

Supporting Information

Ternary PEO/PVDF-HFP-based polymer electrolytes for Li-ion batteries

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Table S1. Vibrations assigned in IR spectra.

Pure PVDF-HFP	Pure PEO	PEO/PVDF- HFP	Pure LiTFSI and pure EMITFSI	tPE PEO _y PVDF-HFP _z y : z						Vibration	Reference	
				25:75	30:70	35:65	40:60	45:55	50:50			
Wavenumber [cm⁻¹]												
	2886	2882		2920	2922	2925	2927	2933	2927	C-H stretching vibration of PEO	[45]	
	1359	1352		absent in ternary polymer electrolytes						CH ₂ wagging vibration of PEO	[29,46]	
	1342	1342		absent in ternary polymer electrolytes						CH ₂ bending vibration of PEO	[29,46]	
	1100	1097		absent in ternary polymer electrolytes						symmetric and asymmetric C-O-C stretching of PEO	[29,46]	
	952	953		955	955	956	956	953	956	C-O stretching vibration of PEO	[45]	
	841	842		838	838	838	837	836	836	CH ₂ rocking vibration of PEO	[45,46]	
	3024, 2985			absent in ternary polymer electrolytes						asymmetric and symmetric C-H stretching vibration of PVDF-HFP	[43]	
	1400	1403		absent in ternary polymer electrolytes						deformation vibration of the C-H bond of PVDF-HFP	[16]	
	974	975		absent in ternary polymer electrolytes, at 972 weak peak in 45:55						crystalline phase (alpha phase) of PVDF-HFP, C-F stretching	[43]	
	795	795		absent in ternary polymer electrolytes						crystalline phase (alpha phase) of PVDF-HFP, CF ₃ stretching vibration	[43]	
	762	762		absent in ternary polymer electrolytes						crystalline phase (alpha phase) of PVDF-HFP, CH ₂ rocking vibration	[43]	
	613	613		absent in ternary polymer electrolytes						crystalline phase (alpha phase) of PVDF-HFP, mixed mode of CF ₂ bending and CCC skeletal vibration	[43]	
	532	532		absent in ternary polymer electrolytes						crystalline phase (alpha phase) of PVDF-HFP, wagging vibrations of CF ₂ group	[43]	
	871	872		879	879	879	879	879	879	amorphous phase (beta phase) of PVDF-HFP	[43]	

841				overlapping with CH ₂ rocking vibration of PEO						amorphous phase (beta phase) of PVDF-HFP	[43]	
				3200-3100						C-H stretching vibration of imidazolium ring	[43]	
				1346	1346	1347	1347	1348	1352	1348	asymmetric SO ₂ stretching vibration of LiTFSI and EMIMTFSI	[43]
				1326	1327	1327	1327	1328	1329	1329	C-SO ₂ -N vibration of the LiTFSI and EMIMTFSI	[43]
				1193	1183	1183	1182	1182	1179	1181	symmetric stretching vibration of CF ₃ group of LiTFSI and C-H vibration of imidazolium ring of IL EMIMTFSI	[43]
				1133	1131	1131	1131	1132	1132	1133	C-SO ₂ -N vibration of LiTFSI and EMIMTFSI	[43]
				1051	1052	1052	1052	1052	1053	1052	asymmetric S-N-S stretching vibration of conducting salt and IL	[43]
				740	742	742	741	741	739	741	overlap of CF ₃ bending vibration with the combination of C-S of the EMIMTFSI and S-N stretching vibration of LiTFSI	[43]
				600	600	600	600	600	599	600	asymmetric SO ₂ bending vibration of LiTFSI and EMIMTFSI	[6]
				612	612	612	612	612	612	612	Ring op asymmetric bending vibration of EMIM ⁺	[6]
				571	570	570	570	570	570	570	asymmetric bending vibration of CF ₃ group of LiTFSI	[43]

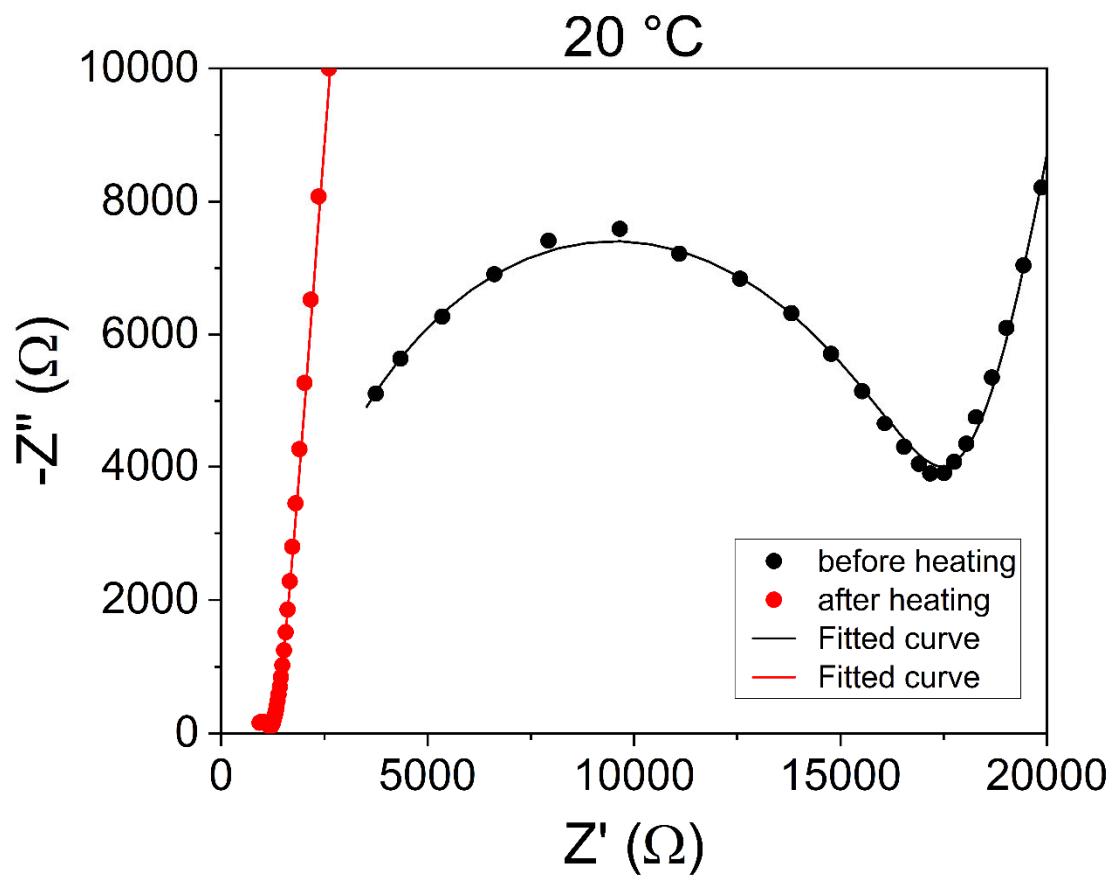


Figure S1. Nyquist plots of tPE PEO₃₅PVDF-HFP₆₅ at 20 °C recorded for materials before and after heating to 80 °C.

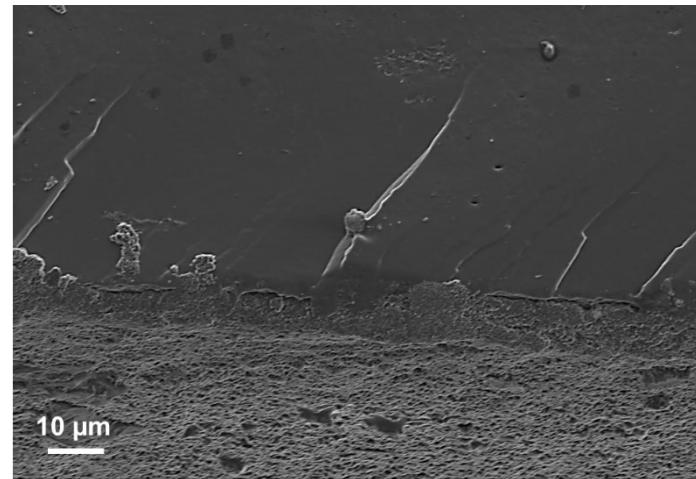
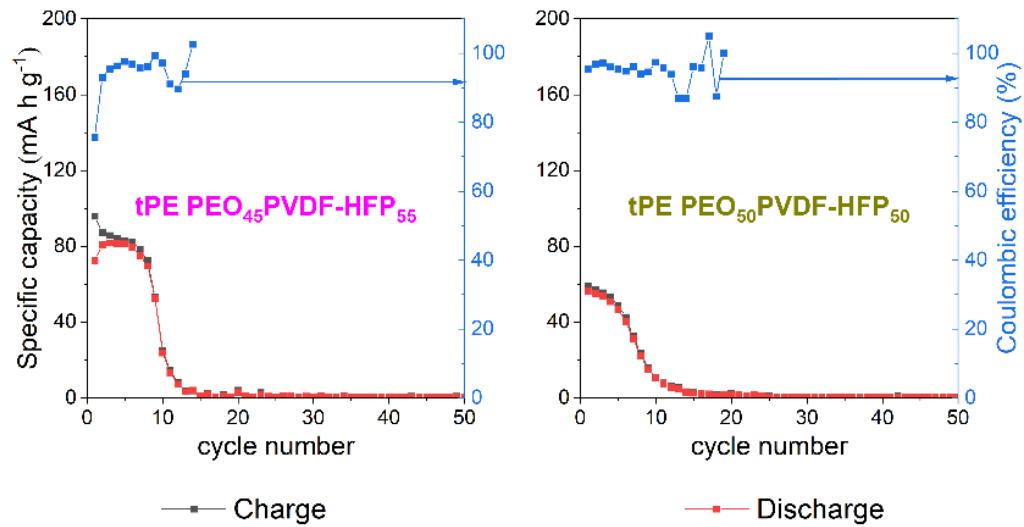


Figure S2. Left: Cycle stability of full cells LTO/tPE PEO_yPVDF-HFp/LFP with a high mass fraction of PEO at 0.1C at 25 °C. Right: SEM image of the sandwich LTO/tPE PEO₅₀PVDF-HFP₅₀/LFP after cycling, reflecting low mechanical integrity.

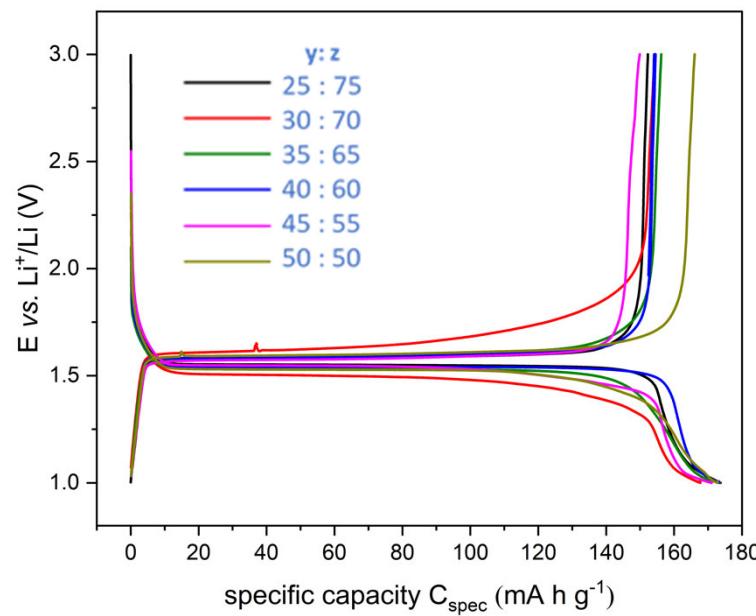
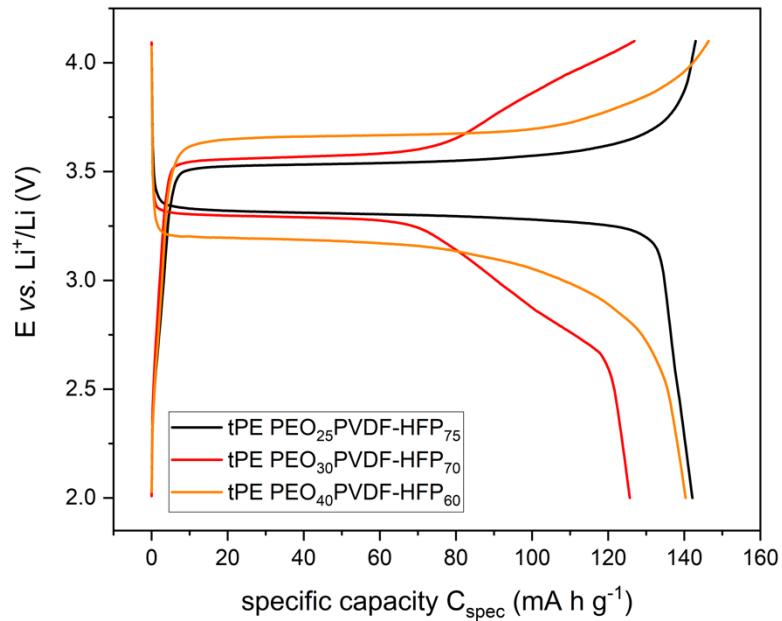


Figure S3. Charge-discharge curves of half cells LFP/ tPE PEO_yPVDF-HFP_z/Li (left) and LTO/ tPE PEO_yPVDF-HFP_z/Li (right) with various electrolyte compositions at 60 °C.

Literature

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