

Electronic Supporting information (ESI)

Voltammetric Detection of Caffeine in Beverages at Nafion/Graphite Nanoplatelets Layer-by-Layer Films

Sandra Hernandez-Aldave ¹, Afshin Tarat ², James D. McGettrick ³ and Paolo Bertoncello ^{1,4,*}

[Synthesis process.mp4](#)

Figure S1. Video showing the procedure for the fabrication of Nafion/GNPs LbL films.

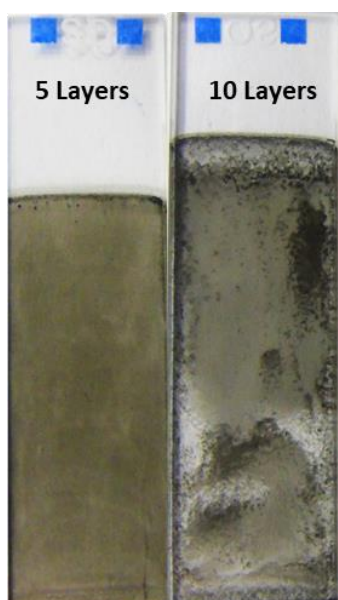


Figure S2. Optical image of Nafion/GNPs films (5 and 10 layers) at 0.8% wt of GNPs.



Figure S3. Optical image of Nafion/GNPs LbL films (left) and pristine GNP LbL films (right).

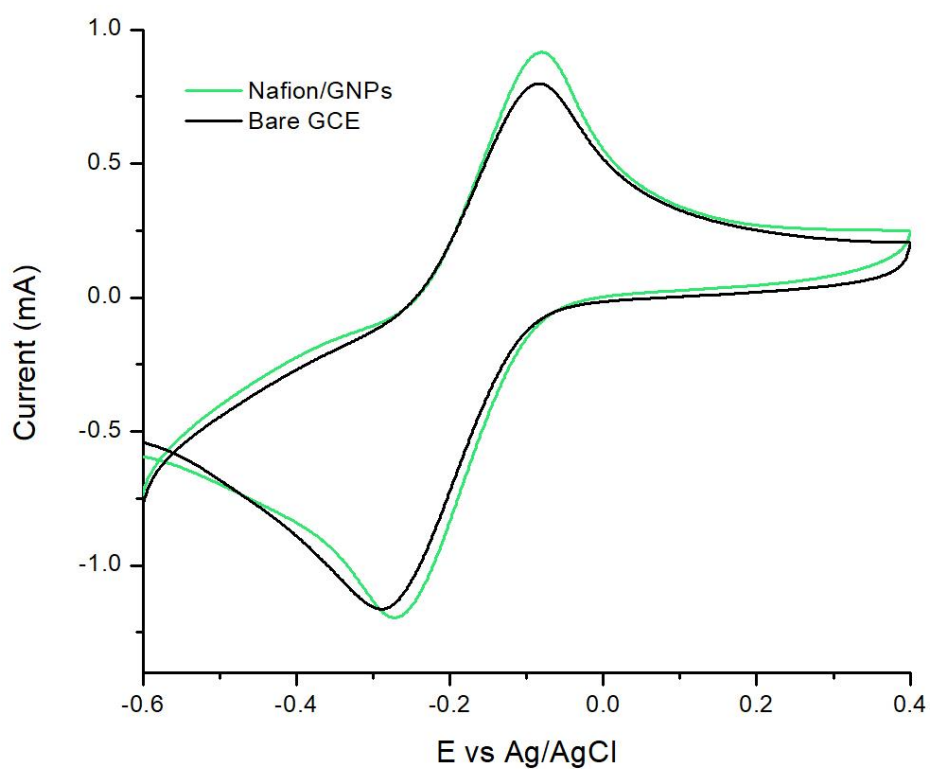


Figure S4. CV of bare GCE (black line) and 5 Nafion/GNPs LbL film (green line) recorded in 0.1 M NaCl supporting electrolyte and 5 mM $\text{Ru}[(\text{NH}_3)_6]^{3+}$; scan rates 50 mV s^{-1} .

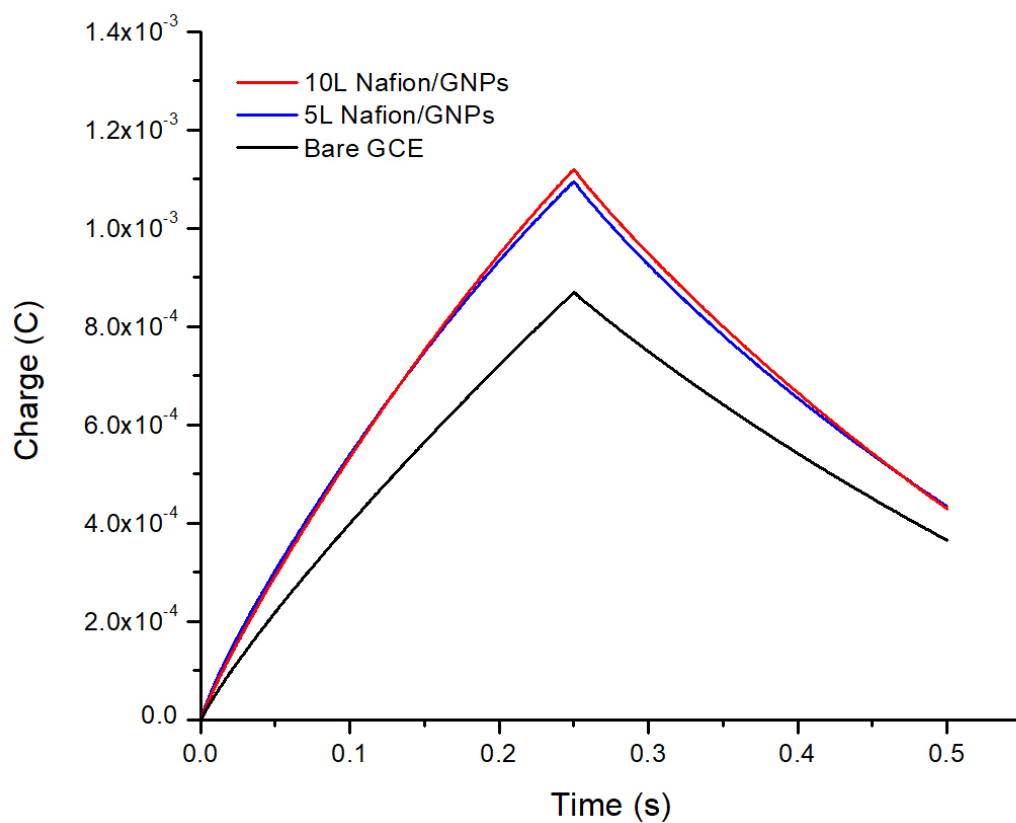


Figure S5. Chronocoulometric plot of bare GCE (black line), 5 Nafion/GNPs LbL films (blue line) and 10 Nafion/GNPs LbL films (red line) recorded in 0.1M NaCl containing 5mM $\text{Ru}[(\text{NH}_3)_6]^{3+}$.

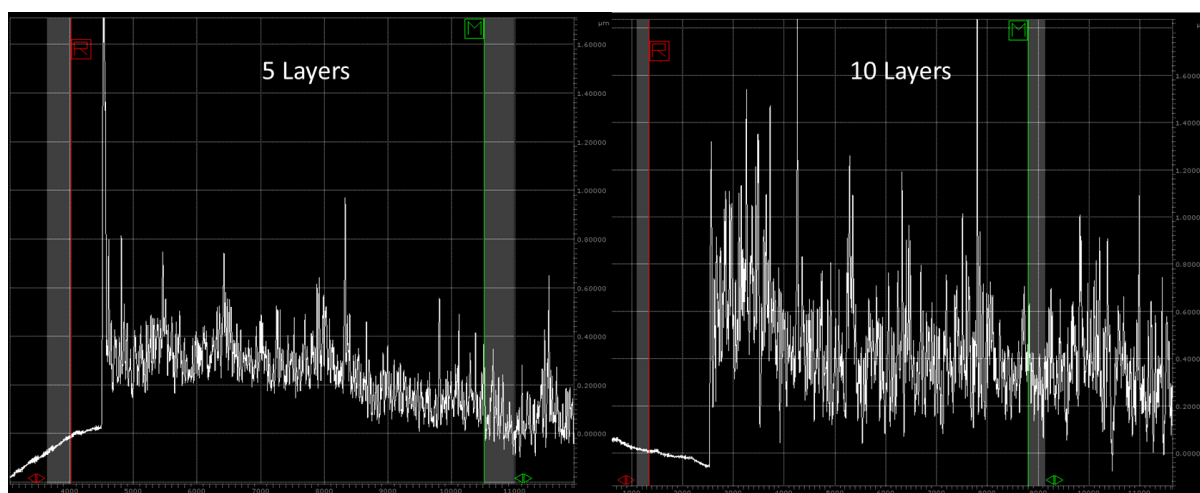


Figure S6. Profilometer patterns of Nafion/GNPs LbL films.

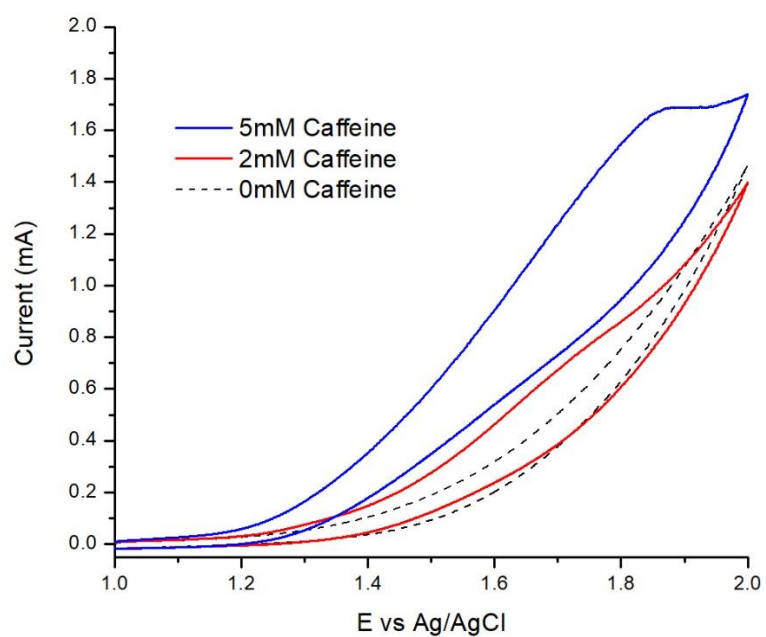


Figure S7. CVs of bare glassy carbon recorded in 0.1 M NaCl supporting electrolyte (pH 2) containing 5 mM caffeine; scan rates 50 mV s⁻¹.

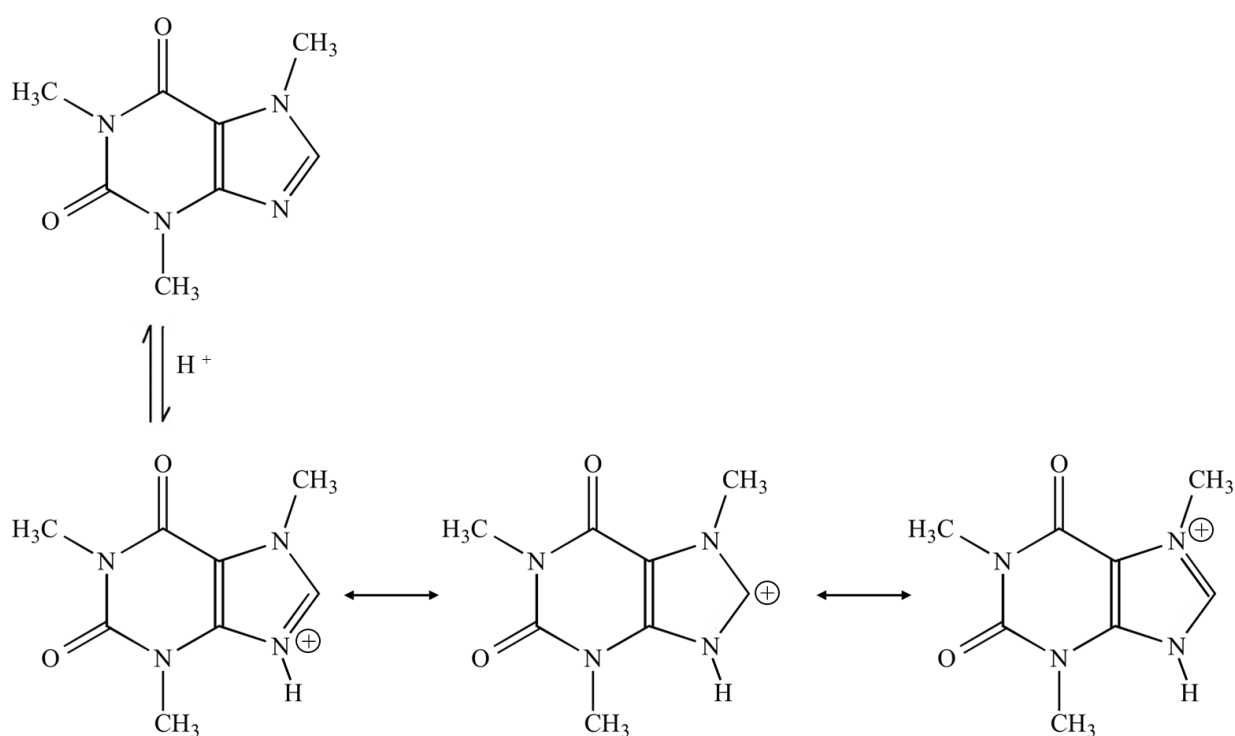


Figure S8. Protonated chemical structures of caffeine [1,2].

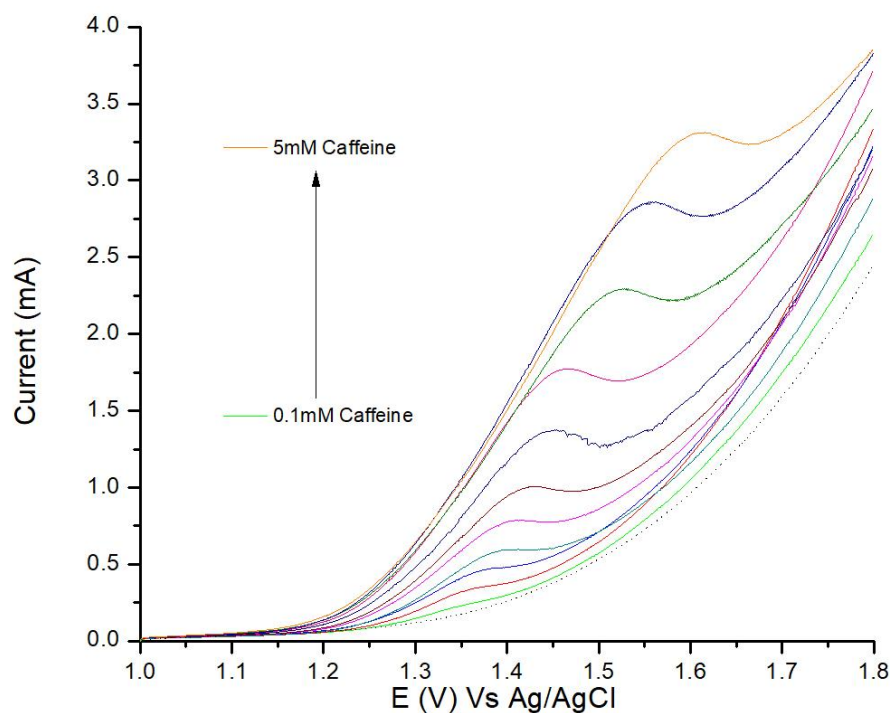


Figure S9. CVs of 5 Nafion/GNPs LbL films recorded in 0.1 M NaCl supporting electrolyte (pH 2) at different concentrations of caffeine from 0.1 mM to 5 mM; scan rate, 50 mV s⁻¹.

References

1. Bahrami, H.; Tabrizchi, M.; Farrokhpour, H. Protonation of caffeine: A theoretical and experimental study. *Chemical Physics* **2013**, *415*, 222-227, doi:<https://doi.org/10.1016/j.chemphys.2013.01.022>.
2. Oestreich-Janzen, S. Chemistry of Coffee, Change History: March 2013. S Oestreich-Janzen has updated the text throughout in this chapter. In *Reference Module in Chemistry, Molecular Sciences and Chemical Engineering*, Elsevier: 2013; <https://doi.org/10.1016/B978-0-12-409547-2.02786-4>.