


Review

Balance of Nitrate and Ammonium in Tropical Soil Conditions: Soil Factors Analyzed by Machine Learning

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Abstract: The nitrogen/N dynamic is complex and affected by soil management (i.e., residue accumulation and correction/fertilization). In soil, most of the N is combined with organic matter (organic forms), but the N forms absorbed by plants are ammonium/ NH_4^+ and nitrate/ NO_3^- (inorganic forms). The N recommendation for agriculture crops does not observe the N available in the soil (organic or inorganic), indicating a low efficiency in nitrogen management in soil. Based on the hypothesis that the stocks of NO_3^- and NH_4^+ can be used as indicative of N status in soil but with high variation according to soil factors (soil uses and management), the objective of the study was to (i) analyze the balance of nitrate and ammonium in tropical soil with different uses and management and (ii) use machine learning to explain the nitrogen dynamic in soil and the balance of nitrate and ammonium. The results showed that soil N stocks and pH promoted the formation of three clusters with the similarity between Cluster 1 (clay texture) and Cluster 2 (loam texture), represented by higher contents of nitrate as a result of high nitrification rate and lower contents of ammonium in soil. Cluster 3 (sand texture) was isolated with different N dynamics in the soil. In agricultural soils, the content of NO_3^- tends to be higher than the content of NH_4^+ . There is a high nitrification rate in clay soil explained by higher organic matter and clay content that promotes soil biology. Based on the results of machine learning, for clay and loam soil, the contents of NO_3^- can be used as indicative of N status as a final result of nitrification rate and higher variation in soil. However, in sandy soil, NO_3^- can not be used as indicative of N status due to N losses by leaching.



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1. Introduction

Nitrogen (N) is a nutrient that plants need in the greatest quantities and most frequently limits plant growth and crop yields. Agricultural soils do not provide sufficient N to achieve higher crop production due to low organic matter stocks, mainly in tropical soils. An alternative to increase the level of N in the soil is human intervention with the application of N as fertilizers (organic or mineral). Positive effects of N fertilizers in sugarcane (*Saccharum officinarum*) [1], corn (*Zea mays* L.) [2,3], rice (*Oryza sativa*) [4], and other crops are demonstrated in the literature. In the last years, there has been a demand for nitrogen between 36,000 and 40,000 tons in agricultural lands, representing a constant increment [5]. Also, there is a tendency to increase N consumption in the next years with an increment of areas to produce food, fiber, and energy.

The main N mineral sources are found as amide, ammonium (NH_4^+), and nitrate (NO_3^-). The amide is the base of N fertilizers more commonly used in the world, such as urea ($\text{CO}(\text{NH}_2)_2$), with 45% of N in the formulation. When applied in soil, the amide is converted to nitrate or ammonium by urease (enzyme), which hydrolyzes the urea to convert into NH_4^+ and NO_3^- releasing an OH^- and promoting an increase in soil pH and N loss by volatilization of ammonia (NH_3) [6]. However, even with this N loss, urea is the N mineral fertilizer most used due to the smaller market nitrogen cost per kg.

Nitrate and ammonium sources are the second basic sources of N in agriculture, with a natural flux between ammonium and nitrate in the soil [7]. The source of nitrate usually applied in agricultural soil is calcium nitrate with a higher N use efficiency due to the absence of N losses by NH_3 . In general, the mineral N fertilizers present a low N use efficiency (i.e., an average of 50% in the production of grains [8,9] and an average of 17–41% in the production of sugarcane [10]). In soil, the N losses are associated with ammonia volatilization [6], nitrous oxide emission [11], and nitrate leaching [12]. Alternative N sources are requested to increase the N use efficiency with the adequate use of the 4R principles (right source, right rate, right time, and right place) to achieve efficient use of resources with harmony between the economy, environment, and society.

In soil, the N dynamic is complex and is affected by soil management (i.e., residue accumulation, soil correction, and fertilization). The main soil N reserves are mainly found as organic N, explaining the direct influence of organic matter on soil N dynamics [13]. For example, about 90% of N is organically combined on the soil surface [14]. Ferraz-Almeida et al. [15] showed that in agriculture and forested soil, the pools of labile and recalcitrant carbon and nitrogen are higher on the soil surface due to the accumulation of residues. On the soil surface, there is a high correlation between the pools of carbon and nitrogen [14,16,17]. Li et al. [18] classified the N in soil according to availability (labile pool and stable pool) and chemical components (i.e., hydrolyzable and non-hydrolyzable, ammonia, amino acid, and amino sugar). The labile pool and a recalcitrant pool are two important constituents of total organic carbon; (i) the labile pool is classified as the most sensitive C pool, (ii) while the recalcitrant fraction is characterized as the least sensitive and least altered by organic C gains or losses [14,15]. Wendling et al. [16] also classified the soil N in the labile pool and stable pool to explain the inputs and variations of N in agricultural soil. The main forms of organic N are the amino acid, followed by non-hydrolyzable, and amino sugar [18]. In contrast, the main forms of inorganic N in the soil are found as NO_3^- and NH_4^+ . Even when applied as organic forms, the organic N is converted into NO_3^- and NH_4^+ for plant absorption. There is no clear evidence of the absorption of organic N by plants, which explains the need to convert organic forms into NO_3^- into NH_4^+ .

Based on the hypothesis that the stocks of NO_3^- and NH_4^+ can be used as indicative of N status in soil but with high variation according to soil factors (soil uses and management), the objective of the study was to (i) analyze the balance of nitrate and ammonium in tropical soil with different uses and management and (ii) use machine learning to explain the nitrogen dynamic in soil and the balance of nitrate and ammonium.

2. Material and Methods

2.1. Study Characterization

The design of the study was based on the contents of ammonium and nitrate in cultivated soil with soybean (*Glycine max*), wheat (*Triticale* sp.), corn (*Zea mays*), sugarcane (*Saccharum officinarum* L.), and forest (Cerrado) with different soil textures (clay, sandy, and loam) in tropical conditions. Soils were classified as sandy, loam, and clay textural classes according to the textural triangle. The FAO soil classification system was used to classify the soils as Arenosols (sandy), Ferralsols (clay and loam), and Cambisol (loam).

The soils were collected in the field and analyzed in the laboratory, monitoring the nitrogen (N-NO_3^- and N-NH_4^+) and soil pH. The soil pH was determined by extraction using CaCl_2 solution (0.01 M) and determined by combining electrodes. The N-NO_3^- and N-NH_4^+ were determined by high-pressure liquid chromatography [19].

The dataset presented a total of 1278 observations distributed for sandy (342 observations), loam (342 observations), and loam soil (594 observations). Part of this dataset was obtained from the literature [20–23], and the other part from our laboratory archive. All data were selected based on the same soil management with alteration just on soil uses.

Data on N recommendations and expected productivity for sugarcane, corn, and wheat productions were monitored in the manual of fertilization recommendations in

Brazil [24]. The expected and recommended N ratio (*ERN*) was calculated based on expected productivity and N recommendation using Equation (1).

$$ERN = EP/RN \quad (1)$$

where *ERN* = expected and recommended N ratio, *EP* = expected productivity, and *RN* = recommended N.

2.2. Data Analysis

The quality of the dataset was tested using data normality (Shapiro–Wilk test; $p < 0.05$) and homogeneity of variance (Bartlett test; $p < 0.05$). The variability of the studied properties was previously evaluated using descriptive statistics calculating the average, variation (%), and minimum and maximum values.

The data on NO_3^- , NH_4^+ , and pH were analyzed through machine learning using the k-means clustering algorithm (unsupervised learning). The data were separated into 3 clusters by the Silhouette method, which is based on the comparison of tightness and separation. The distances between the observations and clusters were calculated with the Euclidean distance. A total of 1278 observations were tested in the k-means clustering algorithm. This algorithm groups points into clusters by minimizing the distances between each point and its cluster's centroid. Previous studies have used this algorithm with excellent results to determine the influence of land management and use on soil nutrient supply [13].

Subsequently, the variables were also subjected to multivariate exploratory analysis by hierarchical clustering. A similarity matrix was constructed with the Euclidean distance and the connections of the clusters were conducted with the Ward method. In this method, the distance between two groups is defined as the sum of squares between the two groups through all variables. The Euclidean distance among access to the set of variables was calculated, distinguishing between soil textures and exposing the group structure contained within the data in a dendrogram.

The correlation between soil texture and the processes of nitrification was calculated based on the distribution of sand, silt, and clay using the Correlation of Pearson ($p < 0.5$). To explain the N dynamic in the soil, schemes were added focusing on (i) the distribution of water and air in the soil profile (aerobic and anaerobic conditions) and (ii) the process of nitrification divided into two steps (conversion from ammonium to nitrite, and conversion from nitrite to nitrate).

All statistical analysis was performed in R (version 4.0.0; R Foundation for Statistical Computing, Vienna, Austria) and Python (version 3.8.3; Python Software Foundation, Wilmington, DE, USA).

3. Results and Discussion

3.1. Cluster Formation

The dataset of soil textures, N stocks, and pH promoted the formation of a one-dimensional plane represented by dimension 1 (total of 28.8) and dimension 2 (39.4%) and a total of 68.2% of the variability. The dataset was separated into three clusters accounting for a total sum of squares of 68.2%, separated into Cluster 1 (sum of squares = 578.24), Cluster 2 (sum of squares = 232.21), and Cluster 3 (sum of squares = 237.77), with a total of observations respectively of 252 (Cluster 1); 126 (Cluster 2); 126 samples (Cluster 3) in each cluster (Figures 1 and 2 and Table 1).

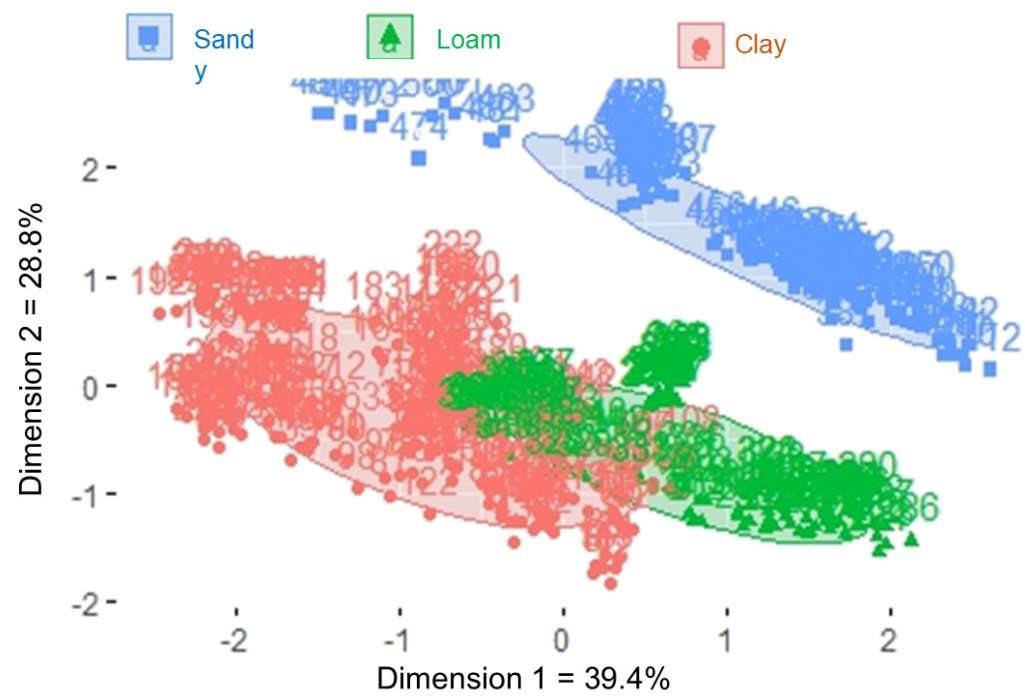


Figure 1. Cluster formation with data of soil texture, N stocks, and pH analyzed by machine learning (K-means). A total of 1278 observations were tested. In Figure, the number represents the identification of observation.

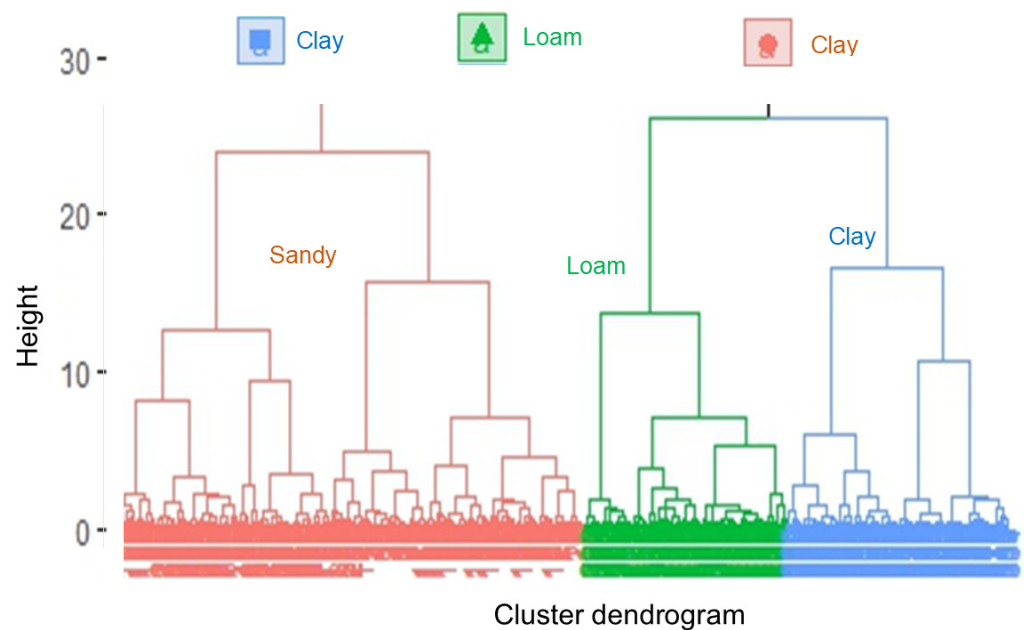


Figure 2. Cluster dendrogram with data on soil texture, N stocks, and pH analyzed by machine learning (K-means). A total of 1278 observations were tested.

The sandy soil was isolated in Cluster 1 with higher values of NH_4^+ (57.0 mg kg^{-1}) and lower values of NO_3^- ($=36.4 \text{ mg kg}^{-1}$) with pH ranging from 5.9 to 6.8 (Table 1). In the Cluster dendrogram, the sandy soil was also isolated from the other soil textures (clay and loam soil), Figure 2.

Table 1. Average, minimum, and maximum variation of ammonium (NH₄⁺) and nitrate (NO₃⁻) in tropical soil with sandy, loam, and clay textures.

Soil Textures	Average	Minimum	Maximum	Variation, %
NH ₄ ⁺ , mg kg ⁻¹				
Clay	31.0	24.9	207.7	±182.8 (±88%)
Loam	46.2	35.2	155.0	±119.7 (±77%)
Sandy	57.0	2.7	135.0	±132.2 (±98%)
NO ₃ ⁻ , mg kg ⁻¹				
Clay	104.2	5.1	104.0	±98.9 (±95%)
Loam	94.5	6.8	108.3	±101.5 (±96%)
Sandy	36.4	4.7	103.2	±98.5 (±95%)
Soil pH				
Clay	5.9	0.1	6.6	±6.4 (±92%)
Loam	7.0	0.1	7.6	±7.5 (±91%)
Sandy	5.9	0.1	6.8	±6.8 (±91%)
Cluster results				
	Cluster 1	Cluster 2	Cluster 3	
NH ₄ ⁺	-0.25	0.12	0.39	
NO ₃ ⁻	0.39	0.19	-0.99	
pH	-0.46	1.4	-0.47	
Sum of squares	578.24	232.21	237.77	
Number	252	126	126	

Cluster formation with data on soil texture, N stocks, and pH analyzed by machine learning (K-means). A total of 504 observations were tested.

Clay and loam soils present similar characteristics and can be classified as soils with a high nitrification rate demonstrated by higher averages of NO₃⁻ (clay = 104.2 mg kg⁻¹; loam = 94.5 mg kg⁻¹), which was converted from the ammonium to nitrate (Table 1).

The high variation in the contents of ammonium and nitrate in sandy, loam, and clay soil also demonstrated the intensity of the nitrification rate. These results are explained by higher organic matter and clay contents that promote soil biological activities, mainly in clay soil (Table 1).

3.2. Expected and Recommended N

The N recommendations for main crop production are based on the expected productivity of crops without monitoring of N stocks in the soil. For sugarcane with an expected productivity of 100 Mg of stalk/ha⁻¹, there is a recommendation of 120 kg N/ha⁻¹, while for corn and wheat with the expected productivity of 3 and 3 Mg of stalk/ha⁻¹, there is a recommendation of 60 and 40 kg N/ha⁻¹, respectively (Figure 3).

The efficiency of N use in sugarcane is higher compared to corn and wheat, where to produce 0.8 Mg/ha⁻¹ of stalks is requested 0.8 kg N/ha⁻¹. For corn and wheat, the same N recommendation produces an average of 0.1 Mg/ha⁻¹ of grains (Figure 1). These data indicate that there is a high N consumption for grain production with lower efficiency of use compared with sugarcane in Brazil. The high N consumption for grain production is also used for corn production in the United States of America, with an N consumption between 50 and 70 kg N ha⁻¹ [7]. The N dynamic and stocks in the soil are challenges due to the high influence of organic matter and the variation of chemical forms of nitrogen [25,26].

The accumulation of crop residues on the soil surface can also act as a source of N to successive crops, mainly to crop residues with high N content (i.e., soybean, and bean). Ferraz-Almeida et al. [25] showed a mean increase in 90% of the total soil N with the accumulation of soybean residue in the soil. Infield, Risi et al. [27], and Torres et al. [28] also showed a positive impact of cover crops (soybean and sunn hemp) in corn production as a result of N content in soil from the legume residues. However, organic nitrogen is not usually countable in the N recommendation of nitrogen in agriculture. Based on this

perspective, this study aims to review the latest advances in the balance of nitrate and ammonium in soil and see how far the soil characteristics help to understand that balance.

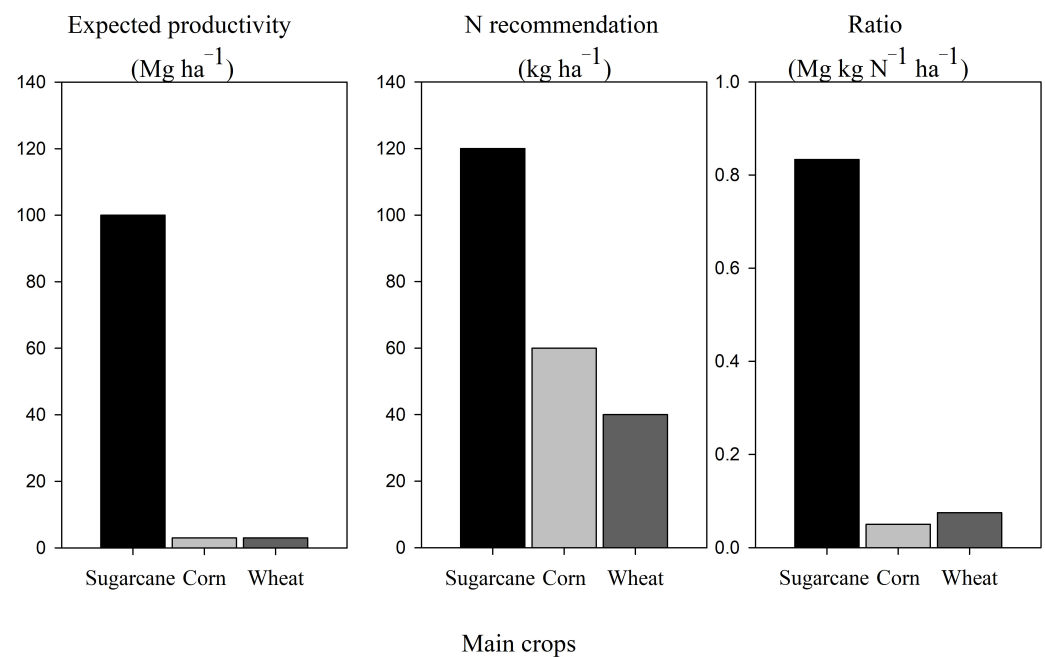


Figure 3. Expected productivity, N recommendation, and ratio (productivity/N recommendation) of sugarcane, corn, and wheat. N recommendation is the average in planting and cover.

3.3. Routes of Nitrate and Ammonium in Soil

The nitrate (NO_3^-) and ammonium (NH_4^+) balance in the soil is controlled by the nitrification process with the conversion of ammonium to nitrate by soil microorganisms (Figure 4).

The nitrification can be separated into two steps in soil: (i) the conversion of ammonium to nitrite and (ii) the conversion of nitrite to nitrate. The bacteria that carry out nitrification can be separated into two groups: ammonia-oxidizing bacteria (AOB, such as *Nitrosomonas*, *Nitrosococcus*, and *Nitrospira*) typically oxidize NH_3 to NO_2^- , and nitrite-oxidizing bacteria (NOB; such as *Nitrobacter*, *Nitrococcus*, *Nitrolanceletus*, *Nitrospina*, *Nitrospira*), with *Nitrosomonas europaea* being the most studied in ammonia oxidation (Figure 4).

Nitrification is classified as a common soil process with a rapid (days and weeks) conversion of NH_4^+ to NO_3^- [29]. In nitrification, both the first and second steps require aerobic conditions with adequate supplement of water in the soil (Figure 2). In the first step, there is the production of 4H^+ that promotes acidification in soil. Qiu et al. [30] showed that soil acidification is derived from the conversion of ammonium to nitrite and the absorption of cation over anion uptake by plants during nitrification. Both soil reactions are considered the major processes that lead to soil acidification.

The NH_4^+ and NO_3^- are ions in soil solution and act on soil colloids (clay and organic matter). The NH_4^+ is a cation (positive charge) that can be fixed on negatively charged clay surfaces and functional groups of organic matter [31]. On the other hand, NO_3^- is an anion (negative charge) that, generally, can be leached in the soil causing environmental damage and decreasing the N efficiency use.

Tropical soils are characterized by the predominance of positive charge on the clay surface, promoting the immobilization of NO_3^- . For example, de Oliveira Junior et al. [32] demonstrated a low NO_3^- -leaching in sugarcane areas in Brazil due to the predominance of positive charges in a tropical oxisol with a high N's extraction by the crop. Nyamangara et al. [33] measured the NO_3^- -leaching in sandy soil from Africa's tropical

and subtropical regions and noticed a low N loss by leaching with mineral application mineral and organic fertilizers.

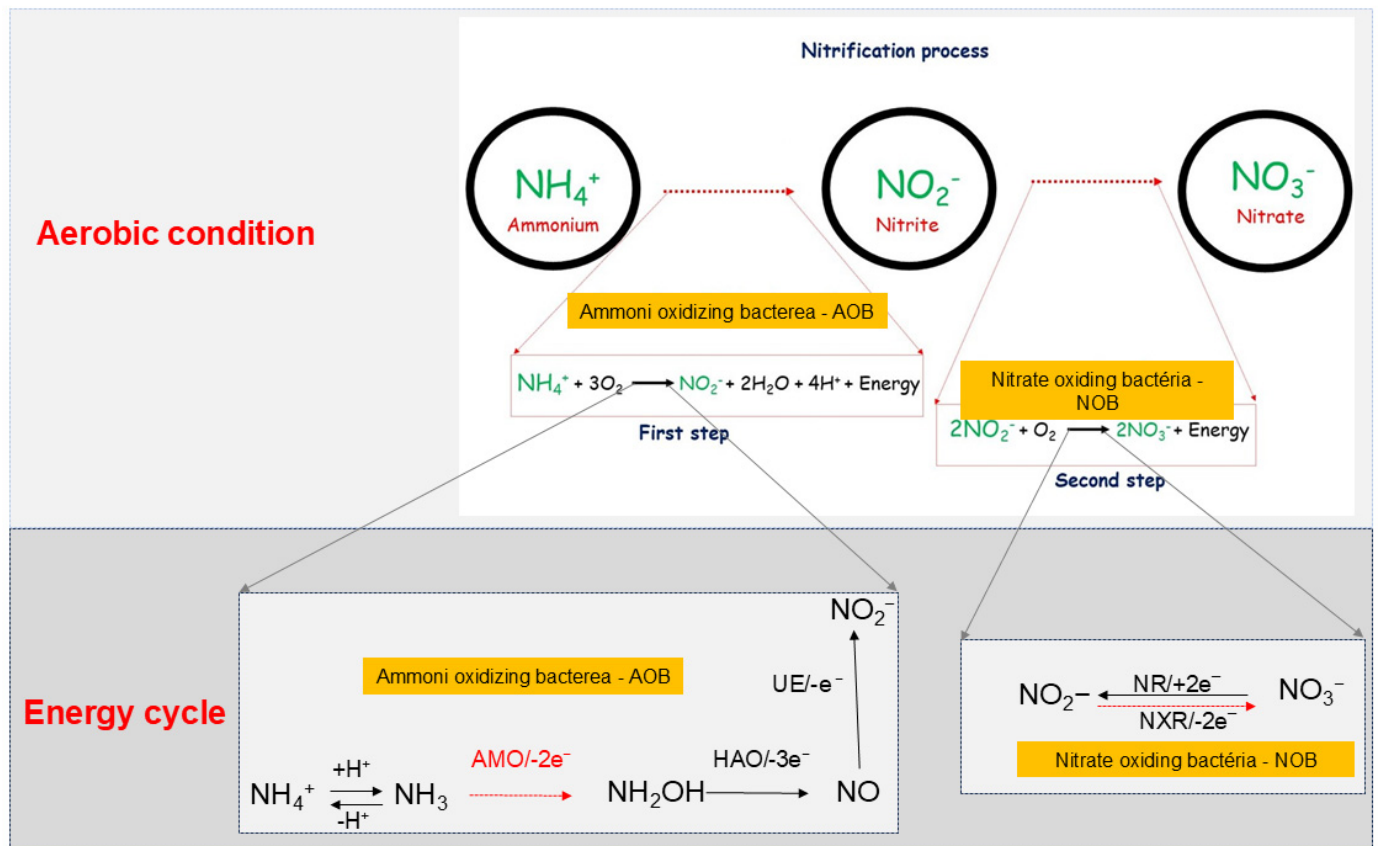


Figure 4. The process of nitrification is divided into two steps: (i) conversion of ammonium (NH_4^+) to nitrite (NO_2^-) and (ii) conversion of nitrite (NO_2^-) into nitrate (NO_3^-) in soil. Arrows represent the transformation in the nitrogen cycle.

The decrease of N losses with adding organic matter is probably associated with the increase in soil charge; thus, the organic matter is an important organic colloid in soil with positive and negative charges. Commonly, there is a high NO_3^- -leaching in humid temperate regions with the application of manure and inorganic fertilizers [34]. In tropical soil, phosphor can also be adsorbed on the surface of clay due to the predominance of positive charges on the clay surface and low organic contents [35,36].

3.4. Soil Water on Nitrate and Ammonium Balance

From a volume perspective, the pore volume is one of the soil factors that control the process of nitrification and denitrification in soil. When there is an aerobic condition with the adequate distribution of water and air in pores (up to 50% of the soil pore space filled with water), nitrification occurs caused by the proliferation of aerobic microorganisms that oxidize the NH_4^+ into NO_3^- . In contrast, the anaerobic condition is caused by the increase of water in the soil, decreasing the oxygen in pores (>75% of the soil pore space filled with water) and promoting the proliferation of anaerobic microorganisms with the production of nitrous oxide (N_2O), Figure 5.

In both routes, there are N losses by N oxides (N_2O , NO), but during nitrification, the N lost as N_2O rarely exceeds 2–4% of the total N nitrified [37]. Therefore, the major pathway to the emission of N_2O is the anaerobic conditions with an N loss between 60 and 80% of the NO_3^- lost as N_2O , NO, and N_2 [11]. Denitrifying bacteria comprise about 0.1 to 5% of the total soil microorganism population, identified as denitrifying microorganisms

(i.e., bacteria *Thiobacillus denitrificans*, *Alcaligenes eutropha*, and *Paracoccus denitrificans*), characterized as heterotrophic and belong to Proteobacteria and autotrophic. According to IPCC [38], there is a direct N_2O emission with the application of mineral N fertilizer, with the emission of approximately 10 kg N_2O for every ton of nitrogen fertilizer applied.

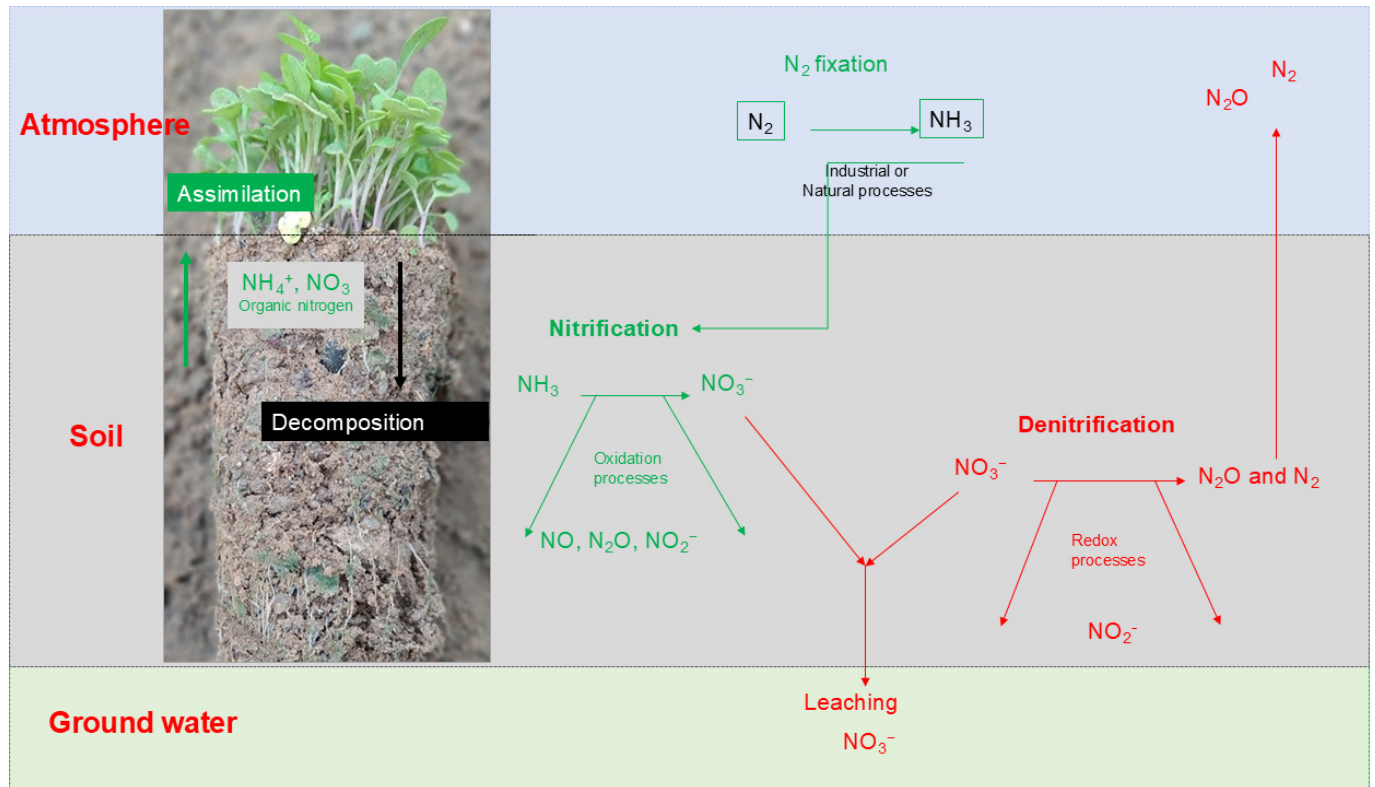


Figure 5. Distribution of water and air volumes: (i) aerobic condition and conversion of ammonium (NH_4^+) to nitrate (NO_3^-) in the soil; (ii) anaerobic condition and conversion of nitrate (NO_3^-) to nitrous oxide (N_2O) in soil. Arrows represent the transformation in the nitrogen cycle.

Agriculture also is responsible for global greenhouse gas emissions, with approximately 14–17% of global anthropogenic emissions [39]. In agriculture, the main source of N_2O emission is from synthetic fertilizers, manure, and crop residues left on soil, with an average of 40% of total agricultural emissions [26].

In soil, the higher N_2O peaks occur when there is an anaerobic condition caused by the water (H_2O) and N in the soil, fulfilling two conditions: (i) low content of O_2 (with approximately 70% of porous material filled by water) and (ii) a sufficiently high stock of NO_3^- [11]. Therefore, anaerobic conditions are requested for N_2O emission, but it is controlled by organic matter mineralization rate and N mineral fertilizers [40]. Reay et al. [41] showed that the mitigation of anthropic N_2O emission is requested to improve nitrogen-use efficiency in agricultural production, mainly in future scenarios with an increase in food demand and N consumption. Also, there is direct N_2O emission from the soil; however, it is considered low when there is a low N content in the soil.

The high accumulation of crop residues on the soil surface also impacts the N use efficiency: (i) residues with a high C:N ratio (i.e., sugarcane and pasture residues) decrease the N use efficiency due to the N immobilization and losses; (ii) residues with low C:N ratio can be considered as N sources due to high N content and rapid mineralization [20]. Studies by Joris et al. [42] showed that in sugarcane areas with residue accumulation the N immobilized can be considered as slow N sources when there is a constant N input. The increase in net mineralization of high C:N residues by long-term synthetic N fertilization is well documented in the literature [43,44]. The input of N in the soil also tends to stimulate

soil organic matter decomposition, which is called the soil priming effect, characterized as a strong short-term change of organic matter decomposition that can accelerate or retardate residue decomposition and release or immobilize a large amount of carbon and nitrogen [45].

3.5. Nitrate and Ammonium Contents in Agricultural Soil

In tropical soil, the content of NO_3^- was higher than the content of NH_4^+ , with an average increase of 35% represented by an average of 11.3 mg kg^{-1} (NO_3^-) and 7.2 mg kg^{-1} (NH_4^+) with all soil uses, except for the sugarcane areas (Figure 6).

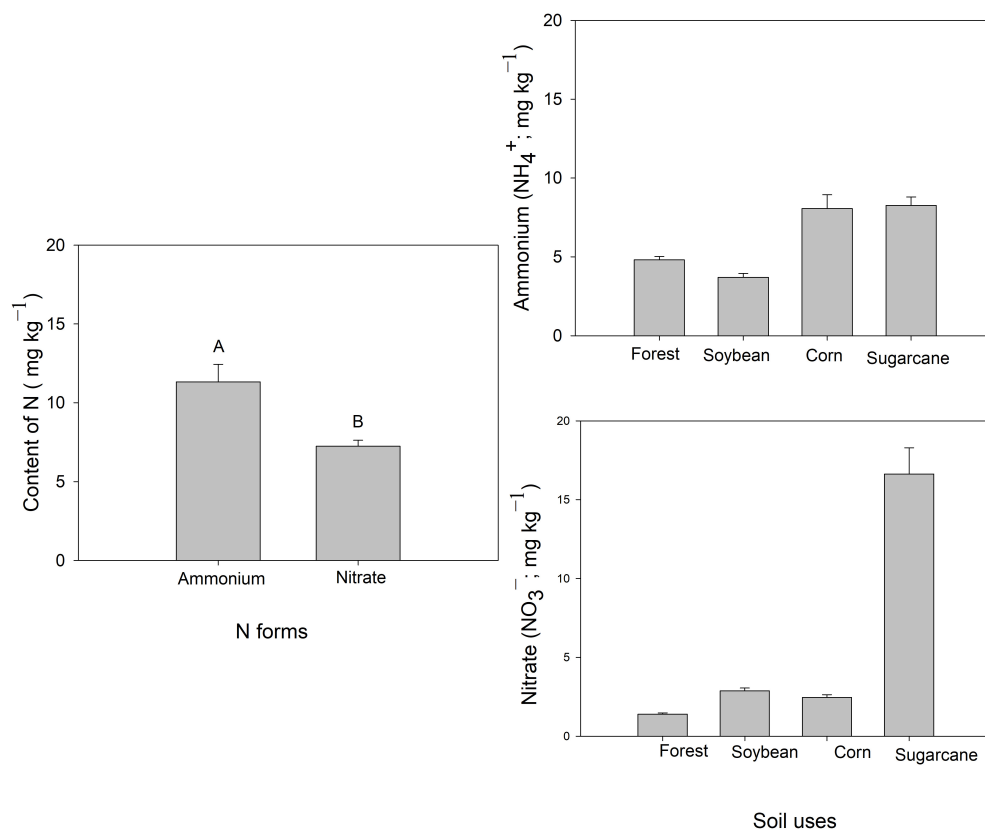


Figure 6. Contents of ammonium (NH_4^+) and nitrate (NO_3^-) in soil with the planting of soybean, corn, sugarcane, and forest in tropical conditions. Averages were compared using the t test ($p < 0.05$), when presented with capital letters there is a significant difference.

Sugarcane presented higher contents of NO_3^- (average = 16.5 mg kg^{-1}) compared with NH_4^+ (average = 8.2 mg kg^{-1}). In soybean, the contents of NH_4^+ ($=3.7 \text{ mg kg}^{-1}$) were higher than NO_3^- ($=2.9 \text{ mg kg}^{-1}$), as well as the contents of NH_4^+ for corn (average; $\text{NH}_4^+ = 8.0 \text{ mg kg}^{-1}$; $\text{NO}_3^- = 2.5 \text{ mg kg}^{-1}$), and forest areas (average; $\text{NH}_4^+ = 4.8 \text{ mg kg}^{-1}$; $\text{NO}_3^- = 1.2 \text{ mg kg}^{-1}$), Figure 4.

Ma et al. [46], monitoring the stocks of ammonium and nitrate in forest soils of Mount Wuyi, China, also noticed a higher nitrate content than ammonium in soil due to the metabolization of tannin in plant leaf litter caused by the secondary plant that reduced the rate of NH_4^+ more than for NO_3^- in soil. In contrast, Li et al. [47] showed that the NH_4^+ concentration in soil tended to be higher than the NO_3^- concentration in the forest, while in the mixed forest, there was an inverse with a higher NO_3^- compared to NH_4^+ concentration.

González-Pedraza et al. [20], monitoring the seasonality of precipitation on contents of NH_4^+ and NO_3^- in a Dry Tropical Forest Area in the Western Llanos of Venezuela, showed there was a significant difference in the content of NH_4^+ with higher content in

the end rainy season followed by the dry and early seasons. In contrast, the content of NO_3^- was higher in the dry season, followed by the end and early rainy seasons. Do Vale et al. [21] noticed that in the sugarcane area, the concentration of NO_3^- was higher (up to 14%) than the NH_4^+ concentration with the application of ammonium nitrate (35% of N). These studies demonstrate that the abundance of NH_4^+ and NO_3^- is related to organic matter content, the N inputs in soil, and the dynamic of soil use, explaining these variations in content and predominance of NO_3^- or NH_4^+ in soil.

The stocks and balance of NH_4^+ and NO_3^- in the soil also are explained by the N absorption of plants. Plants can absorb the N as NH_4^+ and NO_3^- , but plants can absorb a limited proportion of NH_4^+ due to toxicity [48]. For example, in hydroponic production, the balance between NH_4^+ and NO_3^- is essential because NH_4^+ is a better alternative to control pH in the nutrient solution. Andriolo et al. [49] showed that higher NH_4^+ concentrations (higher than 9–12%) in the nutrient solution promoted plant water flux restrictions and reduced roots and the number of leaves of lettuce plants.

Some plants are most sensitive to NH_4^+ toxicity (i.e., citrus species [50], tomato [51], potato [52]), while other plants are considered more adapted to NH_4^+ (i.e., domesticated species such as rice). Sugarcane is an example of a plant that prefers NH_4^+ as a source of N with a positive impact on plant development [53]. Rashti et al. [54] and Boschiero et al. [55] showed a difference in N assimilation as NH_4^+ and NO_3^- in sugarcane, requiring different amounts of energy and reduction of chloroplast in the plant. Robinson et al. [56] showed that sugarcane presented a low capacity to store nitrate in shoots, indicating rapid assimilation of nitrate in shoots and low nitrate uptake. Boschiero et al. [57] showed that the uptake of NH_4^+ is faster than NO_3^- , explaining the rapid N assimilation by sugarcane and suggesting that the term “preference” should be used cautiously to avoid misinterpretation. Comparing the N mineral sources, the application of ammonium nitrate (both NH_4^+ and NO_3^-) presented a higher N utilization (60%) than urea to sugarcane development [55].

Corn is an example that can absorb both ammonium and nitrate forms. In contrast, the application of a source of N as nitrate promotes a superior response on growth compared to ammonium due to the deleterious effect of ammonium on the growth and condition of the root system [43]. Uptake and assimilation of ammonium-N and nitrate-N influenced the growth of corn directly. When both nitrate-N and ammonium-N were supplied, the two forms presented similar uptake rates, but the ammonium is used preferentially to synthesize amino acids and protein [57]. Tills and Alloway [48] noticed that the weight gain of shoots and roots was decreased in the order of $\text{NO}_3^- > \text{NH}_4^+/\text{NO}_3^- > \text{NH}_4^+$. Some agriculture crops (i.e., rice) request management systems to apply NH_4^+ to avoid N losses by oxide nitroso.

3.6. Correlation between Soil Textures and the Processes of Nitrification

There was a variation in the nitrification caused by the soil texture, indicating that the soil texture directly influences the N dynamic in soil. In sandy soil, there was a positive correlation between NH_4^+ and NO_3^- ($r: 0.62; p < 0.05$), indicating that the increase of NH_4^+ contributes to an increase in the content of NO_3^- . In loam and clay soils, the flux is inverse while the increase of NO_3^- reduces NH_4^+ represented by a correlation of -0.12 (loam soil) and -0.34 (clay soil), Figure 7.

In sandy soil, there is a greater movement of NO_3^- with inputs of N mineral fertilizers caused by high mass flows [58]. In addition, NO_3^- movement in sandy soil can be intensified by the excess inputs of water and successive N fertilization, promoting the N losses by leaching [59]. Comparing the dataset from clayey and sandy, the high nitrification rate in clayey soil is explained by higher organic matter and clay content that promote soil biological activities [60].

The alteration of soil pH from nitrification due to the release of H^+ in the soil solution is the first step. Barth et al. [60] showed that in the sand-textured soil with the application of N mineral fertilizer, there was a higher NH_4^+ content with a difference and mean increase

of 65% compared to clay-textured soil. As a result of the balance, the increase of NH_4^+ content in soil promoted a decrease in the content of NO_3^- .

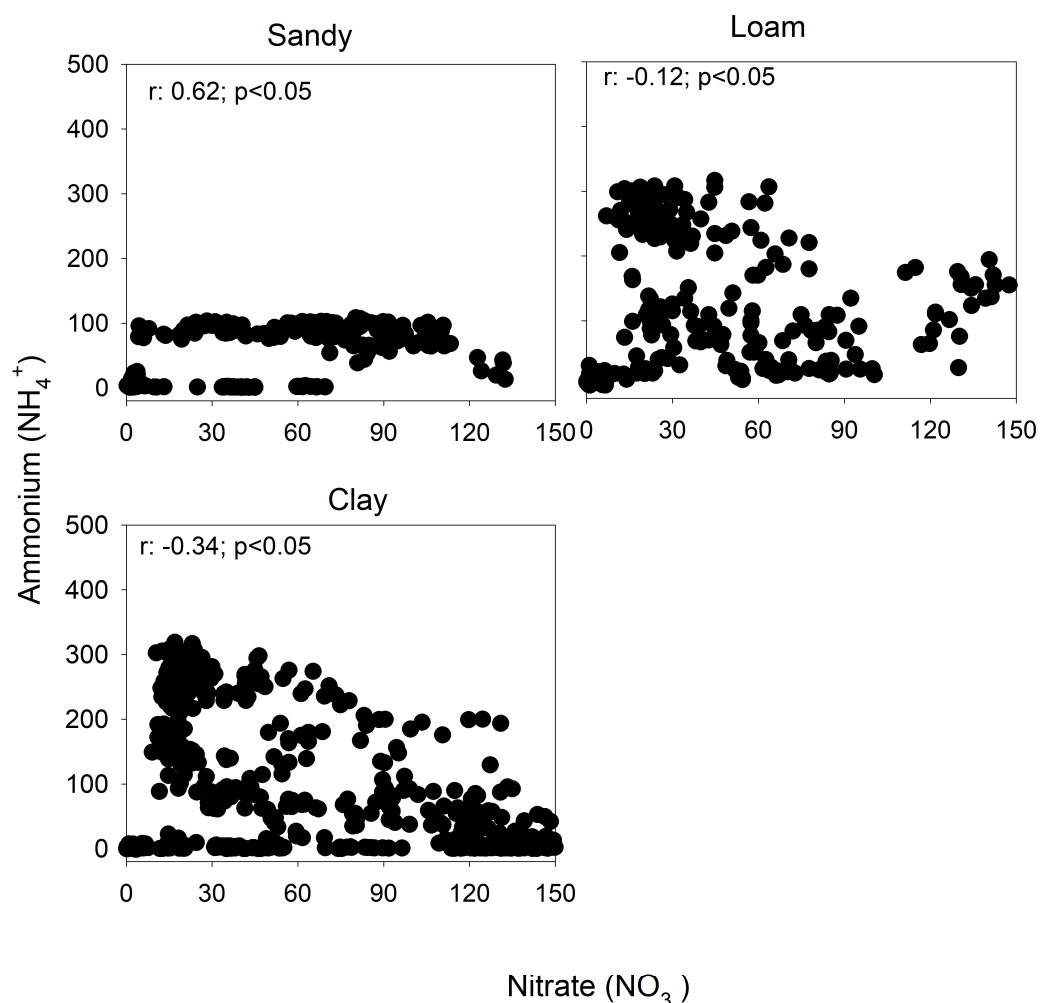


Figure 7. Correlation of ammonium (NH_4^+) and nitrate (NO_3^-) in tropical soil with sandy, loam, and clay textures. Correlations were tested by Pearson correlation using a p of 0.05.

4. Conclusions

Soil textures, N stocks, and pH promoted the formation of a one-dimensional plane represented into three clusters with the similarity between Cluster 1 and Cluster 2 and represented by clay and sand textures with higher contents of nitrate as a result of high nitrification rate and lower contents of ammonium in soil. In agricultural soils, the content of NO_3^- tends to be higher than the content of NH_4^+ . However, the abundance of NH_4^+ and NO_3^- is associated with the organic matter content, the N inputs in soil, the dynamic of soil uses, soil texture, and N absorption by plants. Based on the results of machine learning, for clay and loam soil, the contents of NO_3^- can be used as indicative of N status as a final result of nitrification rate and higher variation in soil. However, in sandy soil, NO_3^- can not be used as indicative of N status due to N losses by leaching. For future studies, algorithms with information on crop management and soil factors are requested due to the great variability in the balance of NH_4^+ and NO_3^- in the soil, which can help to improve N use efficiency and N recommendation in agriculture. The contents of NO_3^- can be used in the algorithms as indicative of N status as a final result of nitrification rate and variation in soil, mainly for the condition of clay and loam soils. The N recommendation in agriculture is based on the expected productivity of crops and does not consider the contents of nitrogen, organic, and/or inorganic, which demonstrate a low efficiency for

the use of nitrogen. Therefore, the increase in the N use efficiency with the adequate use of the 4R principles (right source, right rate, right time, and right place) contributes to the adequate use of resources with harmony between the economy, environment, and society.

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