

Supplementary Information

Regio-selective carboxylated cellulose nanofibrils from dissolving pulp: C6 via TEMPO-mediated oxidation and C2,C3 via periodate-chlorite oxidation

Mengzhe Guo¹, James Ede², Christie Sayes³, Jo Anne Shatkin², Nicole Stark⁴, You-Lo Hsieh^{1*}

¹Biological and Agricultural Engineering, Chemical Engineering, University of California at Davis, Davis, CA 95616-8722

²Vireo Advisors, LLC, PO Box 51368, Boston, MA, 02130

³Environmental Science, Baylor University, Waco, TX 76798-7266

⁴USDA Forest Service, Forest Products Laboratory, Madison, WI 53726-2398

*Email: ylhsieh@ucdavis.edu; Tel: +1 530 752 0843

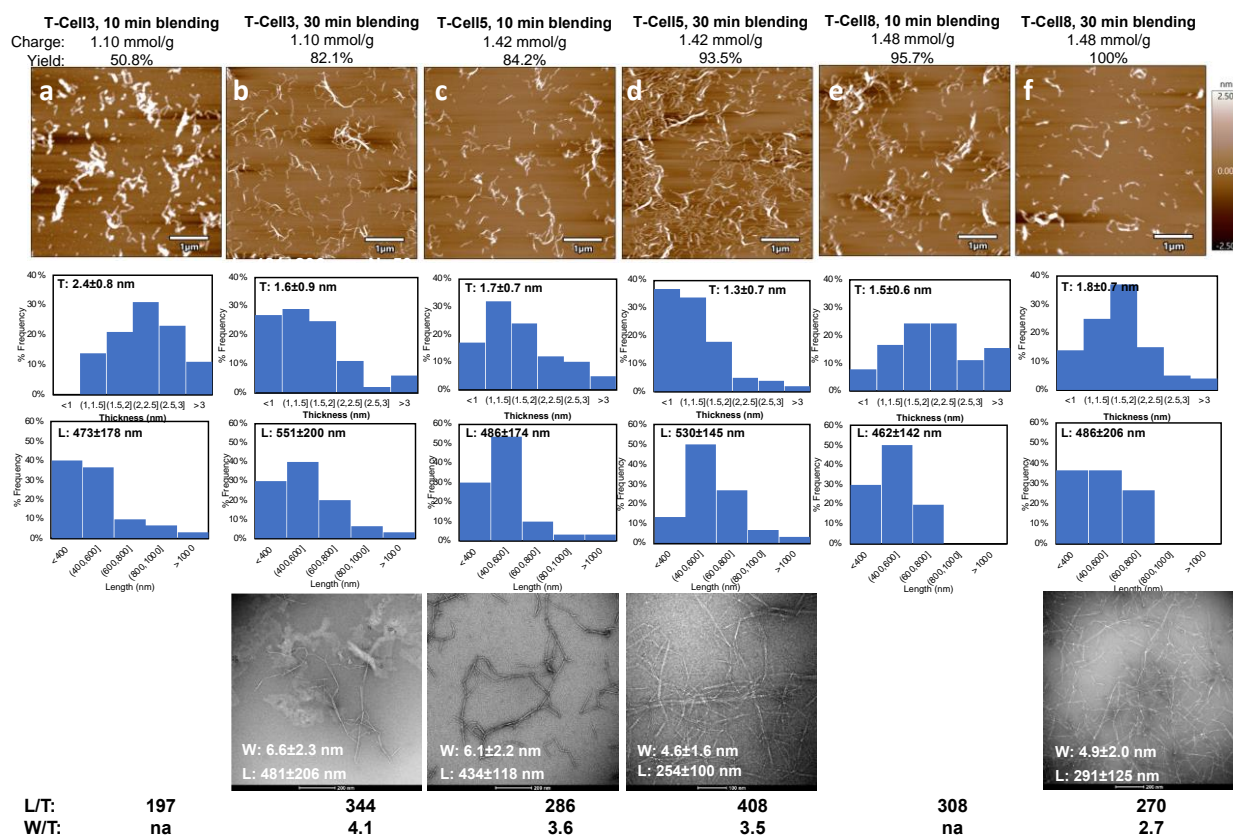


Figure S1. Yields and charge of T-CNFs from blending (5k rpm) of (a) Cell3 (10 min), (b) Cell3 (30 min), (c) Cell5 (10 min), (d) Cell5 (30 min), (e) Cell8 (10 min), and (f) Cell8 (30 min). AFM were imaged on mica with corresponding thickness (N=100) and length (N=30) distribution at 0.0002 w/v%.

1.1 Sodium Hypochlorite Titration

NaI (1.26 g, 8.4 mmol) was dissolved in acetic acid (4 mL) and mixture was diluted to 100 mL with purified water. NaI solution (100 mL) was divided into two equal 50 mL solutions with 1 mL NaClO solution (ca. 13.5 w/v%) added to each for dark red color solution. 0.1 M Na₂S₂O₃ solution was drop-wise (0.1 mL) added into NaI solution until red color faded and total volume was recorded.

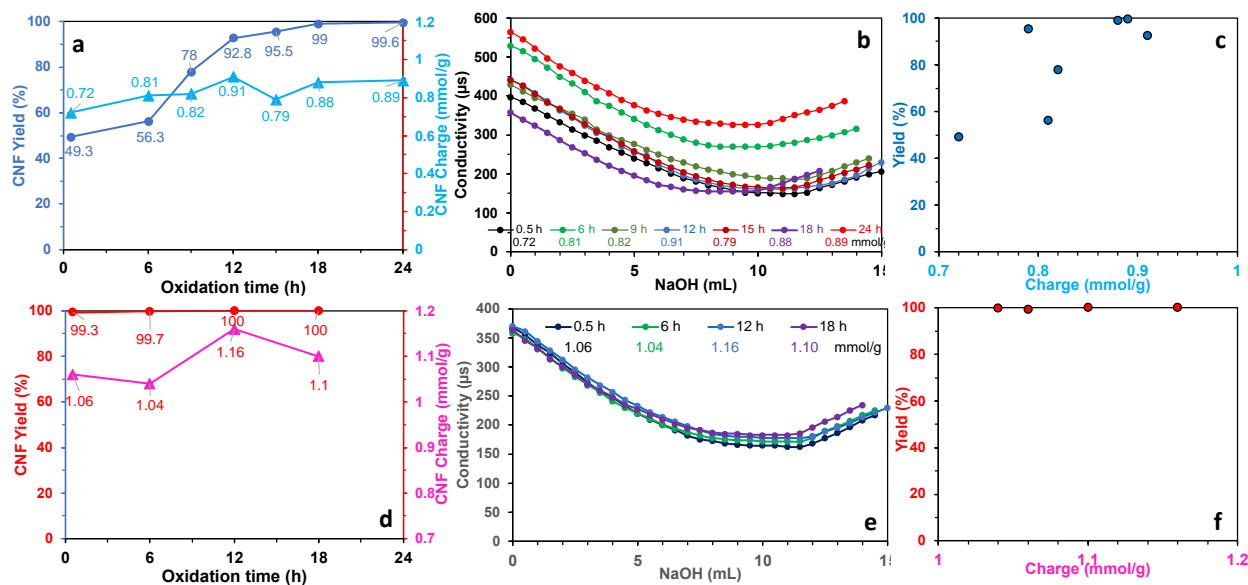


Figure S2. PC-CNF from sequential periodate-chlorite oxidation followed by blending (37,000 rpm, 30 min) at (a,d,c) 0.5:1 or (d,e,f) 0.75:1 NaIO₄/AG primary NaIO₄ oxidation (55 °C, 4 h) and varying secondary NaClO₂ (1:1 NaClO₂/AG) oxidation time followed by 30 min blending: (a,d) yield and charge, (b,e) conductivity, (c,f) yield.