

## Electronic Supplementary Information

### **Analysis of *in vivo* Plant Volatiles Using Active Sampling and TD-GC×GC-TOFMS**

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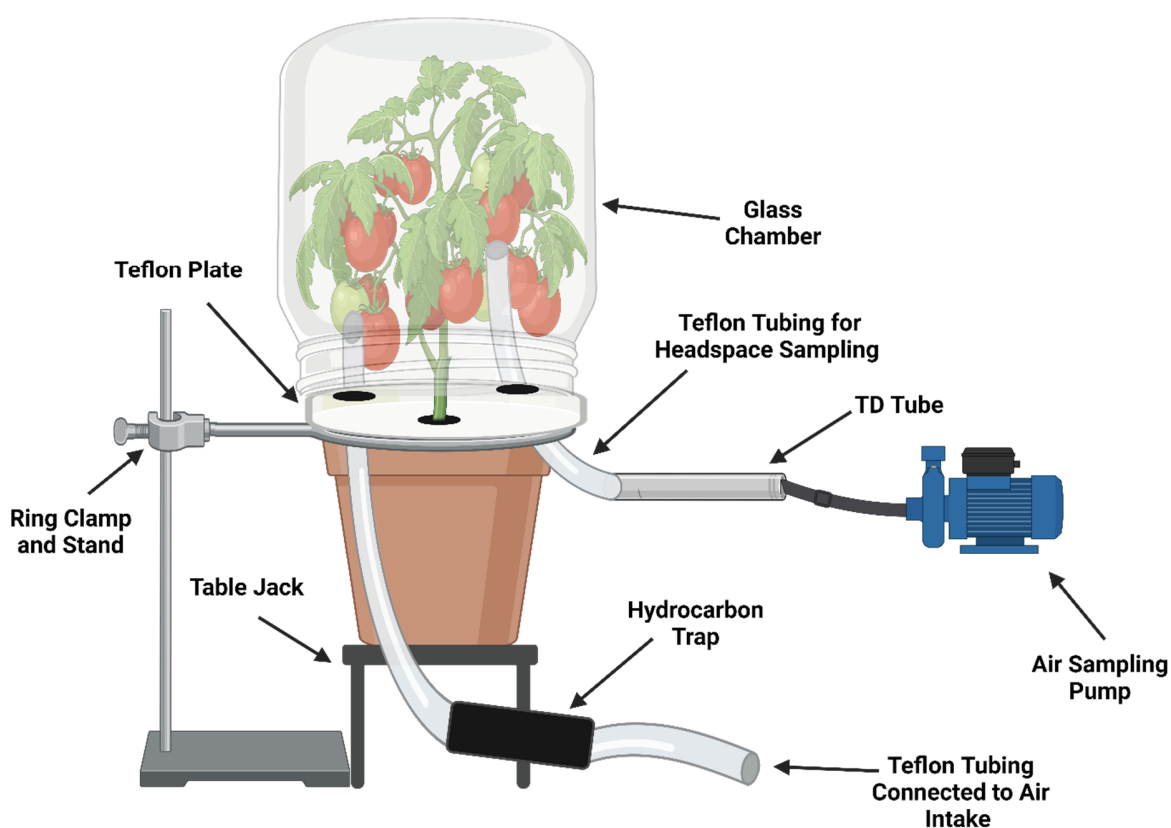


Figure S1. *in vivo* active plant sampling system with a thermal desorption tube.

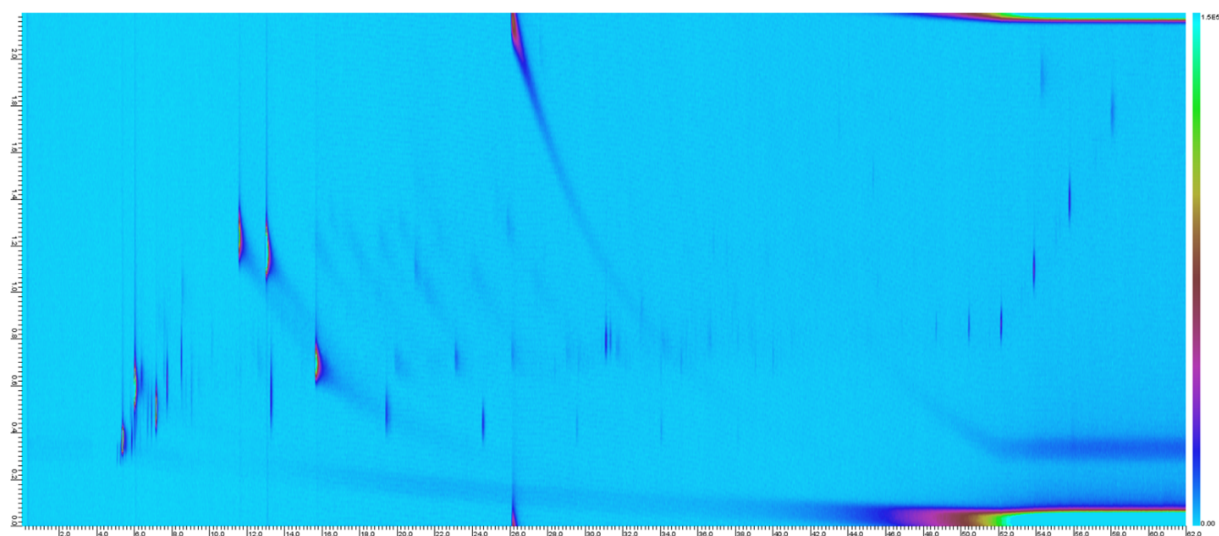


Figure S2. GC×GC-TOFMS TIC contour plot of the clean *in vivo* plant sampling system.

## **Horwitz Calculation**

A solution containing n-nonane-d<sub>20</sub> was used to estimate the precision of the method based on the Horwitz Ratio. A solution of n-nonane-d<sub>20</sub> diluted in methanol was prepared at a concentration of 10 ng/μL. 4μL of this solution was manually spiked onto a thermal desorption tube using a 10 μL syringe. The Horwitz Ratio calculation requires the predicted relative standard deviation (PRSD<sub>R</sub>), which is calculated based on analyte concentration (C), expressed as a dimensionless w/w ratio.

$$PRSD_R(\%) = 2C^{-0.15}$$

Calculation of C, based on the concentration of n-nonane-d<sub>20</sub> and the density of methanol at the laboratory temperature proceeds as follows:

for n-nonane-d<sub>20</sub>:

$$\left(\frac{10 \text{ ng}}{1.0 \text{ } \mu\text{L}}\right) \times (4.0 \text{ } \mu\text{L}) \times \left(\frac{1 \text{ g}}{10^9 \text{ ng}}\right) = 4.0 \times 10^{-8} \text{ g}$$

for methanol:

$$\left(0.7913 \frac{\text{g}}{\text{mL}}\right) \times (4.0 \text{ } \mu\text{L}) \times \left(\frac{1 \text{ mL}}{10^3 \text{ } \mu\text{L}}\right) = 3.1652 \times 10^{-3} \text{ g}$$

thus,

$$C = \frac{4.0 \times 10^{-8} \text{ g}}{3.1652 \times 10^{-3} \text{ g}} = 1.2637 \times 10^{-5}$$

and,

$$PRSD_R(\%) = 2(1.2637 \times 10^{-5})^{-0.15} = 10.9 \%$$