

Polypyrrole/ α -Fe₂O₃ Hybrids for enhanced Electrochemical Sensing

Performance towards Uric Acid

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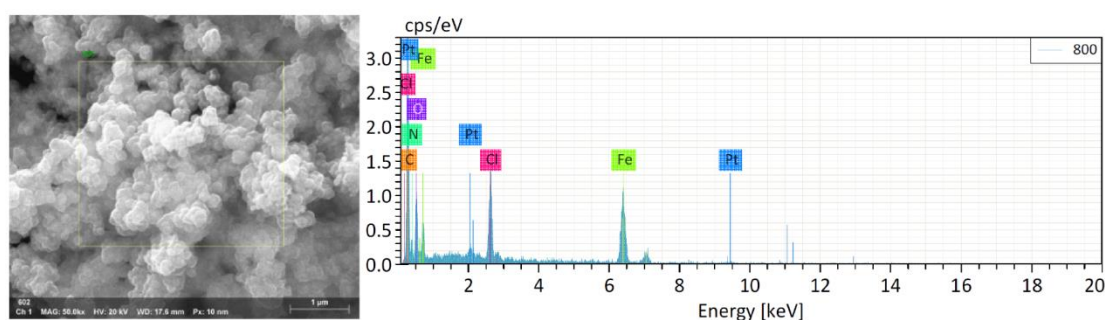


Figure S1. EDS mapping of PPy/ α -Fe₂O₃ composite.

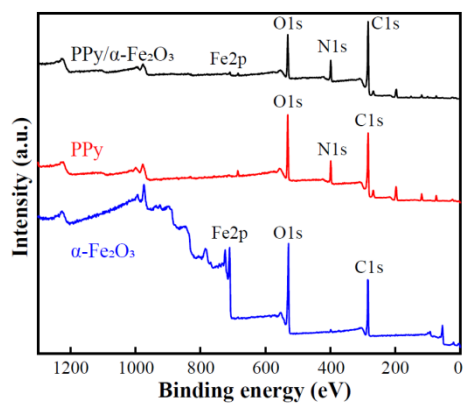


Figure S2. The XPS spectra of α -Fe₂O₃, PPy, and PPy/ α -Fe₂O₃ composite.

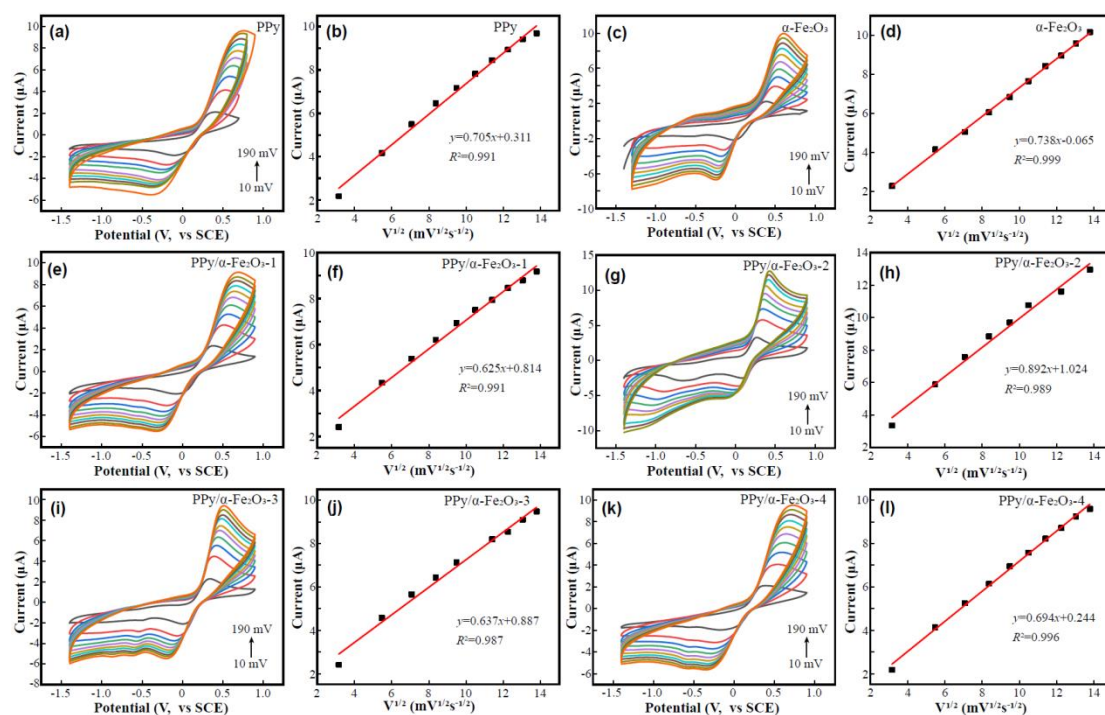


Figure S3. CV curves and the dependence plots of anodic peak current vs square root of scan rate of (a-b) PPY, (c-d) $\alpha\text{-Fe}_2\text{O}_3$, (e-f) PPY/ $\alpha\text{-Fe}_2\text{O}_3$ -1, (g-h) PPY/ $\alpha\text{-Fe}_2\text{O}_3$ -2, (i-j) PPY/ $\alpha\text{-Fe}_2\text{O}_3$ -3, and (k-l) PPY/ $\alpha\text{-Fe}_2\text{O}_3$ -4 in 0.1 M KCl solution containing 5 mM $[\text{Fe}(\text{CN})_6]^{3-/4-}$.

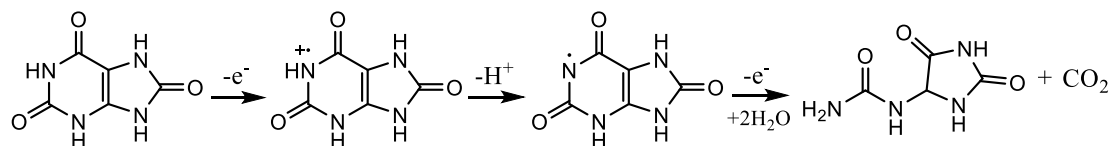


Figure S4. The probable reaction mechanisms of UA on PPY/ $\alpha\text{-Fe}_2\text{O}_3$ electrode.

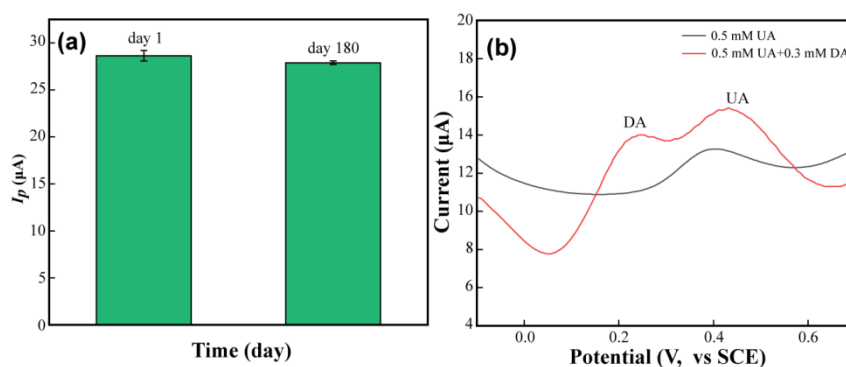


Figure S5. (a) Storage stability of the PPY/ $\alpha\text{-Fe}_2\text{O}_3$ electrode tested by CV for 180 days. (b) DPV curves of the PPY/ $\alpha\text{-Fe}_2\text{O}_3$ electrode were obtained in a solution containing 0.5 mM UA, and 0.3 mM DA their mixtures, respectively.