

SO₂ Detection over a Wide Range of Concentrations: An Exploration on MOX-Based Gas Sensors

Arianna Rossi ^{1,*}, Elena Spagnoli ¹, Alan Visonà ¹, Danial Ahmed ¹, Marco Marzocchi ², Vincenzo Guidi ¹ and Barbara Fabbri ^{1,*}

¹ Department of Physics and Earth Sciences, University of Ferrara, Via Saragat 1/C, Ferrara 44122, Italy; elena.spagnoli@unife.it (E.S.); alan.visona@edu.unife.it (A.V.); danial.ahmed@unife.it (D.A.); vincenzo.guidi@unife.it (V.G.)

² Sacmi Imola S.C., Olfactory Systems, Via Selice Provinciale, 17/a, Imola 40026, Italy; marco.marzocchi@sacmigroup.com

* Correspondence: arianna.rossi@unife.it (A.R.); barbara.fabbri@unife.it (B.F.)

Sensor Fabrication

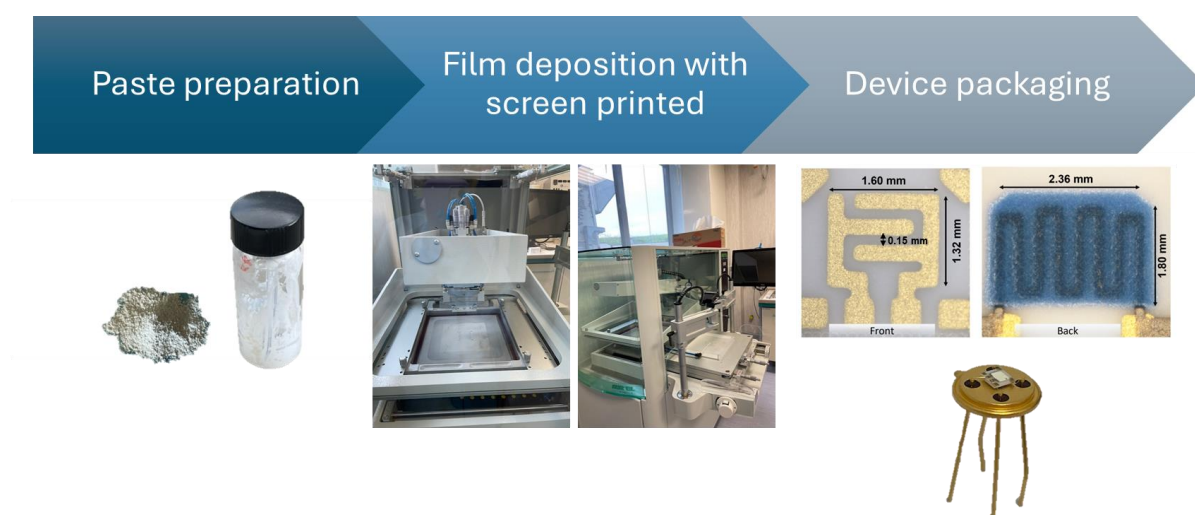


Figure S1. Schematic representation of sensor fabrication.

Electrical characterization

Table S1. The parameters of the calibration fit shown in **Figure 2** for ZnO, WO₃, SnO₂:Pd, SnO₂:Au, SnO₂, and SnO₂:Pt sensors.

Sensor	Equation asymptotic $y = a - b \cdot c^x$			
	a	b	c	R^{2*}
WO ₃	0.45 ± 0.059	2.59 ± 6.57	0.018 ± 0.095	0.89
ZnO	3.17 ± 0.18	7.02 ± 0.816	0.24 ± 0.058	0.99
SnO ₂	8.19 ± 0.20	6.36 ± 0.33	0.66 ± 0.040	0.99
SnO ₂ :Au	16.58 ± 1.53	19.04 ± 1.38	0.755 ± 0.032	0.99
SnO ₂ :Pd	11.29 ± 0.83	12.52 ± 0.74	0.75 ± 0.030	0.99
SnO ₂ :Pt	2.01 ± 0.022	289.51 ± 1301.79	0.0054 ± 0.0244	0.99

* Expresses the goodness of the fit. As for the SnO₂:Ag-sensor, the trend could not be defined.

Table S2. Load resistances of sensors during selectivity measurements.

Sensor	Relation: $G_S = \frac{1}{R_S} = -\frac{V_{out}}{R_f \cdot V_{in}}$
	<i>Load resistance</i>
WO ₃	13 kΩ and 160 kΩ for NO ₂ measurement
ZnO	160 kΩ
SnO ₂	3.6 kΩ
SnO ₂ :Au	6.2 kΩ
SnO ₂ :Pd	130 kΩ
SnO ₂ :Pt	2.7 MΩ
SnO ₂ :Ag	750 kΩ

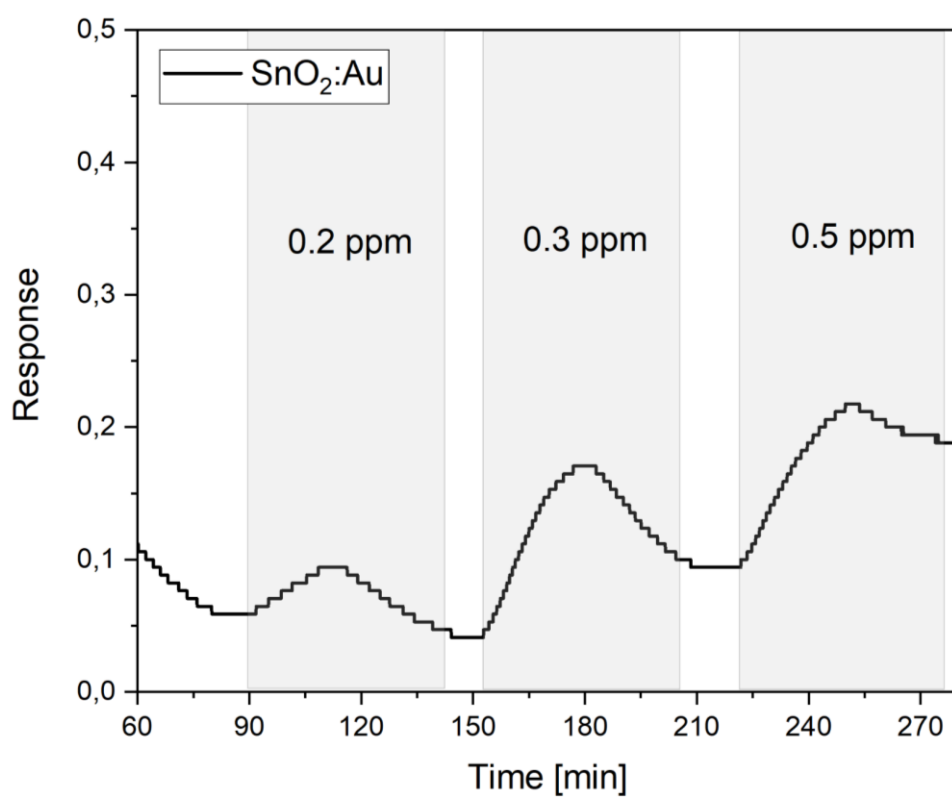


Figure S2. Response to three different concentrations of SO₂ around the theoretical LOD of SnO₂:Au sensor.

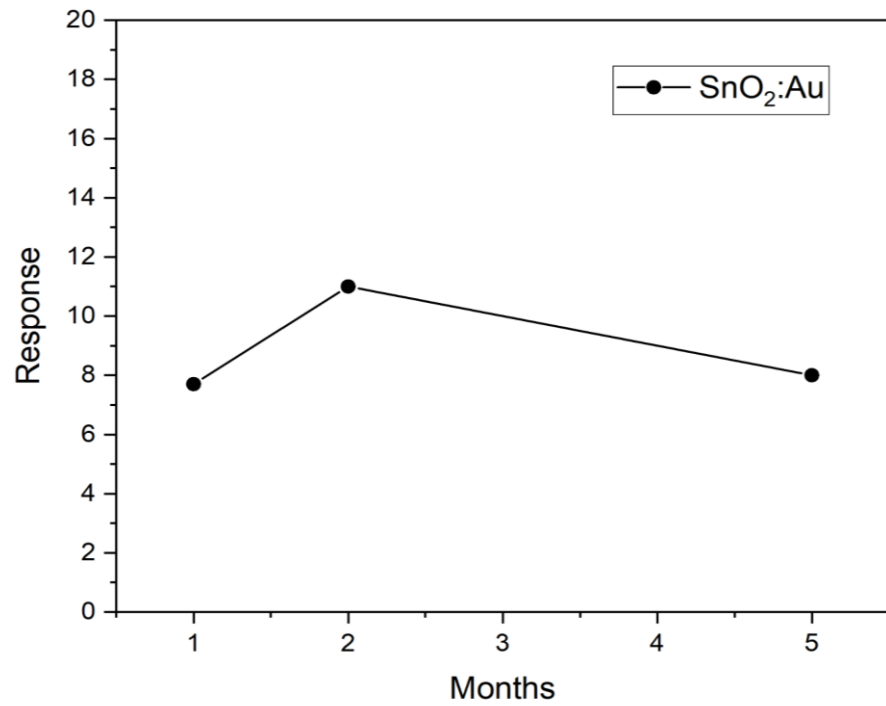


Figure S3. Response to 3 ppm of SO_2 obtained for $\text{SnO}_2:\text{Au}$ sensor tested with three different measurements over a period of five months.

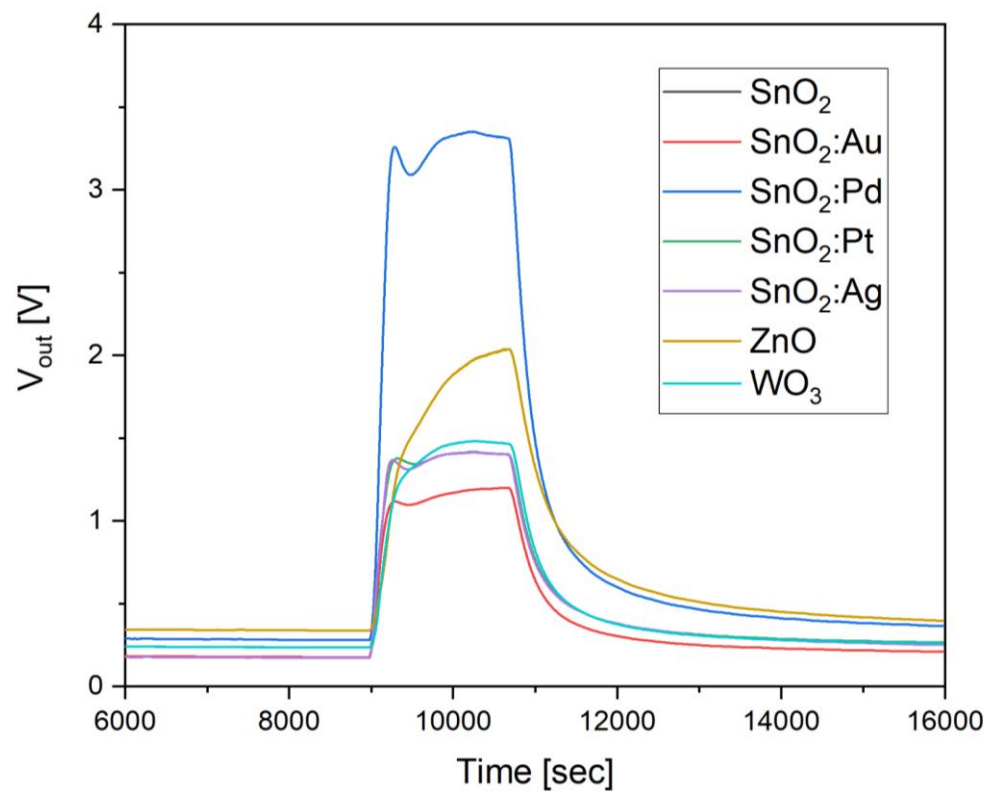


Figure S4. V_{out} response vs. 0.25 ppm of DMDS.

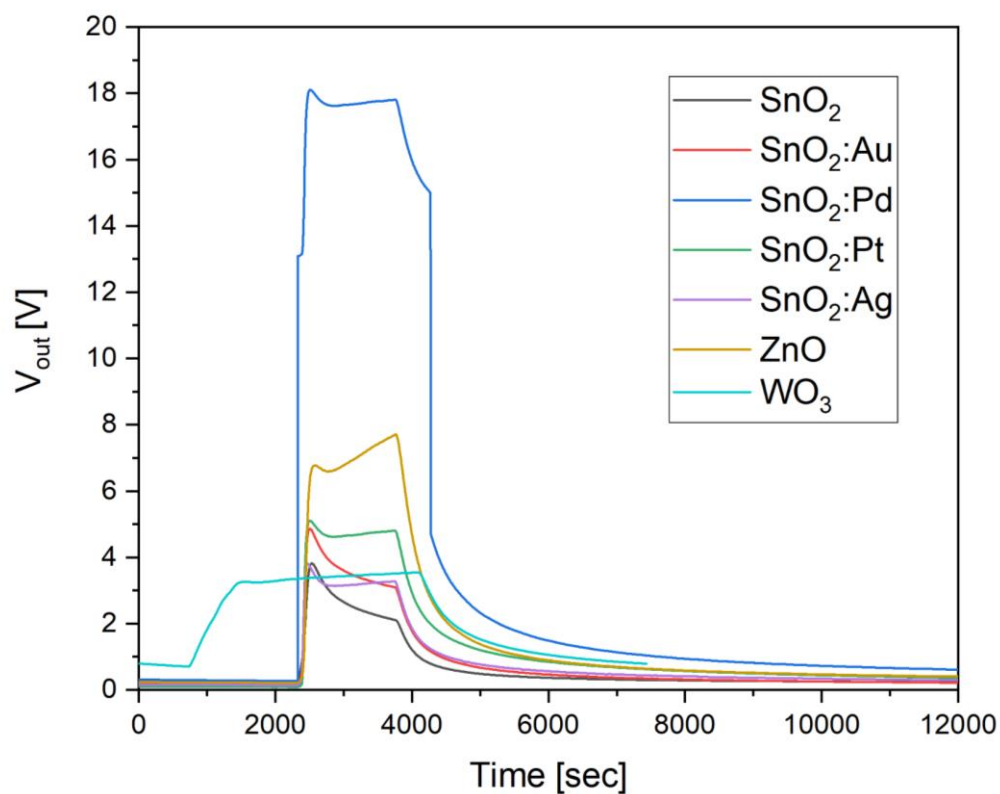


Figure S5. V_{out} response vs. 5 ppm of ethanol. In the case of WO_3 , the injection of the gas was different due to a different day of measurement.

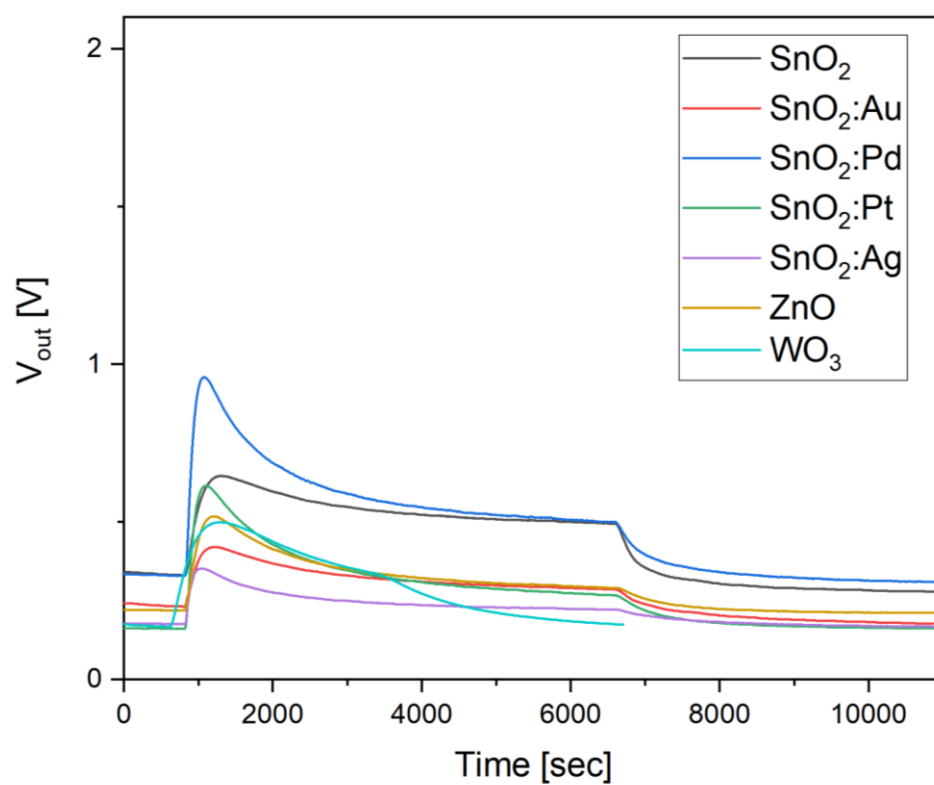


Figure S6. V_{out} response vs. 5000 ppm of CO_2 . In the case of WO_3 , the injection of the gas was different due to a different day of measurement.

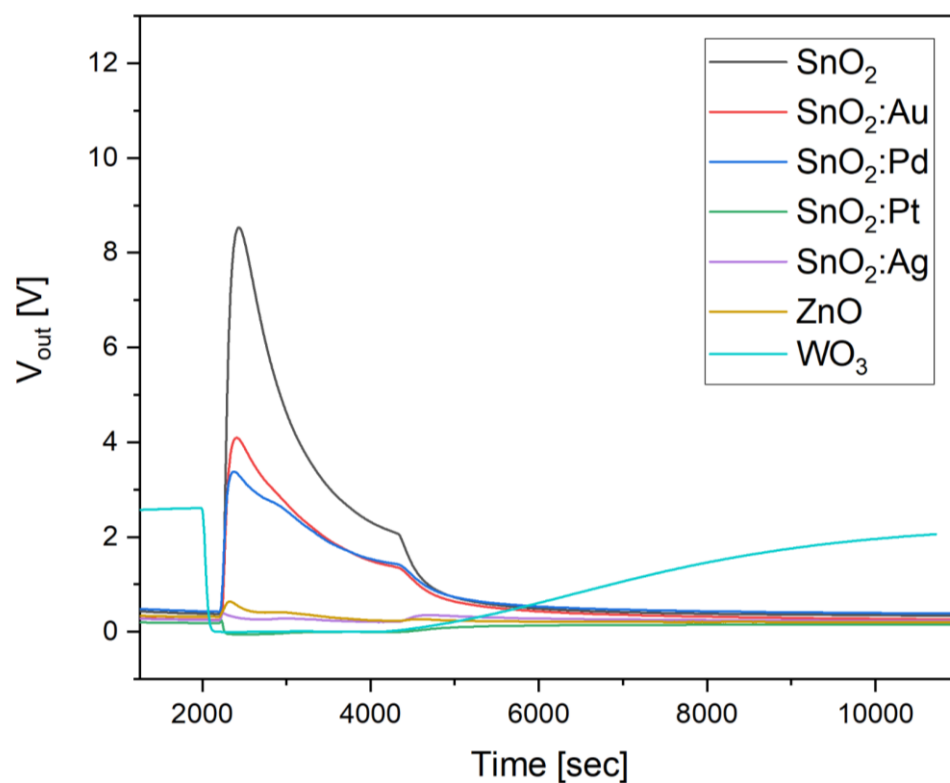


Figure S7. V_{out} response vs. 3 ppm of NO_2 . In the case of WO_3 , the injection of the gas was different due to a different day of measurement.

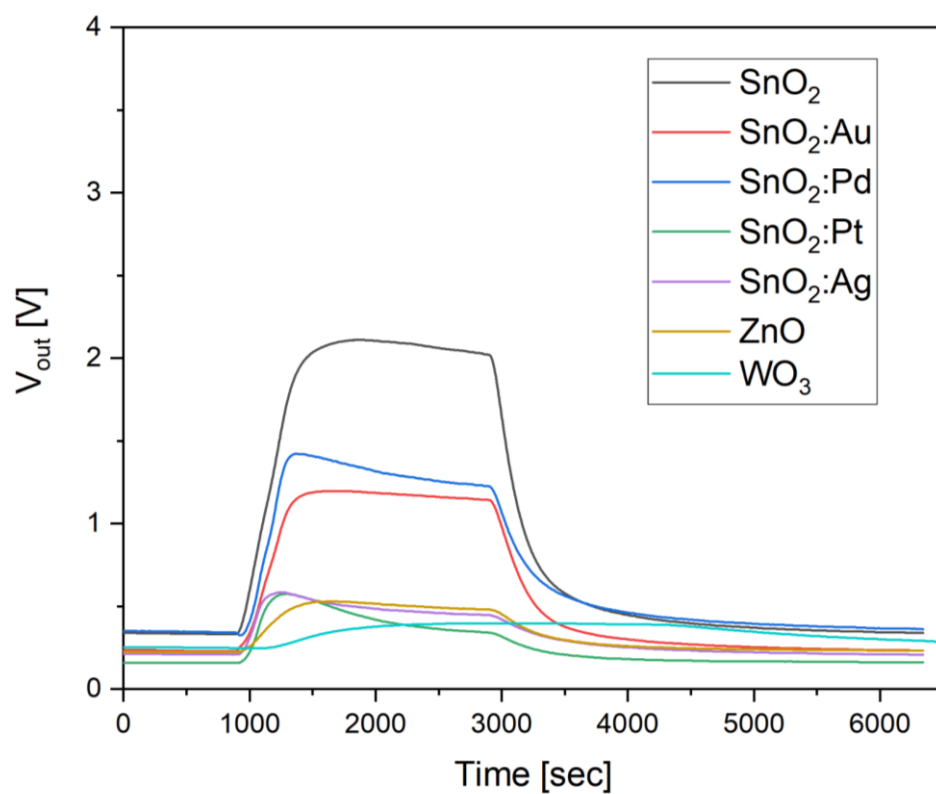


Figure S8. V_{out} response vs. 0.5 ppm of benzene. In the case of WO_3 , the injection of the gas was different due to a different day of measurement.

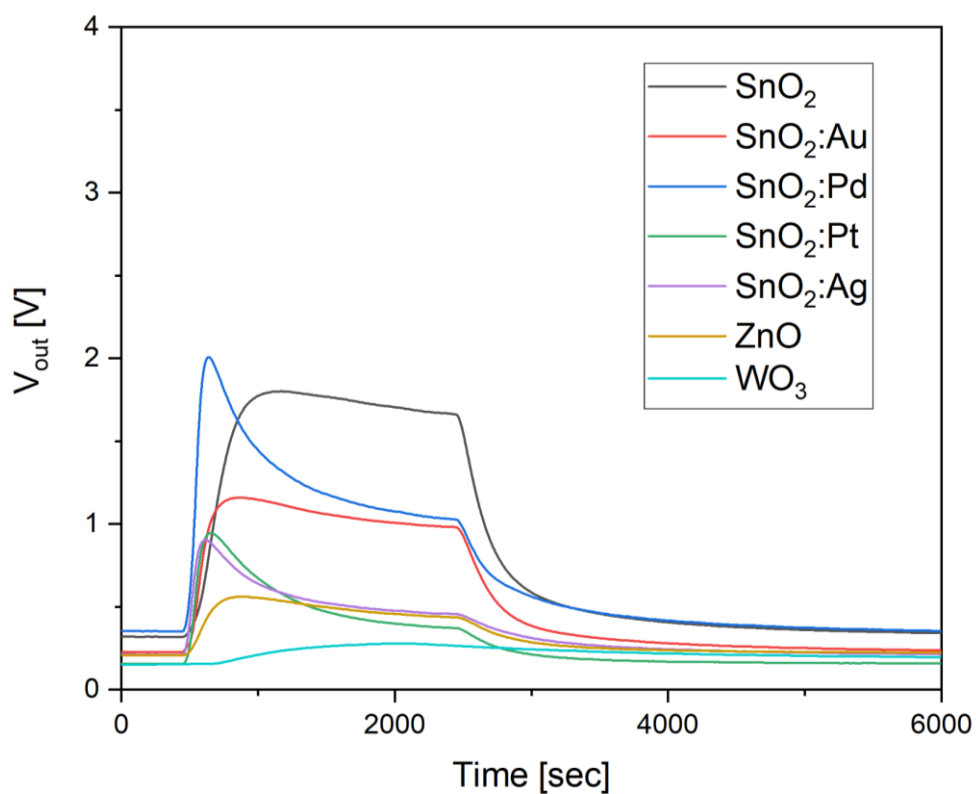


Figure S9. V_{out} response vs. 25 ppm of CO. In the case of WO_3 , the injection of the gas was different due to a different day of measurement.

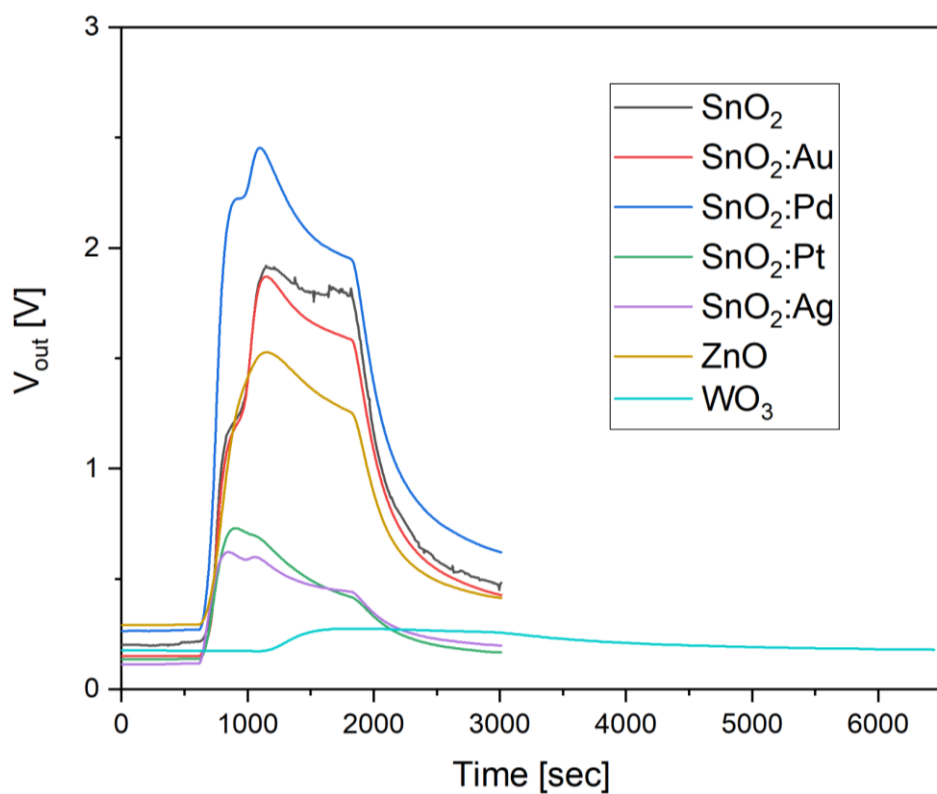


Figure S10. V_{out} response vs. 5 ppm of SO_2 . In the case of WO_3 , the injection of the gas was different due to a different day of measurement.