Supporting information (S1): Chemical synthesis procedure of KryAz600 surfactant.

## **General Procedures**

All purchased reagents were used without further purification. HFE7100 was dried on molecular sieves 4Å. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a JEOL 400 spectrometer and referenced to the resonance of the solvent used. Infrared (IR) spectra were recorded with a Nicolet Nexus FT-IR spectrometer equipped with an ATR-Germanium unit and are reported as wavenumbers (cm<sup>-1</sup>). High-resolution mass spectra (HRMS) were performed on a Bruker maXis mass spectrometer by the "Fédération de Recherche" ICOA/CBM (FR2708) platform.

## Synthetic scheme

$$\begin{array}{c} NO \\ NO \\ NO \\ CO_2 tBu \end{array} \begin{array}{c} H_2N - NH_2 \\ ACOH \\ \hline \end{array} \begin{array}{c} H_2N - NH_2 \\ ACOH \\ \hline \end{array} \begin{array}{c} H_2N - NH_2 \\ ACOH \\ \hline \end{array} \begin{array}{c} 1 \\ CO_2 tBu \end{array} \end{array} \begin{array}{c} 1 \\ CO_2 tBu \end{array}$$

## Synthetic procedures

1: To a solution of t-butyl 4-nitrosobenzoate (95 mg, 0.46 mmol) in AcOH (4 mL) was added 4-phenylenediamine (49 mg, 0.45 mmol, 1 equiv). After stirring at 60°C for 12 h, the mixture was evaporated under reduced pressure. The residue was dissolved in  $CH_2CI_2/H_2O$  (1/1, 50 mL) and the layers were separated. The aqueous layer was extracted with  $CH_2CI_2$  (3 x 10 mL), the combined organic layers were washed with a saturated aqueous solution of NaHCO<sub>3</sub>, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated under reduced pressure. The crude product was purified by column chromatography (silica gel-PE/EA: 7/3) to afford 1 (37 mg, 0.12 mmol, 27%) as an orange solid;  $^1H$  NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  ppm 8.10 (d, J = 8.2 Hz, 2H), 7.84 (m, 4H), 6.75 (d, J = 8.7 Hz, 2H), 4.14 (br s, 2H, NH<sub>2</sub>), 1.63 (s, 9H);  $^{13}C$  NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  ppm 165.6, 155.5, 150.3, 145.7, 132.7, 130.5, 125.7, 122.1, 114.7, 81.4, 28.4; HRMS (ESI): calc. for  $C_{17}H_{20}N_3O_2^+$ , [M+H]<sup>+</sup> m/z 298.1550, found m/z 298.1550.

**KryAz-COOtBu:** To a solution of Krytox® (2.22 g, 0.32 mmol) in anhydrous HFE7100 (7 mL) under argon was added oxalyl chloride (0.2 mL, 10 equiv). After stirring at 60°C for 12 h, the mixture was evaporated under reduced pressure. To a solution of the crude product in anhydrous HFE7100 (8 mL) under argon was added a solution of 1 (95 mg, 0.32 mmol, 1 equiv) and Et<sub>3</sub>N (0.2 mL) in anhydrous THF (8 mL). After stirring at r.t. for 12 h, the supernatant was removed and the orange oil was washed with THF (3 x 15 mL) to afford **KryAz-COOtBu** (1.62 g, 0.24 mmol, 74%) as an orange oil. **IR (ATR):** 1705, 1693 cm<sup>-1</sup>.

**KryAz-COOH:** To a solution of **KryAz-COOtBu** (1.62 g, 0.24 mmol) in HFE7100 (5 mL) was added trifluoroacetic acid (100  $\mu$ L, 1.35 mmol, 5.6 equiv). After stirring at r.t. for 12 h, the mixture was evaporated under reduced pressure. The crude product was washed with H<sub>2</sub>O (4 x 5 mL), acetone (2 x 5 mL) and CH<sub>2</sub>Cl<sub>2</sub> (2 x 5 mL) to afford **KryAz-COOH** (1.43 g, 0.21 mmol, 89%) as an orange oil. **IR (ATR):** 1784, 1708, 1690 cm<sup>-1</sup>.

**KryAz600:** To a solution of **KryAz-COOH** (930 mg, 0.138 mmol) in anhydrous HFE7100 (6 mL) under argon was added oxalyl chloride (225 μL, 1.38 mmol, 10 equiv). After stirring at 60°C for 12 h, the mixture was evaporated under reduced pressure. The crude residue in solution in anhydrous HFE7100 (6 mL) was added to a solution of *O*-methyl-undecaethylene glycol (80 mg, 0.15 mmol, 1.2 equiv) and  $\rm Et_3N$  (37 μL, 0.28 mmol, 2 equiv) in anhydrous THF (6 mL). After stirring at r.t. for 12 h, the supernatant was removed and the yellow oil was washed with  $\rm H_2O$  (2 x 5 mL) acetone (2 x 5 mL) and  $\rm CH_2Cl_2$  (2 x 50 mL) to afford **KryAz600** (920 mg, 0.127 mmol, 92%) as a yellow oil. **IR (ATR):** 1707, 1693 cm<sup>-1</sup>.

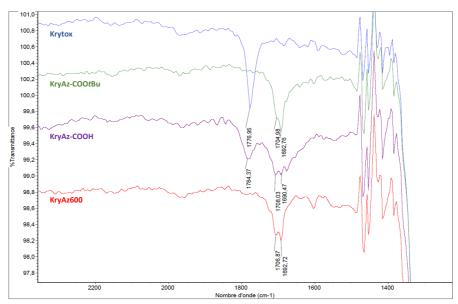


Figure S1. Portion of the infrared spectra of the different surfactants synthesized.