

Supporting information

NaBH₄-Poly(ethylene oxide) Composite Electrolyte for All-Solid-State Na-Ion Batteries

*Xiaoxuan Luo,^a Kondo-Francois Aguey-Zinsou^{*b}*

^a MERLin, School of Chemical Engineering, The University of New South Wales, Sydney NSW 2052, Australia,

^b MERLin, School of Chemistry, University of Sydney, Sydney NSW 2006, Australia.

* f.aguey@sydney.edu.au

Table of Figures

Figure S1. (a) Nyquist plots of Hy-NaBH₄-20PEO from 25°C to 65°C. (b) magnification of the dash box in Figure S1(a)).....2

Figure S2. Current -time curves following DC polarization of Hy-NaBH₄-20PEO with 10 mV polarization voltage at 45 °C. (b) The Nyquist plot of Hy-NaBH₄-20PEO before and after DC polarization. The equivalent circuit was shown as insert.3

Figure S3. O1s XPS spectra of Hy-NaBH₄-20PEO.....4

Figure S4. FTIR of PEO5

Table S1 the lattice pararm of NaBH₄ in Hy-NaBH₄-PEO calculated from XRD by Retvield analysis.....6

Table S2. The degreed of crystallinity (χ_c) of PEO in Hy-NaBH₄-PEO calculated by using melting enthalpy in DSC.....6

Table S3. Na⁺ transference number ($tNa +$) of Hy-NaBH₄-20PEO.....6

Table S4. Summarized of PEO composite sodium solid state electrolytes 7

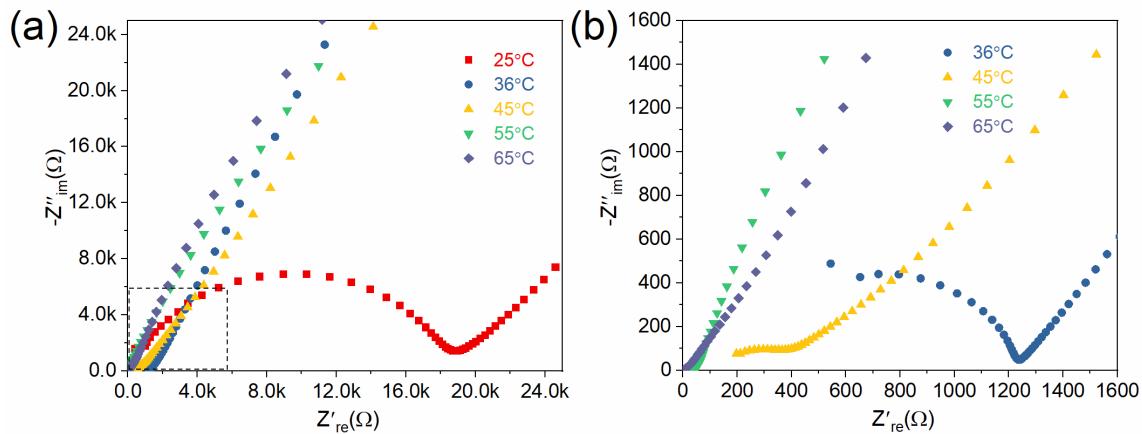


Figure S1. (a) Nyquist plots of Hy-NaBH₄-20PEO from 25°C to 65°C. (b) magnification of the dash box in Figure S1(a).

At room temperature, the Nyquist plot consists of a semicircle and line tail consistent with that of the reported borohydride-based electrolytes.¹ R is the intercept of semicircle with the X axis. With temperatures increasing, the magnitude of the semicircle gradually decreases.

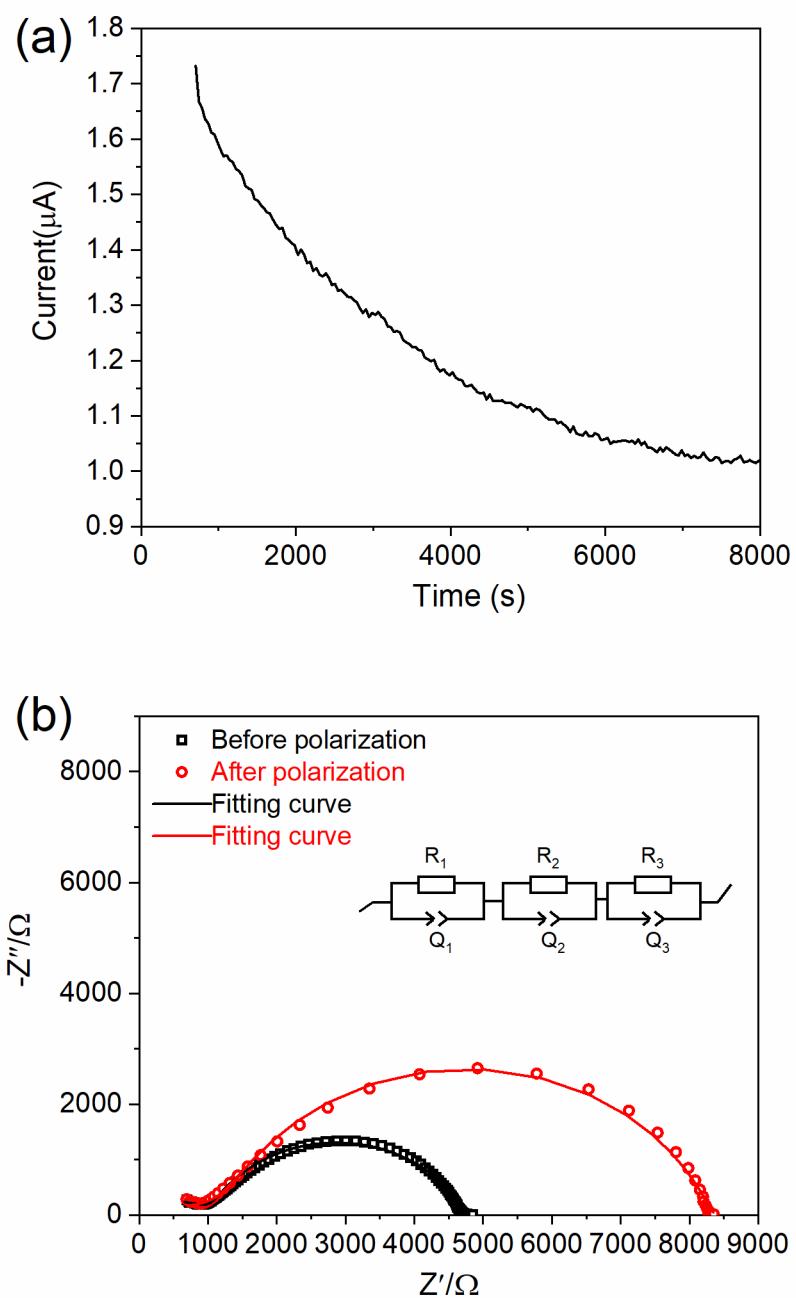


Figure S2. Current -time curves following DC polarization of Hy-NaBH₄-20PEO with 10 mV polarization voltage at 45 °C. (b) The Nyquist plot of Hy-NaBH₄-20PEO before and after DC polarization. The equivalent circuit was shown as insert.

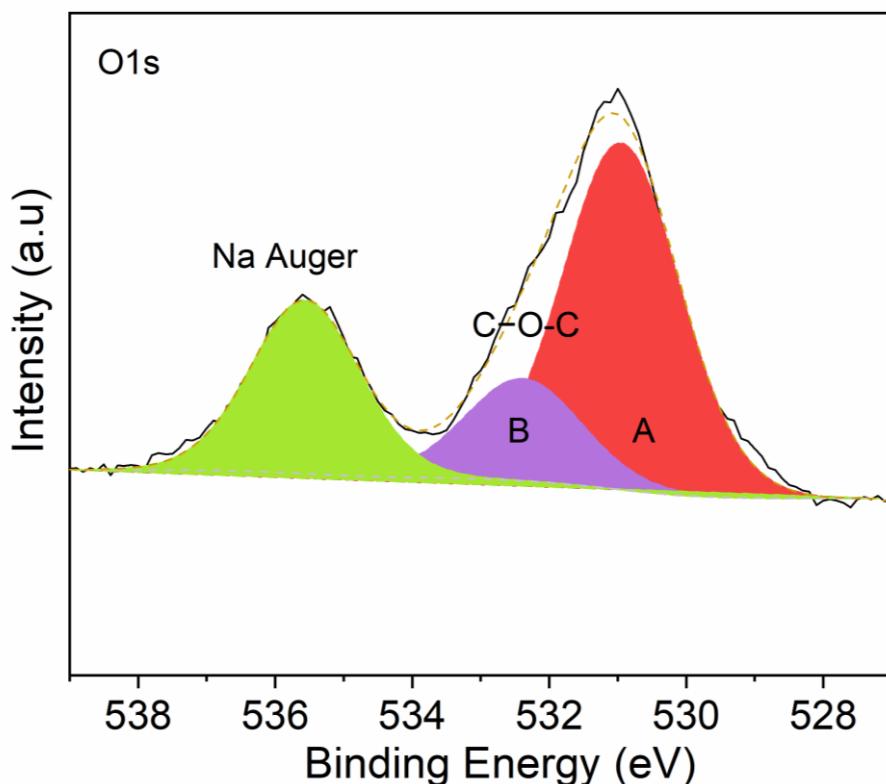


Figure S3. O1s XPS spectra of Hy-NaBH₄-20PEO.

The peak B located at 532.3 eV is assigned to the C-O-C from PEO.² The peak A may relate to the boron-oxide species,³ which is consistent with 191.2 eV in B1s spectrum (Figure 3a).

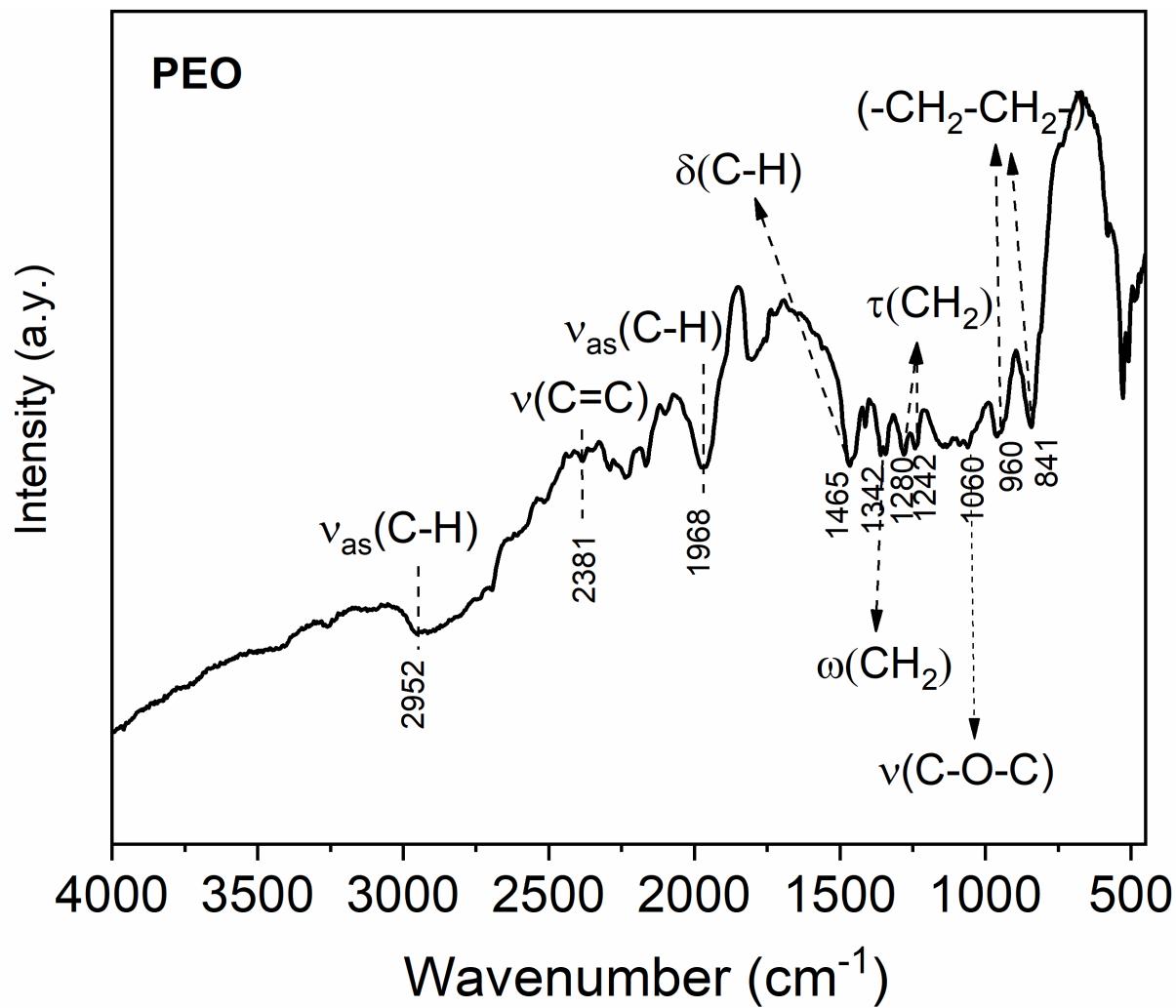


Figure S4. FTIR of PEO

Table S1 the lattice parameter of NaBH₄ in Hy-NaBH₄-PEO calculated from XRD by Retviel analysis

Materials	Lattice parameter a ($\pm 0.001 \text{ \AA}$)
Unmodified NaBH ₄	6.164
Hy- NaBH ₄ -5PEO	6.163
Hy- NaBH ₄ -10PEO	6.157
Hy- NaBH ₄ -20PEO	6.160
Hy- NaBH ₄ -30PEO	6.159

Table S2. The degree of crystallinity (χ_c) of PEO in Hy-NaBH₄-PEO calculated by using melting enthalpy in DSC.

Materials	Degree of crystallinity (χ_c)
PEO	35.5%
Hy- NaBH ₄ -5PEO	10.7%
Hy- NaBH ₄ -10PEO	18.8%
Hy- NaBH ₄ -20PEO	20.8%
Hy- NaBH ₄ -30PEO	21.6%

Table S3. Na⁺ transference number (t_{Na^+}) of Hy-NaBH₄-20PEO.

Materials	$I_{(t=0)}/\mu A$	$I_{(t=\infty)}/\mu A$	$R_{(t=0)}/\Omega$	$R_{(t=\infty)}/\Omega$	t_{Na^+}
Hy-NaBH ₄ -20PEO	1.73	1.0	3630	6569	0.54

$$t_{Na^+} = \frac{I_\infty (\Delta V - I_0 R_0)}{I_0 (\Delta V - I_\infty R_\infty)}$$

Table S4. Summarized of PEO composite sodium solid state electrolytes

Materials	Ionic conductivity (S cm ⁻¹)	Transference number	Ref
PEO-NaPF ₆	6.3×10^{-4} @ 80 °C	0.58	4
PEO–NaClO ₄ – Na ₃ Zr ₂ Si ₂ PO ₁₂	7.8×10^{-4} @ 75 °C	NA	5
Na ₃ PS ₄ –PEO	9.5×10^{-4} @ 25 °C	NA	6
Na ₃ Zr ₂ Si ₂ PO ₁₂ – PEO ₁₂ –NaFSI	2.2×10^{-5} @ 25 °C	NA	7
PEO ₂₀ :NaTFSI	1.1×10^{-3} @ 80 °C	0.39	8
Hy-NaBH ₄ -20PEO	1.6×10^{-3} @ 45 °C	0.54	This work

References

- Roedern, E.; Kühnel, R.-S.; Remhof, A.; Battaglia, C., Magnesium Ethylenediamine Borohydride as Solid-State Electrolyte for Magnesium Batteries. *Sci. Rep.* **2017**, *7*, 46189.
- Eriksson, T.; Andersson, A. M.; Bishop, A. G.; Gejke, C.; Gustafsson, T.; Thomas, J. O., Surface Analysis of LiMn₂O₄ Electrodes in Carbonate-Based Electrolytes. *J. Electrochem. Soc.* **2002**, *149* (1), A69.
- Pilli, A.; Jones, J.; Chugh, N.; Kelber, J.; Pasquale, F.; LaVoie, A., Atomic layer deposition of BN as a novel capping barrier for B₂O₃. *J. Vac. Sci* **2019**, *37* (4).
- Zhang, Q.; Lu, Y.; Yu, H.; Yang, G.; Liu, Q.; Wang, Z.; Chen, L.; Hu, Y.-S., PEO-NaPF₆ Blended Polymer Electrolyte for Solid State Sodium Battery. *J. Electrochem. Soc.* **2020**, *167* (7), 070523.

5. Yu, X.; Xue, L.; Goodenough, J. B.; Manthiram, A., A High-Performance All-Solid-State Sodium Battery with a Poly(ethylene oxide)–Na₃Zr₂Si₂PO₁₂ Composite Electrolyte. *ACS Materials Letters* **2019**, *1* (1), 132-138.
6. Xu, X.; Li, Y.; Cheng, J.; Hou, G.; Nie, X.; Ai, Q.; Dai, L.; Feng, J.; Ci, L., Composite solid electrolyte of Na₃PS₄-PEO for all-solid-state SnS₂/Na batteries with excellent interfacial compatibility between electrolyte and Na metal. *Journal of Energy Chemistry* **2020**, *41*, 73-78.
7. Zhang, Z.; Zhang, Q.; Ren, C.; Luo, F.; Ma, Q.; Hu, Y.-S.; Zhou, Z.; Li, H.; Huang, X.; Chen, L., A ceramic/polymer composite solid electrolyte for sodium batteries. *J. Mater. Chem. A* **2016**, *4* (41), 15823-15828.
8. Serra Moreno, J.; Armand, M.; Berman, M. B.; Greenbaum, S. G.; Scrosati, B.; Panero, S., Composite PEO_n:NaTFSI polymer electrolyte: Preparation, thermal and electrochemical characterization. *J. Power Sources* **2014**, *248*, 695-702.