

Carbon membranes prepared from poly(furfuryl alcohol-furfural) precursors: effect of FeCl₃ additive

Supplementary material file

Table S1. *H₂, CO₂ and CH₄ permeance and corresponding H₂/CH₄ and H₂/CO₂ selectivity for the C1-TS membrane (corresponding Figs. 9 and 10).*

#	Coating cycles	Pyrolysis temperature (°C)	Permeance (mol m ⁻² s ⁻¹ Pa ⁻¹)			Selectivity	
			H ₂	CO ₂	CH ₄	H ₂ /CH ₄	H ₂ /CO ₂
3		500	nm*	nm	6,94x10 ⁻⁸	nm	nm
4		500	5,12x10 ⁻⁸	nm	2,32x10 ⁻⁸	2,20	nm
5		500	1,93 x10 ⁻⁸	nm	3,79x10 ⁻⁹	5,08	nm
6		500	1,34 x10 ⁻⁸	nm	2,57x10 ⁻⁹	5,23	nm
7		500	1,23 x10 ⁻⁸	nm	2,05x10 ⁻⁹	5,98	nm
8		500	1,24 x10 ⁻⁸	1,46x10 ⁻⁸	1,77x10 ⁻⁹	7,01	0,85
8		700	nm	nm	2,02x10 ⁻⁸	nm	nm
9		700	4,27 x10 ⁻⁸	3,95x10 ⁻⁸	3,62 x10 ⁻⁹	11,81	1,08
10		700	1,80 x10 ⁻⁸	1,18x10 ⁻⁸	6,57x10 ⁻¹⁰	27,34	1,52
11		700	8,00 x10 ⁻⁹	1,76x10 ⁻⁹	1,60x10 ⁻¹⁰	49,98	4,54
12		700	2,04 x10 ⁻⁹	1,42x10 ⁻¹⁰	2,87x10 ⁻¹¹	70,80	14,38
12		800	9,31x10 ⁻⁹	1,47x10 ⁻⁹	4,62x10 ⁻¹⁰	20,15	6,3
13		800	1,35x10 ⁻⁹	9,2x10 ⁻¹¹	1,16x10 ⁻¹⁰	11,67	14,67
14		800	6,41x10 ⁻¹⁰	1,4x10 ⁻¹¹	1,7x10 ⁻¹¹	38,20	45,13
14		900	1,52x10 ⁻⁹	1,4x10 ⁻¹⁰	1,68x10 ⁻¹⁰	9,00	10,66
15		900	1,20x10 ⁻⁹	7,2x10 ⁻¹¹	8,1x10 ⁻¹¹	14,88	16,5
16		900	7,77x10 ⁻¹⁰	2,2x10 ⁻¹¹	1,8x10 ⁻¹¹	42,32	36,12
16		1000	4,53x10 ⁻¹⁰	5,5x10 ⁻¹¹	8,4x10 ⁻¹¹	5,42	8,3
17		1000	2,91x10 ⁻¹⁰	3,2x10 ⁻¹¹	5,3x10 ⁻¹¹	5,55	9,13

* nm: not measured

Table S2. H_2 , CO_2 and CH_4 permeance and corresponding H_2/CH_4 and H_2/CO_2 selectivity for the C2-Fe membrane (corresponding Figs. 11 and 12).

	Pyrolysis temperature	Permeance ($\text{mol m}^{-2} \text{s}^{-1} \text{Pa}^{-1}$)			Selectivity	
Coating cycles	(°C)	H_2	CO_2	CH_4	H_2/CH_4	H_2/CO_2
3	500	nm*	nm	$3,62 \times 10^{-8}$	nm	nm
4	500	$2,34 \times 10^{-8}$	nm	$4,65 \times 10^{-9}$	5,03	nm
5	500	$1,12 \times 10^{-8}$	nm	$8,23 \times 10^{-10}$	13,59	nm
6	500	$8,02 \times 10^{-9}$	$3,21 \times 10^{-9}$	$3,73 \times 10^{-10}$	21,50	2,50
7	500	$6,33 \times 10^{-9}$	$2,32 \times 10^{-9}$	$2,62 \times 10^{-10}$	24,14	2,72
8	500	$6,14 \times 10^{-9}$	$2,69 \times 10^{-9}$	$1,80 \times 10^{-10}$	33,98	2,28
8	700	nm	nm	$1,46 \times 10^{-8}$	nm	nm
9	700	$2,19 \times 10^{-8}$	$1,29 \times 10^{-8}$	$2,56 \times 10^{-9}$	8,58	1,70
10	700	$1,16 \times 10^{-8}$	$4,46 \times 10^{-9}$	$5,20 \times 10^{-10}$	22,25	2,59
11	700	$6,30 \times 10^{-9}$	$1,57 \times 10^{-9}$	$1,65 \times 10^{-10}$	38,16	4,01
12	700	$2,71 \times 10^{-9}$	$3,54 \times 10^{-10}$	$1,13 \times 10^{-10}$	23,92	7,66
12	800	$1,76 \times 10^{-8}$	nm	$3,84 \times 10^{-9}$	4,58	nm
13	800	$3,54 \times 10^{-9}$	nm	$7,10 \times 10^{-10}$	4,99	nm
14	800	$3,10 \times 10^{-9}$	$3,82 \times 10^{-10}$	$4,50 \times 10^{-10}$	6,89	8,11
14	900	$1,03 \times 10^{-8}$	$1,86 \times 10^{-9}$	$2,62 \times 10^{-9}$	3,94	5,56
15	900	$5,94 \times 10^{-9}$	$9,91 \times 10^{-10}$	$1,52 \times 10^{-9}$	3,91	5,99
16	900	$1,99 \times 10^{-9}$	$2,37 \times 10^{-10}$	$3,75 \times 10^{-10}$	5,31	8,39
17	900	$1,24 \times 10^{-9}$	$1,07 \times 10^{-10}$	$1,49 \times 10^{-10}$	8,34	11,58
18	900	$4,72 \times 10^{-10}$	$8,2 \times 10^{-12}$	$1,3 \times 10^{-11}$	35,31	57,88
18	1000	$5,76 \times 10^{-9}$	$1,28 \times 10^{-9}$	$1,47 \times 10^{-9}$	3,92	4,49
19	1000	$2,34 \times 10^{-9}$	$3,66 \times 10^{-10}$	$4,94 \times 10^{-10}$	4,73	6,39

* nm: not measured