

Supplementary Materials: Impact of the Subunit Arrangement on the Nonlinear Absorption Properties of Organometallic Complexes with Ruthenium(II) σ -Acetylide and Benzothiadiazole as Building Units

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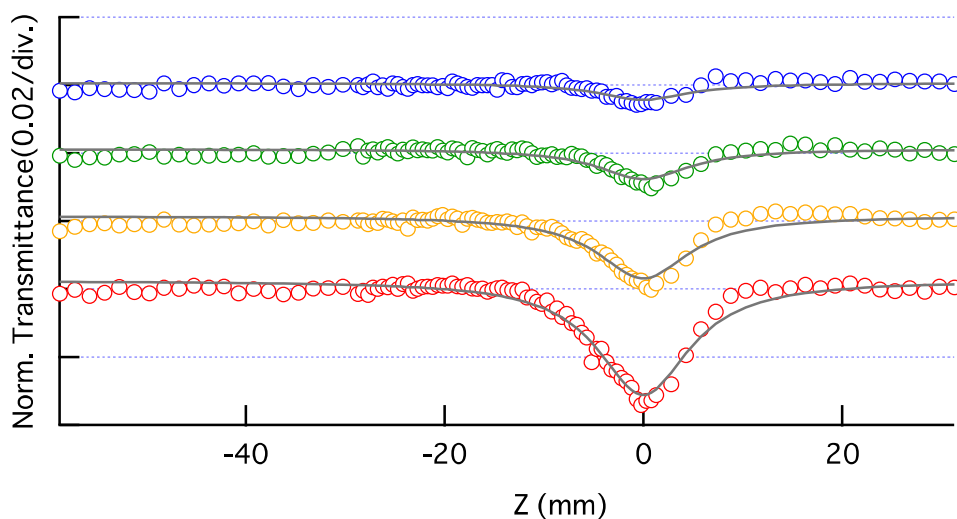


Figure S1. Open-aperture Z-scan traces at different incident powers (0.19, 0.23, 0.32, 0.42 mW from top to bottom) of **Ru-1** in dichloromethane excited at 800 nm. Solid curves are theoretical fits.

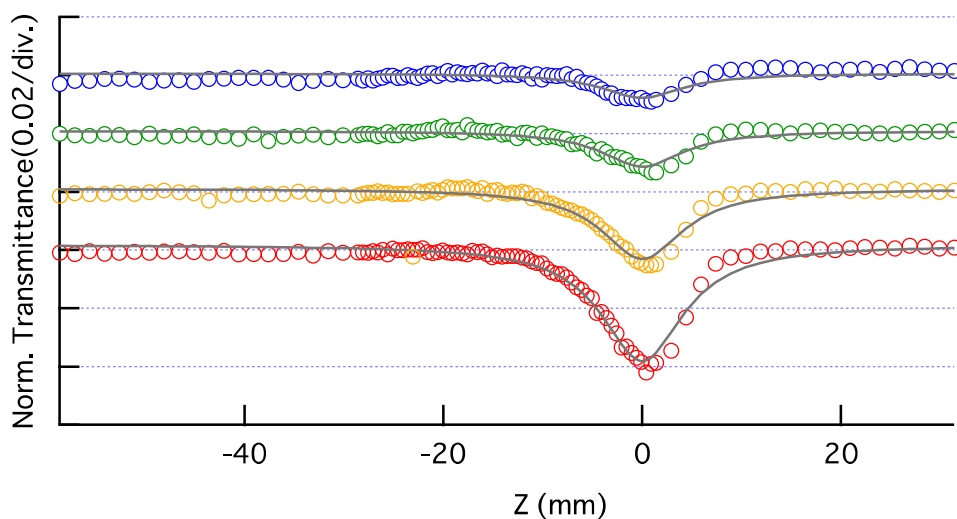


Figure S2. Open-aperture Z-scan traces at different incident powers (0.18, 0.22, 0.29, 0.39 mW from top to bottom) of **Ru-1** in dichloromethane excited at 840 nm. Solid curves are theoretical fits.

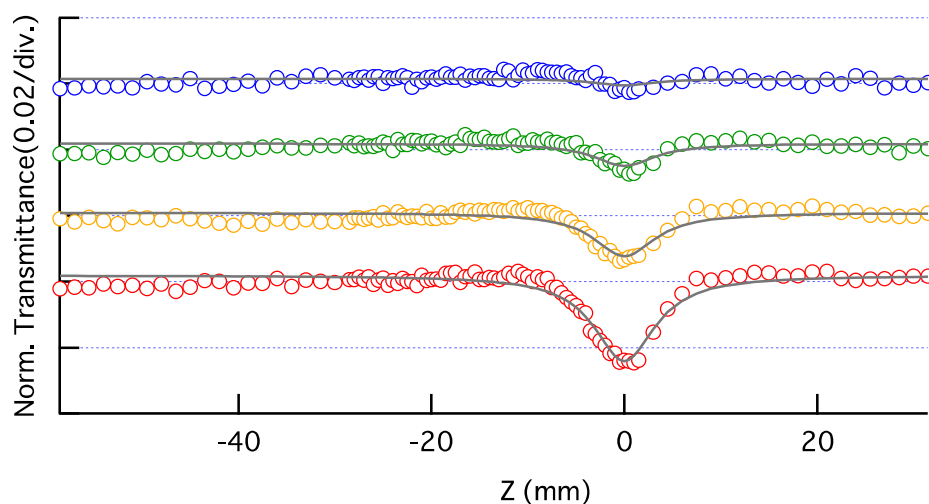


Figure S3. Open-aperture Z-scan traces at different incident powers (0.22, 0.25, 0.34, 0.44 mW from top to bottom) of **Ru-1** in dichloromethane excited at 960 nm. Solid curves are theoretical fits.

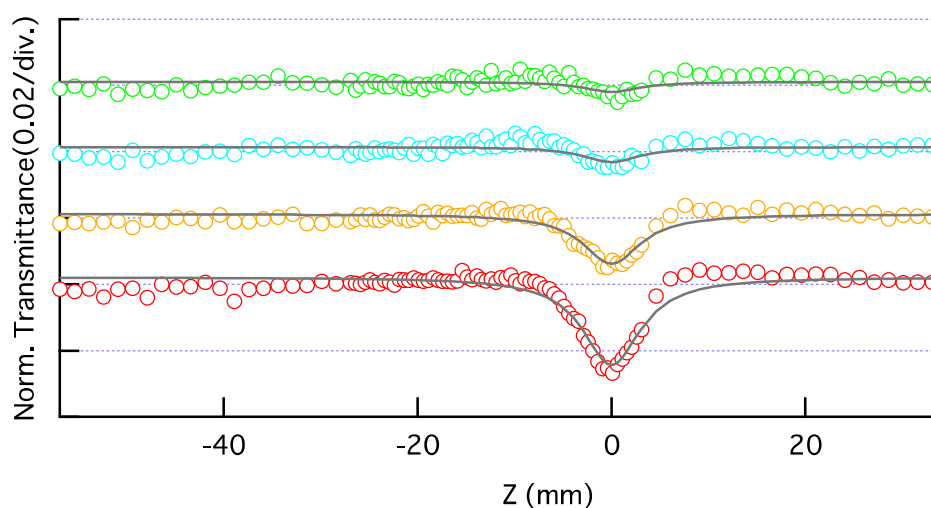


Figure S4. Open-aperture Z-scan traces at different incident powers (0.25, 0.29, 0.39, 0.50 mW from top to bottom) of **Ru-1** in dichloromethane excited at 970 nm. Solid curves are theoretical fits.

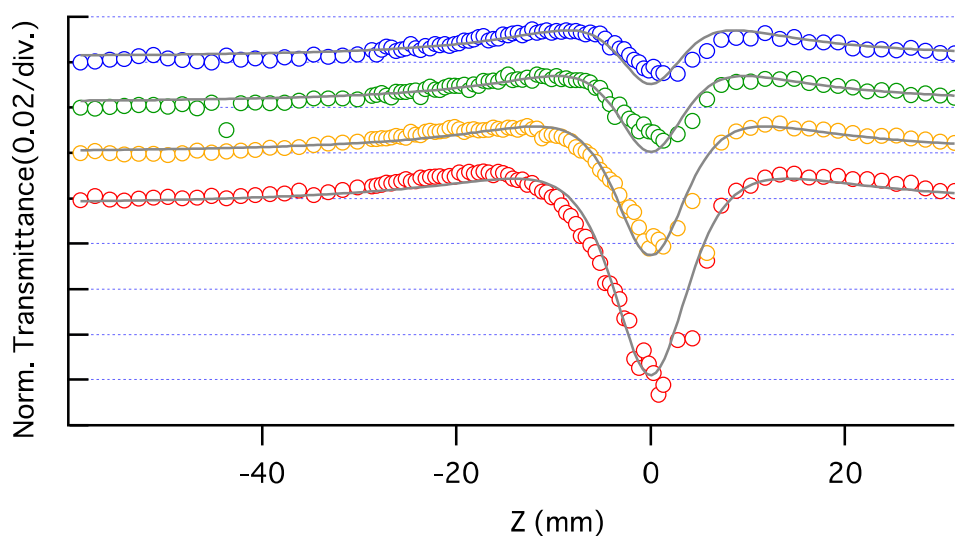


Figure S5. Open-aperture Z-scan traces at different incident powers (0.19, 0.22, 0.31, 0.42 mW from top to bottom) of **Ru-2** in dichloromethane excited at 800 nm. Solid curves are theoretical fits assuming the saturable absorption.

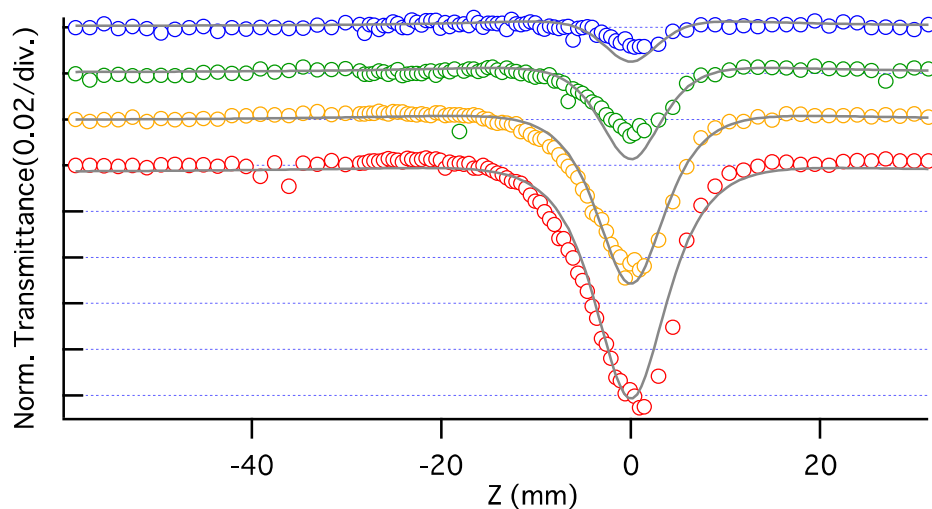


Figure S6. Open-aperture Z-scan traces at different incident powers (0.11, 0.28, 0.29, 0.39 mW from top to bottom) of **Ru-2** in dichloromethane excited at 840 nm. Solid curves are theoretical fits assuming the saturable absorption.

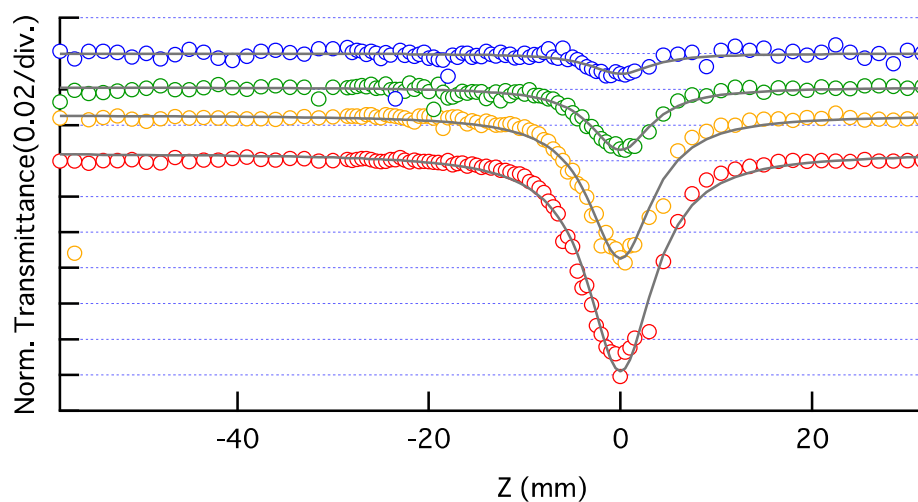


Figure S7. Open-aperture Z-scan traces at different incident powers (0.12, 0.22, 0.38, 0.44 mW from top to bottom) of **Ru-2** in dichloromethane excited at 960 nm. Solid curves are theoretical fits.

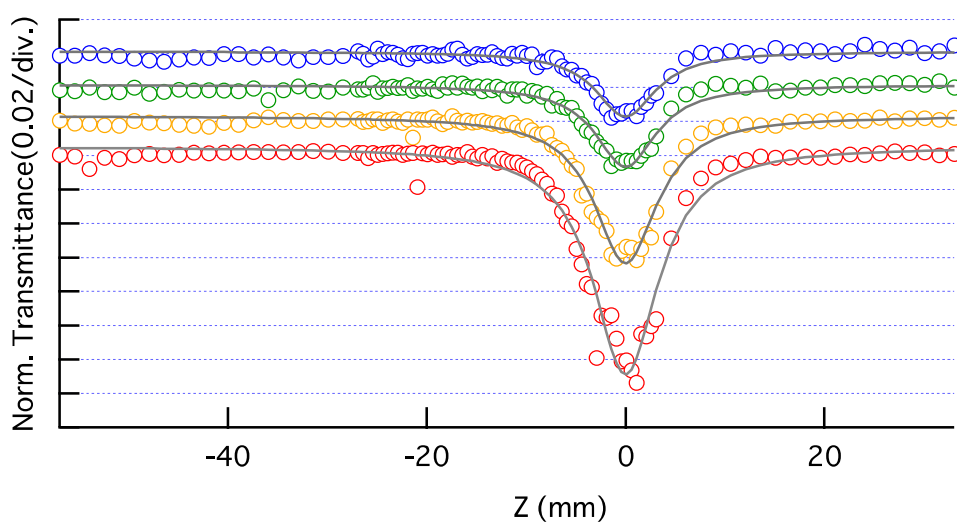


Figure S8. Open-aperture Z-scan traces at different incident powers (0.25, 0.29, 0.39, 0.50 mW from top to bottom) of **Ru-2** in dichloromethane excited at 970 nm. Solid curves are theoretical fits.

Table S1. Summary of the TD-DFT calculations of transition energy, wavelength, oscillator strength, electronic configuration and the contribution of the 12 lowest excited states of Ru-1 and Ru-2 calculated at the B3LYP/6-31G(d) (lanl2DZ for Ru) level.

| Ru-1 (HOMO: orbital 383; LUMO: orbital 384) | | | |
|--|-----------|-----------|------------|
| Excited State 1 | 1.7217 eV | 720.11 nm | $f=0.8670$ |
| | 382 → 385 | | -0.11458 |
| | 383 → 384 | | 0.68509 |
| | 383 → 385 | | 0.11653 |
| Excited State 2 | 1.7979 eV | 689.60 nm | $f=0.0465$ |
| | 383 → 384 | | -0.11921 |
| | 383 → 385 | | 0.69191 |
| Excited State 3 | 2.0192 eV | 614.03 nm | $f=0.0776$ |
| | 382 → 384 | | 0.67523 |
| | 382 → 385 | | -0.16890 |
| Excited State 4 | 2.0321 eV | 610.14 nm | $f=0.1895$ |
| | 382 → 384 | | 0.17403 |
| | 382 → 385 | | 0.66917 |
| Excited State 5 | 2.4556 eV | 504.91 nm | $f=0.0118$ |
| | 381 → 384 | | 0.69470 |
| Excited State 6 | 2.4867 eV | 498.59 nm | $f=0.0133$ |
| | 381 → 385 | | 0.69298 |
| Excited State 7 | 2.6025 eV | 476.40 nm | $f=0.0004$ |
| | 380 → 386 | | 0.16191 |
| | 383 → 386 | | 0.66606 |
| Excited State 8 | 2.7333 eV | 453.61 nm | $f=0.0092$ |
| | 379 → 385 | | 0.14558 |
| | 380 → 384 | | 0.67217 |
| Excited State 9 | 2.7618 eV | 448.93 nm | $f=0.0201$ |
| | 379 → 384 | | 0.15514 |
| | 380 → 385 | | 0.67070 |
| Excited State 10 | 2.9157 eV | 425.23 nm | $f=0.0010$ |
| | 381 → 386 | | 0.36859 |
| | 382 → 386 | | 0.57089 |
| Excited State 11 | 3.0374 eV | 408.19 nm | $f=1.2070$ |
| | 382 → 388 | | -0.15883 |
| | 383 → 387 | | 0.66262 |
| Excited State 12 | 3.1037 eV | 399.47 nm | $f=0.0227$ |
| | 378 → 384 | | 0.10993 |
| | 378 → 385 | | -0.15095 |
| | 379 → 384 | | 0.63485 |
| | 380 → 385 | | -0.12978 |
| | 383 → 388 | | -0.11123 |

Table S1. (continued)

| Ru-2 (HOMO: orbital 577; LUMO: orbital 578) | | | |
|--|-----------|-----------|------------|
| Excited State 1 | 1.6468 eV | 752.86 nm | $f=0.8802$ |
| | | 577 → 578 | 0.70356 |
| Excited State 2 | 2.1033 eV | 589.46 nm | $f=0.0318$ |
| | | 575 → 578 | -0.11271 |
| | | 576 → 578 | 0.69293 |
| Excited State 3 | 2.1631 eV | 573.19 nm | $f=0.0151$ |
| | | 575 → 578 | 0.69108 |
| | | 576 → 578 | 0.11265 |
| Excited State 4 | 2.3624 eV | 524.83 nm | $f=0.0004$ |
| | | 574 → 578 | 0.69978 |
| Excited State 5 | 2.5364 eV | 488.83 nm | $f=0.0003$ |
| | | 573 → 578 | 0.23534 |
| | | 575 → 580 | 0.12445 |
| | | 576 → 580 | 0.30107 |
| | | 577 → 580 | 0.55134 |
| Excited State 6 | 2.5435 eV | 487.46 nm | $f=0.0044$ |
| | | 573 → 578 | 0.65459 |
| | | 576 → 580 | -0.10929 |
| | | 577 → 580 | -0.19261 |
| Excited State 7 | 2.5987 eV | 477.09 nm | $f=0.0002$ |
| | | 573 → 579 | 0.13038 |
| | | 576 → 579 | -0.38141 |
| | | 577 → 579 | 0.54917 |
| Excited State 8 | 2.7338 eV | 453.52 nm | $f=0.0003$ |
| | | 573 → 579 | -0.10552 |
| | | 575 → 579 | 0.61281 |
| | | 575 → 617 | -0.11202 |
| | | 576 → 579 | -0.17899 |
| | | 577 → 579 | -0.18236 |
| Excited State 9 | 2.8537 eV | 434.47 nm | $f=0.0942$ |
| | | 572 → 578 | 0.66408 |
| | | 577 → 581 | -0.20256 |
| Excited State 10 | 2.9065 eV | 426.58 nm | $f=1.3085$ |
| | | 572 → 578 | 0.20280 |
| | | 577 → 581 | 0.65108 |
| Excited State 11 | 2.9266 eV | 423.65 nm | $f=0.0087$ |
| | | 574 → 580 | 0.66474 |
| | | 574 → 618 | 0.13483 |
| Excited State 12 | 3.0571 eV | 405.57 nm | $f=0.0171$ |
| | | 571 → 578 | 0.68731 |