

Supplementary Materials

Optimization by Central Composite Experimental Design of the Synthesis of Physically Crosslinked Chitosan Spheres

Sara Isabel Zamora Lagos ¹, Jefferson Murillo Salas ¹, Mayra Eliana Valencia Zapata ¹, José Hermínsul Mina Hernández ¹ and Carlos David Grande Tovar ^{2,*}

¹ Escuela de Ingeniería de Materiales, Facultad de Ingeniería, Universidad del Valle, Calle 13 No. 100-00, Santiago de Cali 760032, Colombia; sara.zamora@correounivalle.edu.co (S.I.Z.L.); Jefferson.murillo@correounivalle.edu.co (J.M.S.); valencia.mayra@correounivalle.edu.co (M.E.V.Z.); jose.mina@correounivalle.edu.co (J.H.M.H.)

² Programa de Química, Facultad de Ciencias, Universidad del Atlántico, Carrera 30 número 8-49, Puerto Colombia 081008, Colombia

* Correspondence: carlosgrande@mail.uniatlantico.edu.co; Tel.: +57-5-3599484

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Table S1. Central composite experimental design for the CS spheres synthesis.

Table S2. Statistical models obtained for CS spheres diameter.

Figure S1. Pareto chart of the standardized effects. Response: Diameter. $RH\ 52\% \pm 0.1\%$

Figure S2. Pareto chart of the standardized effects. Response: Diameter. Room R.H.

Figure S3. Diameter contour plot for CS spheres. Conditions: Needle 2, $RH\ 10\% \pm 0.1\%$.

Figure S4. Diameter contour plot for CS spheres, Conditions: Needle 1, R.H. $52\% \pm 0.1$.

Figure S5. Diameter contour plot for CS spheres, Conditions: Needle 2, R.H. $52\% \pm 0.1$.

Figure S6. Diameter contour plot for CS spheres, Conditions: Needle 1, Room R.H.

Figure S7. Diameter contour plot for CS spheres, Conditions: Needle 2, Room R.H.

Figure S8. Optimization chart, Conditions: Needle 1, R.H. $10\% \pm 0.1\%$.

Figure S9. Optimization plot, Conditions: Needle 2, R.H. $10\% \pm 0.1\%$.

Figure S10. Optimization plot, Conditions: Needle 1, R.H. $52\% \pm 0.1\%$.

Figure S11. Optimization plot, Conditions: Needle 2, R.H. $52\% \pm 0.1\%$.

Figure S12. Optimization plot, Conditions: Needle 1, Room R.H.

Figure S13. Optimization plot, Conditions: Needle 2, Room R.H.

Figure S14. Chemical chitosan structure.

Figure S15. $^1\text{H-NMR}$ spectrum for CS sheets.

Figure S16. $^1\text{H-NMR}$ spectrum for CS spheres.

Table S1. Central composite experimental design for the CS spheres synthesis.

StdOrder	RunOrder	AAC (Fraction)	CS (Fraction)	NaOH (Fraction)	Needle
5	1	0.018	0.007	0.176	1
36	2	0.030	0.012	0.140	2
12	3	0.030	0.020	0.140	1
37	4	0.030	0.012	0.140	2
8	5	0.042	0.017	0.176	1
30	6	0.050	0.012	0.140	2
39	7	0.030	0.012	0.140	2
27	8	0.018	0.017	0.176	2
1	9	0.018	0.007	0.104	1
17	10	0.030	0.012	0.140	1
6	11	0.042	0.007	0.176	1
13	12	0.030	0.012	0.080	1
32	13	0.030	0.020	0.140	2
34	14	0.030	0.012	0.200	2
9	15	0.010	0.012	0.140	1
24	16	0.042	0.017	0.104	2
31	17	0.030	0.004	0.140	2
15	18	0.030	0.012	0.140	1
38	19	0.030	0.012	0.140	2
19	20	0.030	0.012	0.140	1
21	21	0.018	0.007	0.104	2
3	22	0.018	0.017	0.104	1
29	23	0.010	0.012	0.140	2
40	24	0.030	0.012	0.140	2
18	25	0.030	0.012	0.140	1
33	26	0.030	0.012	0.080	2
11	27	0.030	0.004	0.140	1
20	28	0.030	0.012	0.140	1
16	29	0.030	0.012	0.140	1
25	30	0.018	0.007	0.176	2
14	31	0.030	0.012	0.200	1
4	32	0.042	0.017	0.104	1
2	33	0.042	0.007	0.104	1
23	34	0.018	0.017	0.104	2
28	35	0.042	0.017	0.176	2
26	36	0.042	0.007	0.176	2
22	37	0.042	0.007	0.104	2
10	38	0.050	0.012	0.140	1
7	39	0.018	0.017	0.176	1
35	40	0.030	0.012	0.140	2

Table S2. Statistical models obtained for CS spheres diameter.

Conditions: R.H 10%% ± 0.1%	
Needle 1	$Diameter = 15,82 + 16,4 AAC - 1239 CS - 67,8 NaOH$ $- 769 AAC * AAC + 33381 CS * CS$ $+ 265,1 NaOH * NaOH + 3024 AAC * CS$
Needle 2	$Diameter = 17,11 + 3,3 AAC - 1239 CS - 75,0 NaOH$ $- 769 AAC * AAC + 33381 CS * CS$ $+ 265,1 NaOH * NaOH + 3024 AAC * CS$
Conditions: R.H 52%% ± 0.1%	
Needle 1	$Diameter = 20,18 + 9,30 AAC - 1646 CS - 83,7 NaOH$ $+ 49657 CS * CS + 320,2 NaOH * NaOH$
Needle 2	$Diameter = 20,89 + 9,30 AAC - 1646 CS - 90,0 NaOH$ $+ 49657 CS * CS + 320,2 NaOH * NaOH$
Conditions: Room R.H	
Needle 1	$Diameter = 28,30 + 26,69 AAC - 2633 CS - 109,0 NaOH$ $+ 80234 CS * CS + 440,2 NaOH * NaOH$
Needle 2	$Diameter = 31,77 - 0,87 AAC - 2633 CS - 129,3 NaOH$ $+ 80234 CS * CS + 440,2 NaOH * NaOH$

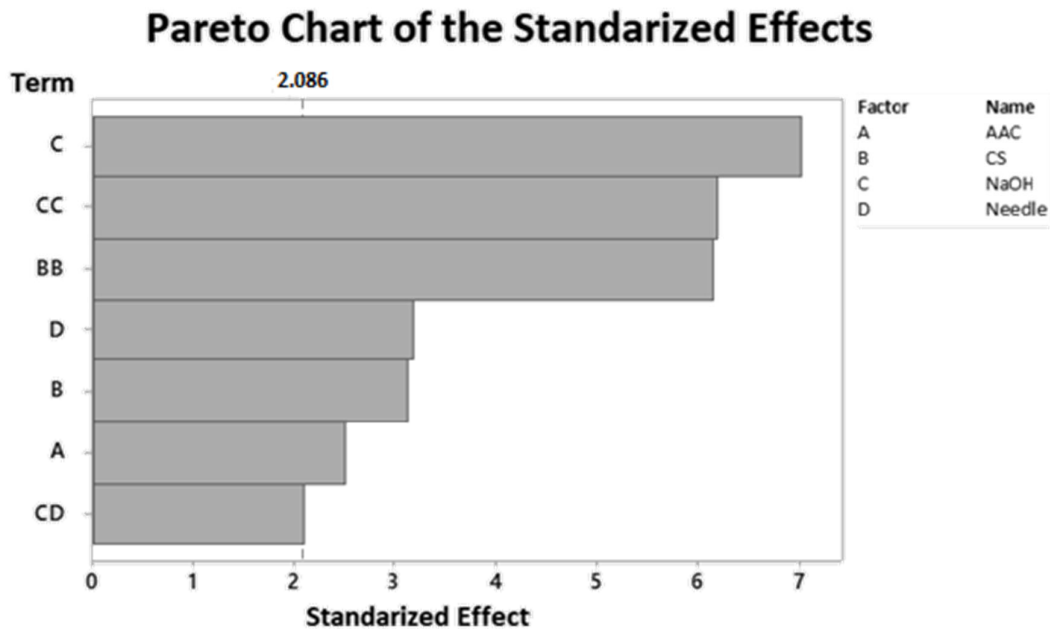


Figure S1. Pareto chart of the standardized effects. Response: Diameter. RH 52% ± 0.1%.

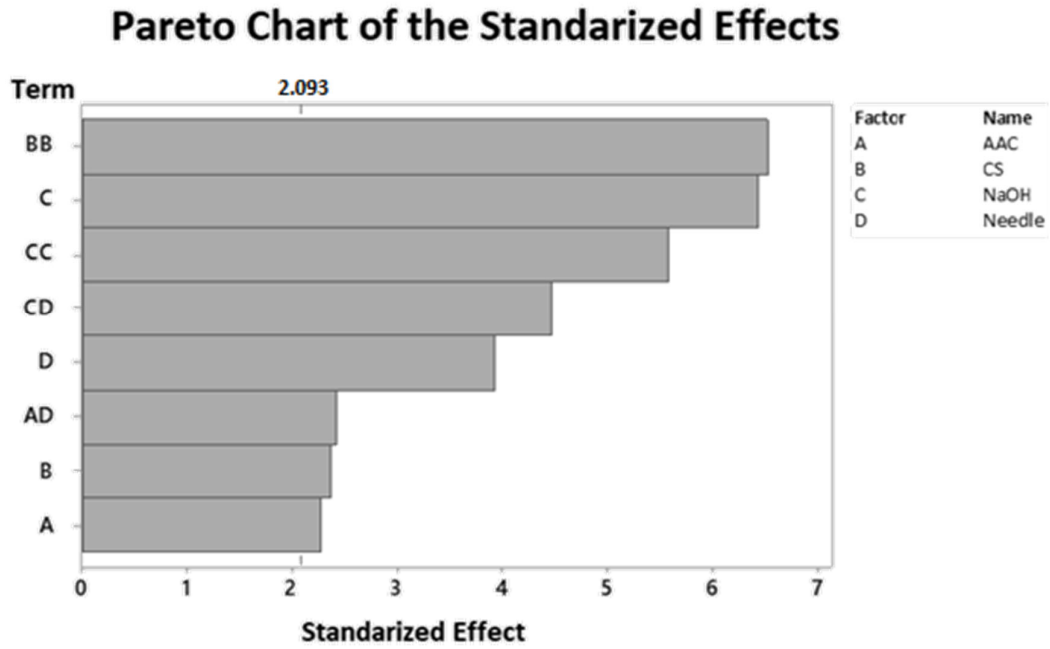


Figure S2. Pareto chart of the standardized effects. Response: Diameter. Room R.H.

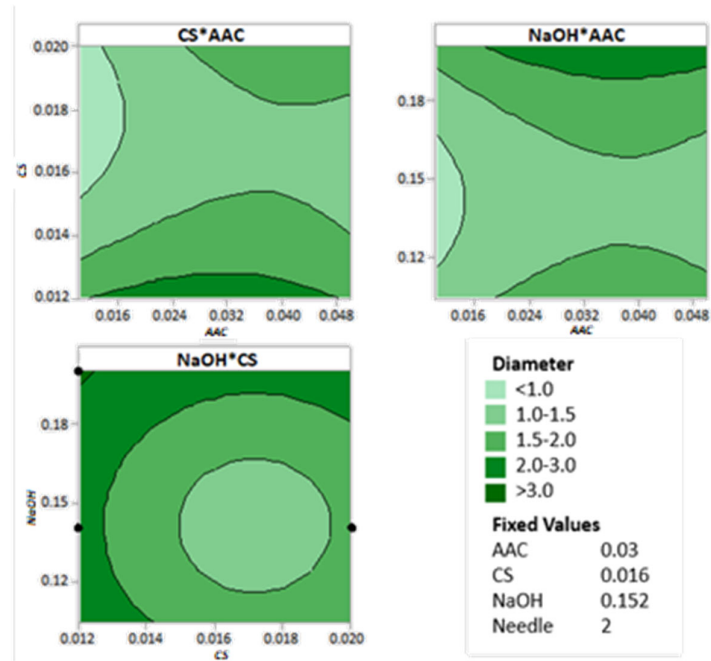


Figure S3. Diameter contour plot for CS spheres. Conditions: Needle 2, R.H. 10% ± 0.1.

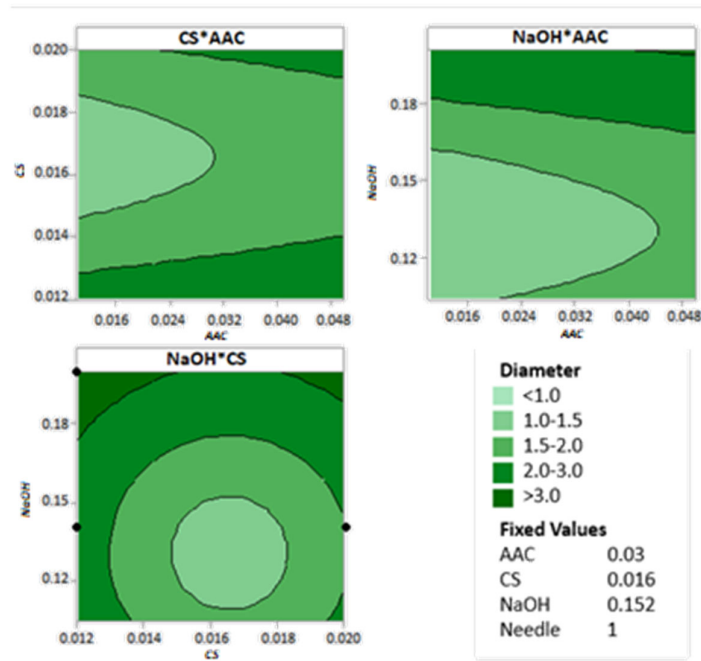


Figure S4. Diameter contour plot for CS spheres, Conditions: Needle 1, R.H. 52% ± 0.1.

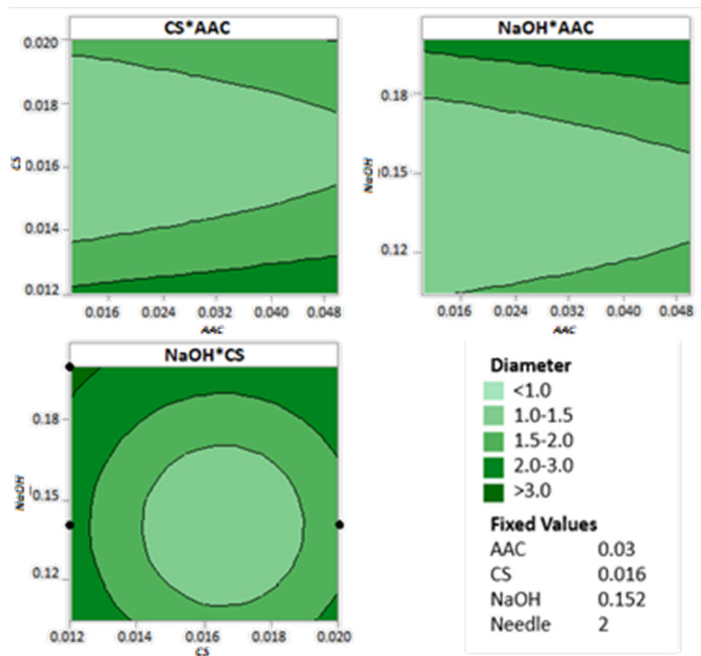


Figure S5. Diameter contour plot for CS spheres, Conditions: Needle 2, R.H. 52% ± 0.1.

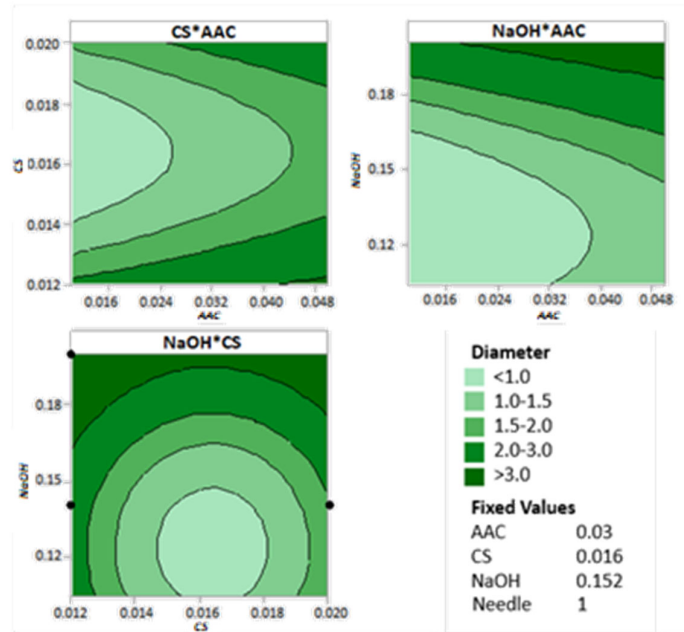


Figure S6. Diameter contour plot for CS spheres, Conditions: Needle 1, Room R.H.

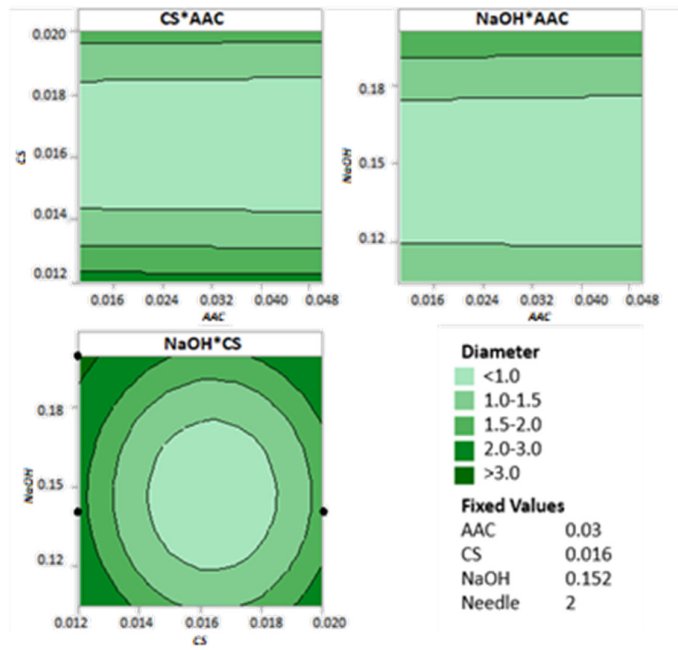


Figure S7. Diameter contour plot for CS spheres, Conditions: Needle 2, Room R.H.

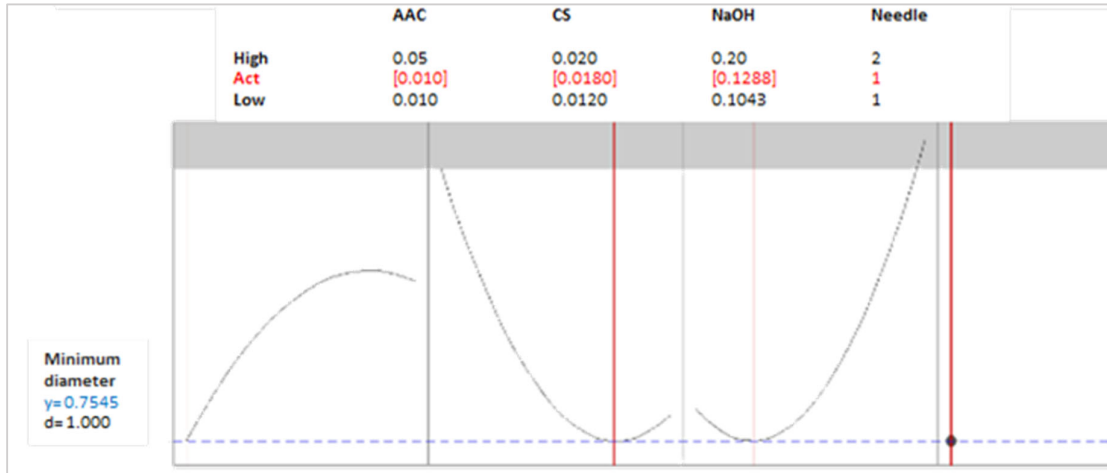


Figure S8. Optimization chart, Conditions: Needle 1, R.H. 10% ± 0.1%.

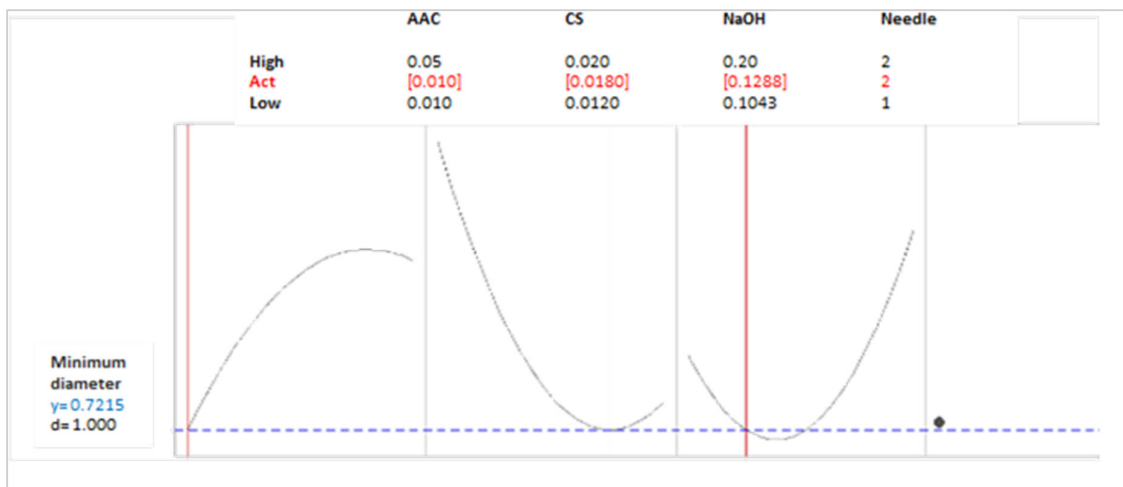


Figure S9. Optimization plot, Conditions: Needle 2, R.H. 10% ± 0.1%.

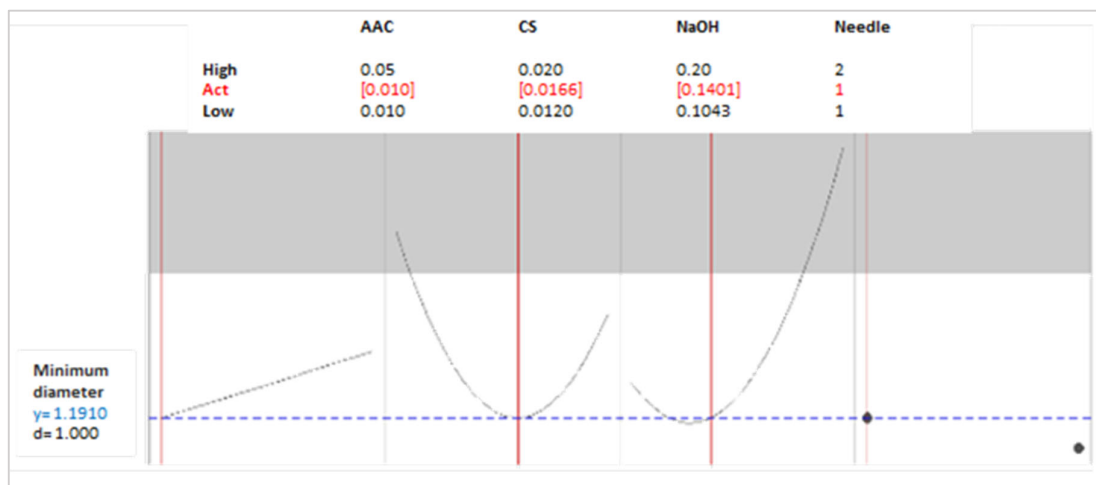


Figure S10. Optimization plot, Conditions: Needle 1, R.H. 52% ± 0.1%.

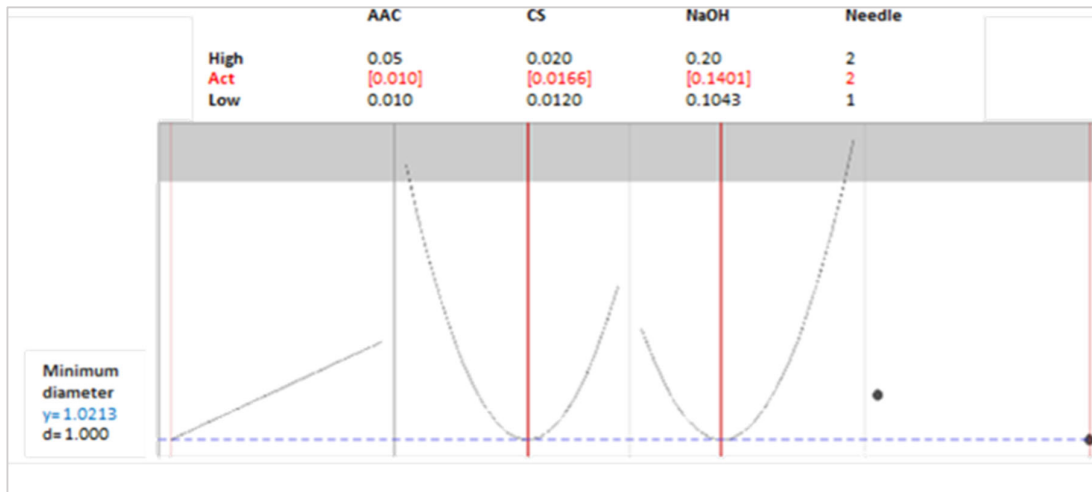


Figure S11. Optimization plot, Conditions: Needle 2, R.H. 52% ± 0.1%.

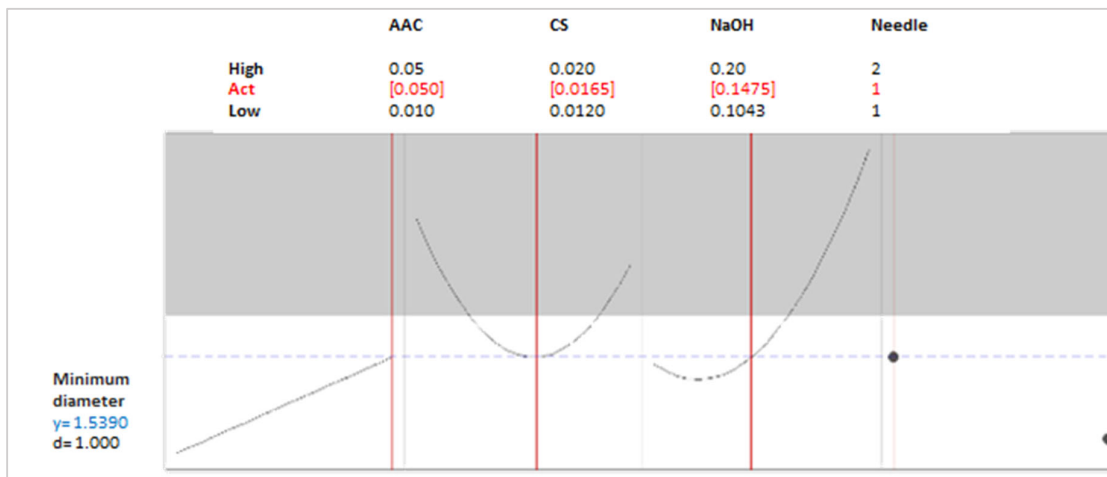


Figure S12. Optimization plot, Conditions: Needle 1, Room R.H.

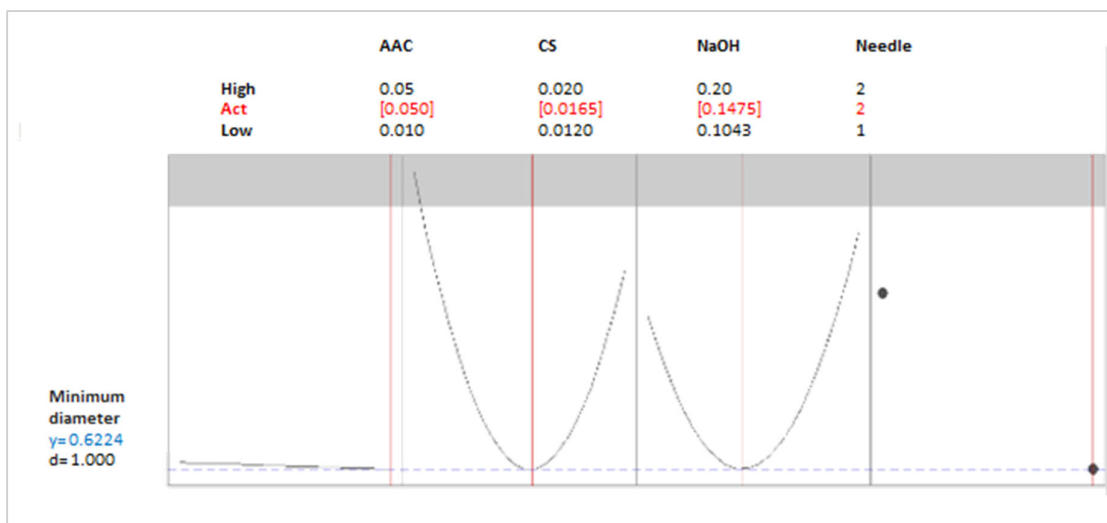


Figure S13. Optimization plot, Conditions: Needle 2, Room R.H.

Proton nuclear magnetic resonance ($^1\text{H-NMR}$)

Based on the chemical structure of chitosan (Figure S13) and $^1\text{H-NMR}$ spectrum for CS sheets and spheres (Figure S14 and S15), it could be observed that there is a peak at 2.08 and 1.59 ppm respectively which represents the 3 protons of the methyl group of N-acetyl glucosamine. The 3.2 and 3.23 ppm peaks present in the spectrum of CS sheets and spheres corresponds to proton number 2 of D-glucosamine, that is, the deacetylated unit. The range between 3.5 and 4 ppm for both spectrums corresponds to protons 2-6. Finally the peak at 4.60 and 4.1 ppm for CS sheets and spheres, corresponds to proton number 1 of D-glucosamine and the peak at 4.88 and 4.39 ppm for CS sheets and spheres corresponds to proton number 1 N-acetyl glucosamine[16]. Using the equation 2 and the results obtained of $^1\text{H-NMR}$, for CS sheets DDA is 75.01%, while for CS spheres DDA corresponds to a 60.78%.

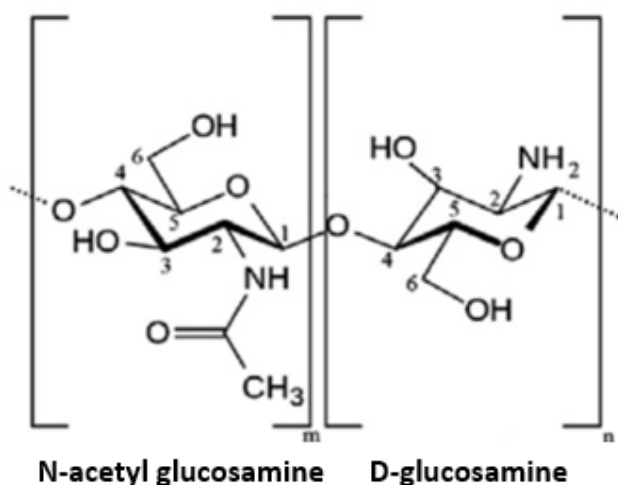


Figure S14. Chemical chitosan structure.[34].

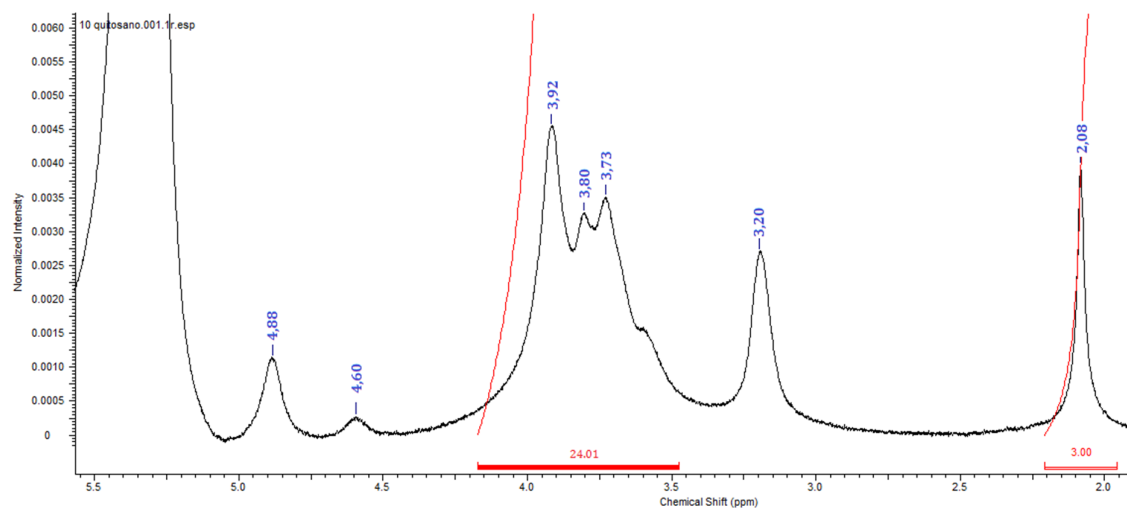


Figure S15. $^1\text{H-NMR}$ spectrum for CS sheets.

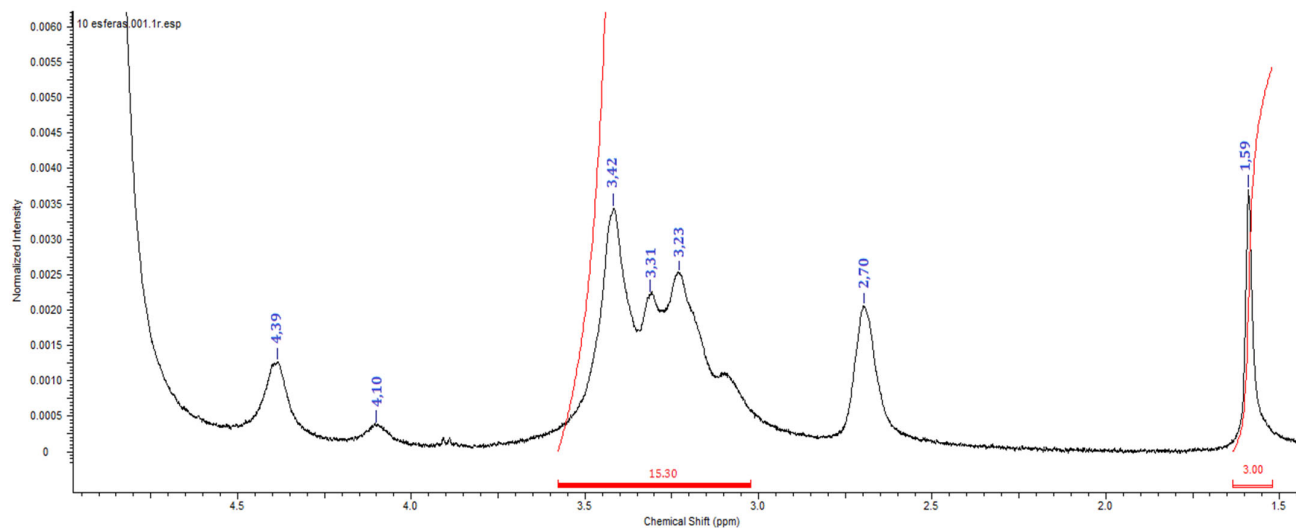


Figure S16. $^1\text{H-NMR}$ spectrum for CS spheres.



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