


## Article

# Monitoring the Ignition of Hay and Straw by Radiant Heat

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**Abstract:** Hay and straw are commonly used materials in agriculture. They are organic materials and, therefore, flammable. This article examines the behaviour of hay and straw when exposed to radiant heat. The objective of this study is to experimentally determine the ignition temperature of hay and straw under the influence of radiant heat. This research investigates the effects of sample type (hay and straw) and sample quantity on the thermal degradation process, temperature increase within the samples, and ignition temperature of the samples as a function of time. The ignition temperature of hay was determined to be higher (407 °C) compared to straw (380 °C).

**Keywords:** hay; straw; radiant heat; ignition temperature

## 1. Introduction

Biomass, as a renewable energy source, is used as fuel [1–5], as a progressive natural insulation material in construction [6–8], and as animal feed and bedding in agriculture [9,10]. Hay and straw, which belong to the biomass group, are primarily utilized in agriculture [11,12]. For a cow weighing about 500 kg (see Figure 1), more than 3 tons of hay will be required during the stay in the stable (from October to April) [13]. Hay and straw are natural materials that are also used as natural thermal insulators [14–16]. However, it is necessary to acknowledge that these materials are flammable and not resistant to heat [17].



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(a)



(b)

**Figure 1.** Demonstration of hay (a) as feed for cows [13] and (b) as wending material [18].

Their basis is dried animal feed, which has a precise technological processing sequence, followed by packaging and storage [17]. The essence of the processing lies in drying and is based on the given moisture content (Table 1). The temperature increase is a consequence of the raw material processing by bacteria (fermentation) or the occurrence of decay. They both pose a fire hazard [19].

**Table 1.** Group of solid fuels according to Decree 258/2007 Coll [20].

Solid Flammable Substance	Characteristics	Moisture	Storage [17]
Dried animal feed (silage)	Mown green grasses	More than 16% and up to 30%	Bales, Haystack, Hayloft, Barn, Hay shed
	Mown green legumes	More than 16% and up to 35%	
Hay	Dried stems of grasses or legumes	Up to 16%	
Straw	Dried stalks of cereal crops	-	

The availability of water is a critical factor for microorganisms to function [21]; the lack of available water on solid substrates is a significant source of microbial stress [22,23]. Microorganisms need water for motility and transportation of nutrients to cells and waste products away from cells [21,24]. Native microbial activity in hay and other agricultural residues has been studied extensively in the context of self-heating to the point of spontaneous combustion.

Many of these studies have noted the importance of available moisture. For example, in hay, little or no microbial activity is expected below 25% to 30% on a wet basis (w.b.) [25–27]. Higher moisture contents should be conducive to more microbial activity because more bacteria would be able to thrive. In hay that was allowed to self-heat, the maximum temperature of a bale increased with increasing initial moisture content [27]. The increased ability to self-heat hay was attributed to higher levels of microbial activity.

Straw and hay are dominantly used as fodder [28,29]. Significant attention is paid to the research of safe storage of the mentioned materials [30–34] and their transport to livestock [35,36].

Some studies are devoted to the energy use of straw [37,38]. Considerable attention is paid to their economic demands [39] in the case of choosing the method of their use [40]. The straw and hay, as crop residues, represent a continuously increasing share of the farmer's income. Agricultural land produces large volumes of these residues, but collecting them requires high costs of labour as well as high transport costs [41].

Sometimes, the process results in straw and hay being presented together. For example, Piekarczy et al. [42] present the results of the elemental composition of the ash produced during the combustion of straw and hay. Test samples of ash were characterized by alkaline pH (10.2) and a substantial overall potassium content K (155.7 g·kg<sup>-1</sup>), calcium Ca (124.0 g·kg<sup>-1</sup>), phosphorus P (15.1 g·kg<sup>-1</sup>), and magnesium Mg (7.3 g·kg<sup>-1</sup>). In addition, the presence of essential micronutrients for plants was found in them (Fe > Mn > Zn > Cu) and small natural content of heavy metals (Cr > Pb > Ni > Cd > Hg), which is not a contraindication for the agricultural use of ash from the plant biomass. Lisowski et al. [43] determined the characteristics of shredded biomass from hay, straw, and their mix in the ratio 1:1 using a sieve separator with oscillatory motion in the horizontal plane. All biomass particle size distributions belonged to the “very poorly sorted” category and were “very fine skewed” and “leptokurtic”. These results report that the particle size has an influence on the mechanical properties of the pellets of straw from wheat, barley, and corn. [43].

Therefore, most hay and straw research activities are focused on hygienic safety [18,44,45].

Hay and straw are used as bedding (Figure 1b). One of the crucial factors in maintaining a healthy dairy herd is having sanitary animal bedding. With bedding being one of the primary sources of exposure to environmental mastitis pathogens, the management of this material is important in maintaining herd health and the economic vitality of the farm [46]. Solan et al. [47] comment that straw is a popular bedding material for a large variety of livestock (e.g., cattle, horses, poultry) because it is quite absorbent, cheap, warm, and easy to maintain. But also, Solan et al. [47] write that straw has been shown to support

the growth of moulds and their spores. These spores can then be released into the air as dust, which can affect livestock and farmers.

Smith et al. (2017) [46] describe basic bedding materials used in New England. There are conventional dairy farmers who use manure solids (MNS; composed or digested) and hay as primary bedding materials, respectively, sawdust, sand, and hay. Straw, wood shavings, wood chips, and woody bedding were also used, but less so than primary bedding materials.

Hay is green forage preserved through natural drying or supplemental drying. In addition to fresh green forage—grass—hay is the most natural and suitable feed for all types of livestock [11,48]. Hay represents the dried stems of plants preserved through drying and partial fermentation. Hay serves as a long-term supply of high-quality dry bulk feed if appropriately stored according to regulations [12]. Hay is most commonly used for feed, and poorer quality hay may be used for bedding purposes. Hay is one of the more expensive beddings. It is quite absorbent and, once soiled, begins to decompose quickly, producing an odour [49]. It is a dried form of plant food, with water content ranging from 10–12% [50]. Once dried, the water content in hay decreases to 9–10% [11,12]. It is stored in haylofts, which typically have a slatted floor with tunnels where heated ventilators blow air to expedite the drying process. Grass hay for ruminant feed can be stored outdoors if protected, and outdoor storage may provide additional flexibility in the transportation and supply of grass hay bedding [51].

Straw is the dried stalk of cereal crops obtained after the threshing. Straw typically consists of dried stalks of a single cereal crop. In Slovakia, barley, oat, wheat, and rye straw are commonly used [17]. Straw is most frequently used as bedding for livestock [9], as a substrate for mushroom cultivation [19], or as an insulating material [52,53]. All these types of straw have a tendency to self-ignite thermally and microbiologically [18]. The literature [50] states that straw is a flammable material that is easily ignited by sparks and hot surfaces. Straw as a weeding material is a soft, dry by-product of small grains and is commonly used. It is easy to handle, carbonaceous for a compost pile, and readily available in most areas [49].

Statistical data on fire incidents obtained from the Fire Technology and Expertise Institute of the Ministry of Interior of the Slovak Republic confirm the occurrence of fires in agriculture [54]. Although the number of fires shows a declining trend, agricultural fires maintain a relevant percentage share (5–11%) of the total annual fire incidents in the Slovak Republic [16].

The risk of fire in agricultural crops was considered when developing the Slovak standard [20]. Solid flammable substances (Table 1) include fodder, hay, straw, and other dry, mowed stem plants, solid fuels, extracted woody biomass, and woody biomass processed into various product assortments (timber, wood chips, sawdust, cellulose-based pellets, and briquettes) (§2 of Decree 258/2007 Coll.) [20].

The fact that straw and hay are flammable materials is an important but not a priority consideration. The flammability of the mentioned materials is starting to be observed to a greater extent, especially thanks to the increase in the number of natural fires [16]. Natural fires are related to the increase in the temperature of the environment in the summer season, as well as the consequence of climate change [55]. The causes of fires in haylofts and warehouses can be hot surfaces and mechanisms present, e.g., combine harvesters [56,57].

The potential ignition of hay and straw is dependent on external conditions. The critical parameter is the ignition temperature at which ignition occurs, depending on the duration of the heat source's activity. Hot surfaces are part of the technological elements used in agriculture. Their surface temperature can exceed the minimum ignition temperature and pose a risk of fire ignition.

The aim of the article is to experimentally determine the ignition temperature of hay and straw due to radiant heat. The influence of the sample type (hay and straw) and the selected sample mass (1, 2, and 3 g) on the course of the thermal degradation, temperature increase inside the sample, and ignition temperature over time were observed. Significant

differences in thermal degradation and ignition temperature based on the type of fuel (hay and straw) were sought.

## 2. Materials and Methods



### 2.1. Experimental Samples

For the purposes of the experiment, samples of hay and straw obtained from an agricultural farm were used (Table 2) [20].

**Table 2.** Characteristics of samples.

Samples (Fodder)	Hay	Straw
Moisture (%) determined gravimetrically	11	10
Moisture (%) according to 258/2007 Act No. [20]	9–10	10

Sample before the experiment		
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The samples were stored in bales and used for the operation of the cattle barn.

### 2.2. Experimental Methods

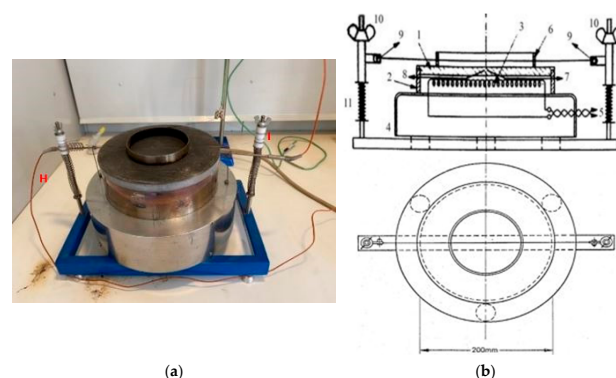
The methodology consisted of two steps:

1. Monitoring the thermal degradation of samples based on gradual heating of the sample by radiant heat and tracking degradation points with identification of temperature and time during the degradation processes;
2. Determination of the minimum ignition temperature using isothermal testing using a hot-plate according to EN 50281-2-1:1998 [58].

In both cases, measurements were repeated 3 times for 1, 2 and 3 g weight of samples.

#### 2.2.1. Methodology for Monitoring the Thermal Degradation of Hay and Straw

The experiment was carried out using a technical device called a hot-plate (Figure 2). Samples of hay and straw were subjected to gradual heating. The temperature of the heated plate and its increase over time were determined following the methodology described by Marková et al. [16]. The obtained temperature–time curve [16] also served as the basis for measuring the ignition temperature according to EN 50281-2-1:1998 [58].



**Figure 2.** (a) Hot-plate device; (b) Scheme of the apparatus for determining the ignition temperatures of settled dust EN 50281-2-1:1998 [58]. Legend: 1—heated plate; 2—frame; 3—heating element; 4—base of the heating element; 5—outlet for connecting the heating element to the power source and control; 6—ring for creating the layer of dust; 7—thermocouple in the plate for regulation; 8—thermocouple in the plate for temperature recording; 9—thermocouple for temperature recording in the layer of dust; 10—height adjustment of the thermocouple using screws; 11—spring.




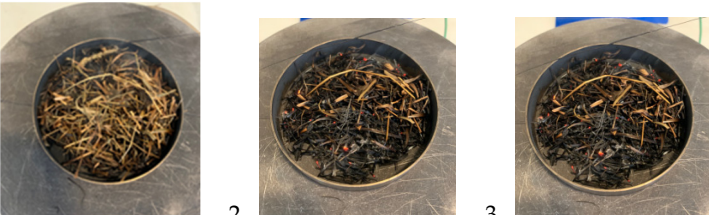
### 2.2.2. Determination of the Minimum Ignition Temperature by Isothermal Heating Using a Hot-Plate according to EN 50281-2-1:1998 [58]

The minimum ignition temperature of the organic layer was determined by isothermal heating of the sample placed on an electrically heated metal plate (Figure 2a,b). The minimum ignition temperature is defined as the lowest temperature of the heated plate's surface at which at least one of the following phenomena can be observed during the test:

1. Glowing, smouldering, or flaming combustion;
2. The temperature–time curve recorded by the thermocouple, which is placed at the centre of the sample layer, continuously rises in comparison to the temperature of the isothermally heated plate;
3. The temperature measured in the sample layer is 250 °C higher than the temperature of the heated plate.

The experiment verified the occurrence of the first two phenomena described earlier. The minimum ignition temperature was determined for all samples (Figure 2 and Table 3).

**Table 3.** Description of the behavior of hay and straw samples when exposed to radiant heat.

Straw				
Process Order	$T_{straw}$ (°C)	$t_{ex}$ (min)/(s) *	Visual Observations during Measurement	$T_{igniton}$ °C
1.	69.1	6 (360)	Odour noted	385.33 ± 13.2
2.	91.4	8.5 (525)	Smoking process appeared	
3.	142.6	11 (825)	Carbonization of the lower stems of the tested sample	
4.	145.2	16 (975)	Carbonization of the edges of the tested sample, increasing smoke intensity	
5.	173.2	17.5 (1050)	Ignition and formation of flames	
				
Hay				
Process Order	$T_{hay}$ (°C)	$t_{ex}$ (min)/(s) *	Visual Observations during Measurement	$T_{igniton}$ °C
1.	111.3	8 (480)	Smoke, thermal degradation	406.6 ± 5.1
2.	160.8	13.5 (810)	Carbonization of the layer on the surface of the hot-plate	
3.	185.4	16.75 (1005)	Carbonization of the edges of the samples and gradual degradation of the entire surface, smouldering process observed	
4.	192.6	18 (1080)	Ignition occurs	
				

\*  $t_{ex}$  (min)/(s)—real experimental time, critical time, which identified steps of thermal degradation.

A detailed description of individual steps of the experiment is provided by Balog et al. [59].

One thermal thermocouple measures the actual temperature of the heated metal plate, and second thermal thermocouple measures the temperature of tested sample, which is located 5 mm above the plate. The experimental results were obtained using installed thermocouples, which measured the following:

- Surface temperature of the hot-plate  $T_{hot}$  (marked by the red letter H in Figure 2a);
- Temperature inside the hay ( $T_{hay}$ ) and straw ( $T_{straw}$ ) sample (marked by the red letter I in Figure 1a).

The weight (1, 2, and 3 g) and hence the thickness of the sample in the testing device hot-plate were gradually increased. The results identify changes in the sample by determining the temperature inside the sample, the temperature of the plate that caused the change, and the chronological sequence of events (Table 3).

### 3. Results and Discussion

The conducted experiments provide interesting results. A description of the behaviour of the hay and straw layers during their thermal exposure to a radiant heat source is presented in Table 3.

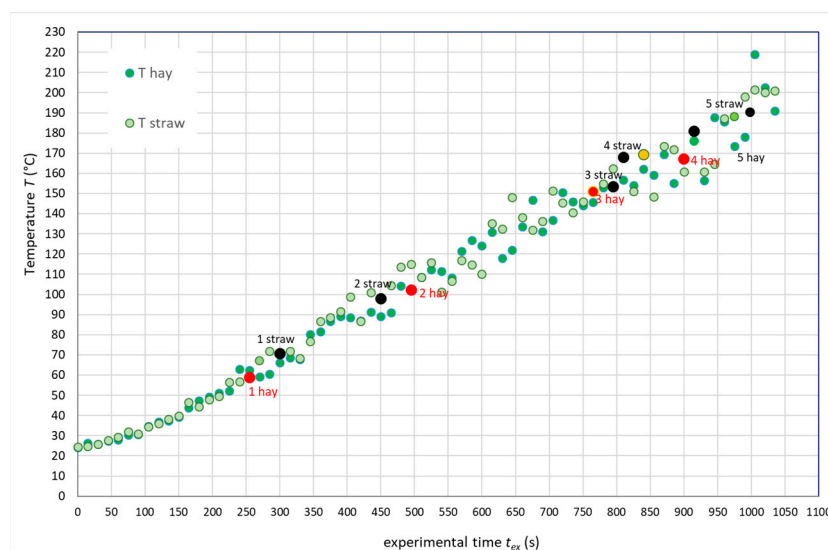
The moment of ignition is recorded as  $T_{ignition}$  and  $T_{hot}$  in Table 4. The temperature values inside the samples are recorded as  $T_{hay}$  and  $T_{straw}$ . The data are supplemented with photographic documentation. The sample weight was continuously increased by 1 g. The change in sample weight did not affect the course of the experiments, confirming the findings presented by Marková et al. [10,16]. Based on the repeated experiments, specific stages of the sample's behaviour during its thermal degradation were identified. These stages include the development of odour, smoking, carbonization of the bottom layer touching the hot-plate, carbonization of the sample edges, and ignition (Tables 3–5). The course of thermal degradation of straw was more pronounced in the identified stages compared to hay (Table 3).

**Table 4.** Experimentally determined temperatures of hay and straw degradation processes as a function of time.

Samples Monitored Parameters *	Hay			Straw		
	$T_{hot}$ (°C)	$T_{hay}$ (°C)	$t_{ex}$ (s)	$T_{hot}$ (°C)	$T_{straw}$ (°C)	$t_{ex}$ (s)
1. Process: Odour	-	62.1 ± 5.1	-	110–160	68.9 ± 1.1	305 ± 43.0
2. Process: Smoke	220–280	105.9 ± 5.2	505 ± 69.9	160–200	97.5 ± 5.8	454 ± 61.5
3. Process: Carbonization of the bottom layer of the sample	340–360	150.2 ± 7.6	765 ± 44.1	360–400	169.4 ± 19.27	800 ± 18.7
4. Process: Carbonization of the edges of the sample	400–430	175.6 ± 6.9	905 ± 50.1	400–410	179.4 ± 27.5	815 ± 30.8
5. Process: Ignition and burning	430–450	183.8 ± 9.2	1050 ± 24.5	410–430	189.9 ± 25.6	960 ± 63.6
Ignition temperature		406.6 ± 5.1			385.33 ± 13.2	
Ignition temperature according to EN 50281-2-1:1998 [58].		407			380	

\* Explanation of abbreviations:  $T_{hot}$  (°C)—temperature of the hot-plate surface,  $T_{hay}$  (°C) and  $T_{straw}$  (°C)—temperatures measured in the samples,  $t_{ex}$  (s)—real experimental time.

The course of thermal degradation of hay and straw is significantly comparable. Differences arise in the thermal time evaluation of their degradation (Tables 3 and 4). The comparison is based on the generated temperature–time curves, i.e., the dependence of experimentally determined temperature ( $T_{hay}$  and  $T_{straw}$ ) (Table 4) on time (Figure 3).



**Figure 3.** Temperature increases inside the samples of hay ( $T_{hay}$ ) and straw ( $T_{straw}$ ) as a function of time. Legend: black color—points of straw thermal degradation processes, red color—points of hay thermal degradation processes.

The ignition temperatures of hay and straw determined by two different procedures are significantly comparable (Table 4).

Experts are still debating which of the materials (hay or straw) poses a higher risk of fire ignition [60]. The results show an overall delayed process of thermal degradation for hay compared to straw at higher temperatures. The first phase of degradation (labelled as 1 hay and 1 straw in Figure 3) is characterized by odour. The odour of burning hay and straw starts at temperatures above 60 °C, but with a 30 s delay for straw (Figure 3). The findings are consistent with the measures implemented in practice [17,60] and with regulation 258/2007 Coll [20]. References [17,60] present a temperature of 65 °C as the beginning of the dangerous thermal zone. The unpleasant odour intensified with increasing temperature in both types of samples.

Smoke production was recorded at a temperature of 100 °C inside the samples. The smoke had a white colour. In terms of time, straw started smoking earlier, at 450 s. Hay started smoking at 505 s. As the surface and sample temperature increased, the smoking intensified (Figure 3).

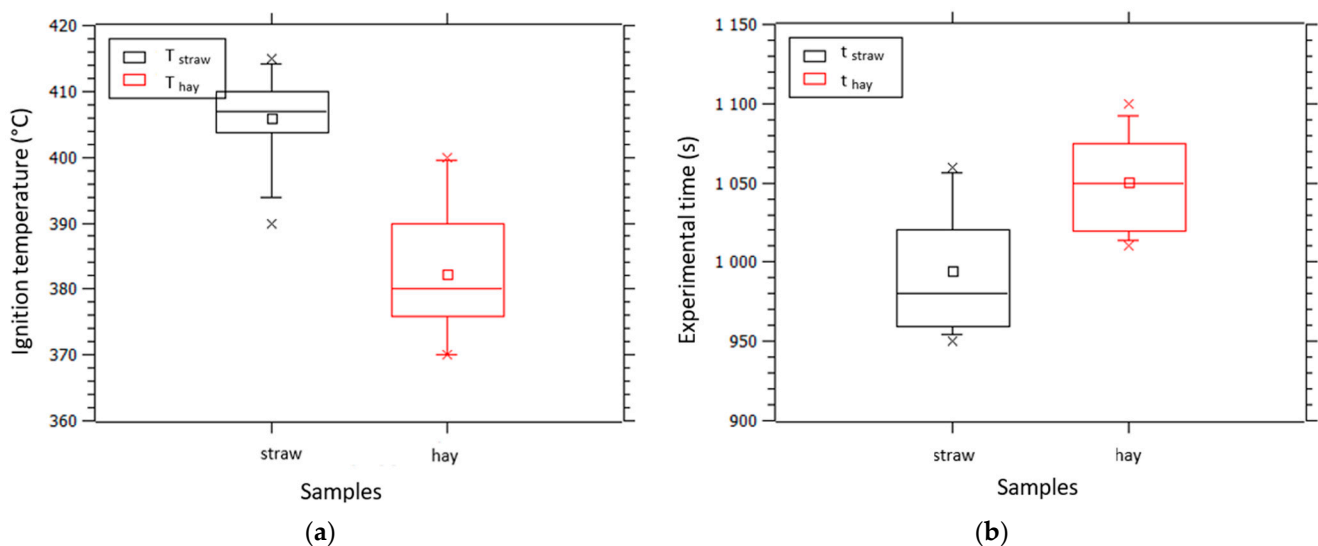
In the straw sample, carbonization of the bottom layer of straw stems occurred at a temperature of 169.4 °C (Table 4) at 800 s on average. The carbonization process of the bottom layer of the hay sample occurred at an average temperature of 150.2 °C at 780 s (14 min) at a hot-plate temperature of 330.2 °C. The carbonization process occurred in the edges of the straw sample at an average temperature of 175.6 °C at 960 s (16 min) with the hot-plate temperature ranging from 360–400 °C. The fourth process begins earlier in straw (400–410 s) at a temperature of 179.4 °C and a hot-plate temperature of 400–430 °C (Figure 4).

With the gradual increase in surface and sample temperature, the smoke intensified until the ignition point was reached, resulting in ignition in the tested straw sample. Subsequently, the smoke intensity decreased until it completely disappeared, leaving only glowing charred residues and stems of the tested samples (Figure 5), positioned 10 mm above the hot surface (Figure 4 and Table 2).

Throughout the entire degradation process, the critical temperatures of hay and straw were comparable, with hay showing degradation at an earlier stage and at lower temperatures compared to straw. Subsequent events occurred earlier in straw and at lower temperatures than in hay. The risk of fire cannot be considered higher for hay or straw in terms of the time-related development of thermal degradation or temperatures of thermal degradation in partial processes (Figure 3, Table 4).



**Figure 4.** Samples after ignition: (a) 2 g of hay sample, measurement at 1020 s; (b) 2 g of straw sample, measurement at 1050 s.



**Figure 5.** Box plots—graphical dependencies (a) of species (straw and hay) and weight on the minimum ignition temperature and (b) of species on the time course of the thermal degradation.

The literature [48] states a critical temperature of 80 °C at which the spontaneous combustion process of hay and straw occurs. Several authors [3,7] report a higher risk of fire for hay compared to straw. The authors of [17] attribute this to the increase in the internal temperature of the hay bale, which does not decrease but, on the contrary, creates an ideal environment for the proliferation of thermophilic bacteria. Consequently, the temperature of the hay bale rises up to 77 °C (reported as the temperature of spontaneous combustion) [17].

The determined experimental results of ignition temperatures are higher for straw. Flachbart and Svetlík [61] also present higher ignition temperature values for straw (Table 5). The ignition temperatures of hay and straw samples (Table 2) show differences in values.

The search for a significant effect was performed using the statistical program Qplot. Produced histograms (box plots) did not confirm a significant influence of weight and sample type on the minimum ignition temperature of hay and straw (Figure 5a), as well as on the time-related development of thermal degradation of the samples (Figure 5b).

Box plots (Figure 5) show faster straw degradation at lower temperatures than hay.

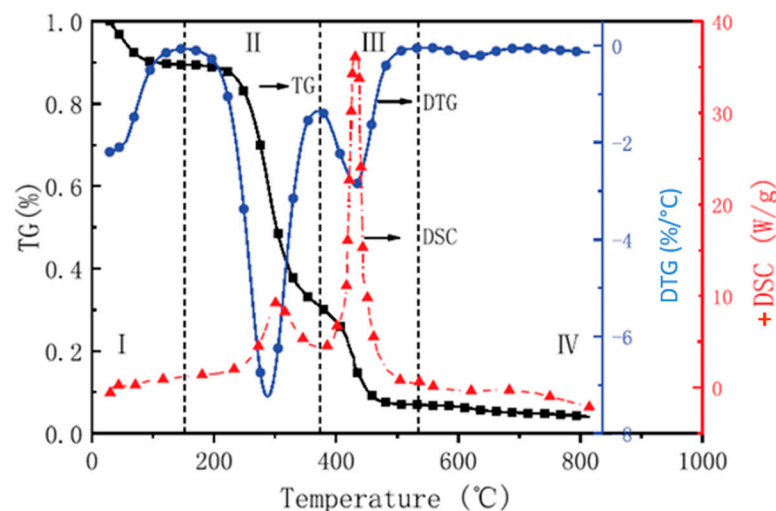
The thermal degradation of hay can be compared with the research of Xie et al. [62]. The TGA-DTG-DSC curves of rape straw under air and CO<sub>2</sub> atmosphere and a heating rate



of  $10\text{ }^{\circ}\text{C}\cdot\text{min}^{-1}$  (Figure 6) identify comparable temperatures of thermal degradation of the hay with our experimental results.

**Table 5.** Mutual comparison Ignition temperatures of hay and straw.

Ignition Temperature	Hay	Straw
Experimentally determined temperature ( $^{\circ}\text{C}$ )	$406.6 \pm 5.1$	$385.33 \pm 13.2$
Temperature experimentally determined according to EN 50281-2-1:1998 [58] ( $^{\circ}\text{C}$ )	407	380
Temperature according to [19] ( $^{\circ}\text{C}$ )	310	330
Temperature according to [61] ( $^{\circ}\text{C}$ )	230	310



**Figure 6.** TGA-DTA-DSC curves of rape straw under a heating rate of  $10\text{ }^{\circ}\text{C}\cdot\text{min}^{-1}$  [62].

The DTG curve identifies the temperatures at which the thermal degradation of the sample is maximal. Thermal degradation of hay is two-stage. The first stage of hay degradation is at a temperature of  $290\text{ }^{\circ}\text{C}$ , and the second is at  $470\text{ }^{\circ}\text{C}$ . But the DSC curve, which shows identical thermal degradation stages (and temperatures) as TG/DTG curves, identifies the maximum heat release rate in the second stage. It is possible to assume that hay can ignite in the second stage ( $470\text{ }^{\circ}\text{C}$ ).

#### 4. Conclusions

Based on the obtained experimental results, the following conclusions can be made:

- The minimum ignition temperature of hay according to EN 50281-2-1:1998 [58] is  $407\text{ }^{\circ}\text{C}$ ;
- During exposure to radiant heat, the critical temperatures of hay and straw were comparable, except for the initial phase, where hay degradation started earlier at a lower temperature and in a shorter time interval compared to straw;
- It is not possible to unequivocally determine which of the mentioned materials poses a greater risk of fire;
- The significant effect of weight and sample type on the minimum ignition temperature of hay and straw, as well as on the time-related development of thermal degradation of the samples, was not confirmed.

Existing studies on this research topic have primarily focused on investigating the impact of different forms of bedding materials on the living conditions of livestock. Hay and straw are standard bedding materials, but there are also alternative forms (manure, peat, sand) that are used in agriculture practice. In this article, we analyzed the fire-technical properties of hay and straw as an alternative form of bedding. We are aware of some limitations of this study. First, this study was conducted on a small number of samples obtained from an agricultural farm. Second, we only used one initiation source (radiant

heat). The innovation of this contribution is the original experimental data and comparison of the thermal degradation of hay and straw.

**Author Contributions:** Conceptualization, I.M. and I.T.; methodology, Z.G.; software, I.M.; validation, I.T., I.M. and Z.G.; formal analysis, J.J.; investigation, Z.G. and I.M.; resources, I.T., J.J. and I.M.; data curation, I.M. and J.J.; writing—original draft preparation, I.M.; writing—review and editing, J.J. and I.T.; project administration, J.J.; funding acquisition, I.M. All authors have read and agreed to the published version of the manuscript.

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