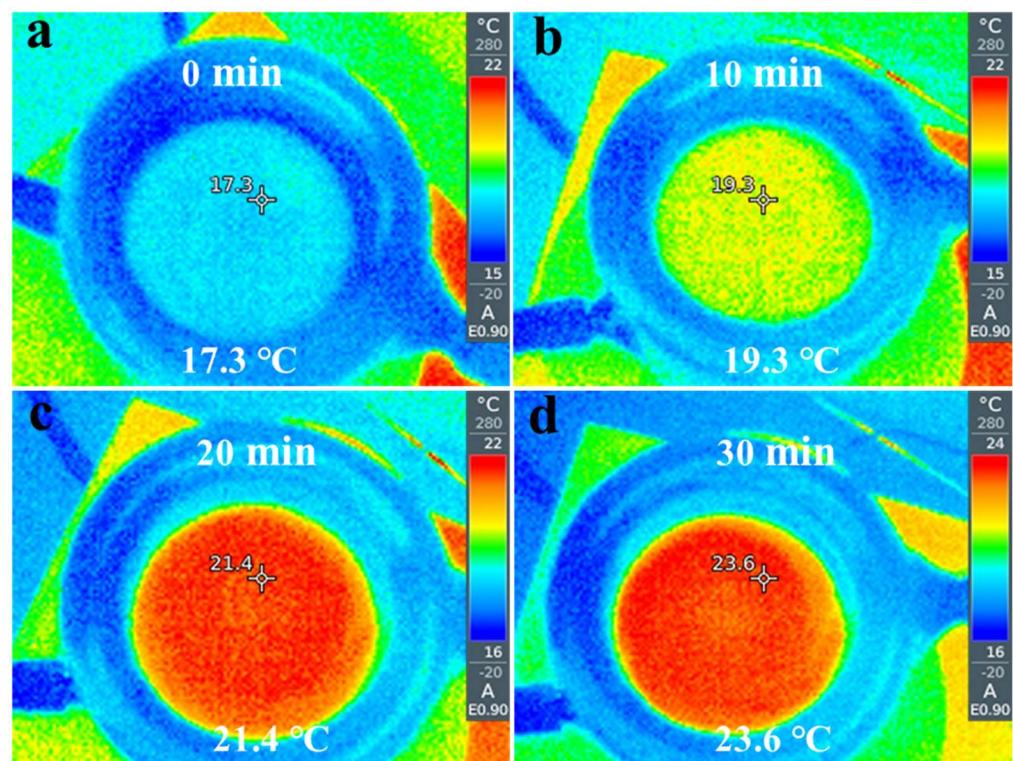


Figure S6. IR images of the pure water under illumination with a 300 w xenon lamp as a function of time.

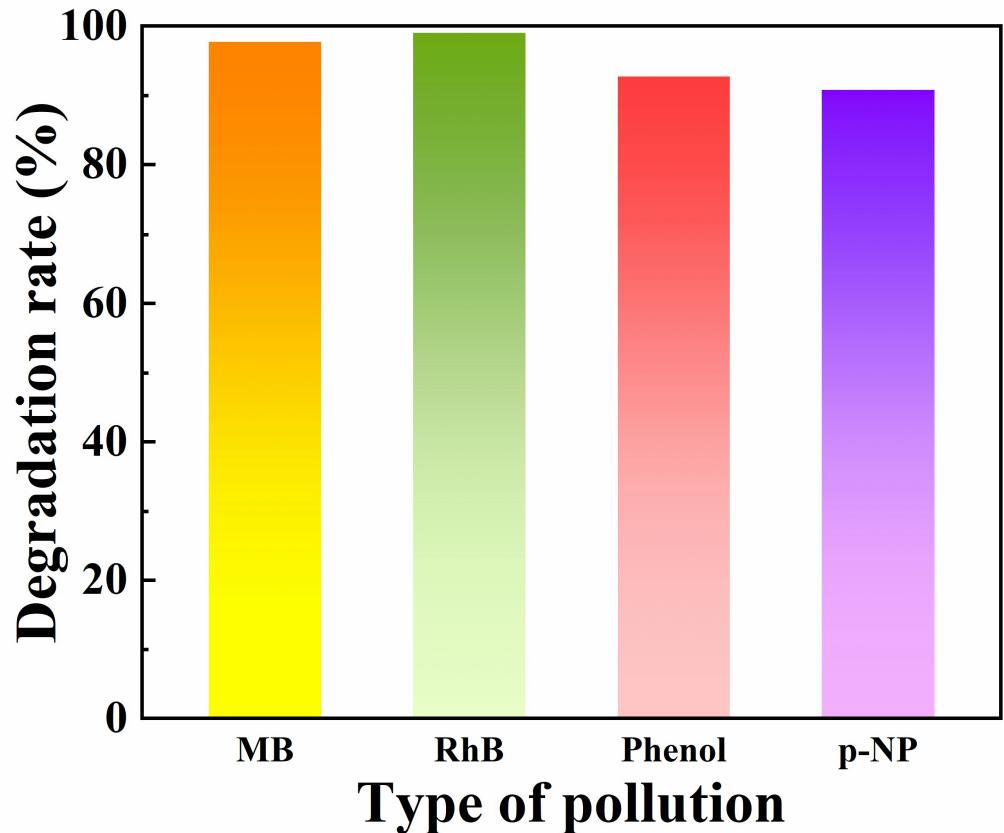


Figure S7. Photocatalytic degradation of other pollutants by $\text{Ag}/\alpha\text{-Fe}_2\text{O}_3/\text{g-C}_3\text{N}_4$.

Table S1. Comparison of the performance of $\text{Ag}/\alpha\text{-Fe}_2\text{O}_3/\text{g-C}_3\text{N}_4$ and other similar photocatalysts for H_2 evolution.

Photocatalysts	Light source	H_2 evolution rate ($\mu\text{mol g}^{-1} \text{h}^{-1}$)	Ref.
$\text{Ag}/\alpha\text{-Fe}_2\text{O}_3/\text{g-C}_3\text{N}_4$	Xe lamp (300W)	3125	This work
$\text{MoS}_2/\text{CdS}/\text{g-C}_3\text{N}_4$	300 W Xe lamp ($\lambda > 420 \text{ nm}$)	956	[8]
KCCN	Xe lamp ($\lambda > 420 \text{ nm}$)	557	[9]
$\alpha\text{-Fe}_2\text{O}_3/\text{CdS}/\text{g-C}_3\text{N}_4$	Xe lamp (1000 W)	165	[10]
$\alpha\text{-Fe}_2\text{O}_3/\text{g-C}_3\text{N}_4(\text{Pt})$	Xe lamp (300W, $\lambda > 420 \text{ nm}$)	5000	[11]
$\text{ZnIn}_2\text{S}_4/\text{Ti}_3\text{C}_2$	Xe lamp (300 W, $\lambda > 420 \text{ nm}$)	979	[12]
$\text{g-C}_3\text{N}_4/\text{WO}_3$ -carbon microsphere	Xe lamp (300W, $\lambda > 420 \text{ nm}$)	2500	[13]
hollow core-shell $\text{TiO}_2/\text{g-C}_3\text{N}_4$	Xe lamp (300 w)	809	[14]

Table S2. Comparison of the photocatalytic performance of Ag/ α -Fe₂O₃/g-C₃N₄ and other similar photocatalysts for TC degradation.

Photocatalysts	Light source	TC Removal rate	Ref.
Ag/ α -Fe ₂ O ₃ /g-C ₃ N ₄	Xe lamp (300W, $\lambda > 420$ nm)	93.6%	This work
TiO ₂ -P25	Xe lamp (300 W, $\lambda = 350$ nm)	94.8%	[1]
CuInS ₂ /Bi ₂ MoO ₆	Xe lamp (300 W, $\lambda > 420$ nm)	84.7%	[2]
Ag ₂ CO ₃ /Bi ₄ O ₅ I ₂ /g-C ₃ N ₄	Xe lamp (300 W, $\lambda > 420$ nm)	82.2%	[3]
β -Bi ₂ O ₃ @g-C ₃ N ₄	Xe lamp (250 W, $\lambda > 420$ nm)	80.2%	[4]
g-C ₃ N ₄ /ZrO _{2-x} NTs	Xe lamp (300 W, $\lambda > 420$ nm)	90.6%	[5]
CDs/gC ₃ N ₄ /MoO ₃	Xe lamp (350W, $\lambda > 420$ nm)	88.4%	[6]
g-C ₃ N ₄ /carbon nanotubes/Bi ₂₅ FeO ₄₀	Xe lamp (500 w, $\lambda > 420$ nm)	87.9%	[7]

References

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