

# Engineering the Dimensional Interface of BiVO<sub>4</sub>-2D Reduced Graphene Oxide (RGO) Nanocomposite for Enhanced Visible Light Photocatalytic Performance

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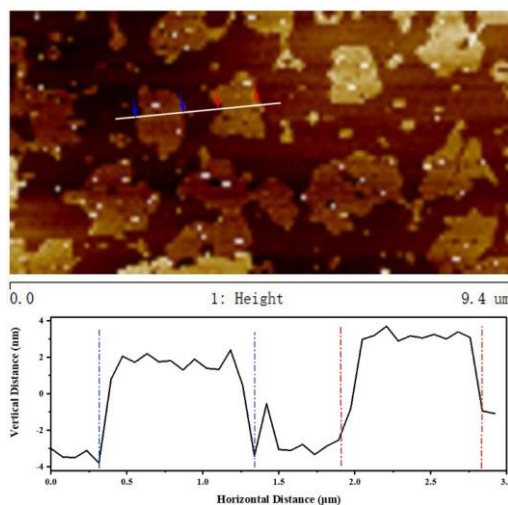
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(7 Pages, 10 Figures)

**Preparation of BiVO<sub>4</sub> nanoparticles.** 2.91 g Bi (NO<sub>3</sub>)<sub>3</sub> 5H<sub>2</sub>O and 0.7204 g NH<sub>4</sub>VO<sub>3</sub> were added into a beaker containing 30 mL 1 M HNO<sub>3</sub> solution. After stirring for a few minutes, 3 g (NH<sub>2</sub>)<sub>2</sub>CO was gradually added into the above solution and then placed in a water bath at a temperature of 80 °C for 24 h. The solution was centrifuged to obtain the yellow sediment and washed with deionized water. The BiVO<sub>4</sub> nanoparticles were dried at 60 °C in an oven and then calcined at 400 °C for 1 h.

**Preparation of BiVO<sub>4</sub> nanoparticles /RGO Composites.** The BiVO<sub>4</sub> nanoparticles and GO solution (1g L<sup>-1</sup>) were dissolved into 50 mL ultrapure water and stirred to ensure complete mixing. Then, the obtained solution was transferred to Teflon-lined autoclave, and maintained at 160 °C for 6 h in an oven. The above product was washed with alcohol and ultrapure water several times, centrifuged, dried at 60 °C in an oven.



**Figure S1.** Atomic force microscopy images of the 2D BiVO<sub>4</sub> nanosheets.

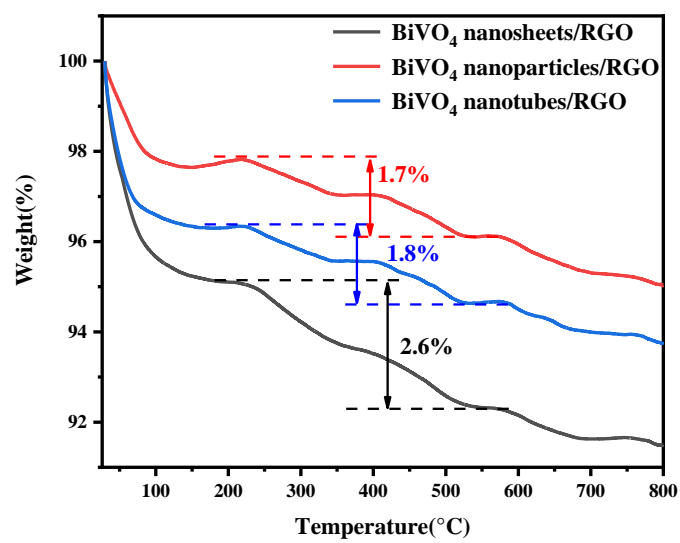


Figure S2. Thermo gravimetric analysis (TGA) of the BiVO<sub>4</sub>/RGO composites.

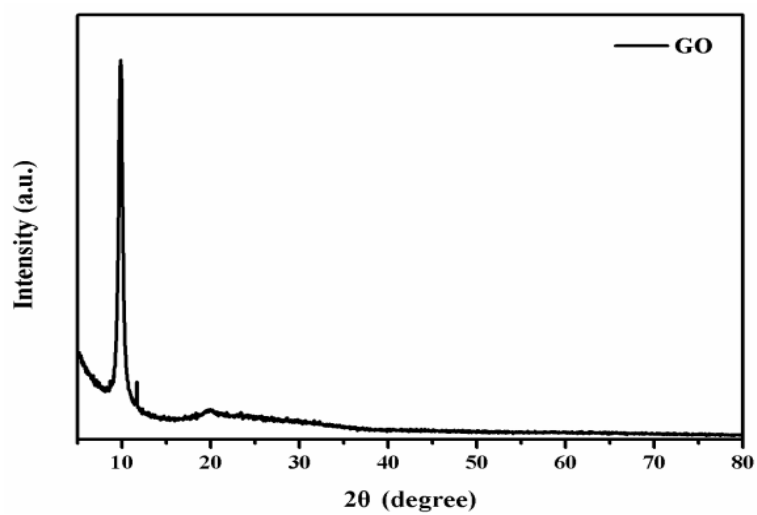


Figure S3. XRD patterns of the prepared graphene oxide (GO).

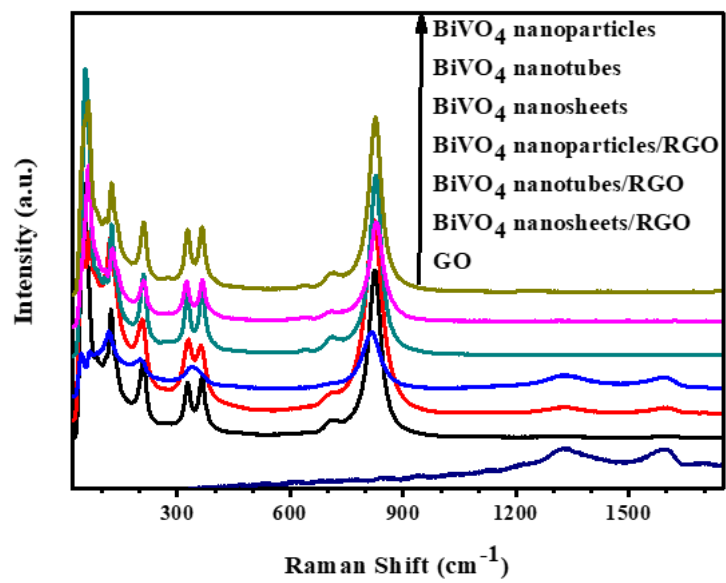


Figure S4. Raman spectra of BiVO<sub>4</sub> and BiVO<sub>4</sub>/RGO composites.

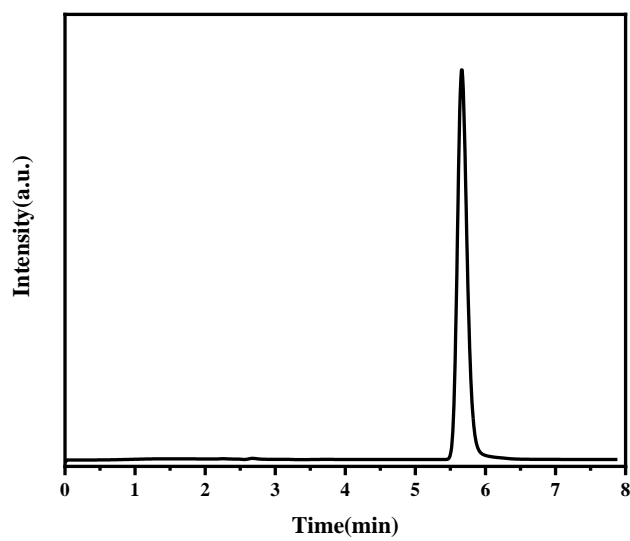
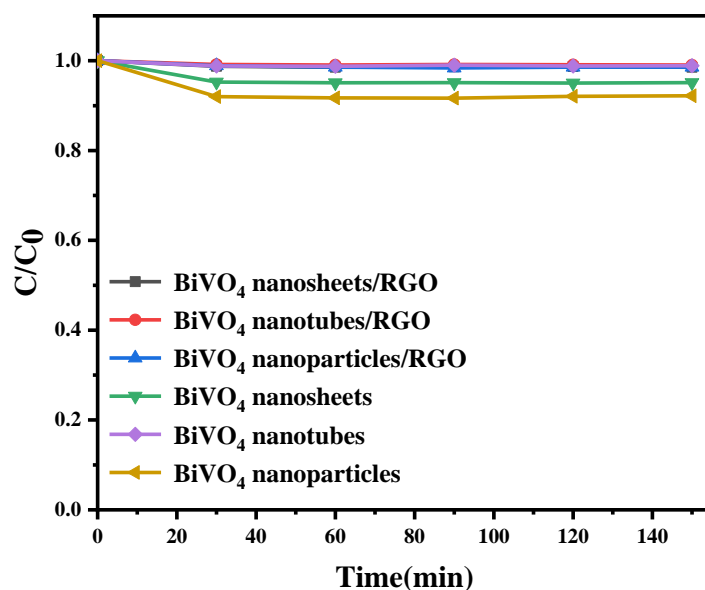


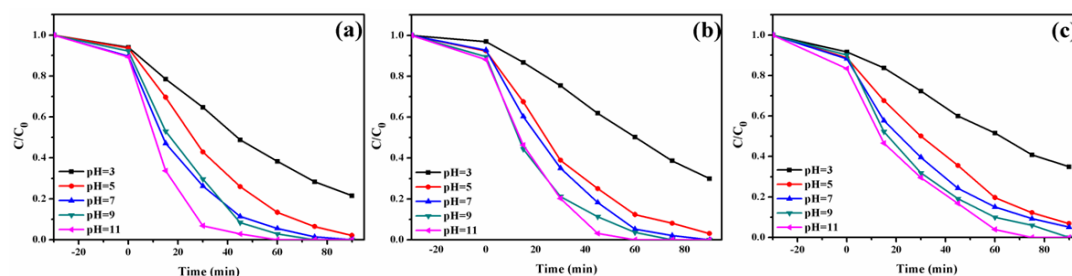
Figure S5. HPLC chromatograms of acetaminophen standard sample.



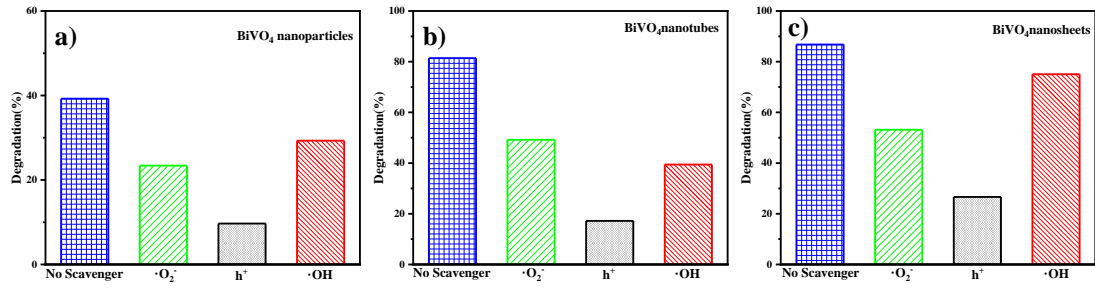
**Figure S6.** The adsorptive performance of acetaminophen over the BiVO<sub>4</sub> and BiVO<sub>4</sub>/RGO composites without visible light irradiation.

**Table S1.** The pseudo-first order kinetic equation and the rate constant (*k*) of BiVO<sub>4</sub> and BiVO<sub>4</sub>/RGO composites.

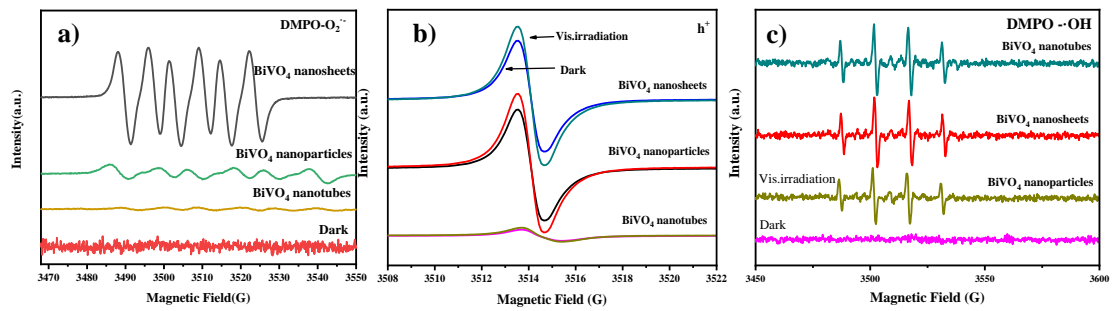
Samples	First-order kinetic equation	<i>k</i> (min <sup>-1</sup> )	R <sup>2</sup>
BiVO <sub>4</sub> nanosheets/RGO	$-\ln(C/C_0) = 0.0141x - 0.0472$	0.0141	0.9976
BiVO <sub>4</sub> nanotubes/RGO	$-\ln(C/C_0) = 0.0122x - 0.0684$	0.0121	0.9963
BiVO <sub>4</sub> nanoparticles/RGO	$-\ln(C/C_0) = 0.0107x - 0.0292$	0.0107	0.9990
BiVO <sub>4</sub> nanosheets	$-\ln(C/C_0) = 0.008x - 0.0142$	0.0080	0.9986
BiVO <sub>4</sub> nanotubes	$-\ln(C/C_0) = 0.0084x - 0.0737$	0.0084	0.9915
BiVO <sub>4</sub> nanoparticles	$-\ln(C/C_0) = 0.0077x - 0.0889$	0.0077	0.9924



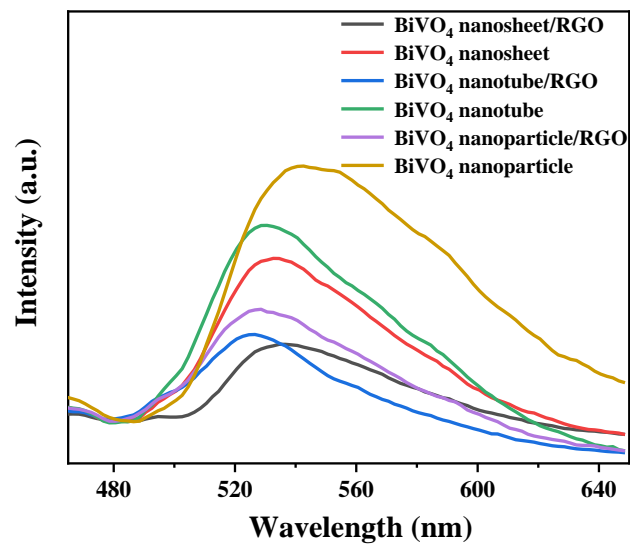
**Figure S7.** Photocatalytic degradation of RhB over photocatalysts under different pH conditions:(a) BiVO<sub>4</sub> nanosheet /RGO; (b) BiVO<sub>4</sub> nanotube /RGO;(c) BiVO<sub>4</sub> nanoparticle /RGO.



**Figure S8.** Free radical inhibition experiment of BiVO<sub>4</sub> samples: (a) BiVO<sub>4</sub> nanoparticles; (b) BiVO<sub>4</sub> nanotubes; (c) BiVO<sub>4</sub> nanosheets.



**Figure S9.** Electron spin resonance spectra of radical in BiVO<sub>4</sub> samples under visible light: (a) DMPO-O<sub>2</sub><sup>·-</sup>, (b) h<sup>+</sup> and (c) DMPO-·OH.



**Figure S10.** PL spectra of BiVO<sub>4</sub> and BiVO<sub>4</sub>/RGO samples.