



# Article Effects of Different Ethanol/Diesel Blending Ratios on Combustion and Emission Characteristics of a Medium-Speed Diesel Engine

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Abstract: In order to better evaluate the effects of ethanol/diesel blends on engine combustion and emission characteristics, we developed an engine cylinder model using the software CONVERGE combined with the program CHEMKIN. The model was validated experimentally. A modified chemical kinetic mechanism was used to calculate the combustion process of diesel fuel and ethanol for the diesel engine, including 154 reactions and 68 species. Furthermore, the influence of different ethanol proportions on diesel engine combustion and emission characteristics, including power, brake specific fuel consumption, brake thermal efficiency, cylinder pressure, cylinder temperature, nitrogen oxide (NO<sub>x</sub>), carbon monoxide (CO), and soot emissions, was also investigated. Our results showed that cylinder pressure and temperature increased with increased ethanol content. When the ethanol content increased to 20% at 100% load, the cylinder pressure increased by 0.46%, and the thermal efficiency increased by 3.63%. However, due to the lower calorific value of ethanol, the power decreased by 4.12%, and the brake specific fuel consumption increased by 4.23%. In addition, the ethanol/diesel blends significantly reduced CO and soot emissions. Compared with diesel, soot and CO emissions from the D80E20 at 100% load reduced by 63.25% and 17.24%, respectively. However, NO<sub>x</sub> emission increased by 1.39%.

**Keywords:** diesel engine; ethanol/diesel blended fuel; CONVERGE; combustion and emission characteristics

# 1. Introduction

Diesel engines are extensively used in both agriculture and industry due to their high thermal efficiency [1], good fuel efficiency [2], and higher power [3]. Although the diesel engine has brought prosperity to the world economy, it also faces more serious problems such as energy shortages and environmental pollution [4]; this has been a major scientific research topic for diesel engine researchers seeking ways to save energy and reduce emissions [5]. In order to reduce human dependence on fossil energy and adapt to increasingly stringent emission regulations, research on the application of renewable and clean fuels in energy consumption continues to receive extensive attention [6]. Biomass fuels represented by ethanol show great potential in reducing diesel consumption and exhaust gas emission. Some even completely replace fossil fuels and are applied to diesel engines [7,8].

Ethanol has a certain calorific value and sufficient combustion; it can also be used as a clean oxygenated fuel [9]. Since it is inexpensive and easy to obtain, ethanol is suitable as an additional fuel blended with diesel [10]. The combustion flame propagation speed of ethanol is fast [11], the combustion efficiency is high, the combustion quality is good [12], and the incomplete combustion products CO and hydrocarbon (HC) are fewer [13]. Compared with diesel, ethanol has a higher latent heat of vaporization. Vaporization can reduce



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**Copyright:** © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). the maximum combustion temperature in the cylinder and reduce  $NO_x$  emission [14]. Simultaneously, the pressure in the cylinder reduces, the air intake and charge coefficient increase, and the diesel power performance improves [15].

As an alternative fuel to petroleum, ethanol is a hot spot in contemporary research. Many studies show that [16–21] oxygenated fuel has many advantages over traditional fuel in optimizing emissions. Ethanol is a typical oxygenated fuel [22]. Ethanol preparation uses cellulose and other wastes as raw materials that belong to renewable energy. Biofuel ethanol decreases harmful emissions such as carbon dioxide  $(CO_2)$  and particles matter (PM), CO, and HC in motor vehicle exhaust [23], which is conducive to improving the ecological environment [24]. As an oxygenated renewable fuel, ethanol has been widely studied and is suitable for improving diesel engine emissions and replacing fossil fuels [25]. Dongle et al. [26] studied the impact of fuel stratification on the combustion characteristics of diesel/ethanol dual-fuel engines. Their results showed that ethanol prolonged the ignition delay period. Pedrozo et al. [27] studied the impact of diesel/ethanol dual fuel on engine combustion characteristics and emissions. According to their results, the net indicated that the thermal efficiency of the dual-fuel engine was higher than the traditional diesel engine, with an average effective pressure range between 0.6 and 2.4 MPa. In addition, the use of the Miller cycle further improved net efficiency and reduced NO<sub>x</sub> emissions from diesel/ethanol dual-fuel engines.

In summary, the blending of ethanol and diesel can improve the combustion and emission characteristics of the engine significantly. In addition, with the development of computers, numerical engine simulation has become an important means of engine research and development [28,29]. Currently, widely used simulation software include ANSYS Fluent, CONVERGE, AVL-Fire, AVL-BOOST, etc. Fan et al. [30] studied the mixing formation and combustion of direct hydrogen injection and natural gas rotary injection engines using the software ANSYS FLUENT. Their results showed that with the increase in injection time, the hydrogen stratification became more apparent. Zhang et al. [31] used the software CONVERGE to study the influence of different fuel blending ratios on diesel engine spray, combustion, and emission characteristics. Gong et al. [32] simulated the effects of injection and ignition timing on engine combustion, in-cylinder formaldehyde, and unburned methanol emissions during the cold start of a methanol engine using AVL-Fire. Poorghasemi et al. [33] studied the effects of diesel injection strategy on NG/diesel RCCI combustion and emission characteristics using CONVERGE. Their results showed that changing the injection strategy reduces  $NO_x$  emissions. However, there are few studies on the effects of low ethanol concentrations in diesel on fuel concentrations using mediumspeed diesel engines combined with detailed chemical kinetic mechanisms [34,35].

In summary, we used the 3D computational fluid dynamics (CFD) software CON-VERGE combined with the CHEMKIN program to numerically simulate the in-cylinder combustion process of a diesel engine. Our experiments were performed on a mediumspeed diesel engine, and these experiments verified the validity of the simulation model under different operating conditions. Finally, the impact of diesel/ethanol blended fuel on engine combustion and emission characteristics were investigated at different loads (25%, 50%, and 100%) and blending ratios (D100, D95E5, D90E10, D85E15, and D80E20) on diesel/ethanol blended fuels with a high ethanol blending ratio.

### 2. Materials and Methods

# 2.1. Mathematical Model

The commercial computational fluid dynamics (CFD) software CONVERGE is used to numerically simulate a diesel engine. CONVERGE automatically generates high-quality grids and using this software saves time and research costs. CONVERGE has a variety of geometric, fluid, and chemical solvers and provides a variety of options for important submodels such as spray, combustion, turbulence, and emissions. CONVERGE can calculate the engine combustion and emission performance accurately and numerically simulate internal combustion engines.

### 2.1.1. Turbulence Model

There is a complex turbulent environment inside the cylinder of an internal combustion engine. The presence of turbulence increases the thickness of the flame and the combustion and heat release speed [36]. The RNG (Re-normalization group) k- $\varepsilon$  model and the wall function model for turbulent flow were used to numerically predict the heat transfer through the boundary [37]. The expression is as follows [38]:

$$\frac{\partial(\rho k_t)}{\partial t} + div(\rho k U) = div[\alpha_k \mu_{eff} \text{grad } k_t] + \tau_{ij} \cdot S_{ij} - \rho \varepsilon$$
(1)

$$\frac{\partial(\rho\varepsilon)}{\partial t} + div(\rho\varepsilon U) = div[a_{\varepsilon}\mu_{eff}\text{grad }\varepsilon] + C_{1\varepsilon}^{*}\frac{\varepsilon}{k_{t}} - \tau_{ij} \cdot S_{ij} - C_{2\varepsilon}\rho\frac{\varepsilon^{2}}{k_{t}}$$
(2)

with

$$\tau_{ij} = -\rho \overline{u'_i u'_j} = 2\mu_t S_{ij} - \frac{2}{3}\rho k_t \delta_{ij}$$
(3)

and

$$\mu_{eff} = \mu + \mu_t \ \mu_t = \rho C_\mu \frac{k_t^2}{\varepsilon} \tag{4}$$

$$C_{1\varepsilon}^* = C_{1\varepsilon} - \frac{\eta (1 - \frac{\eta}{\eta_0})}{1 + \beta \eta^3} \eta = \frac{k_t}{\varepsilon} \sqrt{2S_{ij} \cdot S_{ij}}$$
(5)

where  $\rho$  is the density, g/cm<sup>3</sup>;  $k_t$  is the turbulent kinetic energy, m<sup>2</sup>/s<sup>2</sup>; U is the fluid velocity, m/s;  $S_{ij}$  is the turbulent source term;  $\varepsilon$  is the turbulent dissipation rate; and  $\mu$  is the molecular viscosity, Pa·s. The Kronecker delta  $\delta_{ij}$  is provided by  $\delta_{ij} = 1$  if i = j and  $\delta_{ij} = 0$  if  $I \neq j$ .  $C_{1\varepsilon} = 1.42$ ,  $C_{2\varepsilon} = 1.68$ ,  $\alpha_k = \alpha_{\varepsilon} = 1.39$ ,  $C_{\mu} = 0.0845$ ,  $\eta_0 = 4.377$ ,  $\beta = 0.012$ .

### 2.1.2. Spray Breakup Model

For the spray model, we selected the widely used KH-RT model in diesel spray. The spray comprises a liquid core area and a gas–liquid mixing area. The KH model describes the breaking phenomenon of the liquid core area, and the KH and RT models describe the breaking phenomenon of droplets in the gas–liquid mixing area [39]. The intact core or breakup length  $L_b$  can be specified as [40]:

$$L_b = C_{bl} \sqrt{\frac{\rho_l}{\rho_g}} d_0 \tag{6}$$

In the range of the broken length  $L_b$ , the crushing process is controlled by the KH crushing model. Outside the range of  $L_b$ , it is jointly described by the KH and RT models. At this point, CONVERGE first uses the RT model to control the droplet breakage; if the breakage cannot be achieved, the KH model is used [39].

$$\tau_{KH} = \frac{B_1}{U} \sqrt{\frac{\rho_1}{\rho_g}} r_0 \tag{7}$$

### 2.1.3. Combustion Model

CONVERGE contains a variety of classic combustion models. In this article, we used the SAGE combustion model [41], which uses a set of input files in the CHEMKIN format to model detailed chemical kinetics. The SAGE combustion mode uses a chemical reaction mechanism including 68 species and 154 reactions. These 68 species contain species related to ethanol composition (including ethanol), and the 154 reaction species contain the oxidation reaction involved in ethanol. Different chemical mechanisms can describe the

combustion of different fuels. According to Turns [42], a multi-step elementary reaction mechanism can be expressed as:

$$\sum_{m=1}^{M} v'_{m,i} x_m = \sum_{m=1}^{M} v''_{m,i} x_m \text{ for } i = 1, 2, \dots, I$$
(8)

where  $v'_{m,i}$  and  $v''_{m,i}$  are the stoichiometric coefficients for the reactants and products for species *m* and reaction *i*, respectively; *I* is the total number of reactions; and  $x_m$  is the chemical symbol for species *m*. The net production rate of species  $\omega_m$  is provided by:

$$\dot{\omega}_m = \sum_{i=1}^{I} v_{m,i} q_i \text{ for } m = 1, 2, \dots, M$$
 (9)

where *M* is the total number of species and

$$v_{m,i} = v_{m,i}'' - v_{m,i}' \tag{10}$$

The reaction rate parameter  $q_i$  of the *i*th reaction is

$$q_i = k_{i,f} \prod_{m=1}^{M} [X_m]^{v'_{m,i}} - k_{i,r} \prod_{m=1}^{M} [X_m]^{v''_{m,i}}$$
(11)

where [Xm] is the molar concentration of species *m*, and  $k_{i,f}$  and  $k_{i,r}$  are the forward and reverse rate coefficients for reaction *i*. In SAGE, the forward rate coefficient is expressed in the Arrhenius form as:

$$k_{i,f} = A_i T^{\beta i} \exp\left(\frac{-E_i}{RT}\right) \tag{12}$$

where  $A_i$  is the pre-exponential factor;  $\beta_i$  is the temperature exponent;  $E_i$  is the activation energy, cal/mol; R is the ideal gas constant; and T is the temperature, K. In addition, the reverse rate coefficient can be specified in an analogous fashion as in Equation (12), or calculated from the equilibrium coefficient  $K_{i,c}$  as:

$$K_{i,c} = \frac{k_{i,f}}{k_{i,r}} \tag{13}$$

The equilibrium coefficient  $K_{i,c}$  is determined by thermodynamic properties,

$$K_{i,c} = K_{i,p} \left(\frac{P_{atm}}{RT}\right)^{\sum_{m=1}^{M} u_{m,i}}$$
(14)

where  $P_{atm}$  is the atmospheric pressure, Pa. The equilibrium constant  $K_{i,p}$  is obtained via

$$K_{i,p} = \exp\left(\frac{\Delta S_i^1}{R} - \frac{\Delta H_i^1}{RT}\right)$$
(15)

The  $\Delta$  refers to the change that occurs in passing completely from reactants to products in the *i*th reaction, specifically

$$\frac{\Delta S_i^1}{R} = \sum_{m=1}^M u_{m,i} \frac{S_m^1}{R}$$
(16)

$$\frac{\Delta H_i^1}{RT} = \sum_{m=1}^M u_{m,i} \frac{H_m^1}{RT} \tag{17}$$

where  $S_{de}$  denotes entropy, J/(mol·K); and  $H_{de}$  denotes enthalpy, kJ/mol.

### 2.1.4. NO<sub>x</sub> Model

According to the working process characteristics of the internal combustion engine cylinder, it is generally believed that thermal NO is the main source of  $NO_x$ . The classic thermodynamic NO generation mechanism is the Zeldovich mechanism [43]. Afterward, a reaction path of a hydroxyl group reacting with a nitrogen atom to generate NO is added to form an extended Zeldovich mechanism. This mechanism is expressed by the following three reactions:

$$O + N_2 \Leftrightarrow NO + N$$
 (18)

$$O_2 + N \Leftrightarrow NO + O$$
 (19)

$$OH + N \Leftrightarrow NO + H$$
 (20)

$$2NO + O_2 \Leftrightarrow 2NO_2 \tag{21}$$

### 2.1.5. Soot Model

Hydrocarbon fuel undergoes a series of physical and chemical processes under high temperature and oxygen deficiency conditions to form soot particles; however, most of the soot will be oxidized before the exhaust valve opens. Developed according to the cylinder's soot generation and oxidation characteristics, a simple empirical model has been successfully used in predicting soot emission from internal combustion engines. This soot model was used in CONVERGE as a semiempirical two-step model. The net generation rate of soot is described as [44]:

$$\frac{dM_{soot}}{dt} = \frac{dM_{ae}}{dt} - \frac{dM_{so}}{dt}$$
(22)

where  $M_{soot}$  is the total amount of soot, kg;  $M_{ae}$  is the amount of soot generation, kg; and  $M_{so}$  is the amount of soot oxidation, kg. The soot generation rate model is shown as follows [44]:

$$\frac{dM_{sf}}{dt} = A_f p_p^{0.5} \exp(-\frac{E_{ae}}{RT}) \cdot M_{fv}$$
(23)

where  $A_f$  refers to the former factor;  $p_p$  is the pressure;  $E_{ae}$  is the activation energy; and  $M_{fv}$  is the quality of fuel vapor. Acetylene (C<sub>2</sub>H<sub>2</sub>), the precursor of soot, is generally used instead of fuel vapor when the chemical reaction mechanism simulates the ignition and combustion process [44].

### 2.2. Establishment of Simulation Model

In this study, the one-dimensional simulation verification of the engine was conducted using AVL-BOOST to determine the initial conditions of the engine. The three-dimensional CFD model of the diesel engine was created in CONVERGE according to the actual shape of the diesel engine. The 3-D CFD simulation model of cylinder is shown in Figure 1.

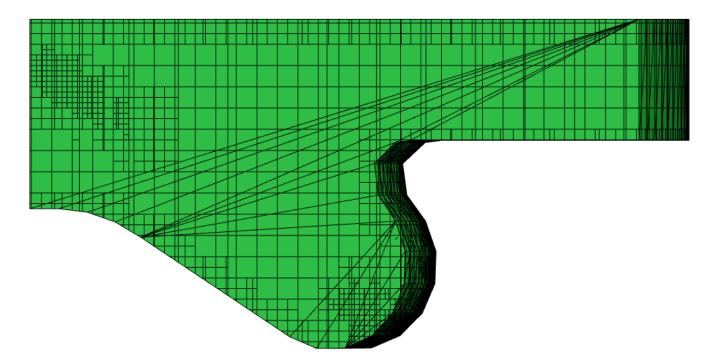


Figure 1. The geometric model of the combustion chamber in CONVERGE environment.

### 2.3. Computational Mesh

According to the geometry of the four-cylinder four-stroke Marine diesel engine block, a dynamic mesh is generated by the symmetrical distribution of eight nozzle holes. Fine mesh partitioning ensures that the model can accurately predict droplet breakage and evaporation. Figure 2 shows the cylinder pressure generated by three grid models when pure diesel is fully loaded. There is no significant difference in-cylinder pressure between the thin and medium grilles. Since the intermediate grid guarantees calculation accuracy and saves calculation time, the optimal intermediate grid was selected. The intermediate grid was considered the best solution for this study.

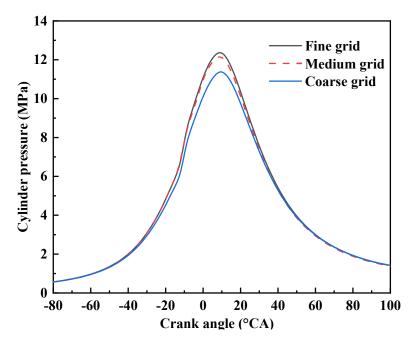


Figure 2. Comparison between in-cylinder pressure curves of different basic grids.

# 3.1. Fuel Properties

Diesel and ethanol were used in this study. Diesel/ethanol blended fuels were formed by blending different ratios of ethanol into diesel fuel, and their combustion and emission characteristics were analyzed. The fuel properties are shown in Table 1. In this article, D100 represents pure diesel, D95E5 represents 95% diesel and 5% ethanol, D90E10 represents 90% diesel and 10% ethanol, D85E15 represents 85% diesel and 15% ethanol, and D80E20 represents 80% diesel and 20% ethanol.

Table 1. The properties of diesel/ethanol blended fuel.

Fuel	Diesel	Ethanol
Chemical formula	-	C <sub>2</sub> H <sub>5</sub> OH
Boiling temperature (°C)	190–280	64.7
Oxygen volume fraction (%)	0.0	34.7
Density (g/m <sup>3</sup> , at 20 $^{\circ}$ C)	0.82-0.86	0.79
Viscosity (MPa·s, at 20 °C)	3.0-8.0	1.2
Flash point ( $^{\circ}$ C)	65–88	13-14
Cetane number (CN)	45-65	8
Low calorific value (MJ $kg^{-1}$ )	42.5-42.8	26.8

# 3.2. Engine Specifications

The test was conducted on a marine medium-speed diesel engine. The main parameters of the diesel engine are shown in Table 2. The schematic diagram of the experimental system is shown in Figure 3.

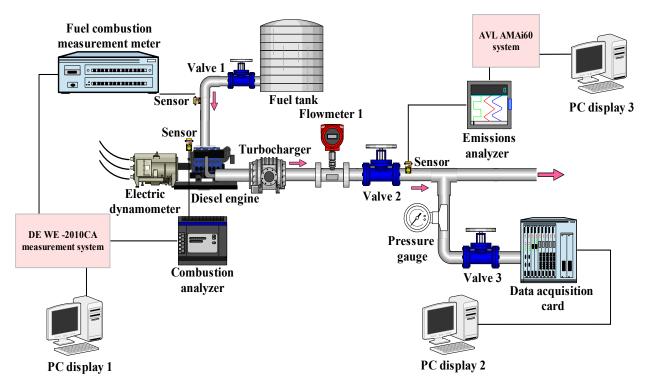


Figure 3. Schematic diagram of engine test.

Туре	Value	Туре	Value
Bore $\times$ stroke (mm)	$190 \times 210$	Compression ratio	14
Number of cylinders	4	Head temperature (K)	553
Engine speed (rpm)	2000	Piston temperature (K)	423
Effective power (kW)	220	Wall temperature (K)	433
Nozzle radius (mm)	0.26	Temperature at IVC (K)	341
Fuel injection holes	8	Pressure at IVC (bar)	1.97
Cylinder diameter (MPa)	2.05	Turbulent kinetic energy (m <sup>2</sup> /s <sup>2</sup> )	62.0271
Connecting rod (mm)	410	Turbulent dissipation $(m^2/s^3)$	17,183.4

Table 2. Engine specifications and boundary conditions.

Total uncertainty of experiment = Square root of [(uncertainty of pressure sensor)<sup>2</sup> + (uncertainty of BSFC)<sup>2</sup> + (uncertainty of BTE)<sup>2</sup> + (uncertainty of Brake power)<sup>2</sup> + (uncertainty of NO<sub>x</sub> emission)<sup>2</sup> + (uncertainty of CO emission)<sup>2</sup> + (uncertainty of Soot emission)<sup>2</sup>] = Square root of [(1.5%)<sup>2</sup> + (1.7%)<sup>2</sup> + (0.53%)<sup>2</sup> + (0.3%)<sup>2</sup> + (0.53%)<sup>2</sup> + (0.53%)<sup>2</sup> + (0.32%)<sup>2</sup> + (2.8%)<sup>2</sup>] = 3.706%

### 3.3. Uncertainty Analysis

Generally, experimental measurement results contain some errors and uncertainties [45]. The uncertainty of experimental results is caused by the selection, observation, and calibration of sensors. The measurement results are used to calculate the required experimental results. The uncertainty percentage of parameters such as BSFC, BTE, and NO<sub>x</sub> can be obtained from Equation (24). The *R* in Equation (25) is a function of the independent variables  $I_1, I_2, \ldots, I_n$ . In Equation (24),  $u_1, u_2, \ldots, u_n$  represents uncertainty in the independent variable and  $U_R$  represents uncertainty in the result.

$$U_{R} = \left\{ \left[ (\partial R/\partial I_{1})u_{1} \right]^{2} + \left[ (\partial R/\partial I_{2})u_{2} \right]^{2} + \dots + \left[ (\partial R/\partial I_{n})u_{n} \right]^{2} \right\}^{1/2}$$
(24)

$$R = \{I_1, I_2, I_3, \cdots I_n\}$$
(25)

4 10

The measuring range and accuracy of the measuring equipment we used in this article are shown in Table 3. The following equation was used to calculate the total experimental uncertainty.

Table 3. Uncertainty of measured parameters.	
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Measurements.	Measuring Range	Accuracy	Uncertainty (%)
Engine speed	1–2000 rpm	±0.2%	$\pm 0.24$
Pressure sensor	0–25 MPa	$\pm 10 \text{ kPa}$	$\pm 0.5$
Exhaust gas temperature	0–1000 °C	±1 °C	$\pm 0.25$
NO <sub>x</sub> emission	0–5000 ppm	$\pm 10 \text{ ppm}$	$\pm 0.53$
CO emission	0–10% vol	$\pm 0.03\%$	$\pm 0.32$
Soot emission	0-9 FSN	$\pm 0.1$ FSN	$\pm 2.8$
BSFC	-	$\pm 5 \text{ g/kW} \cdot \text{h}$	$\pm 1.5$
BTE	-	$\pm 0.5\%$	$\pm 1.7$
Brake power	-	$\pm 0.03$ kW	$\pm 0.3$
Air flow rate	0–33.3 kg/min	$\pm 1\%$	$\pm 0.5$
Fuel flow rate	0.5–100 L/h	$\pm 0.04$ L/h	$\pm 0.5$

# 3.4. Model Validation

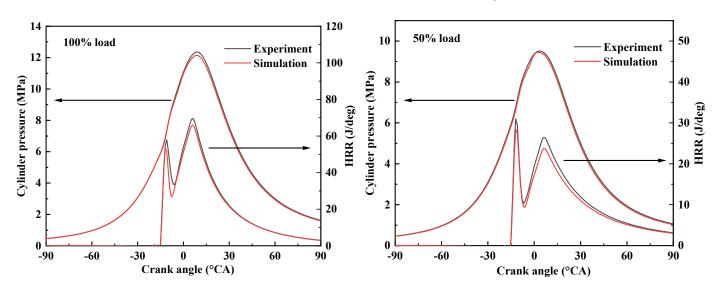
In order to verify the accuracy of the model, the simulation results were compared to the experimental results. During the experiment, an exhaust gas analyzer (Horiba MEXA-1600) was used to measure the generated nitrogen oxide with an error of 1%. A fuel consumption meter (FCMM-2) was used to measure the brake specific fuel consumption.

A combustion analyzer (DEWE-2010CA) was used to monitor the diesel engine's combustion. An Opacimeter (AVL Dismoke-4000) was used to measure the smoke generated. The EFS-IFR600 was used to measure the fuel injection rate with a measurement error of 0.5%. A hydraulic dynamometer was used to measure the load of the diesel engine. An electronic control unit (ECU) system was used to control the electronically controlled diesel engine. In addition, appropriate sensors were used to measure temperature, flow, and pressure. Detailed specifications for the main measuring instruments are shown in Table 4.

Item	Content	Accuracy	Uncertainty (%)
Electric dynamometer	NIDY S22-2/0525-1BV-1	Torque: $\pm 0.5\%$ F.S; Speed: $\pm 1$ r/min	$\pm 0.2$
Dynamometer control system	PUMA OPEN1.4.1	$\pm 0.5\%$ F.S	$\pm 0.2$
Air flowmeter	TOCEIL 20N125	$\pm 1\%$	$\pm 0.15$
Diesel flowmeter	TOCEIL CMFG010	0.11%	$\pm 0.15$
Temperature sensor	Thermojunction type	$\pm 0.5~^\circ\mathrm{C}$	$\pm 0.15$
Pressure sensor	Piezoresistance type	$\pm 0.5\%$ F.S	$\pm 0.5$
Emissions analyzer	AVL AMAi60	$\pm 1.0\%$ F.S	$\pm 0.25$
Combustion analyzer	DEWE-2010CA	/	$\pm 0.25$
Injection measuring instrument	EFS-IFR600	$\pm 0.5\%$	$\pm 0.5$

Table 4. Specifications of main measuring instruments.

The experiment was conducted on a four-cylinder four-stroke diesel engine. Tests were conducted at 200 rpm for 50% and 100% loads, respectively. The diesel engine was fueled with D90E10 blended fuel. The experimental results and simulation results are shown in Figure 4. As seen in Figure 4, the error between the experimental results and the simulation results is minor at less than 5%. Therefore, the model can accurately predict the combustion and emission characteristics of diesel engines.



**Figure 4.** Comparison between experimental and simulation results of engine cylinder pressure and heat release rate.

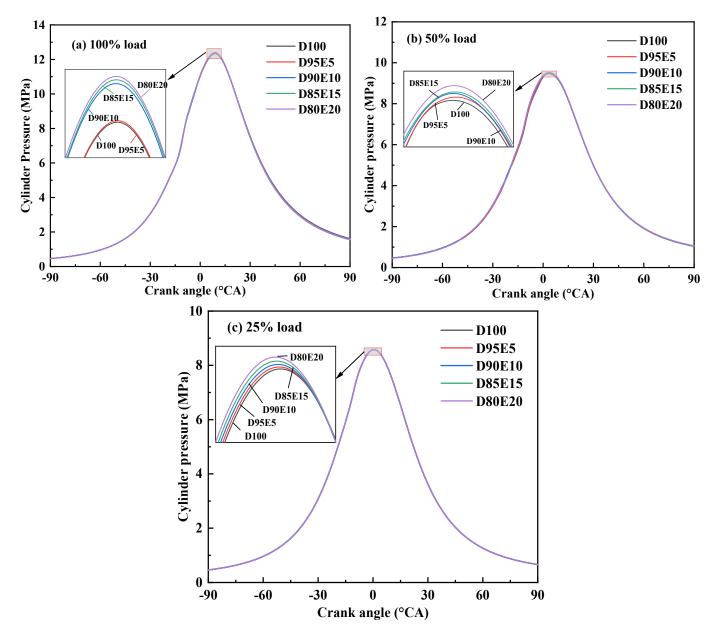
# 4. Results and Discussion

4.1. Engine Combustion Performance

4.1.1. Cylinder Pressure

Figure 5 shows the effects of diesel/ethanol blended fuel with different blending ratios on the engine's cylinder pressure at different loads (including 100% load (see Figure 5a), 50% load (see Figure 5b), 25% load (see Figure 5c)). The cylinder pressure of D80E20 is the

highest, followed by D85E15, D90E10, D95E5, and D100. As the ethanol ratio increases in the blended fuel, the cylinder's maximum combustion pressure increases gradually. For example, when the ethanol ratio in the blended fuel increases to 20%, the cylinder pressure increases by 0.46% because the cetane number of ethanol is low, leading to the extension of the mixed fuel's ignition delay period, the delay of the ignition time, and the increase in the mixture amount formed during the ignition delay period. In addition, ethanol is an oxygenated fuel that provides more oxygen while participating in the incylinder combustion reaction and effectively promotes combustion. Therefore, the pressure in the cylinder increases. Similar conclusions were drawn by Liang et al. [15] and Chen et al. [7].

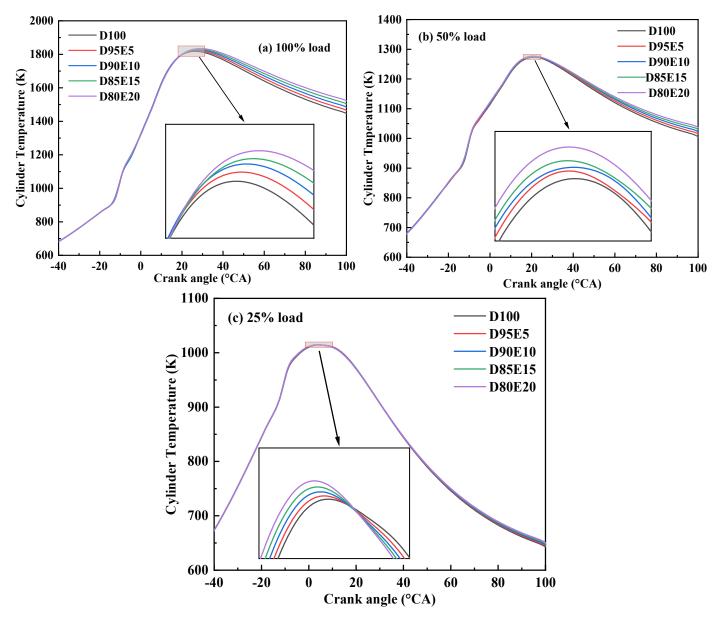


**Figure 5.** Effects of diesel/ethanol blended fuels with different blending ratios on cylinder pressure under different loading conditions.

### 4.1.2. Cylinder Temperature

Figure 6 shows the effects of diesel/ethanol blended fuel with different blending ratios on the engine's cylinder temperature at different loads (including 100% load (see Figure 6a), 50% load (see Figure 6b), 25% load (see Figure 6c)). As the ratio of ethanol in the blended

fuel increases, the maximum cylinder temperature also increases. When the ethanol ratio in the blended fuel increases to 5%, 10%, 15%, and 20%, the peak temperature of the cylinder becomes 0.28%, 0.52%, 0.65%, and 0.94% higher than diesel, respectively; the main reason for this is that ethanol is an oxygenated fuel. As the ethanol ratio increases, the oxygen content of blended fuel also increases, and the decrease in viscosity can better facilitate the atomization of fuel into the combustion chamber. Simultaneously, the higher oxygen content increases the combustion rate of blended fuel, improving the intensity of premixed combustion, and sufficient combustion. Therefore, the cylinder temperature increases. Zhang et al. [35] drew similar conclusions.



**Figure 6.** Effects of diesel/ethanol blended fuels with different blending ratios on temperature under different loading conditions.

Figure 7 shows the distribution field of the temperature in-cylinder at 100% load. Compared with D100, diesel/ethanol blended fuel produces more high-temperature areas because ethanol's lower density and surface tension promote fuel atomization and sufficient combustion.

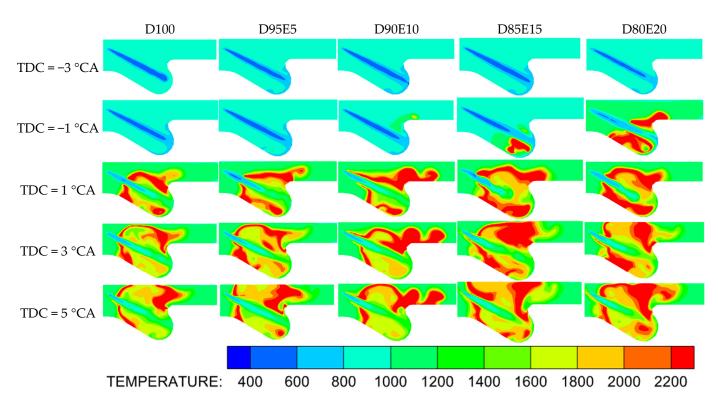


Figure 7. The distribution field of the temperature in-cylinder at 100% load. [K].

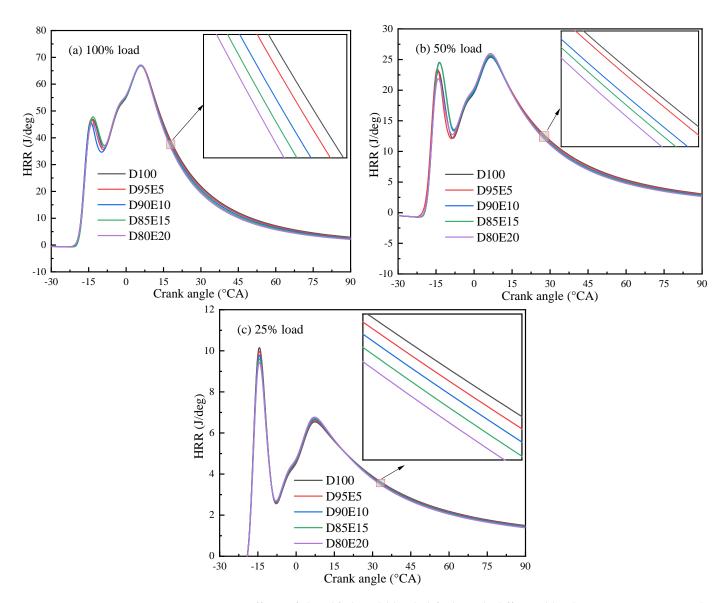
### 4.1.3. Heat Release Rate

Figure 8 shows the effects of diesel/ethanol blended fuel with different blending ratios on the engine's heat release rate(HRR) at different loads(including 100% load (see Figure 8a), 50% load (see Figure 8b), 25% load (see Figure 8c)). The overall HRR of diesel/ethanol blended fuel increases as the ethanol blending ratio increases because of the low cetane number and high latent heat of vaporization in ethanol, which leads to a longer ignition delay period of the blended fuel, increases the amount of combustible mixture formed in the premixing stage, improves the engine combustion conditions, and allows the blended fuel to burn fully. Liu et al. [46] evaluated the effects of E0, E10, and E20 on the engine. Their conclusion also confirmed that HRR increases with the increase in ethanol ratio.

### 4.2. Economic Performance

# 4.2.1. Brake Specific Fuel Consumption

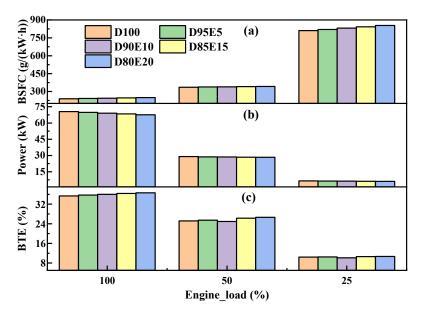
Brake specific fuel consumption (BSFC) is an important index for measuring the economic performance of an engine [47]. Figure 9a shows the effects of diesel/ethanol blended fuel with different blending ratios on the engine's BSFC at different loads. The diesel engine's fuel consumption gradually as the ethanol ratio increases in blended fuel. For example, when pure diesel is used at 25% load, the fuel consumption of the engine is 810.79 g/(kW·h). As the ethanol content in the blended fuel increases to 5%, 10%, 15%, and 20%, the fuel consumption increases to 820.22/(kW·h), 831.88/(kW·h), 842.76/(kW·h), and 853.91/(kW·h), respectively; this is because the calorific value of diesel fuel (42.5 MJ/kg) is higher than the calorific value of ethanol (26.8 MJ/kg), as shown in Table 1. An increase in the ratio of ethanol in blended fuel reduces the total calorific value of blended fuel, leading to an increase in BSFC.



**Figure 8.** Effects of diesel/ethanol blended fuels with different blending ratios on HRR under different loading conditions.

# 4.2.2. Brake Power

Figure 9b shows the effects of diesel/ethanol blended fuel with different blending ratios on the engine's brake power at different loads. The engine's power decreases gradually as the ethanol ratio increases in the blended fuel. Moreover, the higher the ethanol ratio is in the blended fuel, the more apparent the reduction in engine power. Compared with D100, the power of D95E5, D90E10, D85E15, and D80E20 blended fuel reduces by 0.96%, 1.97%, 2.99%, and 4.12%, respectively; this is because the calorific value of ethanol/diesel blended fuel is lower than D100, resulting in lower combustion power than diesel.



**Figure 9.** Effects of diesel/ethanol blended fuels with different blending ratios on (**a**) BSFC, (**b**) power, and (**c**) BTE under different loading conditions.

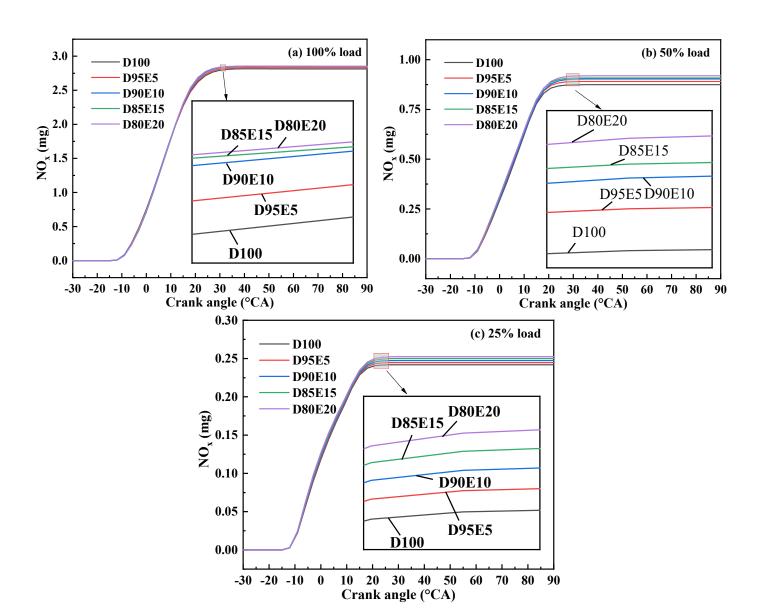
### 4.2.3. Brake Thermal Efficiency

Figure 9c shows the effects of diesel/ethanol blended fuel with different blending ratios on the engine's brake thermal efficiency (BTE) at different loads. BTE decreases with increased ethanol content at low loads and increases as the ethanol ratio increases at high loads because the increase in ethanol ratio in diesel/ethanol blended fuel improves the spray characteristics of the blended fuel. Since ethanol is an oxygenated fuel, it increases the oxygen content of the blended fuel, improving the combustion characteristics of the fuel, and further improving the BTE of the engine. Han et al. [48] investigated the dynamic characteristics of diesel/ethanol dual fuel engines and reached similar conclusions. They confirmed that the combustion efficiency and BTE of ethanol decreased under low load conditions in the dual fuel mode and increased under medium and high load conditions.

### 4.3. Emission Performance

# 4.3.1. NO<sub>x</sub> Emission

Figure 9 shows the effects of diesel/ethanol blended fuel with different blending ratios on NO<sub>x</sub> emission at different loads (including 100% load (see Figure 9a), 50% load (see Figure 9b), 25% load (see Figure 9c)). NO<sub>x</sub> emission increases with increased engine load; because the increase in engine load leads to increased peak temperature, which improves the emission level of NO<sub>x</sub>. In addition, the engine's NO<sub>x</sub> emission increases as the ethanol ratio increases in the blended fuel; this is because ethanol has low density, low viscosity, and high oxygen content characteristics, which improve combustion in the engine cylinder and produce more NO<sub>x</sub> in the local high-temperature area. Compared with the cylinder temperature cloud diagram in Figure 10, NO<sub>x</sub> generation is mainly affected by the temperature in the cylinder. The higher the temperature is, the more NO<sub>x</sub> emissions is, particularly in the concentrated temperature area. In addition, the cetane number of diesel/ethanol blended fuel is lower than D100, which leads to the extension of the ignition delay period, the increase of in-cylinder fuel injection, and premixed combustion, thereby increasing NO<sub>x</sub> emission. Similar conclusions were drawn by Sayin et al. [49] and Alptekin et al. [50].



**Figure 10.** Effects of diesel/ethanol blended fuels with different blending ratios on NO<sub>x</sub> emission under different loading conditions.

### 4.3.2. CO Emission

Figure 11 shows the effects of diesel/ethanol blended fuel with different blending ratios on the engine's CO emission at different loads. Compared with pure diesel, CO emission increases with increased ethanol content at a low load. At high loads, CO emission decreases with increased ethanol content; this is because at low load, the anal temperature is low, and the latent vaporization heat of ethanol is higher than diesel, resulting in more heat absorption in the combustion process of blended fuel, difficult complete combustion, and increased CO emission (see Figure 12). There are large amounts of fuel injection per cycle at high loads and higher temperatures in the cylinder. As an oxygenated fuel, ethanol can provide additional oxygen during combustion, effectively improving the problem of insufficient oxygen in the cylinder, creating sufficient combustion, and inhibiting the generation of CO. This observation is consistent with the conclusion of Le et al. [14].

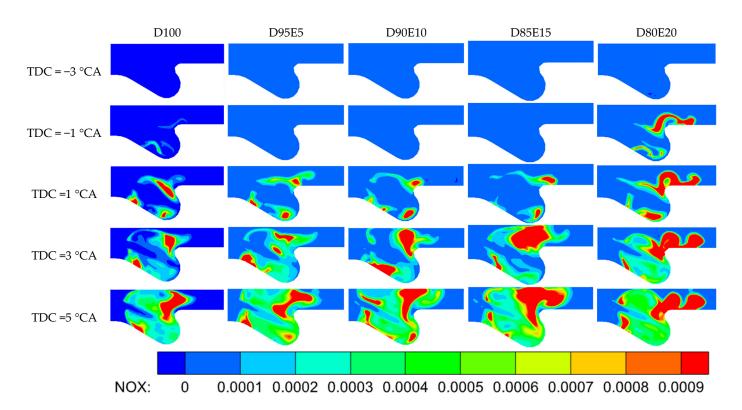
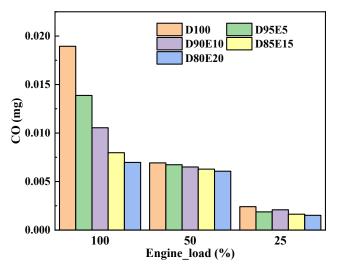


Figure 11. The NO<sub>x</sub> distribution field in the cylinder at 50% load [g].

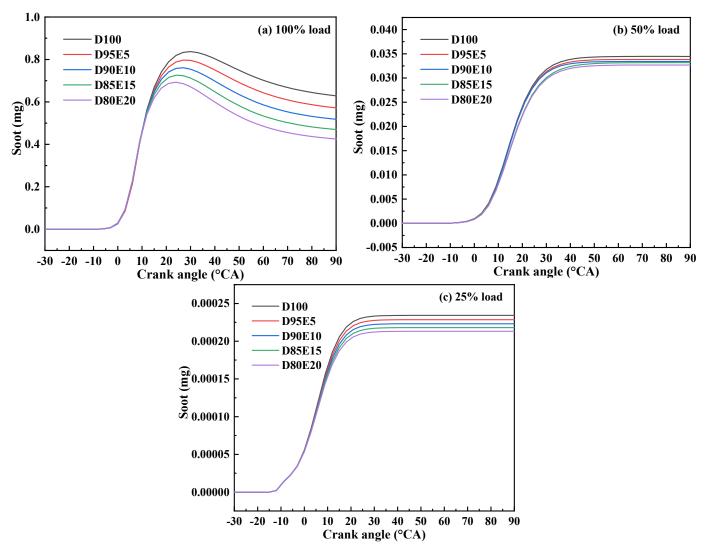


**Figure 12.** Effects of diesel/ethanol blended fuels with different blending ratios on CO emission under different loading conditions.

### 4.3.3. Soot Emission

Figure 13 shows the effects of diesel/ethanol blended fuel with different blending ratios on engine soot emissions at different loads (including 100% load (see Figure 13a), 50% load (see Figure 13b), 25% load (see Figure 13c)). Soot emission increases as the engine load increases because the oxygen in the cylinder is insufficient at higher loads and large amounts of soot form in the high-temperature anoxic area. In addition, soot emission decreases as the ethanol ratio increases in blended fuel. For example, under 100% load conditions, soot emission increases by 4.77%, 9.08%, 13.51%, and 17.24% when the ethanol content in blended fuels increases to 5%, 10%, 15%, and 20%, respectively. This increase is mainly because the combustion temperature in the cylinder is higher at high loads, and

the oil supply per cycle increases, enabling the formation of an over-concentrated mixture. At this time, the mitigation effects of ethanol on the local over-concentrated area in the cylinder and the combustion promotion effects of additional oxygen are more apparent and effectively reduce the soot emission under high load conditions.



**Figure 13.** Effects of diesel/ethanol blended fuels with different blending ratios on soot emission under different loading conditions.

# 5. Conclusions

Due to the global energy crisis [51–56] and environmental problems [57–61], the development of diesel engines is also facing great challenges. In this paper, our main conclusions are as follows:

(1) Diesel/ethanol blended fuels improve the brake thermal efficiency of the engine. Since ethanol is an oxygenated fuel, the oxygen content of the blended fuel increases, and the brake thermal efficiency of the engine improves. For example, when the ethanol content in the blended fuel increases to 20%, the cylinder pressure increases by 0.46%. In addition, compared with D100, the engine power of D80E20 reduces by about 4.12%; this is because ethanol has a high oxygen content and low calorific value;

(2) Diesel/ethanol blended fuel increases BSFC and reduces power. The blended fuels of D95E5, D90E10, D85E15, and D80E20 decrease by 0.96%, 1.97%, 2.99%, and 4.12%, respectively; this is mainly due to the low calorific value of ethanol. However, the BTE of the engine improves due to the increase in oxygen content;

(3) Diesel/ethanol blended fuel will lead to increase of the soot, the carbon monoxide, and the NO<sub>x</sub> emissions. Compared with D100, the CO and soot emissions of D80E20 increase by 63.25% and 17.24%, respectively. As the oxygen content of the blended fuel increases, the combustion is more sufficient, and the soot emission reduces.

In conclusion, adding ethanol to diesel engines can improve combustion and emission characteristics. This trend is worth studying. Further research is needed to obtain more accurate results.

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