

Figure S1 The effect of temperature on dissolved Fe in pore water under different DO conditions: (A) high DO; (B) low DO.

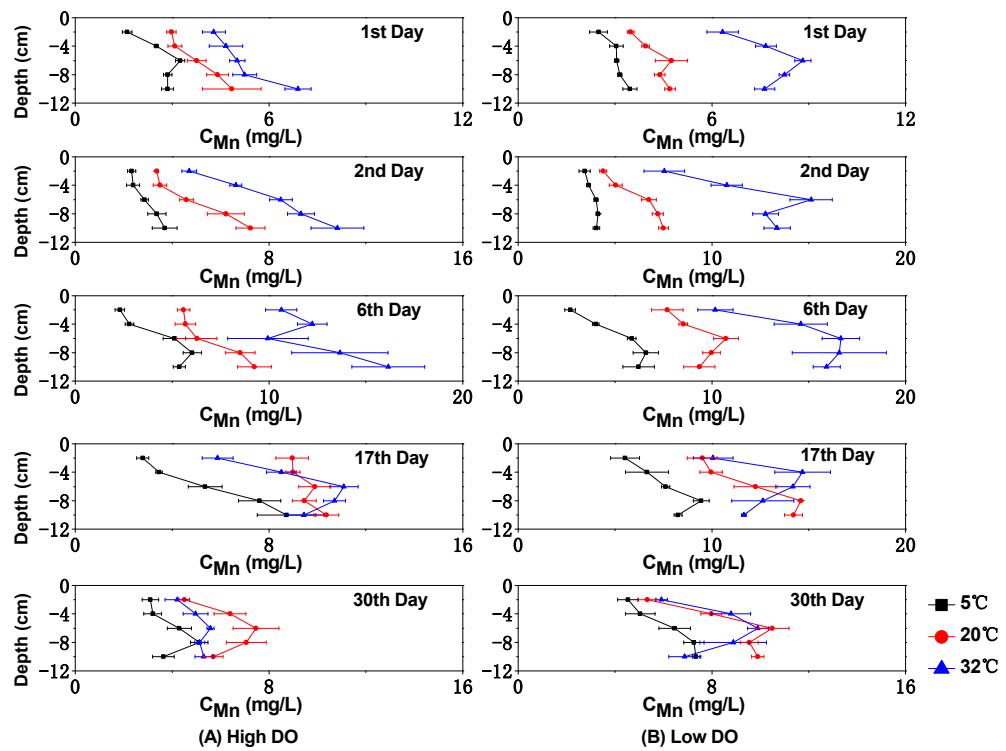


Figure S2 The effect of temperature on dissolved Mn in pore water under different DO conditions: (A) high DO; (B) low DO.

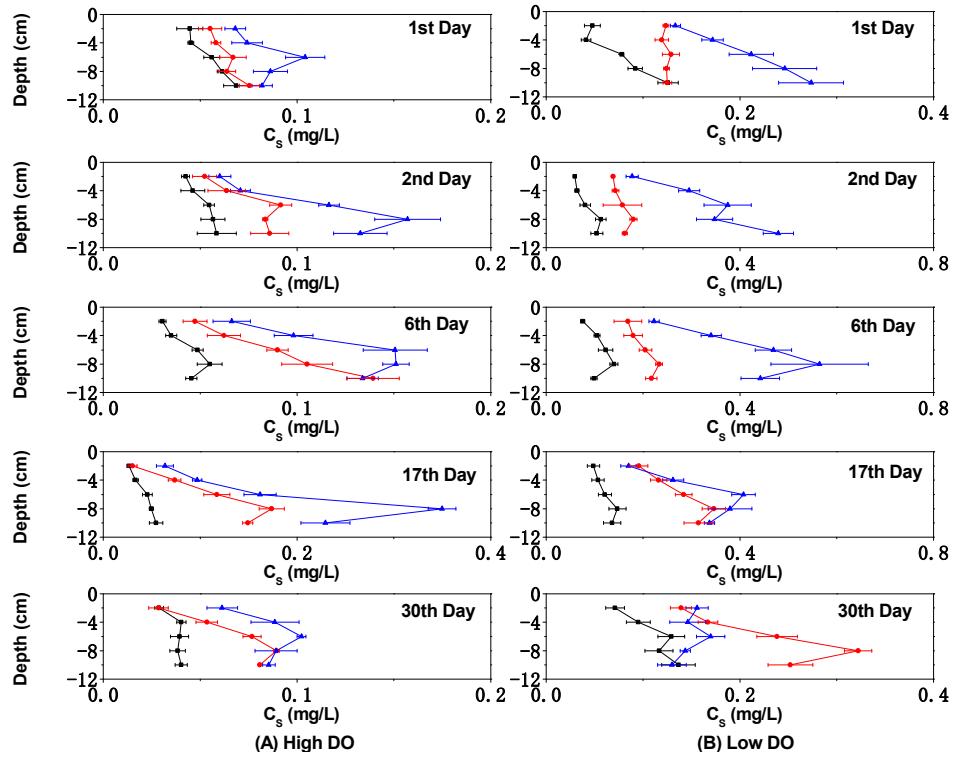


Figure S3 The effect of temperature on dissolved S of pore water under different DO conditions: (A) high DO; (B) low DO.

Table S1 The linear equation of the standard regression curve, the correlation coefficient and LODs of each element

Detection object	Linear equation	r	LOD	Measurement method
As	$y = 0.0302x + 0.0014$	0.9997	0.02 µg/L	ICP-MS
As (V)	$y = 0.7359x - 0.000055$	0.9998	0.06 µg/L	HPLC-ICP-MS
As (III)	$y = 0.6908x - 0.000069$	0.9999	0.05 µg/L	HPLC-ICP-MS
Fe	$y = 3.624x + 0.1607$	0.9998	0.8 µg/L	ICP-MS
Mn	$y = 0.1765x + 0.1763$	0.9997	0.36 µg/L	ICP-MS
Fe (II)	$y = 0.2028x + 0.0092$	0.9997	0.04 mg/L	Phenanthroline colorimetry
S (II)	$y = 0.1538x + 0.0014$	0.9996	0.006 mg/L	Methylene blue method

Table S2 The correlation analysis between the mean values of dissolved As and dissolved Fe, Mn and S in sediment

	High DO	Low DO
5 °C	$C_{Fe} = 0.0591 C_{As} + 0.406$	$C_{Fe} = 0.0658 C_{As} + 0.7148$
	$R^2 = 0.669 (P < 0.01)$	$R^2 = 0.7845 (P < 0.01)$
	$C_{Mn} = 0.2408 C_{As} + 2.0158$	$C_{Mn} = 0.1737 C_{As} + 3.271$
	$R^2 = 0.5044 (P < 0.01)$	$R^2 = 0.7104 (P < 0.01)$
	$C_S = -9E^{-05} C_{As} + 0.0475$	$C_S = 0.0022 C_{As} + 0.0778$
	$R^2 = 0.0016 (P < 0.01)$	$R^2 = 0.4541 (P < 0.01)$
	$C_{Fe} = 0.0724 C_{As} + 0.4089$	$C_{Fe} = 0.0684 C_{As} + 0.9141$
	$R^2 = 0.7049 (P < 0.01)$	$R^2 = 0.7966 (P < 0.05)$
	$C_{Mn} = 0.1683 C_{As} + 3.2731$	$C_{Mn} = 0.1103 C_{As} + 4.5119$
	$R^2 = 0.649 (P < 0.05)$	$R^2 = 0.577 (P < 0.01)$
	$C_S = 0.002 C_{As} + 0.0442$	$C_S = 0.0025 C_{As} + 0.1133$
	$R^2 = 0.4053 (P < 0.01)$	$R^2 = 0.6692 (P < 0.01)$
20 °C	$C_{Fe} = 0.06 C_{As} + 0.5291$	$C_{Fe} = 0.0561 C_{As} + 0.856$
	$R^2 = 0.3329 (P < 0.01)$	$R^2 = 0.3578 (P < 0.01)$
	$C_{Mn} = 0.1072 C_{As} + 3.5542$	$C_{Mn} = 0.0924 C_{As} + 6.4489$
	$R^2 = 0.3932 (P < 0.01)$	$R^2 = 0.3219 (P < 0.01)$
	$C_S = 0.0022 C_{As} + 0.0262$	$C_S = 0.004 C_{As} + 0.0829$
32 °C	$R^2 = 0.4267 (P < 0.01)$	$R^2 = 0.4421 (P < 0.01)$