

Polyoxometalate-Stabilized Silver Nanoparticles and Hybrid Electrode Assembly using Activated Carbon

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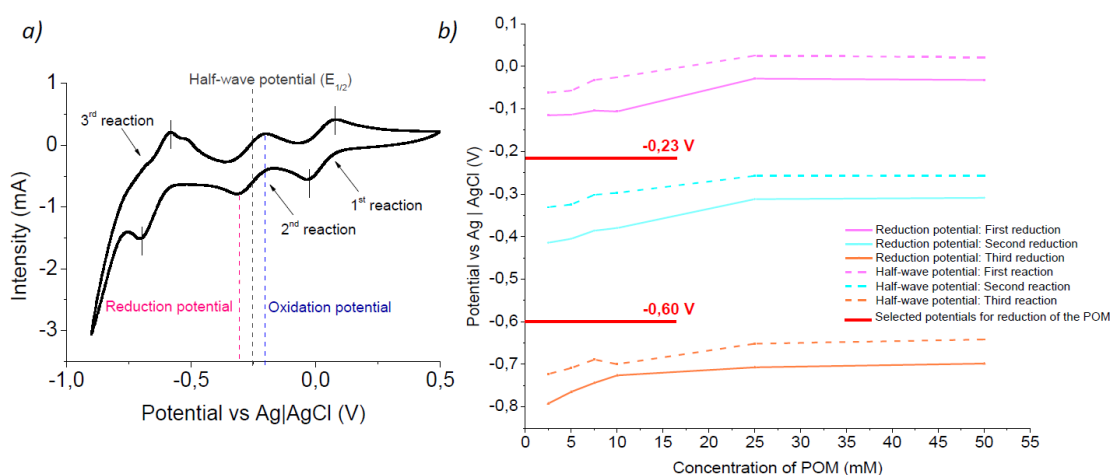


Figure S1. Cyclic Voltammetry results of POM in water. a) Example of a Cyclic Voltammogram of a 50 mM POM solution at 100 mV/s of scan rate. Three redox waves corresponding to the 1, 2 and 3 e^- reduction reactions are observed in agreement with the literature.[55] b) Change of reduction and half-wave potentials with nominal concentration of POM; values were taken from CV measurements at 100 mV/s of scan rate. Selected values for the chronoamperometric study of the POM are marked in red.

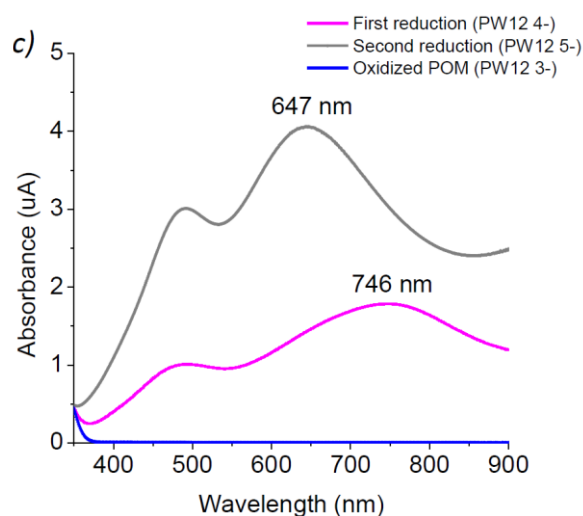


Figure S2. UV-vis spectroscopy signal of both reduced states (PW₁₂⁴⁻ and PW₁₂⁵⁻) and oxidized state of the POM (PW₁₂³⁻).

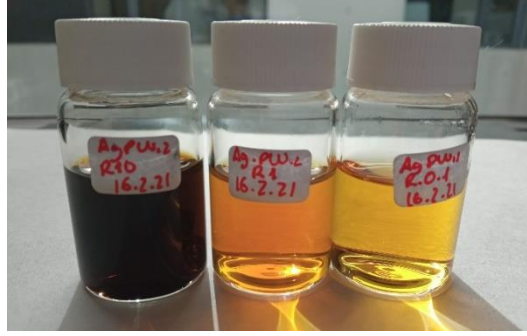


Figure S3. Photography of the reaction mixtures resulting from the synthesis of POM-Ag⁰ NPs using Ag⁺ / *red*POM ratios of 10 (left), 1 (center), and 0.1 (right).

The capacitance C (F) was calculated from the cyclic polarization curves according to the following equation:[67]

$$C_{cell} = \frac{\int I(V)dV}{2v\Delta V} \quad (\text{Equation S1})$$

Where C_{cell} is the capacitance of the device, I is the current, ΔV is voltage window, and v is the scan rate.

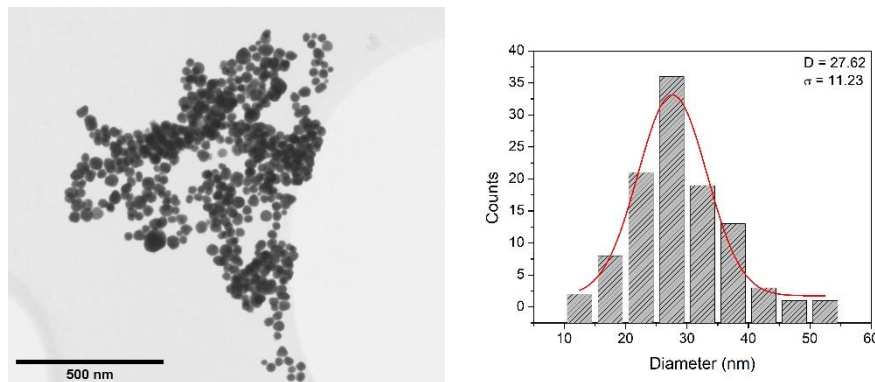


Figure S4. STEM image (left) and size distribution (right) obtained from Gaussian fit to the STEM histograms of POM-Ag⁰ NPs synthesized using: Ag⁺ / *red*POM = 1 and *red*POM = PW₁₂⁵⁻.

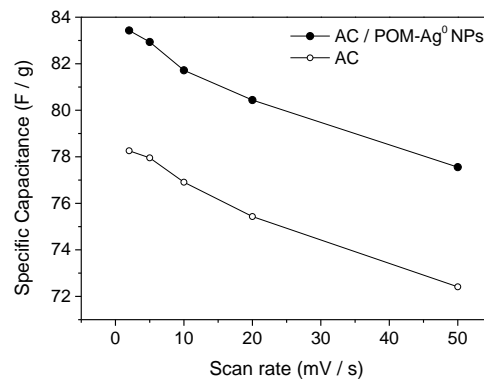


Figure S5. Variation of specific capacitance of bare AC and AC / POM-Ag⁰ NPs symmetric cells with scan rate.

References:

55. Sadakane, M.; Steckhan, E. Electrochemical Properties of Polyoxometalates as Electrocatalysts. *Chemical Reviews* 1998, 98, 219–238, doi:10.1021/cr960403a.
67. Allagui, A.; Freeborn, T.J.; Elwakil, A.S.; Maundy, B.J. Reevaluation of Performance of Electric Double-Layer Capacitors from Constant-Current Charge/Discharge and Cyclic Voltammetry. *Sci Rep-uk* 2016, 6, 38568, doi:10.1038/srep38568.