

Extended Abstract

Morphological Control of Metal Oxide for Semiconductor-Based Gas Sensor [†]

Takeshi Hashishin ^{1,*}, Jian Sun ², Kazuyoshi Sato ³ and Hiroshi Kubota ²

¹ Faculty of Advanced Science and Technology, Kumamoto University, 2-39-1 Kurokami, Chuo-ku, Kumamoto 860-8555, Japan

² Graduate School of Science and Technology, Kumamoto University, 2-39-1, Kurokami Chuo-ku, Kumamoto 860-8555, Japan; 186d8884@st.kumamoto-u.ac.jp (J.S.); kubota@cs.kumamoto-u.ac.jp (H.K.)

³ Graduate School of Science and Technology, Gunma University, 1-5-1 Tenjin-cho, Kiryu, Gunma 376-8515, Japan; kazuyoshi-sato@gunma-u.ac.jp

* Correspondence: hashishin@msre.kumamoto-u.ac.jp

[†] Presented at the 8th GOSPEL Workshop. Gas Sensors Based on Semiconducting Metal Oxides: Basic Understanding & Application Fields, Ferrara, Italy, 20–21 June 2019.

Published: 19 June 2019

Morphological control of metal oxide (MO) is important for enhancement of sensing properties such as sensor response and response-recovery characteristics. Our research group has been developed MO based gas sensors fabricated by WO₃ and SnO₂ nanocrystals synthesized by hydrothermal method for high sensor response to NO₂ or H₂ [1,2]. On the other hand, shuttle-shape SnO₂ showed the sensor response to NO₂ and H₂S at room temperature.

The film sensor with cuboid-shape monoclinic WO₃ nanocrystal (Figure 1a) showed sensor response (R_g/R_a) of 10²–10⁴ to 0.05–1 ppm NO₂ at 200 °C (Figure 1c). In contrast, the sensor with hexagonal-shape hexagonal WO₃ nanocrystal (Figure 1b) showed sensor response lower one order of magnitude than that with cuboid-shape monoclinic WO₃ in above same detection condition (Figure 1d). This difference was related to surface states. XPS spectra of O1s showed that the content of OH⁻ was larger for hexagonal-shape hexagonal WO₃ (O²⁻/OH⁻/H₂O (in %) = 60.0/38.1/1.9, in Figure 1e) than for cuboid-shape monoclinic WO₃ (80.4/4.9/14.7, in Figure 1f). The results suggested that the high content of oxygen adsorbate (O²⁻) on the surface of WO₃ could be contributed to higher sensor response.

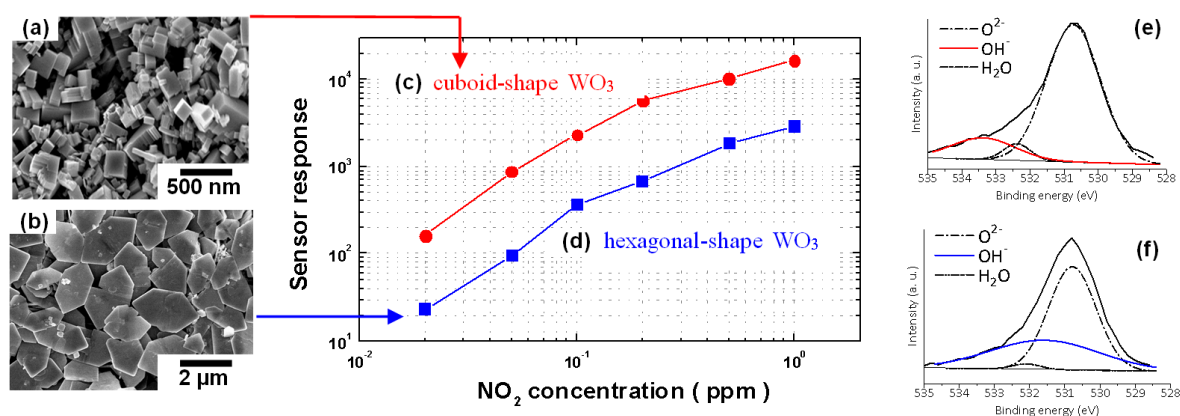


Figure 1. FE-SEM images of (a) cuboid-shape monoclinic WO₃ nanocrystal and (b) hexagonal-shape hexagonal WO₃ nanocrystal. Sensor response as a function of NO₂ concentration for (c) cuboid-shape WO₃ and (d) hexagonal-shape WO₃. XPS spectra of O 1s on the surface of (e) as-prepared cuboid-shape WO₃ and (f) hexagonal-shape WO₃.

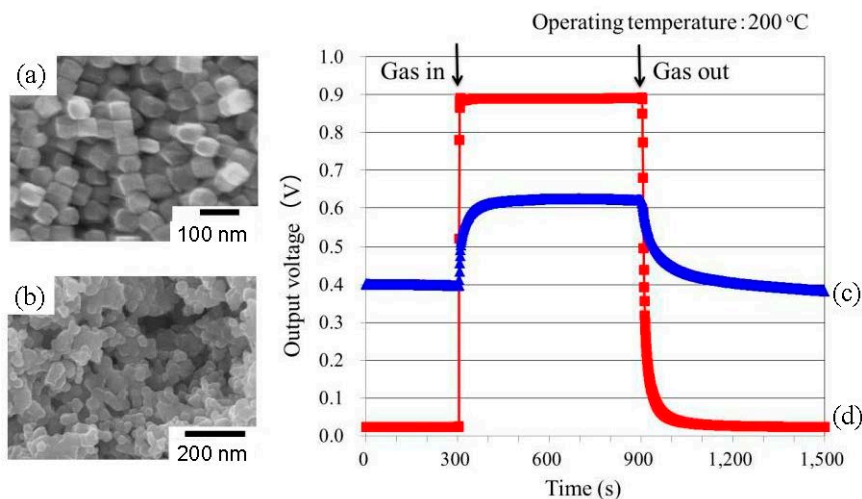


Figure 2. FE-SEM images of (a) SnO₂ nanocubes annealed at 250 °C for 3 h and (b) commercial SnO₂ nanoparticles calcined at 1100 °C. Response- recovery transients of (c) and (d) to respective gas mixture of 1000 ppm H₂ and air.

References

1. Meng, Z.; Fujii, A.; Hashishin, T.; Wada, N.; Sanada, T.; Tamaki, J.; Kojima, K.; Haneoka, H.; Suzuki, T. Morphological and crystal structural control of tungsten trioxide for highly sensitive NO₂ gas sensors. *J. Mater. Chem. C* **2015**, *3*, 1134–1141.
2. Sato, K.; Yokoyama, Y.; Valmalette, J.C.; Kuruma, K.; Abe, H.; Takarada, T. Hydrothermal Growth of Tailored SnO₂ Nanocrystals. *Cryst. Growth. Des.* **2013**, *13*, 1685–1693.



© 2019 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<http://creativecommons.org/licenses/by/4.0/>).