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Effect of Negative Bias of HiPIMS and AIP Hybrid Deposition on Microstructure, Mechanical and Anti-Corrosive Properties of Cr₂N/TiN Multilayer Coatings

Rong Tu ^{1,2,3} , Yang Yuan ¹, Mai Yang ¹, Rui Min ¹, Jiao Jiao ¹, Qizhong Li ^{1,4,*}, Meijun Yang ¹, Baifeng Ji ⁵  and Song Zhang ¹

- ¹ State Key Laboratory of Advanced Technology for Materials Synthesis and Processing, Wuhan University of Technology, 122 Luoshi Road, Wuhan 430070, China; turong@whut.edu.cn (R.T.); yuan_yang@whut.edu.cn (Y.Y.); yangmai@whut.edu.cn (M.Y.); minrui12234@whut.edu.cn (R.M.); materialjiao@whut.edu.cn (J.J.); liyangmeijun@163.com (M.Y.); kobe@whut.edu.cn (S.Z.)
- ² Chaozhou Branch of Chemistry and Chemical Engineering Guangdong Laboratory, Chaozhou 521000, China
- ³ Wuhan University of Technology Advanced Engineering Technology Research Institute of Zhongshan City, Xiangxing Road 6, Zhongshan 528400, China
- ⁴ Hubei Key Laboratory of Advanced Technology for Automotive Components, Wuhan University of Technology, Wuhan 430070, China
- ⁵ Hubei Key Laboratory of Roadway Bridge and Structure Engineering, Wuhan University of Technology, 122 Luoshi Road, Wuhan 430070, China; jbfeng@whut.edu.cn
- * Correspondence: qizhongli@whut.edu.cn



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Abstract: Multi-layered nitride coatings have been widely applied to improve the mechanical and anti-corrosive of metals and/or alloys. Cr₂N/TiN multilayer coatings were prepared by the combination of high-power pulsed magnetron sputtering (HiPIMS) and arc ion plating (AIP). The Cr₂N layer was co-deposited by HiPIMS and AIP, while the TiN layer was deposited by a single HiPIMS. With increasing the negative bias voltage (V_s) on substrate up to -100 V, the number and size of the droplets decreased; the average grain size of the coatings decreased from 9.4 to 7.5 nm and the hardness increased from 21.5 to 25.1 GPa, and the level of the adhesion of the coatings has reached HF1. The coatings obtained at $V_s = -100$ V present the best corrosion resistance in NaCl aqueous solution based on the anodic polarization curves and EIS spectroscopy.

Keywords: Cr₂N/TiN multilayer coatings; HiPIMS/AIP hybrid deposition; bias; hardness; corrosion resistance

1. Introduction

The reliability and safety of marine equipment depend on some key workpieces, which are exposed to seawater corrosion all year round. Thus, it is necessary to put higher demands on their service life and require the development of protective coatings with more corrosion resistance. The coatings containing titanium and chromium are generally effective in terms of corrosion resistance [1,2], whereas conventional single-layer coatings, such as TiN, CrN and TiCrN, are difficult to fulfill extreme and demanding conditions, which require excellent mechanical properties, bonding, and corrosion resistance simultaneously. Obviously, binary thin film materials are far from meeting the ideal requirements. In order to find materials with higher performance, multi-element, multi-layer and nanostructure have gradually entered our vision [3–8].

Multilayer coatings have better performances, such as wear resistance and corrosion resistance, which could be widely used to improve the performance and service life of components in complex environments [9–15]. CrN single-layer coatings have the strongest hindrance to ion release [9], while CrN/TiN multilayer coatings have higher hardness and corrosion resistance [10]. The high density of CrN/TiN multilayer coatings also prevents

the penetration of corrosive solutions into the substrate, thus improving the corrosion resistance [11,12]. In addition, Samim et al. found that ZrN/CrN multilayer coatings have extremely high polarization resistance and different layers can act as a barrier to the penetration of corrosive solutions [13], and the absence of pits or any serious damage on their surfaces is due to the presence of an anti-corrosion barrier that prevents corrosion from occurring [14]. Vengesa et al. reported that CrN/CrAlN multilayer coatings annealed at 700 °C had the highest corrosion resistance, which can be attributed to the presence of Cr₂N as a corrosion-resistant phase [15]. In recent years, the hcp-Cr₂N in chromium nitride binary system has been reported as an effective anti-corrosion phase [16]. Wei et al. reported that Cr₂N is the most effective wear-resistant coating compared to Cr and CrN coatings [17]. Cr₂N coatings have higher hardness [18] and corrosion resistance [19] compared to CrN, and they improve the interlayer bond [20].

The kinds of multilayer nitride coatings have usually been prepared by magnetron sputtering (MS) and multi arc ion plating (AIP). The coatings obtained by MS [21] showed dense microstructure, but low deposition rate and adhesion, while those by AIP [1] had high deposition rate and adhesion, but columnar microstructure embedded many droplets, which may easily cause pitting corrosion. The combination of the MS and AIP is a promising route to overcome the shortcomings of each technique [22]. In recent years, high-power pulsed magnetron sputtering (HiPIMS) aims to obtain the highly dense microstructure of the films by supplying high ionization of the target through an electric power source with high voltage and current in a pulsed mode within several microseconds [23,24]. HiPIMS combined with DC magnetron sputtering has been successfully applied to deposit monolayer TiN coatings with clear and dense nanocrystalline structure and excellent passivation to improve corrosion resistance [25]; however, the deposition rate and quality of the coatings still need to be improved. Atomic layer deposition (ALD) can deposit dense, uniform and conformal oxide films with controllable thickness, but the low deposition rate limits its application. Kong et al. [26] prepared CrN/TiO₂/CrN coatings by HiPIMS combined with ALD technology, which can take into account the advantages of both. Until now, there are few reports about HiPIMS hybrid AIP technology on anti-corrosive properties of hard coatings.

Among the diverse deposition parameters, such as target power, target current, temperature, and bias voltage [27–31], negative bias is a promising key factor to improve the mechanical and anti-corrosive properties by increasing the energy of the sputtered ions. Lin et al. [27] reported the microhardness of the TiN multilayer coatings those prepared by AIP increased with increasing the bias voltage up to 900 V. Lv et al. [29] deposited CrAlN films by nonequilibrium magnetron sputtering, revealing that the preferred orientation changed from (220) to (200) and the deposition rate decreased as V_s increased up to -250 V. Meanwhile, TiAlSiN nanocomposite coatings were deposited by HiPIMS, showing that with increasing bias voltage up to -150 V, the microstructure changed from columnar to equiaxed, grain size and hardness increased, whereas the adhesion decreased from HF2 to HF5 [30].

Therefore, it is important to study the effect of bias-induced high-energy ions on the performance of Cr₂N/TiN multilayer coatings to adapt to the various application conditions. In this work, the effect of bias voltage on the microstructure, nano-hardness, adhesion and corrosion resistance were investigated, which aims to prepare Cr₂N/TiN multilayer coatings under various V_s by the HiPIMS/AIP hybrid technique and study the effect of V_s on the mechanical and anti-corrosive properties of the coatings.

2. Materials and Methods

2.1. Preparation of Cr₂N/TiN Multilayer Coatings

HiPIMS/AIP hybrid technique was constructed to deposit Cr₂N/TiN multilayer coatings on AISI 304 stainless steel and Si (100) substrates by using chromium and titanium targets. As shown in Figure 1, chromium and titanium planar targets ($484 \times 81 \times 12$ mm³) with 99.95% in purity are mounted opposite to each other at a distance of 600 mm (cavity

diameter) for HiPIMS. The chromium circular target ($\phi 124 \times 16 \text{ mm}^3$) with 99.95% in purity for AIP was mounted on the opposite side of the view glass at an associate angle of 90 degrees to the planar target. Before deposition, the sample would be mechanically ground and polished to make the sample surface achieve mirror effect, then it were ultrasonically cleaned with chemical reagents such as acetone and anhydrous ethyl for 20 min before deposition. The substrate was placed within the chamber at a distance of 160 mm from the planar target and 250 mm from the circular target. The revolution period of the substrate holder is 20 s and the rotation period is 5.7 s. The chamber pressure was vacuumed up to $8.0 \times 10^{-3} \text{ Pa}$ by using mechanical pumps.

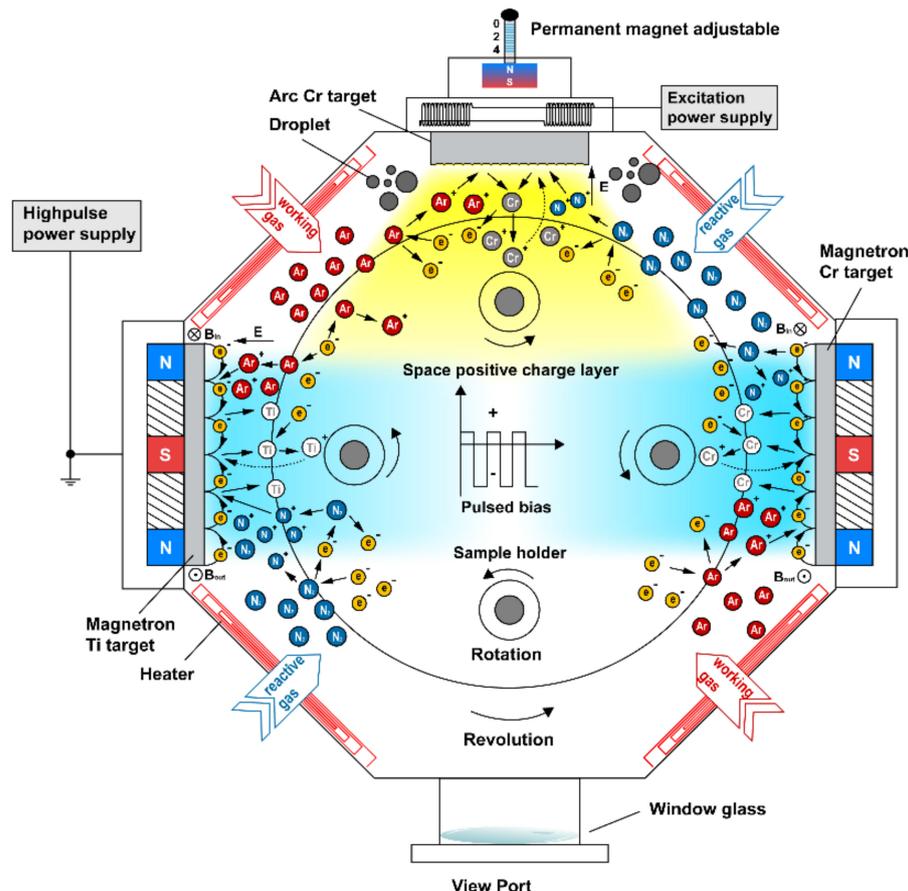


Figure 1. Scheme of stage to pulsed magnetron sputtering and arc ion plating processes.

The substrate surface was sputtering with argon ions to remove surface contaminants for 20 min by applying bias of -700 V to the substrate. The Cr interlayer was first deposited by AIP. $\text{Cr}_2\text{N}/\text{TiN}$ multilayers were alternately composed of the Cr_2N sublayers by the co-deposition of HiPIMS and AIP, and the TiN sublayers by AIP solely under the Ar and N_2 atmosphere. The planar targets are powered by a PDCMS power supply using a TruPlasma DC 4010 power providing with a connected electronic oscilloscope, which could gain the pulsed waveforms of voltage and current imposed to the target and substrate. The discharge voltage of the HiPIMS planar targets gradually stabilized at average voltage (-860 V) from peak voltage (-1500 V) and the target current density is $0.73 \text{ A}\cdot\text{cm}^{-2}$, as shown in Figure 2. $\text{Cr}_2\text{N}/\text{TiN}$ multilayer coatings were deposited with 4 kW Cr planar target power and 130 A Cr circular target current together with 4 kW Ti planar target power, with various V_s ranging from -30 to -150 V . Detailed parameters are shown in Table 1. All coatings were deposited using the same HiPIMS mode with a pulse width of $100 \mu\text{s}$, a frequency of 300 Hz, and a duty cycle of 3%. During each pulse start, the HiPIMS glow causes the bias current to increase and therefore the number of arcing to extend. The bias power provide detects the arc and quickly extinguishes it, that in turn causes the bias power

provide to stop working and also the bias voltage to gradually decrease. The operating pressure was maintained at 0.5 Pa throughout all depositions and the nitrogen flow was 33.3% of the total gas flow (30 sccm).

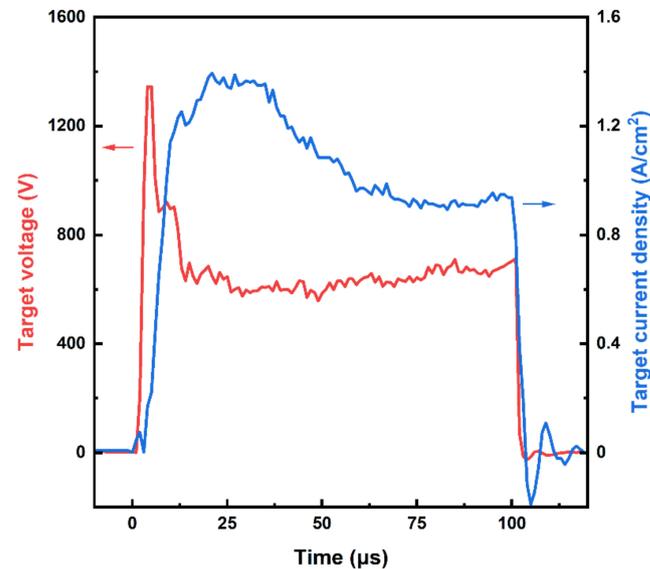


Figure 2. The target voltage and target current density for 100 μs HiPIMS pulse used for $\text{Cr}_2\text{N}/\text{TiN}$ multilayer coatings depositions.

Table 1. Experiment parameters of the $\text{Cr}_2\text{N}/\text{TiN}$ multilayer films deposited by HiPIMS and AIP hybrid technology.

Parameter	Values
Target materials (99.9 at.%)	Cr, Ti
Target-substrate distance of HiPIMS (mm)	160
Target-substrate distance of AIP (mm)	250
Operating pressure (Pa)	0.5
Process temperature ($^{\circ}\text{C}$)	350
Nitrogen partial pressure (Pa)	0.15
Depositing time of CrN/TiN multilayer (min)	120
Depositing time of CrN and TiN sublayers (min)	5/15
Duty cycle (%)	3
DC bias voltage, V_s (V)	−30, −60, −100, −150
Pulse frequency (Hz)	300
Pulse on-time/off-time (μs)	100/3233
Substrate rotation (rpm)	3.0

2.2. Characterization of Coatings

2.2.1. Morphological Characterization

The crystal structures of as-deposited multilayer coatings were characterized by an Empyrean-type X-ray diffraction (XRD, PANalytical B.V., Malvern, UK) scanning from 20° to 90° at $4^{\circ}/\text{min}$ in steps of 0.02° with Cu $K\alpha$ radiation ($\lambda = 0.154 \text{ nm}$). The residual stresses of $\text{Cr}_2\text{N}/\text{TiN}$ multilayer coatings were estimated by $\sin^2\psi$ method [32], which is using a D8-Discover grazing incident X-ray diffraction (GIXRD). The microstructures of the coatings were examined by a GeminiSEM 300 field emission scanning electron microscope (FESEM) from Zeiss, Germany, and a Talos F200S field emission high resolution transmission electron microscope (HRTEM) with 200 kV from FEI, Houston, TX, USA, respectively, and then investigated by the fracture cross-sectional TEM as well as line-scan compositional analyses using the integrated energy dispersive X-ray spectroscopy (EDS, Houston, TX, USA). The SEM surface topography of the sample is processed by software to

obtain the droplet distribution histogram. In order to ensure the reliability of the data, five coating surface photos under 1000 magnification are randomly selected for each sample.

2.2.2. Mechanical Properties

The hardness (H) and elastic modulus (E) of the coatings were measured by TI-980 nano indentation instrument manufactured by Bruker-Hysitron (Billerica, MA, USA). The load was 5 mN, the loading time was 5 s, and the indentation depth was always kept below 10% of the film thickness to minimize the substrate effect. The hardness (H) and effective modulus (E^*) were determined by nanoindentation, and the elastic modulus (E) was calculated by the equation $E^* = E/(1 - \nu^2)$, wherever ν is the Poisson's ratio [22,33]. By taking the average value of Cr_2N Poisson's ratio 0.22 and TiN Poisson's ratio 0.25, it is estimated that the Poisson's ratio of $\text{Cr}_2\text{N}/\text{TiN}$ multilayer coatings is 0.235. Rockwell tests were conducted according to ISO 26443, classifying the strength from HF1 (high adhesive strength) up to HF6 (very low adhesive strength). Rockwell indentation with a C tip (200 μm radius) under a 150 kg load for 15 s was applied to evaluate the adhesion strength.

2.2.3. Anti-Corrosive Properties

The corrosion protection performance of the coatings was tested using an electrochemical workstation CHI660E manufactured by Shanghai Chenhua Instruments Co (Shanghai ChenHua instrument technology Co., Ltd., Shanghai, China). The instrument uses a three-electrode system, with the thin film as the working electrode and the counter electrode as a platinum sheet, and all potentials are related to the saturated calomel electrode (SCE). The pure NaCl reagent and deionized water were used to form a 3.5 wt.% NaCl aqueous solution as the test medium. The multilayer coatings with an exposed area of 1 cm^2 were pressed tightly into the solution at a constant temperature of 25 $^\circ\text{C}$ and the relevant tests are performed sequentially. After 60 min of open circuit potential (OCP) testing, electrochemical impedance spectroscopy (EIS, Shanghai ChenHua instrument technology Co., Ltd., Shanghai, China) was performed at an AC voltage amplitude of 10 mV in the frequency range from 0.01 Hz to 100 kHz. After EIS measurements were repeated three times, dynamic potential polarization test was operated with a sweep rate of 0.001 V/s in the anode direction from -0.5 to 0.5 V. In order to ensure the reliability of the data, all sample tests were carried out in accordance with the above steps.

3. Results and Discussion

3.1. Crystalline Structure and Microstructure of Coatings

Figure 3 shows the crystallographic structures of as-deposited $\text{Cr}_2\text{N}/\text{TiN}$ multilayer coatings at different bias voltages. Metal cubic Cr and hexagonal Cr_2N was identified. The Cr phase was soft, while the Cr_2N phase was hard. The peak near $2\theta = 42.5^\circ$ corresponds to the hcp structure of Cr_2N (111) phase and the relative intensities of Cr_2N phase varies greatly. With the decrease of V_s , the diffraction pattern of Cr_2N with (111) orientation increases, which indicates that a clear β -structure was obtained during these cases [34]. In addition, the compressive stress increases during film deposition because the peak corresponding to hexagonal phase Cr_2N (111) shifts to the reference value at a small angle with the bias, which results in the increase in defect concentration [35]. The results show that at the beginning of CrN_x deposition, the amount of nitrogen is not enough to produce stoichiometric CrN [36]. According to the GIXRD pattern, we use $\sin^2\psi$ means to measure and obtain the residual stress state. As V_s increases from -30 to -150 V, the compressive residual stress are -1.32 , -3.68 , -4.80 , and -6.97 GPa, respectively, which could explain how the (111) peak moves to a lower angle. The increase of ion bombardment promotes the formation of defect concentration, which leads to the increase of residual compressive stress in the coating. With the increase of V_s , these phases are rich in Cr. The TiN sublayer is only 20 nm according to Figure 4, so that it was not detected by X-ray diffractometer. As the V_s increases to -100 V, the average crystallite size decreases from 9.4 to 8.5 and 7.5 nm,

but at -150 V, the average grain size increases to 11.2 nm, which is calculated by Scheler's formula [33].

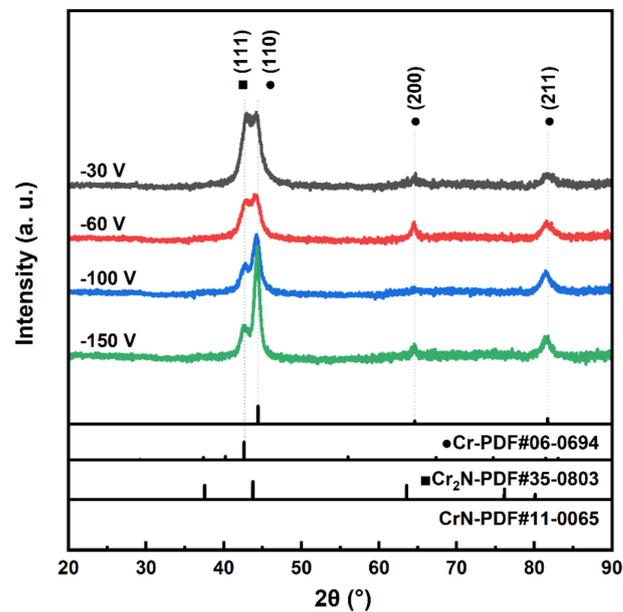


Figure 3. Effect of V_s on crystalline phase.

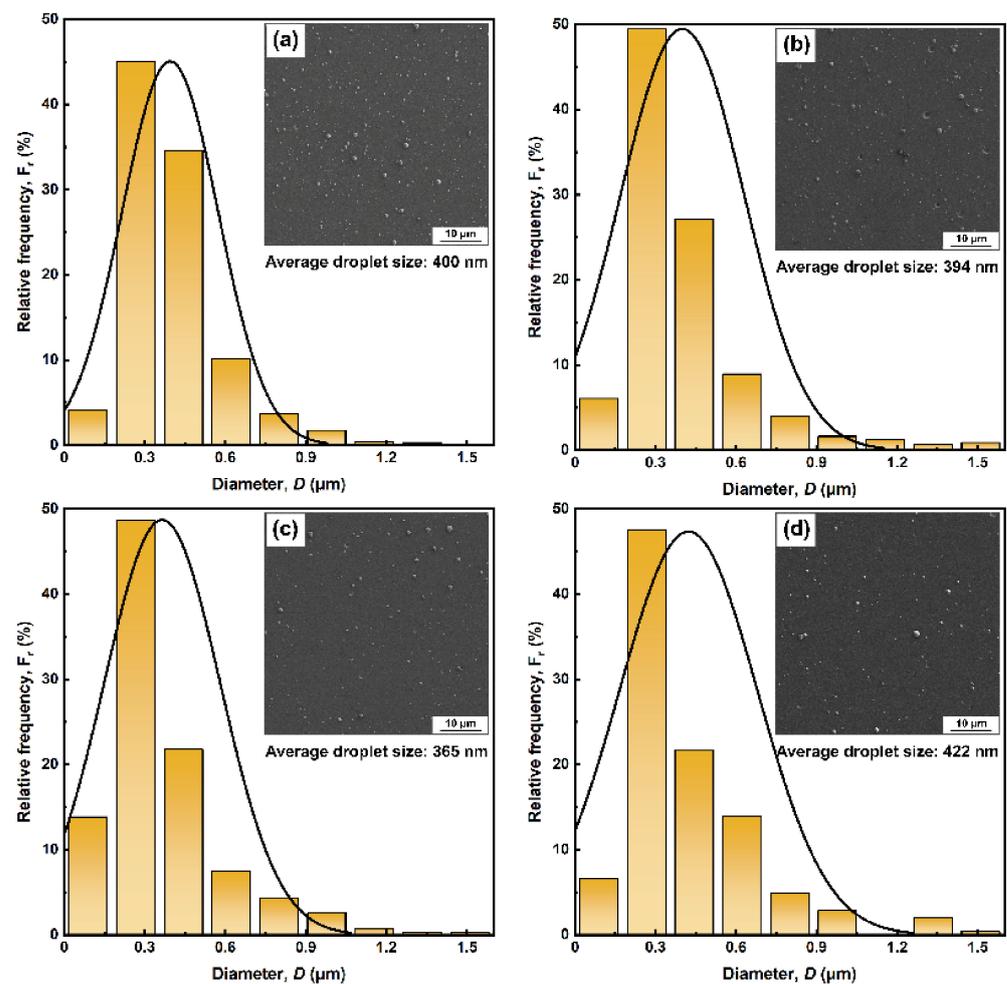


Figure 4. The surface SEM micrograph of the Cr₂N/TiN multilayer coatings and its particle size distribution histogram. (a) -30 V, (b) -60 V, (c) -100 V, and (d) -150 V.

Figure 4 shows the surface SEM micrographs of Cr₂N/TiN multilayers at different V_s and the corresponding histograms of the particle size distribution. With the increase of the V_s , the large particles, i.e., droplets by AIP, gradually decreased from 400 nm at $V_s = -30$ V, 394 nm at $V_s = -60$ V and 365 nm at $V_s = -100$ V. The droplet size becomes large, 422 nm at $V_s = -150$ V. The appropriate ion bombardment could optimize the coating surface; however, with the increasing bias, the further amplified ion bombardment could generate the surface defects.

Figure 5a shows the cumulative distribution curves of particle size on the surface morphology in Figure 4. As the bias voltage increases from -30 to -100 V, the cumulative distribution curve shifts to the left and the large particle size is refined, showing that nearly 80% of the particle size is less than 0.5 μm . The particle size accumulation curve shifts flat to the right and only 63% of the droplet is less than 0.5 μm at $V_s = -150$ V. The inset in Figure 5a shows the first-order differentiation of the cumulative distribution curve with two peaks at 0.62 and 1.39 μm at $V_s = -150$ V, indicating larger bias voltage may lead to larger droplets. Figure 5b shows the ratio of the area occupied by the particles on the coating surface. The droplet area on the sample surface is the smallest at $V_s = -100$ V, which is consistent with the SEM surface morphology.

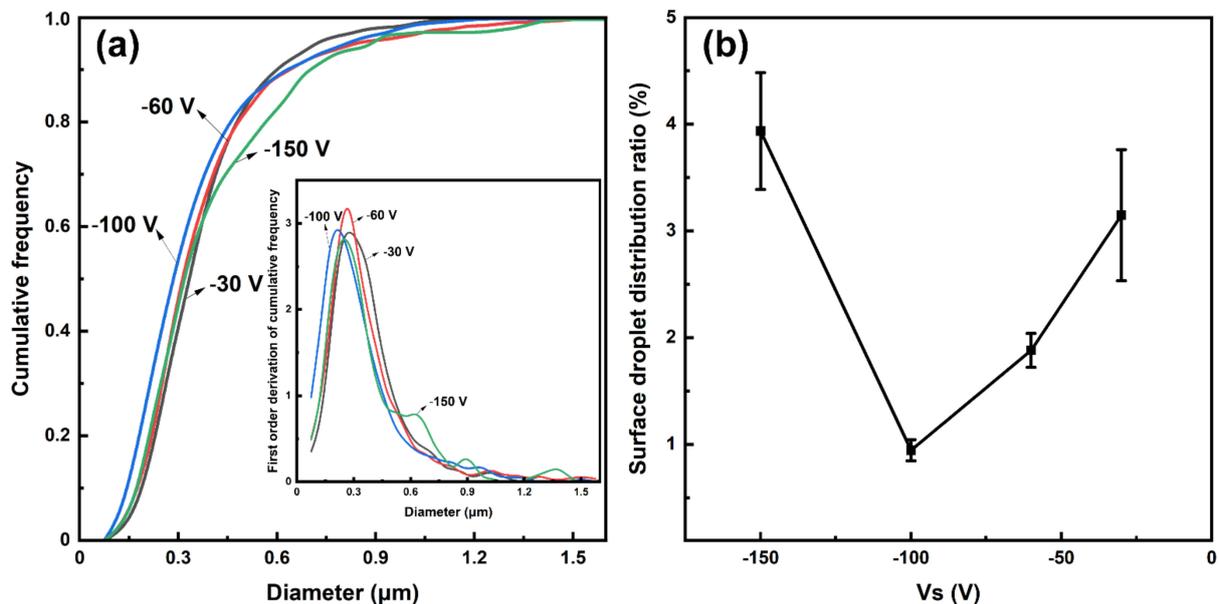


Figure 5. (a) The cumulative particle size distribution curve, and (b) the droplet distribution area on the surface of the sample.

Figure 6 shows the potential and current density of the substrate during the pulse of bias and HiPIMS based on different bias voltages, with a bias operation time of 10 μs , a frequency of 60 kHz, and a duty cycle of 60% . The bias voltages can increase the energy of charged particles in the vacuum plasma, and the energetic particles excited from the target surface bombard the substrate to clean it, thus improving the adhesion of the subsequent deposition film. Meanwhile, it can purify the large particles in AIP. When HiPIMS glowed, the bias voltage remained constant, but the bias current increased and the probability of arcing increased, making the bias-detected arc turn off, which led to the low deposition rate by using HiPIMS solely. During the deposition of Cr₂N layer by HiPIMS/AIP hybrid technique, when negative bias turned on, cation glow onto samples making positive current density increased suddenly; when negative bias turned off, electron glow onto samples making negative current density decreased. When HiPIMS turns on for 100 μs , the potential and the current on the targets increased, making the sample holder current density (I_H) increase because much more cation and electrons formed, but the negative bias was forcibly and gradually decreased to avoid forming an arc on the samples due to the high voltage of high power on targets plus bias on sample holders; when HiPIMS was turned off, the

bias increased gradually and exceeded the preset bias voltage, finally recovering to the preset ones. With increasing V_s from -30 to -100 V, the I_h increased from 5 to 7.5 mA/cm² and all the sample holder potential (P_h) decreased to about -10 V when HiPIMS turns on. The higher P_h and the higher I_h provides the cations' higher moving energy, resulting in the denser microstructure and less droplets. On the other hand, when bias was preset at -150 V, the I_h increased to 10 mA/cm² and the P_h only decreased to -50 V when HiPIMS turns on, and then increased to over -220 V. The overloaded voltage and current may lead to more and larger droplets, which is consistent with the results in Figure 5.

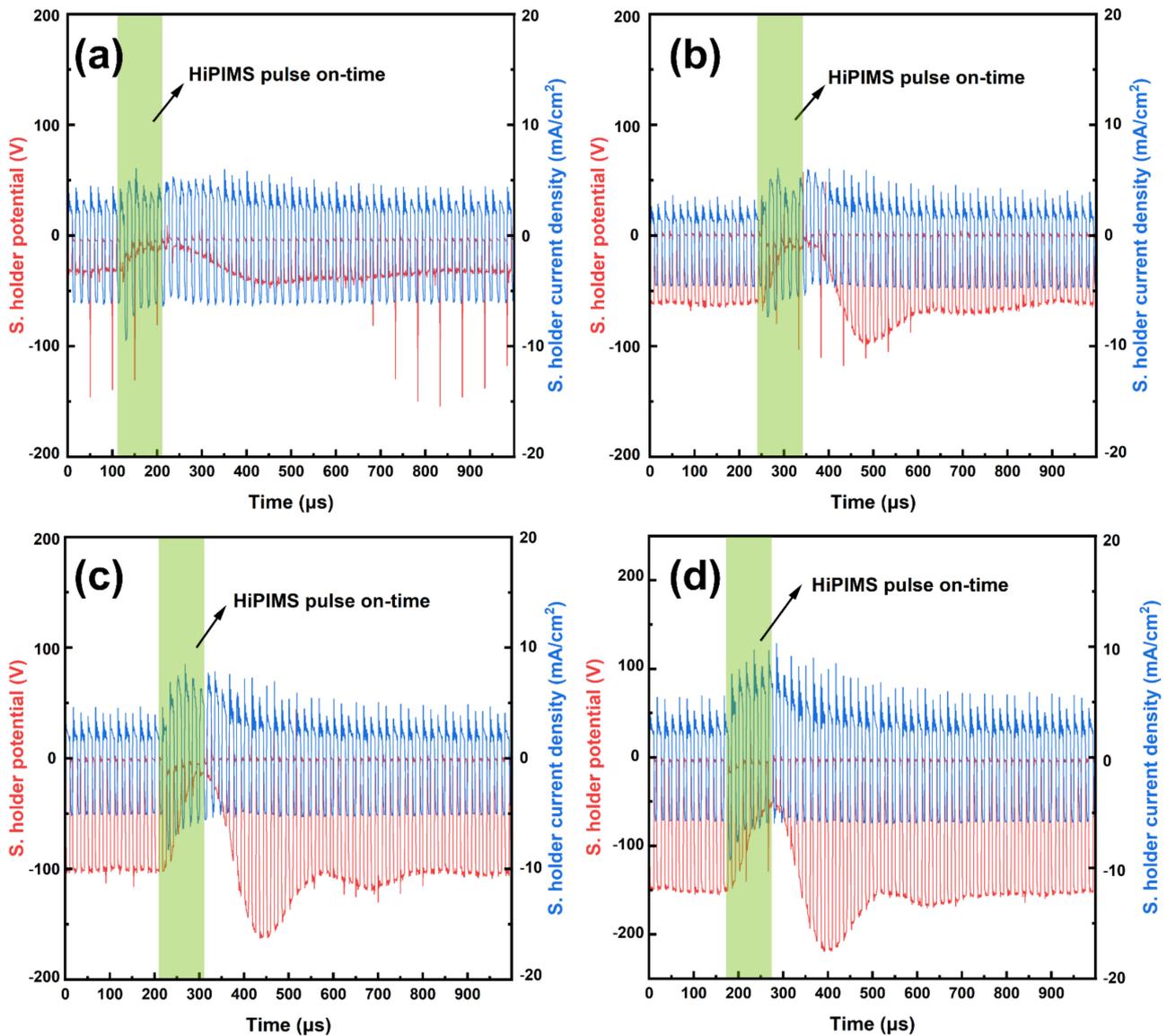


Figure 6. The substrate potential and substrate current density with $V_s = -100$ V. (a) -30 V, (b) -60 V, (c) -100 V, and (d) -150 V.

Figure 7 shows the cross-section of coatings deposited at various V_s under Secondary Electron Images (SEI) mode. The Cr transition layer with similar thickness can be observed in the coatings deposited under different V_s . This is consistent with the deposition parameters of the transition layer in the four groups of experiments, as shown in Table 1. Figure 7 also shows that the Cr transition layer is dense and uniform without obvious microcracks and hole defects, which can improve the interface bonding and have good bonding force with the substrate. At $V_s = -30$ V, the coatings are dense, and the thickness is only 1.0 μm . With increasing V_s from -30 to -100 V, the cross-section became denser, avoiding the for-

mation of columnar crystals. Therefore, HiPIMS and AIP hybrid technique has an obvious effect on the refinement of the coatings [37]. The bias applied to the substrate significantly increased the ionization and energy of the sputtered species, and then enhanced the film growth [28].

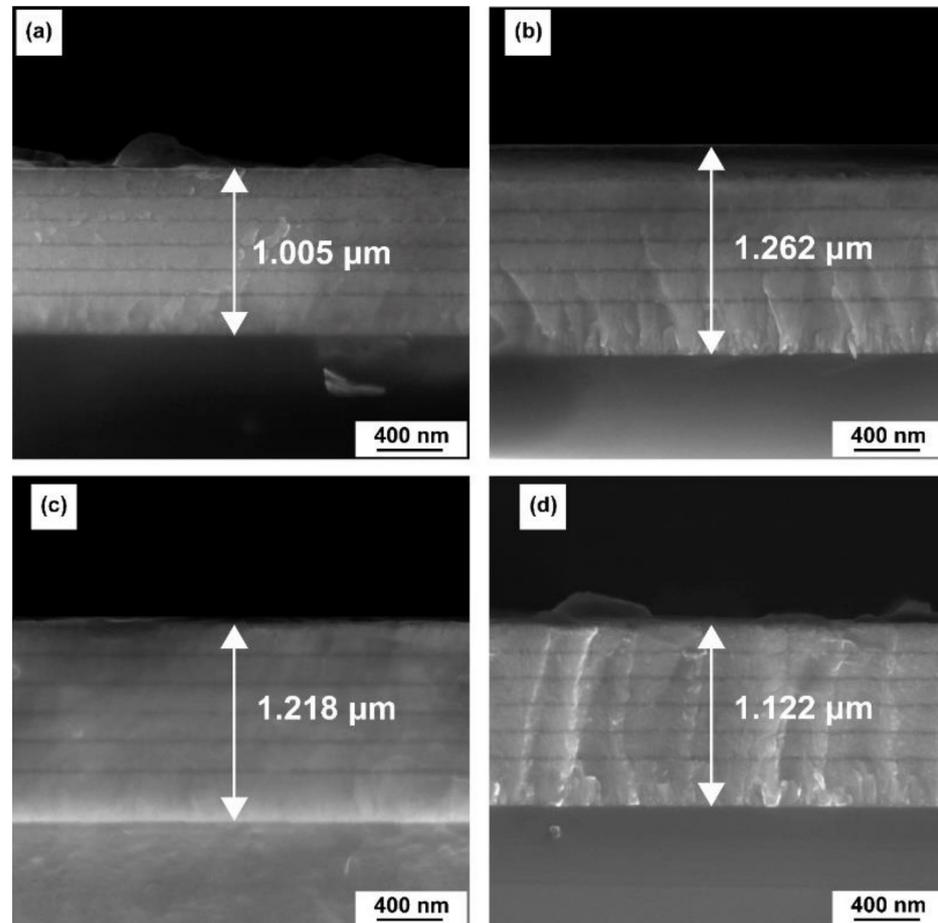


Figure 7. Cross-section of coatings deposited at various V_s under Secondary Electron Images (SEI) mode. (a) -30 V, (b) -60 V, (c) -100 V, and (d) -150 V.

Figure 8 shows the cross-sectional TEM microstructure and elemental distribution of the $\text{Cr}_2\text{N}/\text{TiN}$ multilayer coating prepared at $V_s = -100$ V in bright-field images. The coating thickness is about 1.2 μm with alternating Cr_2N and TiN sublayers. A flat interface was observed in between the Cr, Cr_2N and TiN sublayers with the thickness of 158, 156 and 28 nm, respectively. Figure 8c shows the line sweep of the cross-sectional coating, where the N content in the Cr_2N sublayer is much lower than that in the TiN sublayer. Cr was sputtered by both of HiPIMS and AIP while Ti was sputtered by HiPIMS solely, resulting in the relative concentration of N atoms during the deposition of Cr_2N is much lower than that during the deposition of TiN to form the Cr_2N [38]. Figure 8d shows the high-resolution image of the interface between sublayers. It can be seen that the crystal orientation of the TiN sublayer is (200) preferred orientation, and the metal phase Cr (200) with the same orientation is at the interface with the Cr_2N sublayer. The TiN sublayer plays a certain role in epitaxial growth, and the formed metal Cr layer is conducive to improving the adhesion of coatings [39]. Figure 8e,f show the high-resolution images of Cr and Cr_2N sublayers and the corresponding SAED pattern, respectively. The lattice spacing obtained by Fourier transform shows that the sublayer is a clear mixed phase crystal structure of Cr and Cr_2N layers, which is consistent with XRD results.

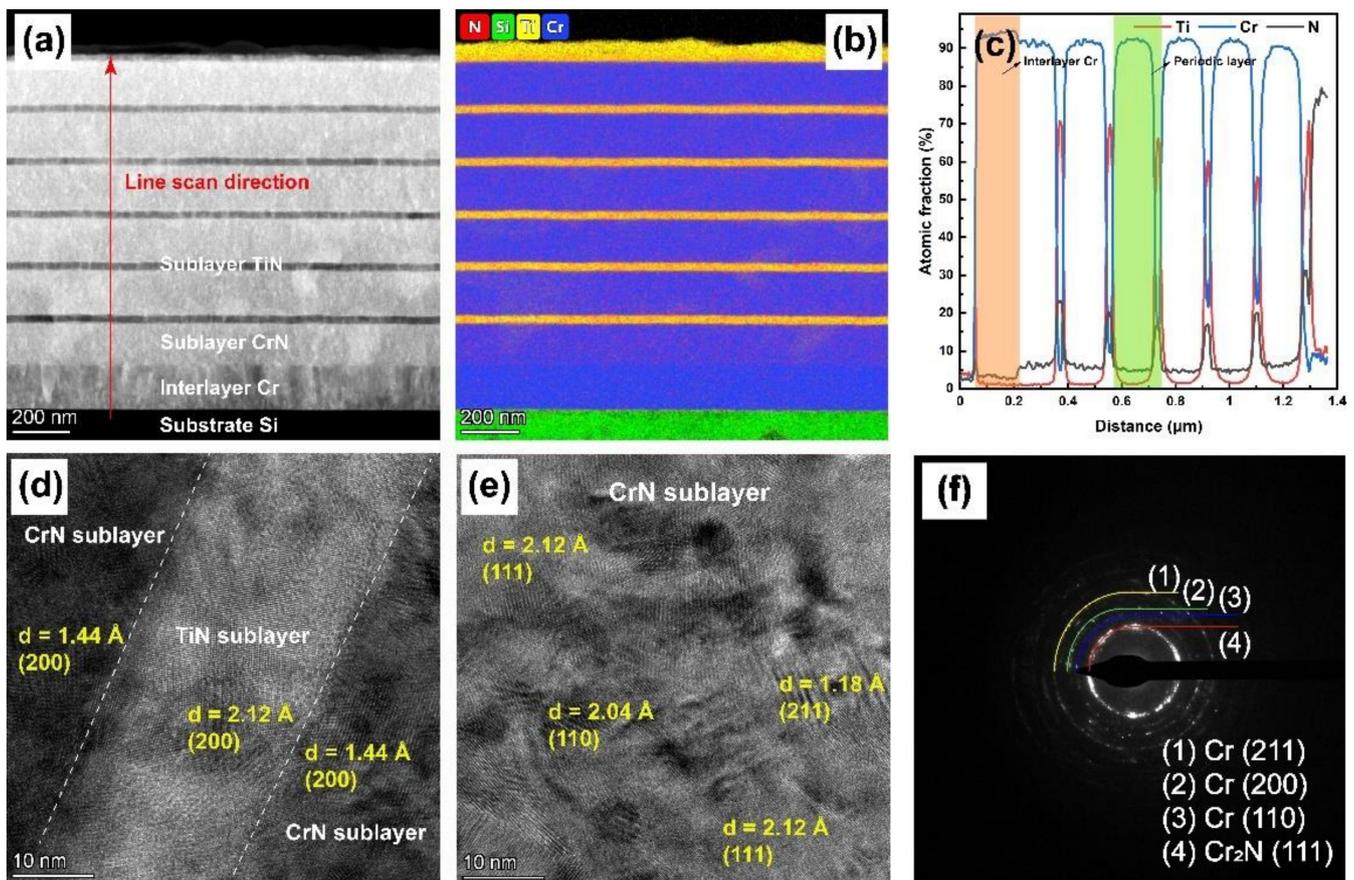


Figure 8. Cross-sectional TEM micrographs (a), elemental mapping (b), line sweep (c), high-magnification of interface (d) and Cr₂N sublayer (e) and selected area diffraction patterns (f).

3.2. Mechanical Properties of Coatings

The mechanical properties, e.g., hardness, elastic modulus and adhesive strength, of the Cr₂N/TiN multilayer coatings were characterized by nanoindentation and Rockwell indentation. Figure 9 shows the nanoindentation loading curves of the Cr₂N/TiN multilayer coatings at various V_s . As shown in Figure 9a, the typical loading-unloading curve of the Cr₂N/TiN coatings deposited with different V_s . The elastic recovery (W_e), which was defined as the ratio of the part of the indentation depth that could be recovered to the maximum indentation depth [40]. As shown in Figure 9b, the indentation depth is inversely proportional to the energy recovery ratio under certain conditions. The inflection point happens at $V_s = -100$ V, and the indentation depth is 75.6 nm with an energy recovery ratio of 61%.

Figure 10 shows the hardness and elastic modulus of samples prepared at various V_s . As shown in Figure 10a, H increased from 21.5 to 25.1 GPa as V_s increased from -30 to -100 V. With further increase in V_s , H and E decreased. The rise in hardness values of samples may lead to the clear interface, compact microstructure, reduced grain size, and increased compressive residual stress [39,40]. Figure 10b shows the H/E^* and H^3/E^{*2} ratios for the Cr₂N/TiN multilayer coatings. H/E^* and H^3/E^{*2} are used to represent resistance to elastic strain damage and resistance to plastic deformation, respectively, which is related to the toughness of the coating [22,26,33,41]. The error of H^3/E^{*2} is less than 0.0001, so the error bar can be ignored. As V_s increases from -30 to -100 V, the H/E^* and H^3/E^{*2} ratios increased from 0.060 to 0.072, and 0.078 to 0.128 GPa, respectively, showing the highest values at $V_s = -100$ V, indicating the highest toughness owing to its clear interfaces, fit grain size and compact microstructure.

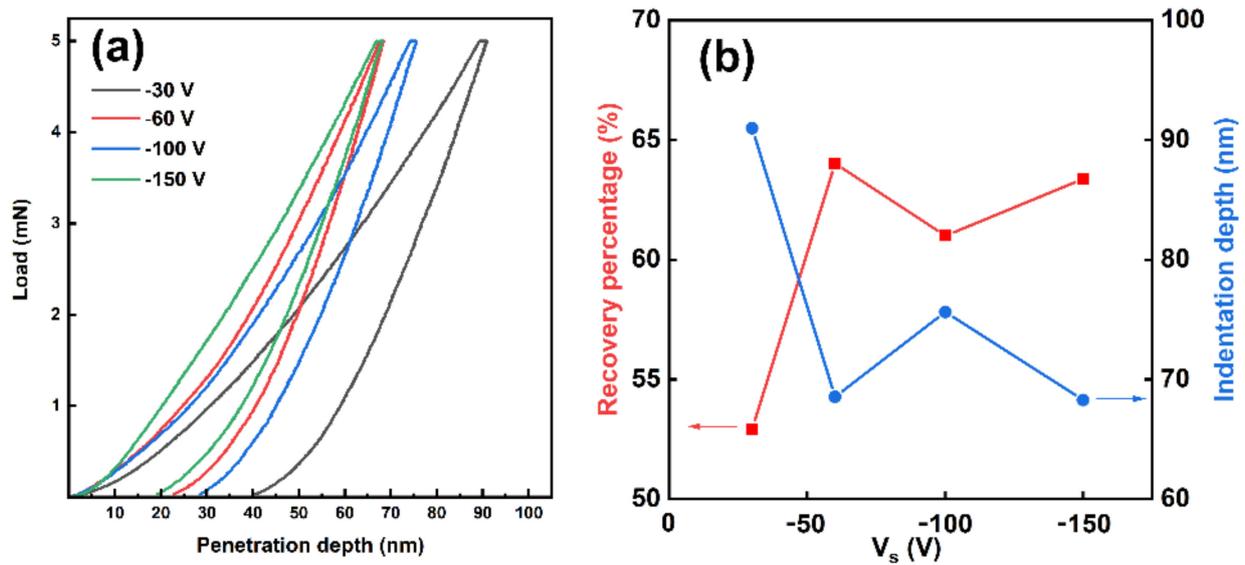


Figure 9. (a) Load–displacement curves with various V_s and (b) corresponding to recovery percentage and indentation depth.

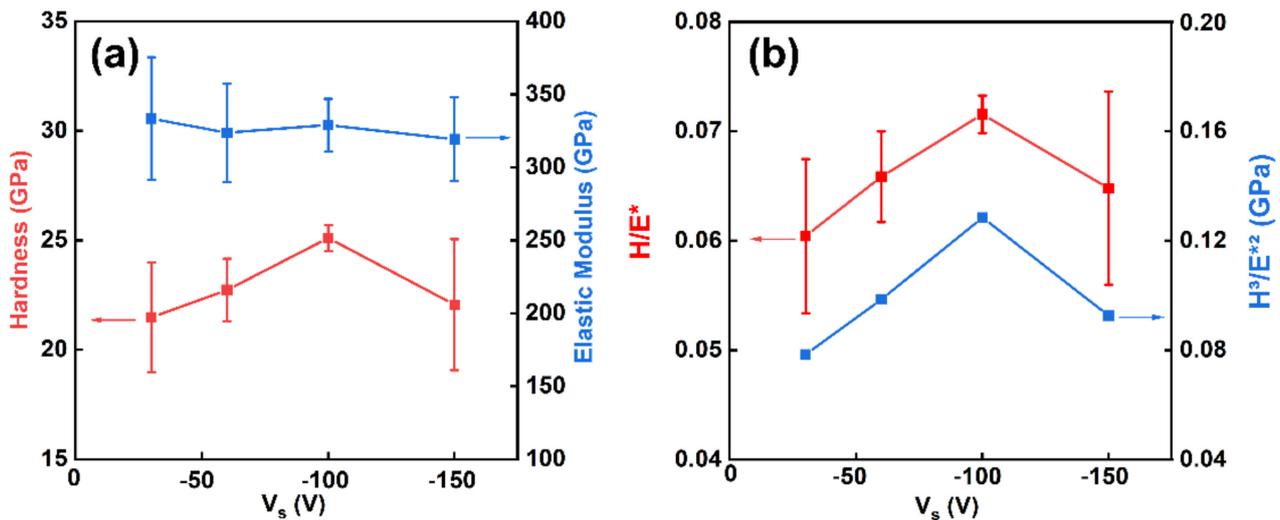


Figure 10. (a) Hardness and elastic modulus and (b) H/E^* and H^3/E^{*2} ratio of coatings deposited at various V_s .

Figure 11 shows the morphology of the Rockwell indentation to evaluate the adhesion of coatings and substrate. When the bias is -30 V, as shown in Figure 11a, a small amount of coating starts to peel off around the indentation, indicating the adhesion of HF3. At $V_s = -60$ and 100 V, as shown in Figure 11b,c, there are only a few cracks and no peeling around the indentation, indicating the adhesion of HF2 and HF1, respectively. Obviously, the H/E^* and H^3/E^{*2} ratios and compressive residual stress in the tested samples would affect the change of the adhesion with bias [22]. The trend of adhesion level was similar to that of the H/E^* and H^3/E^{*2} ratios. With V_s up to -150 V, obvious cracks begin to peel off, or even worse. This is because one of the reasons for the decrease of adhesion is the high compressive residual stress inside the coating [23]. Therefore, the enhancement of the adhesion of the coatings benefits from the complex impact of high H/E^* and H^3/E^{*2} ratios and appropriate compressive residual stress.

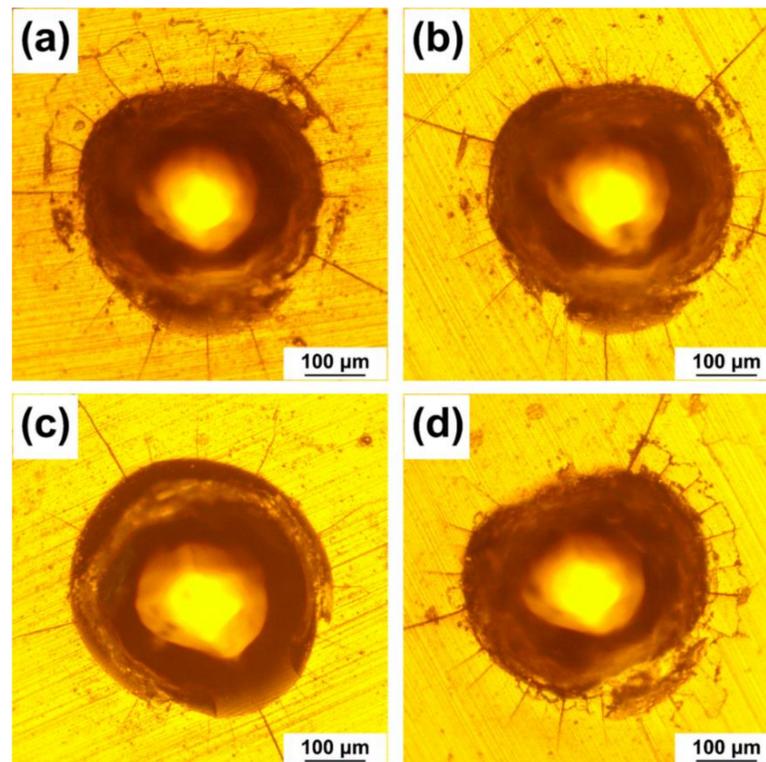


Figure 11. Rockwell indentation images made with 150 kg load for 15 s in Cr₂N/TiN coatings deposited by various V_s (a) -30 V, (b) -60 V, (c) -100 V, and (d) -150 V.

3.3. Anti-Corrosive Properties of Coatings

The OCP and kinetic potential polarization curves were investigated in 3.5 wt.% NaCl aqueous solution to investigate the corrosion behavior of the coated samples. Figure 12 shows the OCP of the Cr₂N/TiN multilayer coatings deposited at various V_s . The OCP rapidly decreased within the first 15 min and then tends to be stable. At $V_s = -30, -60, -100$ and -150 V, the stable OCP were $-0.0946, -0.0425, 0.0237$ and -0.067 V, respectively, representing the best corrosion resistance at $V_s = -100$ V. Figure 13 shows the dynamic potential polarization curves of Cr₂N/TiN multilayer coatings deposited at various V_s . Under different bias conditions, the E_{corr} includes a clear tendency to move to the positive direction, and then the I_{corr} decreased, suggesting that proper bias can enhance corrosion resistance. The coated samples exhibited highest corrosion potential of $E_{\text{corr}} = -60$ mV (SCE) and lowest corrosion current density of $I_{\text{corr}} = 3.57 \times 10^{-8}$ A·cm⁻² at $V_s = -100$ V. The coating experienced the process from self-passivation to over-passivation, and no craters were observed on the coating surface. The corroded surface of the coating prepared at $V_s = -100$ V was the closest to that of the as-deposited coating, whereas some pitting corrosion was observed on the other samples under different bias. The passivation layer on the coating surface has been repaired, which can be seen from the fluctuation of corrosion current in the passivation area in Figure 13 [42]. Therefore, the corrosion rate is related to the microstructures; the smaller grain size and the less droplets, the better corrosion resistance [43]. The polarization resistance (R_p) can be calculated according to Equation (1).

$$R_p = \frac{\beta_a \beta_c}{2.303 i_{\text{corr}} (\beta_a + \beta_c)} \quad (1)$$

where β_a and β_c are the anodic and cathodic slopes in the polarization curves. In Table 2, the coating shows a higher polarization resistance at $V_s = -100$ V, implying the enhanced corrosion resistance. In addition, the reasons for improving the pitting resistance include appropriate residual stress and adhesion of coating/substrate, whereas the corrosion resistance of the coating decreased at $V_s = -150$ V. This is because overtop bias will

enhance the energy of ion bombardment on the sample surface, resulting in surface defects and high residual stress in the film. the decrease in corrosion resistance of coatings prepared at $V_s = -150$ V is due to the introduction of surface defects and high residual stress by enhanced ion bombardment with the increase of bias.

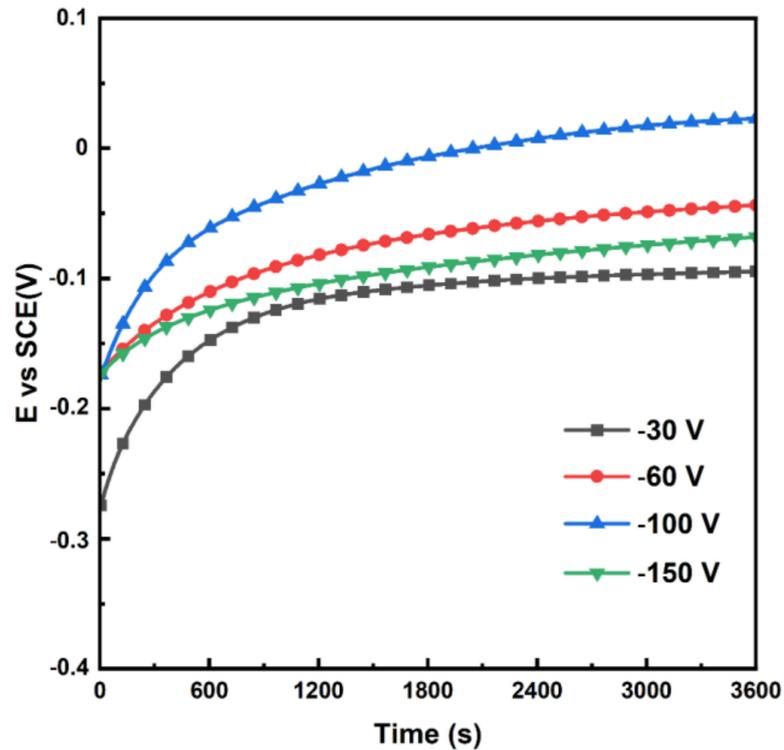


Figure 12. Open circuit potential of coatings deposited at various V_s .

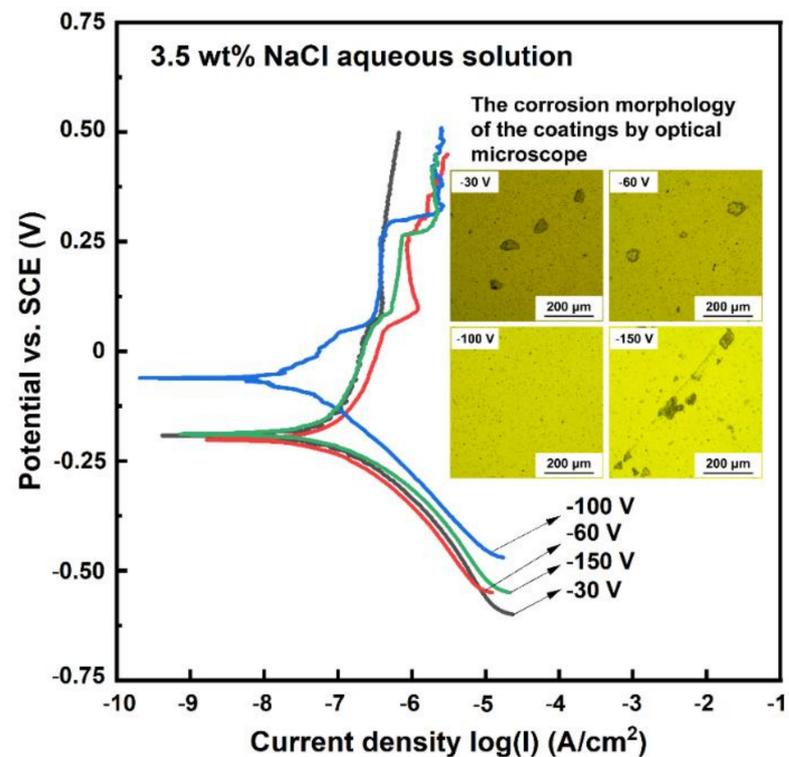


Figure 13. Potentiodynamic polarization curves of coatings deposited at various V_s .

Table 2. Corrosion properties obtained from the dynamic potential polarization curves in 3.5 wt.% NaCl aqueous solution.

V_s (V)	E_{corr} (V)	β_a (V/Decade)	β_c (V/Decade)	I_{corr} ($nA \cdot cm^{-2}$)	R_p ($k\Omega \cdot cm^2$)
-30	-0.19	0.33	0.12	72.76	525.17
-60	-0.20	0.24	0.12	94.51	367.55
-100	-0.06	0.11	0.13	35.73	724.10
-150	-0.19	0.30	0.11	84.65	412.87

Figure 14 shows the corrosion current density and corrosion potential of the coating in this study comparing with those in literatures in 3.5 wt.% NaCl aqueous solution [1,44–52]. The corrosion resistance of coatings prepared by AIP fluctuates more frequently than those by magnetron sputtering because of the defects caused by the large particle droplets, which may make the solution easily breaking through the coating. The Ti/AlTiN coating deposited by Tan et al. [45] through HiPIMS + APA – Arc dual technique has a high corrosion potential and low corrosion current, and has a high protection efficiency for the substrate. The composite technique can be further developed into thin film/coating deposition with unique features and applications. Ma et al. [51] deposited CrN/AlN nanomultilayer coatings by reactive magnetron sputtering, and the nano-multilayer structure has a good “sealing effect” by comparing with the single layer CrN coating, and it is difficult for corrosive solutions to corrode the substrate through the coating. The Cr₂N/TiN multilayer films prepared by HiPIMS/AIP hybrid technique in this study has advanced corrosion resistance, which is related to the dense structure of the coatings by adjusting the bias parameters.

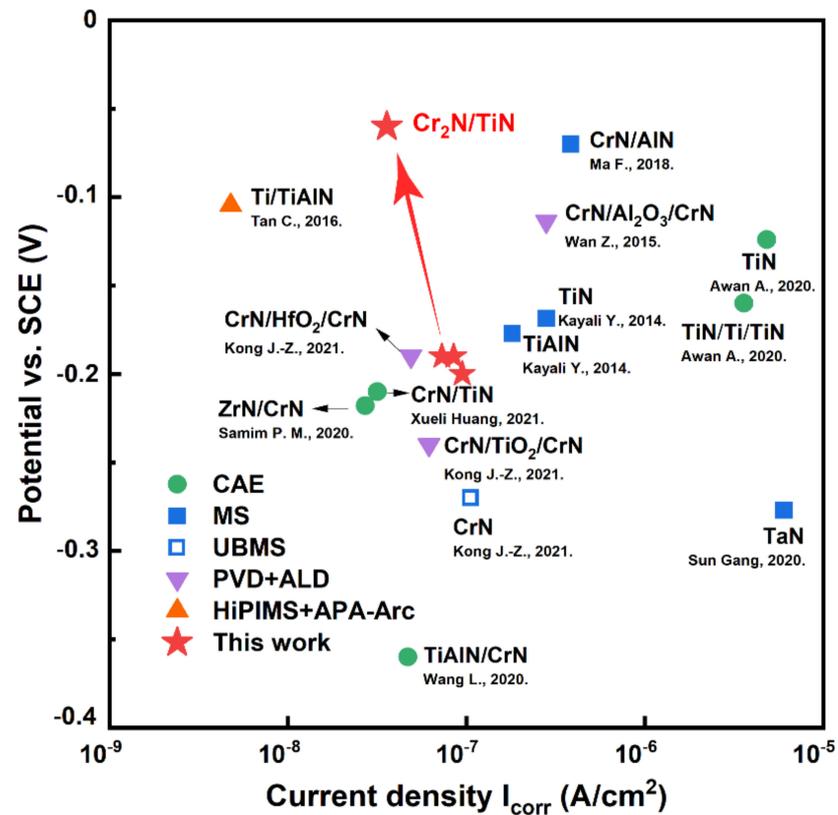


Figure 14. The investigation on corrosion resistance of different coating systems [1,44–52].

Figure 15 shows the Nyquist plots and the corresponding Porter plots of Cr₂N/TiN multilayer coatings deposited at various V_s . The coating prepared at $V_s = -100$ V in Figure 15a has the best curving radius of the capacitive arc and contains a considerably

improved corrosion impedance. Figure 15b shows a Bode plot of impedance phase versus frequency. The coating prepared at $V_s = -100$ V has a wider frequency vary with a phase of over 65° . The phase angles at low and high frequencies provide the impedance and capacitance of the coatings, respectively [2]. The EIS results demonstrate that the bias voltage can improve the corrosion resistance of the coatings, showing the best corrosion resistance at $V_s = -100$ V. The corresponding equivalent circuit was used to analyze the EIS spectrum as shown in Figure 15c. This equivalent circuit model consists of solution resistance (R_s), coating resistance (R_{coat}) and charge transfer resistance (R_{ct}), and the corresponding coating capacitance (CPE_{coat}) and double layer capacitance (CPE_{dl}) [46]. To replace the ideal capacitor, the constant phase element (CPE) parameter is used to describe the non-ideal capacitor. Its ohmic resistance is expressed by Equation (2).

$$Z_{CPE} = \frac{1}{Y_o(j\omega)^n} \tag{2}$$

where ω and n denote the angular frequency (rad/s) and the deviation from an ideal to a non-ideal capacitor, respectively. In Table 3, the R_{ct} of the coatings is relatively high at $V_s = -100$ V, indicating that it provides a lower charge transfer rate. As mentioned above, the bias voltage at -100 V affects the protection of defects such as grain refinement and voids, therefore improving the corrosion resistance of the coatings.

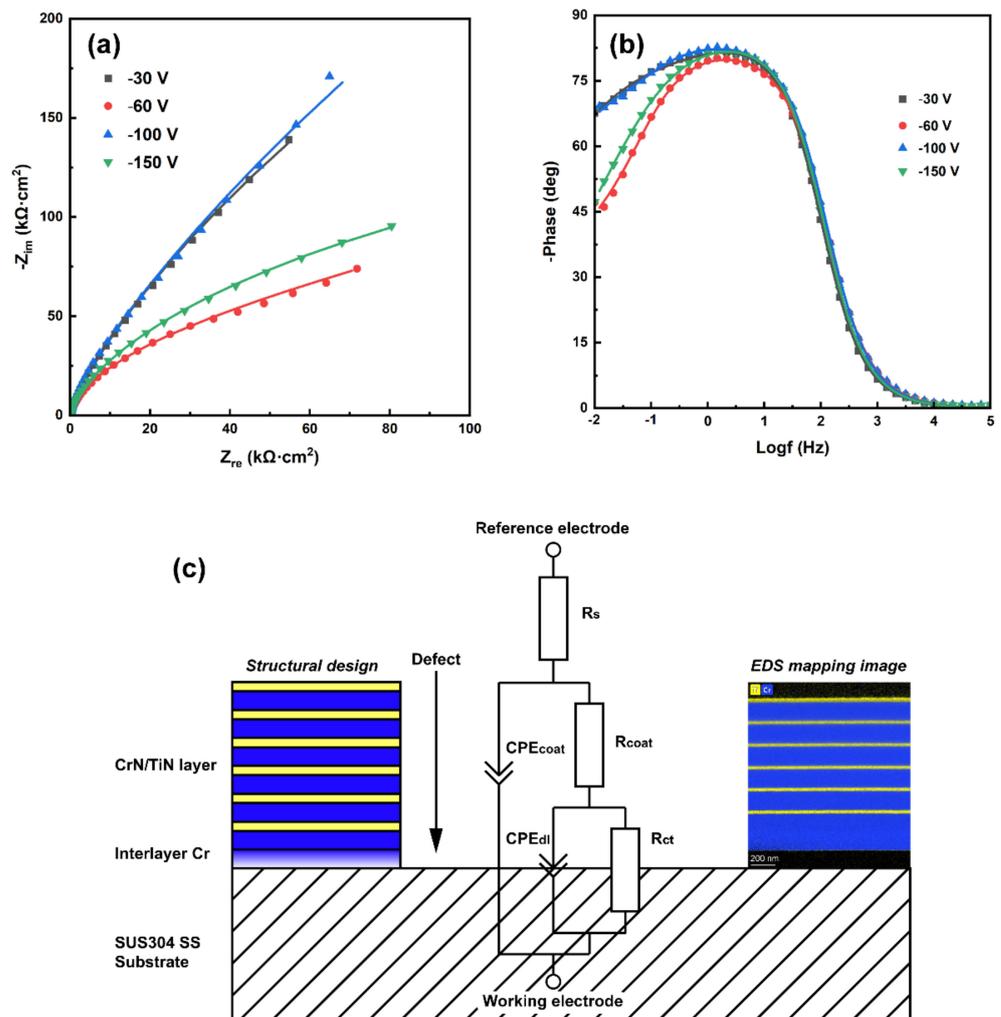


Figure 15. (a) Nyquist and (b) bode plots of the coatings at respective OCP in a 3.5 wt.% NaCl solution and (c) electronic equivalent circuit used in the fitting procedure of the EIS data.

Table 3. EIS fitting parameters of the samples.

V_s (V)	R_s ($\Omega \cdot \text{cm}^2$)	CPE_{coat}	-	R_{coat} ($\text{k}\Omega \cdot \text{cm}^2$)	CPE_{dl}	-	R_{ct} ($\text{M}\Omega \cdot \text{cm}^2$)
-	-	$\text{Yo/S} \cdot \text{cm}^{-2} \cdot \text{s}^n$	n	-	$\text{Yo/S} \cdot \text{cm}^{-2} \cdot \text{s}^n$	n	-
−30	43.68	5.09×10^{-5}	0.94	26.67	1.09×10^{-5}	0.56	2.18
−60	45.38	4.79×10^{-5}	0.92	81.08	5.62×10^{-5}	0.64	0.32
−100	43.96	4.39×10^{-5}	0.93	158.82	1.36×10^{-5}	0.68	1.55
−150	45.75	4.48×10^{-5}	0.94	47.66	1.81×10^{-5}	0.47	0.57

4. Conclusions

$\text{Cr}_2\text{N}/\text{TiN}$ multilayer coatings with an alternate Cr_2N and TiN sublayers were obtained at various bias by the HiPIMS/AIP hybrid technique to provide dense microstructure by HiPIMS and high deposition rate by AIP simultaneously. The Cr_2N thick layer co-deposited by HiPIMS and AIP was separated by the TiN thin layer deposited by HiPIMS solely. With the increase of V_s , Cr (110) gradually presents a preferred orientation, while Cr_2N (111) phase is on the contrary, which means that there is an optimal ratio between the metal cubic Cr and hexagonal Cr_2N in the CrN sublayer. At the same time, the coating gradually acquires a dense CrN sublayer and a transparent organization at the interface of the TiN sublayer. The average grain size decreased from 9.4 to 7.5 nm. The compressive residual stress increased from -1.32 to -6.97 GPa. The H , H/E^* and H^3/E^{*2} of $\text{Cr}_2\text{N}/\text{TiN}$ multilayer coatings reveals the highest values of 25.1 GPa, 0.078 and 0.128 GPa, as well as the adhesion of coatings and substrate at $V_s = -100$ V, indicating the best mechanical properties including hardness, toughness and adhesion. The $\text{Cr}_2\text{N}/\text{TiN}$ multilayer coatings prepared at $V_s = -100$ V also presents the highest corrosion potential and lowest corrosion current density, implying the best anti-corrosive properties. Both of the excellent mechanical and anti-corrosive properties may be resulted from the dense microstructure, suitable internal stress abrupt interface, fine grain size and less droplets, which were obtained at the proper bias voltage.

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