

# **Construction of Advanced S-Scheme Heterojunction Interface Composites of Bimetallic Phosphate MnMgPO<sub>4</sub> with C<sub>3</sub>N<sub>4</sub> Surface with Remarkable Performance in Photocatalytic Hydrogen Production and Pollutant Degradation**

**Ting Cheng<sup>1,2,3</sup>, Jiarui Zhu<sup>3</sup>, Chen Chen<sup>3,\*</sup>, Yulin Hu<sup>4</sup>, Liangliang Wu<sup>3</sup>, Mengyi Zhang<sup>3</sup>, Liwei Cui<sup>3</sup>, Youzhi Dai<sup>5</sup>, Xiao Zhang<sup>1,2,6</sup>, Yuan Tian<sup>3</sup> and Fei Wu<sup>3</sup>**

<sup>1</sup> School of Environmental Ecology, Jiangsu City Vocational College, Nanjing, 210017, China; wncchengting@sina.com (T.C.); zhangxiao7376@sina.com (X.Z.)

<sup>2</sup> Jiangsu Engineering and Technology Centre for Ecological and Environmental Protection in Urban and Rural Water Environment Management and Low Carbon Development, Nanjing 210017, China

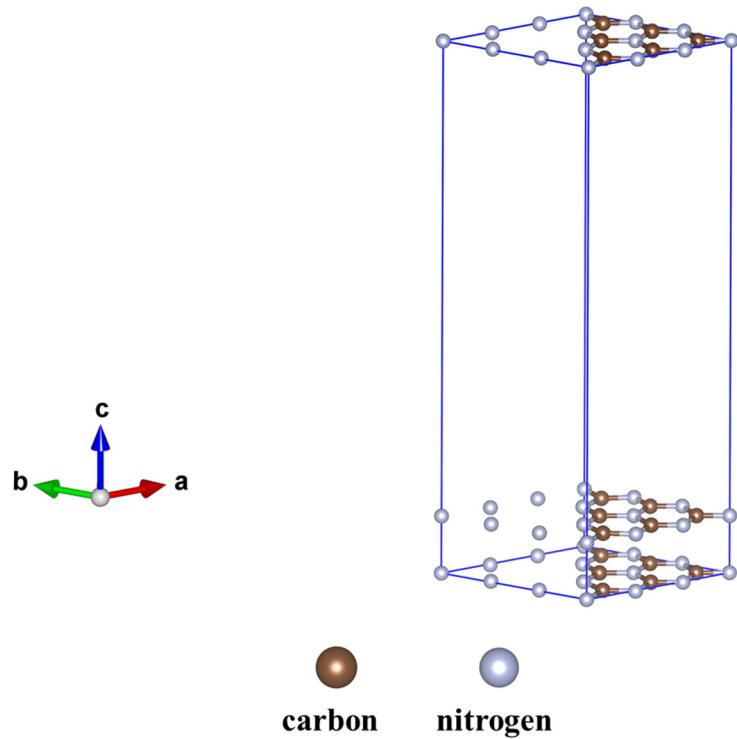
<sup>3</sup> School of Environmental and Chemical Engineering, Jiangsu University of Science and Technology, Zhenjiang 212100, China; zjr000323@163.com (J.Z.); w18852913906@163.com (L.W.); zhangmengyi031029@163.com (M.Z.); cui36984628@163.com (L.C.); ttyy1974.ok@163.com (Y.T.); wufei1224wf@hotmail.com (F.W.)

<sup>4</sup> College of Chemistry and Chemical Engineering, Anshun University, Anshun 561000, China; huyulin1982@163.com

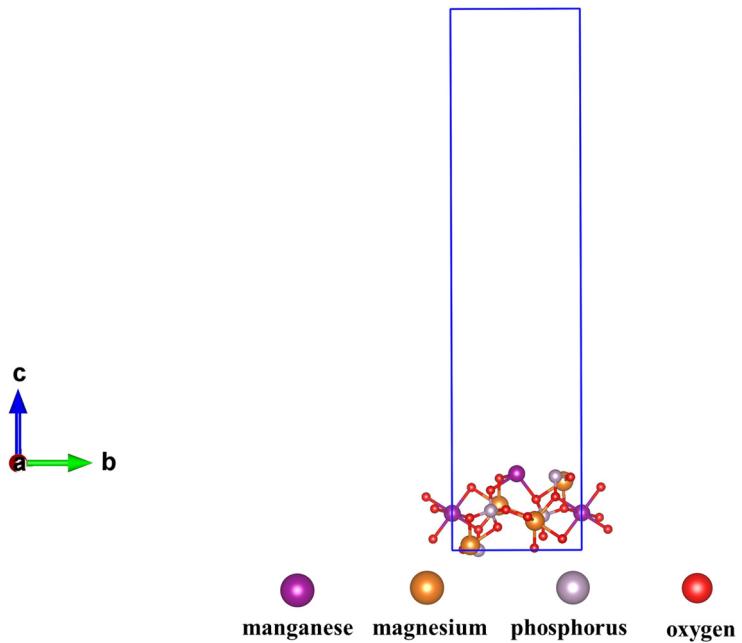
<sup>5</sup> College of Environment and Resource, Xiangtan University, Xiangtan 411105, China; daiyouzhi202@163.com

<sup>6</sup> Nanjing University and Yancheng Academy of Environmental Technology and Engineering, Yancheng 224000, China

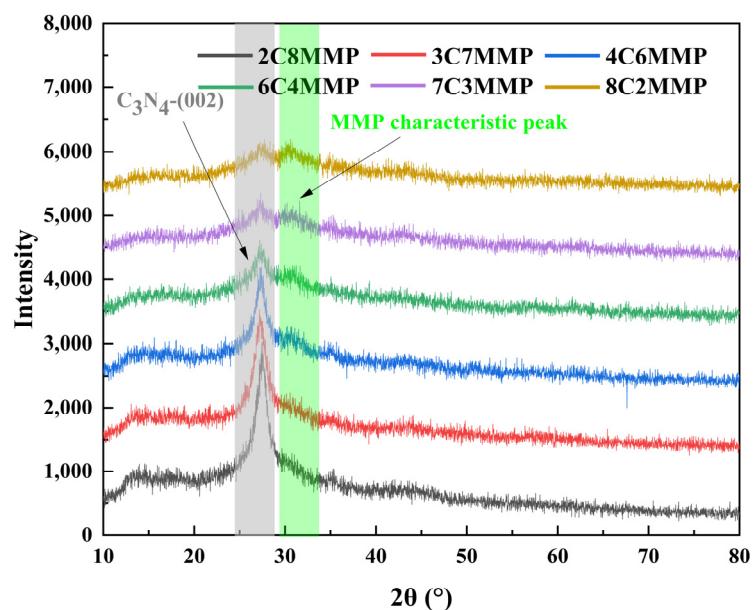
\* Correspondence: chenc@just.edu.cn



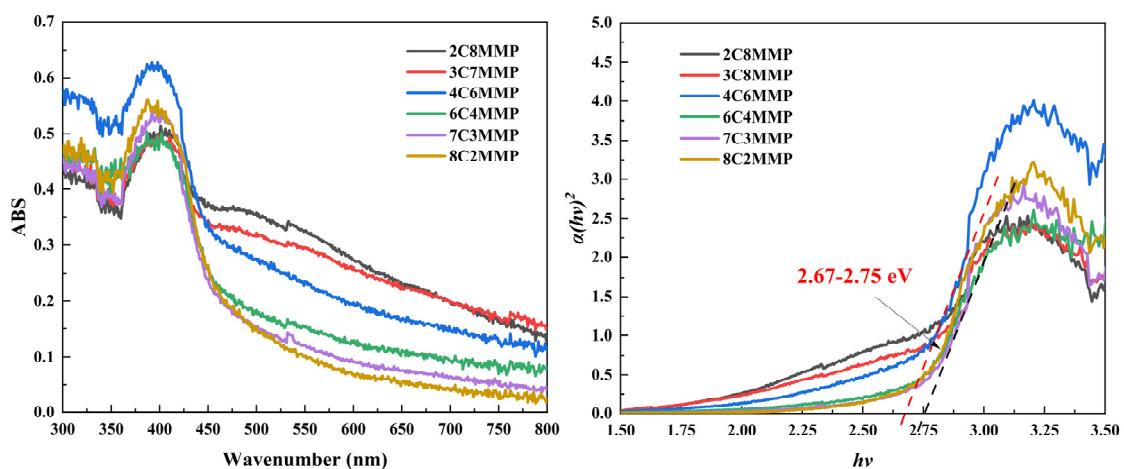
**Figure S1. DFT theoretical work function calculation models of  $\text{C}_3\text{N}_4$ .**



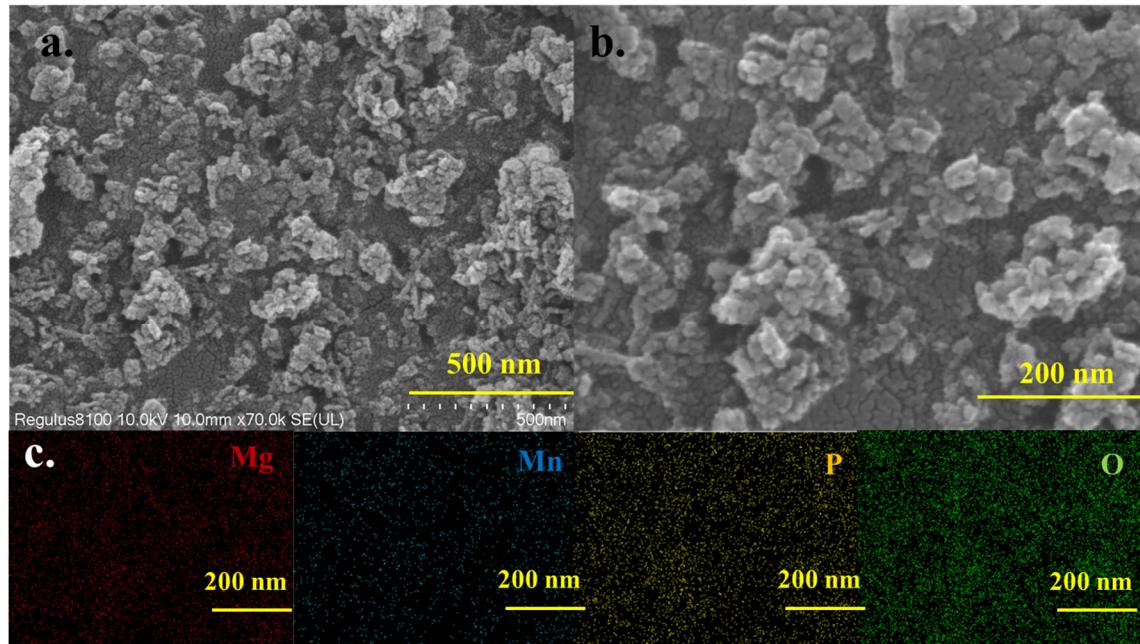
**Figure S2.** DFT theoretical work function calculation models of MgMnP.



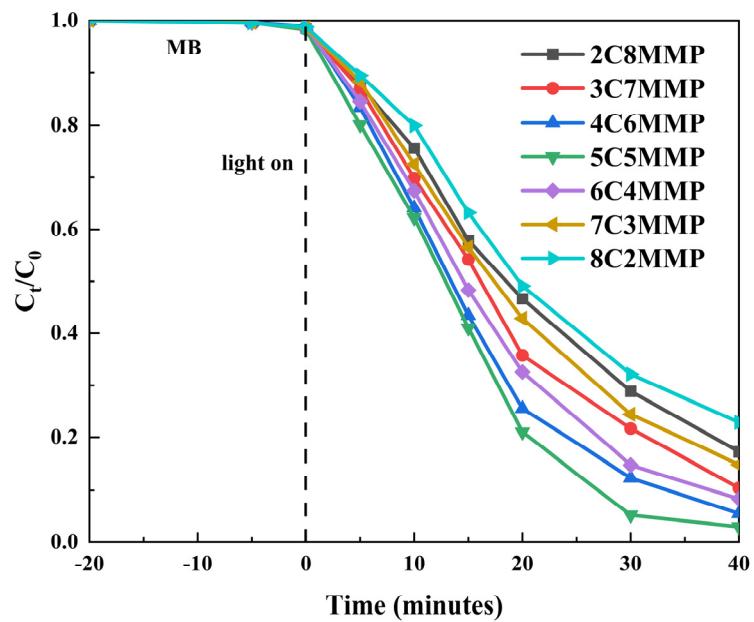
**Figure S3.** The XRD patterns of composite materials with different  $\text{C}_3\text{N}_4$  and  $\text{MgMnP}$  ratios.



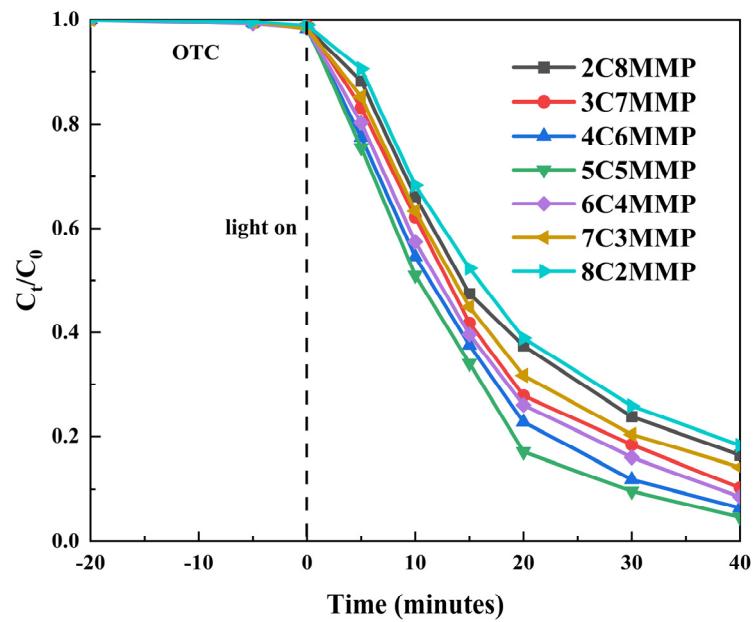
**Figure S4. The UV-Vis diffuse reflectance spectra (a) and band fitting results (b) of composite materials with different  $\text{C}_3\text{N}_4$  and  $\text{MgMnP}$  ratios.**



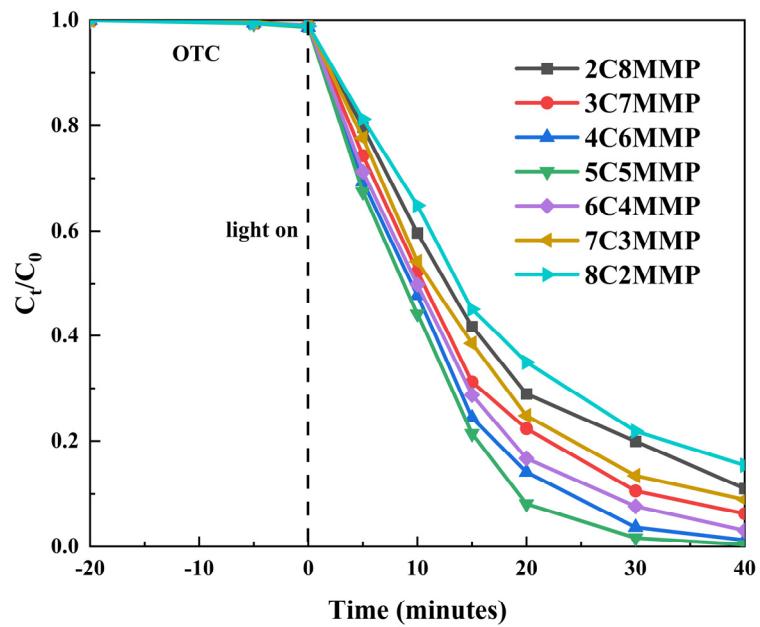
**Figure S5.** The SEM morphology analysis results of MgMnP material (a and b); the SEM-EDX elemental mapping analysis results of MgMnP material (c).



**Figure S6. The photocatalytic degradation curves of MB using composites with different  $\text{C}_3\text{N}_4$  and  $\text{MgMnP}$  ratios.**



**Figure S7.** The photocatalytic degradation curves of OTC using composites with different  $\text{C}_3\text{N}_4$  and  $\text{MgMnP}$  ratios.



**Figure S8.** The photocatalytic degradation curves of TE using composites with different  $\text{C}_3\text{N}_4$  and  $\text{MgMnP}$  ratios.

**Table S1. The photocatalytic degradation efficiency of MB for recent photocatalysts.**

| Catalyst   | Light source                   | Concentration (mg/L) | Catalyst dosage (mg/ml) | Degradation (%)                                      | Reference    |
|--|--------------------------------|----------------------|-------------------------|--|--------------|
| MgFe <sub>2</sub> O <sub>4</sub>                                   | Sunlight                       | 10                   | 20/40                   | 80 (120mins)   | [58]         |
| NiMn <sub>2</sub> O <sub>4</sub> /ZnMn <sub>2</sub> O <sub>4</sub> | 200W tungsten lamp             | 20                   | 10/100                  | 92 (80mins)  | [59]         |
| Ag <sub>3</sub> PO <sub>4</sub> /MnFe <sub>2</sub> O <sub>4</sub>  | Sunlight                       | 15                   | 25/50                   | 98 (82mins)  | [60]         |
| Ni <sub>0.6</sub> Zn <sub>0.4</sub> Fe <sub>2</sub> O <sub>4</sub> | Mercury vapor Lamp             | 20                   | 30 /100                 | 94 (275mins)   | [61]         |
| ZnFe <sub>2</sub> O <sub>4</sub> /ZnO                              | UV lamp                        | 10                   | 100/100                 | 100 (180mins)  | [62]         |
| Co/Ni-MOF@BiOI   | Sunlight                       | 10                   | 80/100                  | 99 (200mins)   | [63]         |
| Ti/TiN/TiON/TiO <sub>2</sub>                                       | 250 W mercury lamp             | 1                    | 20/20ml                 | 69 (60mins)  | [64]         |
| Ag-ZnO/MWCNT CVCs  | 30 W UV lamp<br>40 W white LED | 10<br>5              | 24/80<br>50/100         | 98 (250mins)<br>96.2 (80mins)<br>almost 100 (40mins) | [65]<br>[66] |
| MgMnP/C <sub>3</sub> N <sub>4</sub>                                | 300W Xe lamp                   | 10                   | 20/30                   | almost 100 (40mins)                                  | This work    |

**Table S2. The photocatalytic degradation efficiency of OTC for recent photocatalysts.**

| Catalyst  | Light source   | Concentration (mg/L) | Catalyst dosage (mg/ml) | Degradation (%)     | Reference |
|---|----------------|----------------------|-------------------------|---------------------|-----------|
| CVCs  | 40 W white LED | 20                   | 50/100                  | 80.5 (80mins)       | [66]      |
| Co <sub>3</sub> O <sub>4</sub> /TiO <sub>2</sub> /GO  | 300 W Xe lamp  | 10                   | 50/200                  | 91 (90mins)         | [67]      |
| Bi <sub>2</sub> WO <sub>6</sub> -BiOCl                | 500 W Xe lamp  | 20                   | 30/30                   | 93 (300mins)        | [68]      |
| GTZ   | Visible 300 W  | 10                   | 20/100                  | 100 (180mins)       | [69]      |
| H <sub>2</sub> O <sub>2</sub> /ZnWO <sub>4</sub> /CaO | Sunlight       | 10                   | 50/50                   | 85 (210min)         | [70]      |
| N,Fe-CDs/G-WO <sub>3</sub> -0.6                       | 500 W Xe lamp  | 20                   | 50/50                   | 47.8 (150mins)      | [71]      |
| ZZFDT (S-   | 300 W Xe lamp  | 10                   | 100/100                 | 95 (150mins)        | [72]      |
| TiO <sub>2</sub> /WS <sub>2</sub> /alginate beads     | Sunlight       | 10                   | 50/100                  | 92. 5 (240mins)     | [73]      |
| BiBaO <sub>3</sub> /Ag <sub>3</sub> PO <sub>4</sub>   | 300W Xe lamp   | 20                   | 20/30                   | 100 (40mins)        | [74]      |
| MgMnP/C <sub>3</sub> N <sub>4</sub>                   | 300W Xe lamp   | 10                   | 20/30                   | almost 100 (40mins) | This work |

**Table S3. The photocatalytic degradation efficiency of TE for recent photocatalysts.**

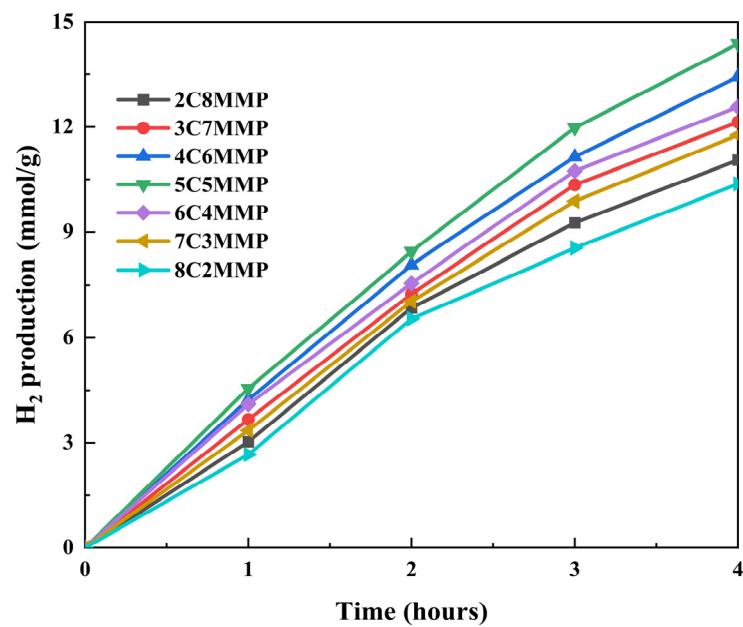
| Catalyst   | Light source      | Concentration (mg/L) | Catalyst dosage (mg/ml) | Degradation (%)     | Reference |
|--|-------------------|----------------------|-------------------------|---------------------|-----------|
| CVCs   | 40 W white LED    | 40                   | 50/100                  | 86 (80min)          | [66]      |
| BiBaO <sub>3</sub> /Ag <sub>3</sub> P <sub>O<sub>4</sub></sub> | 300W Xe lamp      | 20                   | 20/30                   | 100 (40mins)        | [74]      |
| BTO  | 300 W Xe UV lamp  | 40                   | 30/50                   | 64.2 (60mins)       | [75]      |
| NiFe-LDH/CTF   | 300W Xe lamp      | 40                   | 20/100                  | 85.6 (120mins)      | [76]      |
| SrTiO <sub>3</sub> /TiO <sub>2</sub>                           | 300W Xe lamp      | 20                   | 30/25                   | 90 (40mins)         | [77]      |
| PCN-5  | 300W Xe lamp      | 20                   | 30/100                  | 83.8 (60mins)       | [78]      |
| Co-CNk-OH  | Sunlight          | 20                   | 50/100                  | 87.1 (40mins)       | [79]      |
| Cr <sub>2</sub> O <sub>3</sub> /ZrO <sub>2</sub>               | 300W Xe lamp      | 50                   | 100/100                 | 97.1 (120mins)      | [80]      |
| Ce/Bi/BiOCl  | 300 W Xe arc lamp | 40                   | 50/100                  | 97.7 (20mins)       | [81]      |
| MgMnP/C <sub>3</sub> N <sub>4</sub>                            | 300W Xe lamp      | 10                   | 20/30                   | almost 100 (40mins) | This work |

**Table S4.** The kinetic constants of each photocatalytic pollutant degradation system.

| MB     |  |
|--------|--|
| System | Kinetic constant (minute <sup>-1</sup> ) |
| 2C8MMP | 0.0446                                   |
| 3C7MMP | 0.0573                                   |
| 4C6MMP | 0.0752                                   |
| 5C5MMP | 0.0959                                   |
| 6C4MMP | 0.0653                                   |
| 7C3MMP | 0.0494                                   |
| 8C2MMP | 0.0387                                   |

| OTC    |  |
|--------|--|
| System | Kinetic constant (minute <sup>-1</sup> ) |
| 2C8MMP | 0.0471                                   |
| 3C7MMP | 0.0584                                   |
| 4C6MMP | 0.0709                                   |
| 5C5MMP | 0.0793                                   |
| 6C4MMP | 0.0626                                   |
| 7C3MMP | 0.0512                                   |
| 8C2MMP | 0.0448                                   |

| TE     |  |
|--------|--|
| System | Kinetic constant (minute <sup>-1</sup> ) |
| 2C8MMP | 0.0555                                   |
| 3C7MMP | 0.0718                                   |
| 4C6MMP | 0.1136                                   |
| 5C5MMP | 0.1452                                   |
| 6C4MMP | 0.0888                                   |
| 7C3MMP | 0.0632                                   |
| 8C2MMP | 0.0484                                   |



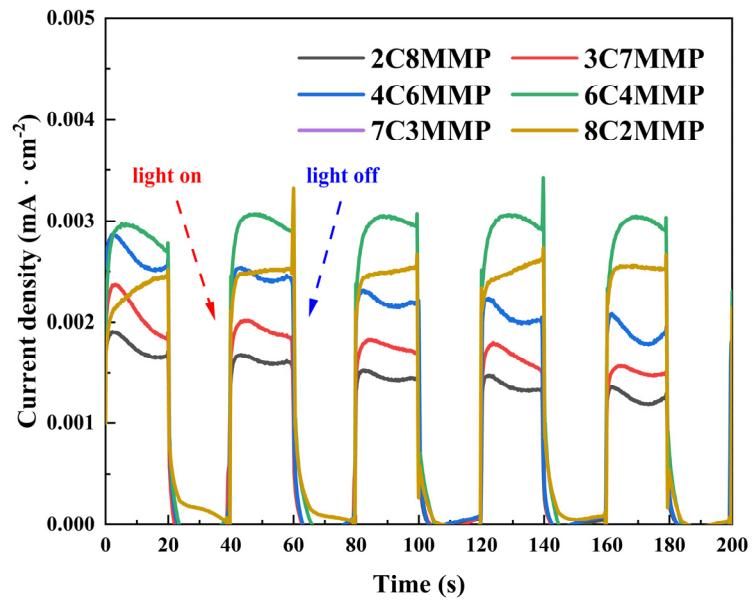
**Figure S9.** The photocatalytic hydrogen evolution curves of composites with different C<sub>3</sub>N<sub>4</sub> and MgMnP ratios.

**Table S5. The apparent H<sub>2</sub> production rate constant of each photocatalytic system.**

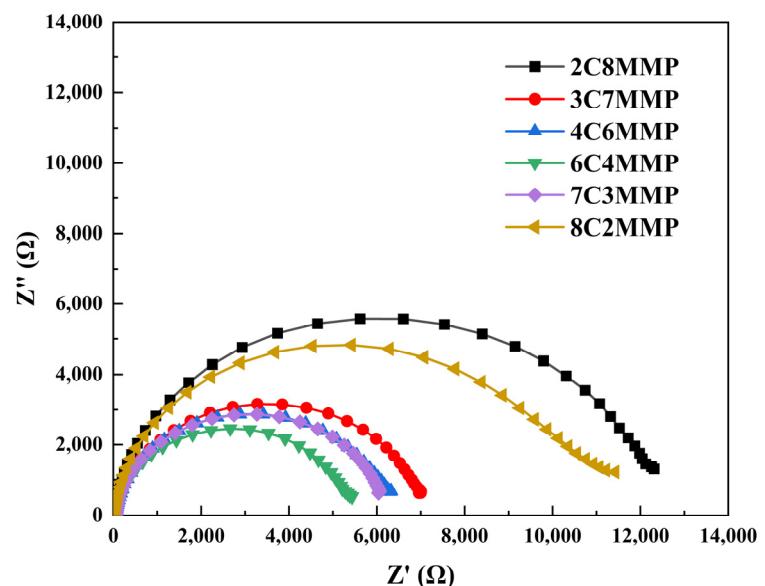
| System | TE | Apparent kinetic constant (mmol·g <sup>-1</sup> ·h <sup>-1</sup> ) |
|--------|----|--|
| 2C8MMP |    | 2.764  |
| 3C7MMP |    | 3.034  |
| 4C6MMP |    | 3.358  |
| 5C5MMP |    | 3.595  |
| 6C4MMP |    | 3.143  |
| 7C3MMP |    | 2.941  |
| 8C2MMP |    | 2.592  |

**Table S6. The hydrogen evolution efficiency of recent photocatalysts.**

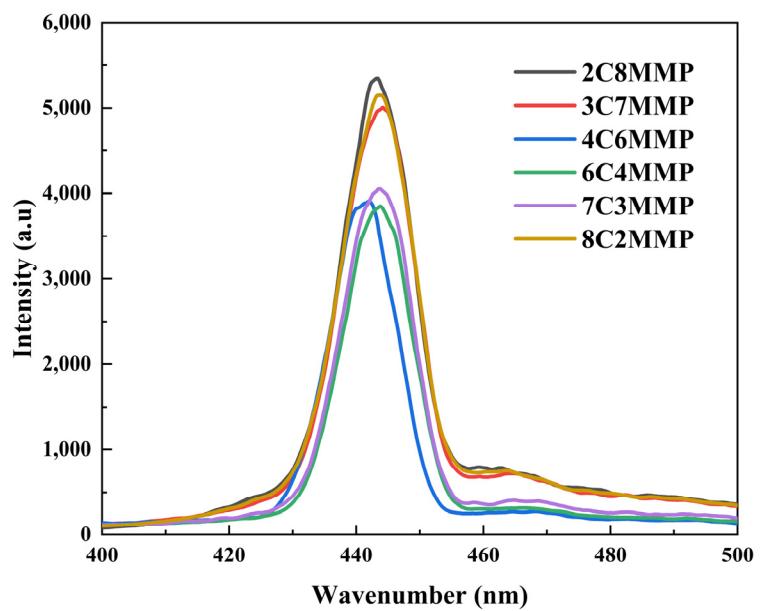
| Catalyst  | Light source   | Apparent H <sub>2</sub> production rate constant<br>(m mol·g <sup>-1</sup> ·h <sup>-1</sup> ) | Reference |
|---|----------------|---|-----------|
| AgCl/CCN  | 300 W Xe lamp  | 0.335   | [82]      |
| Cu-Zn0.5Cd0.5S  | 300 W Xe lamp  | 1.904   | [83]      |
| MoP/a-TiO <sub>2</sub> /Co-ZnIn <sub>2</sub> S <sub>4</sub>                       | 300 W Xe lamp  | 2.96  | [84]      |
| CF/SrTiO <sub>3</sub> /CdS  | 300 W arc lamp | 0.578   | [85]      |
| WS <sub>2</sub> /WSe <sub>2</sub>   | 300 W Xe lamp  | 3.857   | [86]      |
| NiFe <sub>2</sub> O <sub>4</sub> /Cu <sub>2</sub> O                               | 300 W Xe lamp  | 0.004   | [97]      |
| NH <sub>2</sub> -ZSTU   | 300 W Xe lamp  | 0.431   | [88]      |
| Bi <sub>4</sub> Ti <sub>3</sub> O <sub>12</sub> /ZnIn <sub>2</sub> S <sub>4</sub> | 300 W Xe lamp  | 19.8  | [89]      |
| ZnO/Co <sub>3</sub> O <sub>4</sub>  | 350 W Xe lamp  | 0.793   | [90]      |
| MgMnP/C <sub>3</sub> N <sub>4</sub>   | 300W Xe lamp   | 3.595   | This work |



**Figure S10.** The photocurrent analysis results of 2C8MMP, 3C7MMP, 4C6MMP, 6C4MMP, 7C3MMP, and 8C2MMP composite materials.



**Figure S11.** The electrochemical impedance analysis results of 2C8MMP, 3C7MMP, 4C6MMP, 6C4MMP, 7C3MMP, and 8C2MMP composite materials.



**Figure S12.** The solid-state fluorescence analysis results of 2C8MMP, 3C7MMP, 4C6MMP, 6C4MMP, 7C3MMP, and 8C2MMP composite materials.

## References

- [58] HAMMACHE Z, SOUKEUR A, OMEIRI S, BELLAL B, TRARI M. Physical and photo-electrochemical properties of MgFe<sub>2</sub>O<sub>4</sub> prepared by sol gel route: application to the photodegradation of methylene blue [J]. *Journal of Materials Science: Materials in Electronics*, 2019, 30(6): 5375-5382 doi: <https://doi.org/10.1007/s10854-019-00830-2>.
- [59] FAWY K F, ASHIQ M F, ALHARBI F F, MANZOOR S, NISA M U, IBRAHIM M, KHAN M T N, MESSALI M, CHUGHTAI A H, ASHIQ M N. Facile synthesis of NiMn<sub>2</sub>O<sub>4</sub>/ZnMn<sub>2</sub>O<sub>4</sub> heterostructure nanocomposite for visible-light-driven degradation of methylene blue dye [J]. *Journal of Taibah University for Science*, 2024, 18(1): 2302656 doi: <https://doi.org/10.1080/16583655.2024.2302656>.
- [60] ABROSHAN E, FARHADI S, ZABARDASTI A. Novel magnetically separable Ag<sub>3</sub>PO<sub>4</sub>/MnFe<sub>2</sub>O<sub>4</sub> nanocomposite and its high photocatalytic degradation performance for organic dyes under solar-light irradiation [J]. *Solar Energy Materials and Solar Cells*, 2018, 178: 154-163 doi: <https://doi.org/10.1016/j.solmat.2018.01.026>.
- [61] PADMAPRIYA G, MANIKANDAN A, KRISHNASAMY V, JAGANATHAN S K, ANTONY S A. Spinel Ni<sub>x</sub>Zn<sub>1-x</sub>Fe<sub>2</sub>O<sub>4</sub> (0.0 ≤ x ≤ 1.0) nano-photocatalysts: Synthesis, characterization and photocatalytic degradation of methylene blue dye [J]. *Journal of Molecular Structure*, 2016, 1119: 39-47 doi: <https://doi.org/10.1016/j.molstruc.2016.04.049>.
- [62] SHAO R, SUN L, TANG L, CHEN Z. Preparation and characterization of magnetic core-shell ZnFe<sub>2</sub>O<sub>4</sub>@ZnO nanoparticles and their application for the photodegradation of methylene blue [J]. *Chemical Engineering Journal*, 2013, 217: 185-191 doi: <https://doi.org/10.1016/j.cej.2012.11.109>.
- [63] SIDDIQA A, AKHTER T, FAHEEM M, RAZZAQUE S, MAHMOOD A, AL-MASRY W, NADEEM S, HASSAN S U, YANG H, PARK C H. Bismuth-Rich Co/Ni Bimetallic Metal-Organic Frameworks as Photocatalysts toward Efficient Removal of Organic Contaminants under Environmental Conditions [J]. *Micromachines (Basel)*, 2023, 14(5): 899 doi: <https://doi.org/10.3390/mi14050899>.
- [64] MUSLIMOV A, ORUDZHEV F, GADZHIEV M, SELIMOV D, TYUFTYAEV A, KANEVSKY V. Facile Synthesis of Ti/TiN/TiON/TiO<sub>2</sub> Composite Particles for Plasmon-Enhanced Solar Photocatalytic Decomposition of Methylene Blue [J]. *Coatings*, 2022, 12(11): 1741 doi: <https://doi.org/10.3390/coatings12111741>.
- [65] LÓPEZ ALEJANDRO E M, RAMÍREZ MORALES E, ARELLANO CORTAZA M C, MORÁN LÁZARO J P, PÉREZ HERNÁNDEZ G, ROJAS BLANCO L. Synthesis of Ag-modified ZnO/MWCNT nanoparticles and their application as a catalyst in the degradation of methylene blue [J]. *Digest Journal of Nanomaterials and Biostructures*, 2023, 18(3): 941-950 doi: <https://doi.org/10.15251/djnb.2023.183.941>.
- [66] FENG J, RAN X, WANG L, XIAO B, LEI L, ZHU J, LIU Z, XI X, FENG G, DAI Z, LI R. The Synergistic Effect of Adsorption-Photocatalysis for Removal of Organic Pollutants on Mesoporous Cu<sub>2</sub>V<sub>2</sub>O<sub>7</sub>/Cu<sub>3</sub>V<sub>2</sub>O<sub>8</sub>/g-C<sub>3</sub>N<sub>4</sub> Heterojunction [J]. *International Journal of Molecular Sciences*, 2022, 23(22) doi: <https://doi.org/10.3390/ijms232214264>.
- [67] JO W-K, KUMAR S, ISAACS M A, LEE A F, KARTHIKEYAN S. Cobalt promoted TiO<sub>2</sub>/GO for the photocatalytic degradation of oxytetracycline and Congo Red [J]. *Applied Catalysis B: Environmental*, 2017, 201: 159-168 doi: <https://doi.org/10.1016/j.apcatb.2016.08.022>.
- [68] GUO M, ZHOU Z, YAN S, ZHOU P, MIAO F, LIANG S, WANG J, CUI X. Bi<sub>2</sub>WO<sub>6</sub>-BiOCl heterostructure with enhanced photocatalytic activity for efficient degradation of oxytetracycline [J]. *Scientific reports*, 2020, 10(1): 18401 doi: <https://doi.org/10.1038/s41598-020-75003-x>.
- [69] HU X-Y, ZHOU K, CHEN B-Y, CHANG C-T. Graphene/TiO<sub>2</sub>/ZSM-5 composites synthesized by mixture design were used for photocatalytic degradation of oxytetracycline under visible light: Mechanism and biotoxicity [J]. *Applied Surface Science*, 2016, 362: 329-334 doi: <https://doi.org/10.1016/j.apsusc.2015.10.192>.

- [70] RAIZADA P, SHANDILYA P, SINGH P, THAKUR P. Solar light-facilitated oxytetracycline removal from the aqueous phase utilizing a H<sub>2</sub>O<sub>2</sub>/ZnWO<sub>4</sub>/CaO catalytic system [J]. *Journal of Taibah University for Science*, 2018, 11(5): 689-699 doi: <https://doi.org/10.1016/j.jtusci.2016.06.004>.
- [71] NI T, LI Q, YAN Y, WANG F, CUI X, YANG Z, WANG Y, YANG Z, CHANG K, LIU G. N,Fe-Doped Carbon Dot Decorated Gear-Shaped WO<sub>3</sub> for Highly Efficient UV-Vis-NIR-Driven Photocatalytic Performance [J]. *Catalysts*, 2020, 10(4): 416 doi: <https://doi.org/10.3390/catal10040416>.
- [72] XUE L, LIANG E, WANG J. Fabrication of magnetic ZnO/ZnFe<sub>2</sub>O<sub>4</sub>/diatomite composites: improved photocatalytic efficiency under visible light irradiation [J]. *Journal of Materials Science: Materials in Electronics*, 2022, 33(3): 1405-1424 doi: <https://doi.org/10.1007/s10854-021-07568-w>.
- [73] KUMAR R, ANSARI M O, TALEB M A, OVES M, BARAKAT M A, ALGHAMDI M A, AL MAKISHAH N H. Integrated Adsorption-Photocatalytic Decontamination of Oxytetracycline from Wastewater Using S-Doped TiO<sub>2</sub>/WS<sub>2</sub>/Calcium Alginate Beads [J]. *Catalysts*, 2022, 12(12): 16776 doi: <https://doi.org/10.3390/catal12121676>.
- [74] ZHANG X, CHEN C, CHENG T, TIAN Y, WEN M, HOU B, XIN X, PAN F, SHI J. Construction highly efficient p-n heterojunctions composite of BiBaO<sub>3</sub> and Ag<sub>3</sub>PO<sub>4</sub> for visible light driven photocatalytic degradation of tetracycline and oxytetracycline [J]. *Water Resources and Industry*, 2024, 31: 100246 doi: <https://doi.org/10.1016/j.wri.2024.100246>.
- [75] GUO Q, GAO T, PADERVAND M, DU D, ZHAO K, ZHANG Y, JIA T, WANG C. Piezo-Photocatalytic Degradation of Tetracycline by 3D BaTiO<sub>3</sub> Nanomaterials: The Effect of Crystal Structure and Catalyst Loadings [J]. *Processes*, 2023, 11(12) doi: <https://doi.org/10.3390/pr11123323>.
- [76] ZHANG J, CHEN X, CHEN Q, HE Y, PAN M, HUANG G, BI J. Insights into Photocatalytic Degradation Pathways and Mechanism of Tetracycline by an Efficient Z-Scheme NiFe-LDH/CTF-1 Heterojunction [J]. *Nanomaterials (Basel)*, 2022, 12(23): 4111 doi: <https://doi.org/10.3390/nano12234111>.
- [77] CHEN W, ZHAO N, HU M, LIU X, DENG B. Strengthened Removal of Tetracycline by a Bi/Ni Co-Doped SrTiO<sub>3</sub>/TiO<sub>2</sub> Composite under Visible Light [J]. *Catalysts*, 2024, 14(8) doi: <https://doi.org/10.3390/catal14080539>.
- [78] ZHENG K, CHEN J, GAO X, CAO X, WU S, SU J. Photocatalytic degradation of tetracycline by Phosphorus-doped carbon nitride tube combined with peroxydisulfate under visible light irradiation [J]. *Water science and technology*, 2021, 84(8): 1919-1929 doi: <https://doi.org/10.2166/wst.2021.376>.
- [79] HOU D, LUO J, SUN Q, ZHANG M, WANG J. Preparation of Co-CNK-OH and Its Performance in Fenton-like Photocatalytic Degradation of Tetracycline [J]. *Catalysts*, 2023, 13(4): 715 doi: <https://doi.org/10.3390/catal13040715>.
- [80] WEI X, NARAGINTI S, CHEN P, LI J, YANG X, LI B. Visible Light-Driven Photocatalytic Degradation of Tetracycline Using p-n Heterostructured Cr<sub>2</sub>O<sub>3</sub>/ZrO<sub>2</sub> Nanocomposite [J]. *Water*, 2023, 15(20): 3702 doi: <https://doi.org/10.3390/w15203702>.
- [81] GAO T, CHU H, WANG S, LI Z, ZHOU W. Construction of Ternary Ce Metal-Organic Framework/Bi/BiOCl Heterojunction towards Optimized Photocatalytic Performance [J]. *Nanomaterials (Basel)*, 2024, 14(16): 1352 doi: <https://doi.org/10.3390/nano14161352>.
- [82] JIANG Y, YAN X, FU X, GU Q. Enhanced visible-light-driven co-production of H<sub>2</sub> and value-added chemicals over AgCl/crystalline carbon nitride with N defects [J]. *Colloid and Interface Science Communications*, 2022, 48: 100627 doi: <https://doi.org/10.1016/j.colcom.2022.100627>.
- [83] CHEN H, ZHU Y, WU J, PENG M, DENG S, YANG H, YANG J. Cu-doped ZnCdS-based photocatalyst for efficient photocatalytic hydrogen production by photothermal assistance [J]. *Case Studies in Thermal Engineering*, 2024, 61 doi: <https://doi.org/10.1016/j.csite.2024.104970>.

- [84] WU K, SHANG Y, LI H, WU P, LI S, YE H, JIAN F, ZHU J, YANG D, LI B, WANG X. Synthesis and Hydrogen Production Performance of MoP/a-TiO<sub>2</sub>/Co-ZnIn<sub>2</sub>S<sub>4</sub> Flower-like Composite Photocatalysts [J]. *Molecules*, 2023, 28(11): 4350 doi: <https://doi.org/10.3390/molecules28114350>.
- [85] HU Q, NIU J, ZHANG K-Q, YAO M. One-Dimensional CdS/SrTiO<sub>3</sub>/Carbon Fiber Core–Shell Photocatalysts for Enhanced Photocatalytic Hydrogen Evolution [J]. *Coatings*, 2022, 12(9): 1235 doi: <https://doi.org/10.3390/coatings12091235>.
- [86] TIEN T-M, CHUNG Y-J, HUANG C-T, CHEN E L. Fabrication of WS<sub>2</sub>/WSe<sub>2</sub> Z-Scheme Nano-Heterostructure for Efficient Photocatalytic Hydrogen Production and Removal of Congo Red under Visible Light [J]. *Catalysts*, 2022, 12(8): 852 doi: <https://doi.org/10.3390/catal12080852>.
- [87] DOMÍNGUEZ-ARVIZU J L, JIMÉNEZ-MIRAMONTES J A, HERNÁNDEZ-MAJALCA B C, VALENZUELA-CASTRO G E, GAXIOLA-CEBREROS F A, SALINAS-GUTIÉRREZ J M, COLLINS-MARTÍNEZ V, LÓPEZ-ORTIZ A. Study of NiFe<sub>2</sub>O<sub>4</sub>/Cu<sub>2</sub>O p-n heterojunctions for hydrogen production by photocatalytic water splitting with visible light [J]. *Journal of Materials Research and Technology*, 2022, 21: 4184-4199 doi: <https://doi.org/10.1016/j.jmrt.2022.11.026>.
- [88] HU N, CAI Y, LI L, WANG X, GAO J. Amino-Functionalized Titanium Based Metal-Organic Framework for Photocatalytic Hydrogen Production [J]. *Molecules*, 2022, 27(13): 4241 doi: <https://doi.org/10.3390/molecules27134241>.
- [89] WANG J, SHI Y, SUN H, SHI W, GUO F. Fabrication of Bi<sub>4</sub>Ti<sub>3</sub>O<sub>12</sub>/ZnIn<sub>2</sub>S<sub>4</sub> S-scheme heterojunction for achieving efficient photocatalytic hydrogen production [J]. *Journal of Alloys and Compounds*, 2023, 930: 167450 doi: <https://doi.org/10.1016/j.jallcom.2022.167450>.
- [90] TIEN T-M, CHEN E L. A Novel ZnO/Co<sub>3</sub>O<sub>4</sub> Nanoparticle for Enhanced Photocatalytic Hydrogen Evolution under Visible Light Irradiation [J]. *Catalysts*, 2023, 13(5): 852 doi: <https://doi.org/10.3390/catal13050852>.