

Supporting information

"Oscillatory copper deposition on conical iron electrodes in nonuniform magnetic fields"

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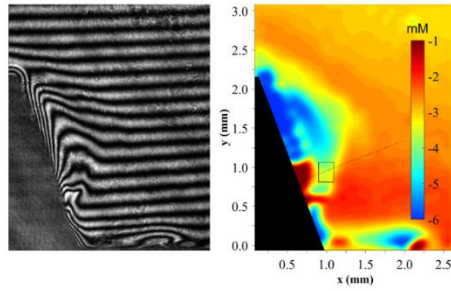


Figure S1.a. Copper deposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 60.0s. Initial CuSO_4 concentration=300 mM; $J=0 \text{ mA/cm}^2$.

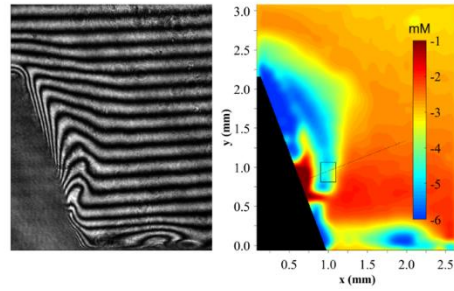


Figure S1.b. Copper deposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 70.0s. Initial CuSO_4 concentration=300 mM; $J=0 \text{ mA/cm}^2$.

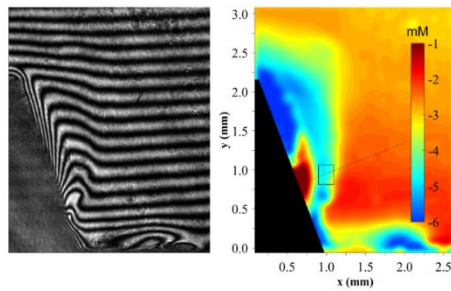


Figure S1.c. Copper deposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 80.0s. Initial CuSO_4 concentration=300 mM; $J=0 \text{ mA/cm}^2$.

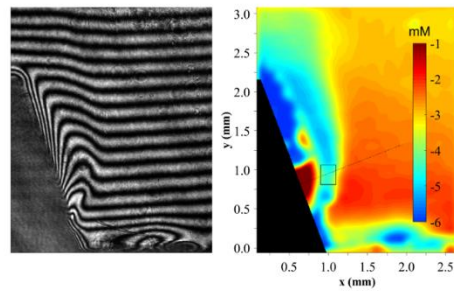


Figure S1.d. Copper deposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 90.0s. Initial CuSO_4 concentration=300 mM; $J=0 \text{ mA/cm}^2$.

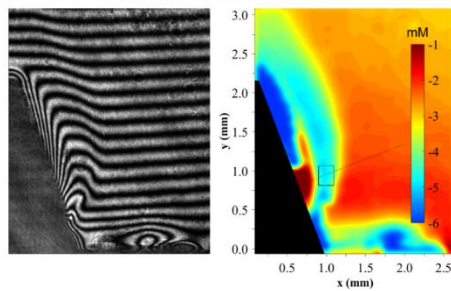


Figure S1.e. Copper deposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 100.0s. Initial CuSO_4 concentration=300 mM; $J=0 \text{ mA/cm}^2$.

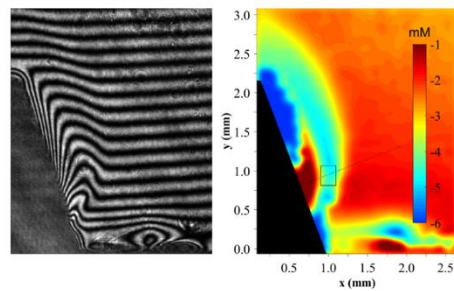


Figure S1.f. Copper deposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 110.0s. Initial CuSO_4 concentration=300 mM; $J=0 \text{ mA/cm}^2$.

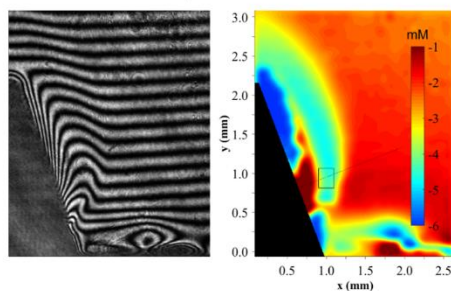


Figure S1.g. Copper deposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 120.0s. Initial CuSO_4 concentration=300 mM; $J=0 \text{ mA/cm}^2$.

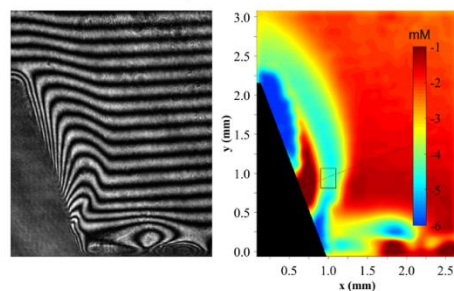


Figure S1.h. Copper deposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 130.0s. Initial CuSO_4 concentration=300 mM; $J=0 \text{ mA/cm}^2$.

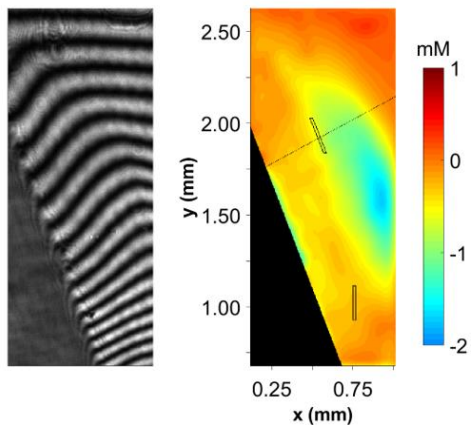


Figure S2.a. Copper electrodeposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 38.0s. Initial CuSO_4 concentration=300 mM; $J=4.0 \text{ mA/cm}^2$.

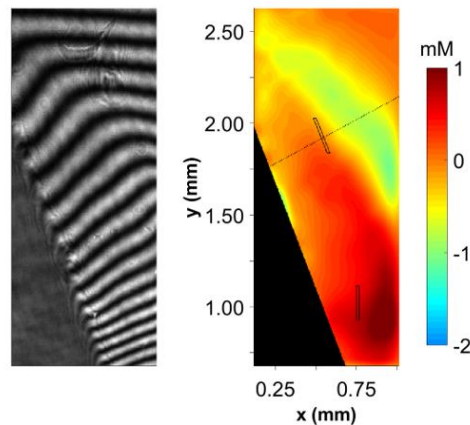


Figure S2.b. Copper electrodeposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 40.0s. Initial CuSO_4 concentration=300 mM; $J=4.0 \text{ mA/cm}^2$.

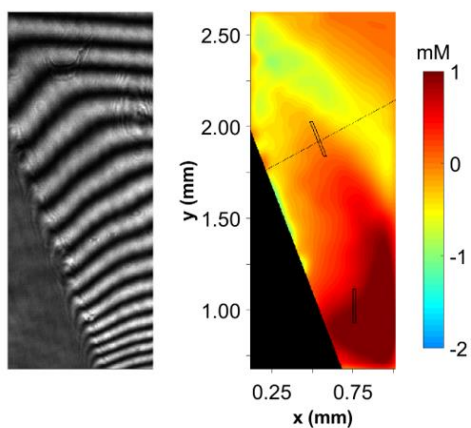


Figure S2.c. Copper electrodeposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 41.0s. Initial CuSO_4 concentration=300 mM; $J=4.0 \text{ mA/cm}^2$.

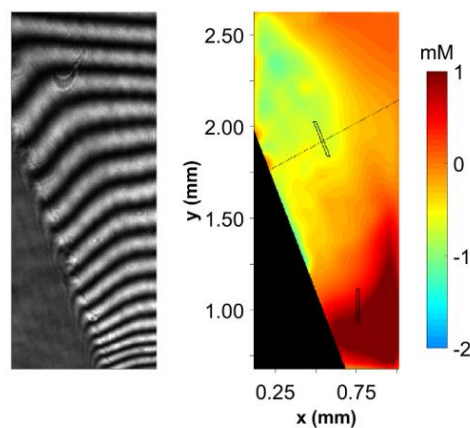


Figure S2.d. Copper electrodeposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 42.0s. Initial CuSO_4 concentration=300 mM; $J=4.0 \text{ mA/cm}^2$.

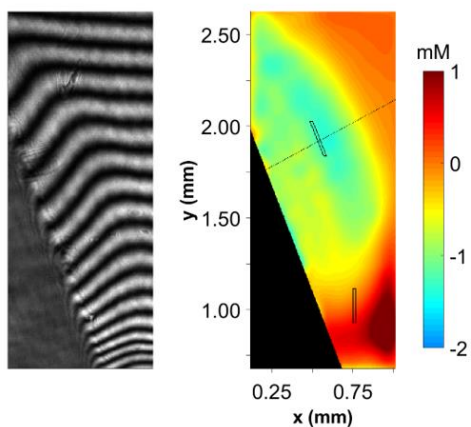


Figure S2.e. Copper electrodeposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 43.0s. Initial CuSO_4 concentration=300 mM; $J=4.0 \text{ mA/cm}^2$.

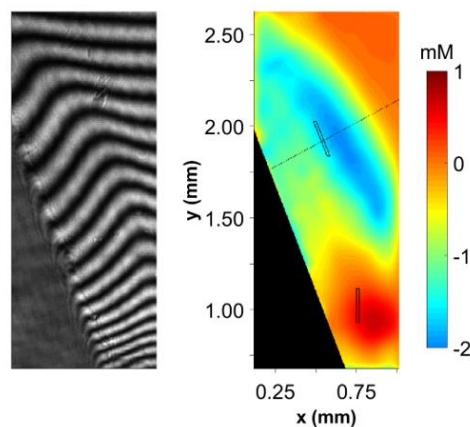


Figure S2.f. Copper electrodeposition on Fe cone (45 Deg): Interferogram and variation of concentration of CuSO_4 at time= 45.0s. Initial CuSO_4 concentration=300 mM; $J=4.0 \text{ mA/cm}^2$.

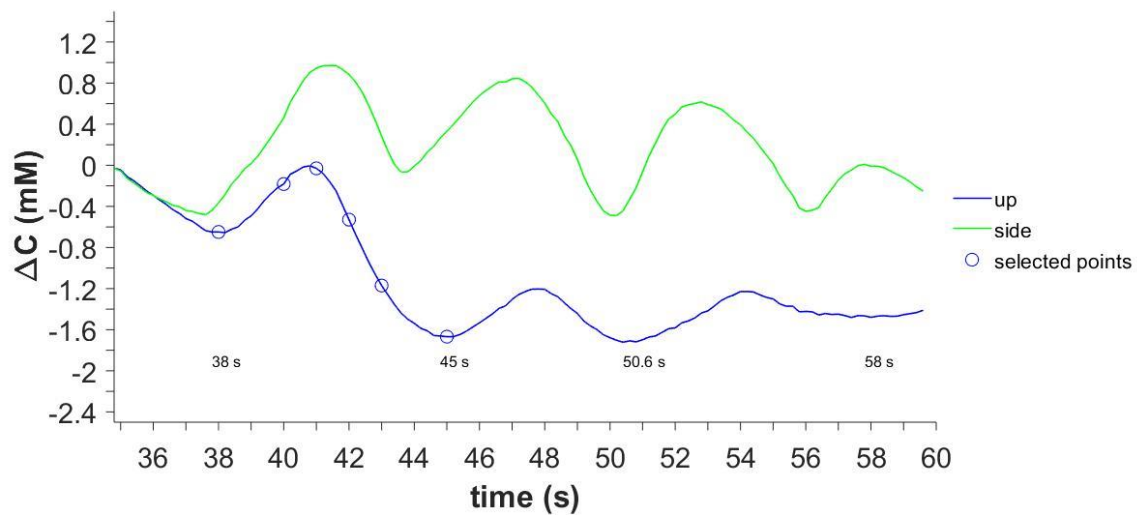


Figure S2.g. Case 4.0 mA/cm^2 . Average of concentration variation in 2 rectangular ROIs $0.02 \times 0.19 \text{ mm}$. Position coordinates (mm): $[x_1=0.52; z_1=1.93]; [x_2=0.8; z_2=1.02]$; period of oscillation: $6.67 \pm 0.95 \text{ s}$